

Studies on Shelf Life Extension of Sugarcane Juice

THESIS

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the Degree of**

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In

AGRICULTURE

(FOOD SCIENCE & TECHNOLOGY)

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2018

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All the assistance and help received during the course of the investigation have been acknowledged by him.

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	CURRICULUM VITAE	

ABBREVIATION

%	=	Percent
i.e.	=	That is
viz.	=	Namely
ml	=	Milliliter
Ppm	=	Part per million
G	=	Gram
RPM	=	Round per minute
p.s.i.	=	Pounds per square inch
et al.	=	Co-workers
w/v	=	weight/volume
°C	=	degree centigrade
T	=	Pasteurization temperature
Min	=	Minute
C	=	Citric acid
KMS	=	Potassium metabisulphite
PS	=	Potassium sorbate
AA	=	Ascorbic acid
Fig.	=	Figure
Mt	=	Million tones
Mh	=	Million hector
Etc	=	et cetra
°B	=	Degrees Brix

INTRODUCTION

Sugarcane (*Saccharum officinarum*) belongs to the family Gramineae. It is one of the most important agro-industrial crops in our country. The total production of sugarcane in India during the year 2017-18 was 330 million tones. Uttar Pradesh and Maharashtra are the two largest sugarcane producing states in the country, accounting for more than 80 percent of the annual crop production (Anonymous 2011). India stands second among other sugarcane growing countries, contributing nearly 20.40% area and 18.60% production. Sugarcanes are largely utilized for production of white sugar crystals, gur, khandasari and sugarcane juice (pattnayak and Misra, 2004).

Fresh Sugarcane juice is popular beverage in Southeast Asia, South Asia, Latin America, and also in other countries where sugarcane is grown commercially. It is known as “Oosacha Ras” or “Ganneka Ras” in Maharashtra in Marathi and Hindi accordingly. It is relished for its sweet taste and flavor. Sugarcane juice provide vitamins A, C, B1, B2, B3, B5, B6 and iron with a high concentration of phytonutrients (including chlorophyll), antioxidants, proteins, soluble fibre and numerous other health supportive compounds (Karthikeyan and Samipillai, 2010).

In the last couple of years, globally, the demand for sugarcane juice has been increased rapidly and as a result of which number of fruit processing industries have come up. These juice-processing industries use conventional technology to produce juice, which are not of high quality in terms of their sensory attributes.

Sugarcane juice is extracted using a crusher, operated either manually or mechanically. It is often freshly extracted and sold in markets by street vendors at public eateries. As sugarcane juice is rich in sugar, it starts fermenting immediately after extraction and also turns brown due to poly phenol oxidease activity (Qudsieh *et al* 2002). Microbial fermentation of the juice turns its taste to sour within few hours of extraction (Yusof *et al* 2000). All these negative changes limit the processing and marketing of sugarcane juice (Eggleston 2002). Development of a preservation method, which can control microbial growth in raw sugarcane juice and maintain its freshness, is

a challenging problem for ensuring its wider distribution and availability. Probably, due to this problem, different groups of researchers have tried to preserve different forms of sugarcane juice using membrane technology (micro-filtration and ultra-filtration), chemical preservatives and thermal processing (Singh *et al* 2002, Taylor 2005). The sugarcane juice clarification by micro-filtration and ultra-filtration with chemical are commercial practice adopted by sugar industries as a continuous non thermal process, which result in significant saving in time, material and labour as well as improves yields and quality of product. However, viable technology of sugarcane juice processing at domestic level is not available.

Preservation of pasteurized sugarcane juice by chemical preservative may be an economical process for food industries. This will provide economy in packaging, transportation, storage and distribution of the sugarcane juice. This calls for thorough investigations whether proposed unit operations for the production of sugarcane juice could be replaced by less expensive unit operation. Very meager information is available on above topic; therefore the proposed research work entitled “**Studies on shelf life extension of Sugarcane juice**” has been planned with following objectives.

Objectives:-

1. To study the effect of various chemical preservatives on chemical constituents of bottled sugarcane juice during storage.
2. To study the effect of chemical preservatives on shelf life of bottled sugarcane juice at refrigerated and ambient room temperature.

REVIEW OF LITERATURE

The past experiences are essential to have an introspective view in the research. Keeping this idea hence forth, an attempt has been made in this chapter to present the available review of literature related to the facts of origin, distribution, area, production and utilization of sugarcane, nutritional value of sugarcane juice, technology of preservation sugarcane juice processing and sensory quality of juices and microbiological and storage stability of juices.

2.1 Origin, distribution, area, production and utilization of sugarcane

Sugarcane, or sugar cane, are several species of tall perennial true grasses of the genus *Saccharum*, tribe Andropogoneae, native to the warm temperate to tropical regions of South and Southeast Asia, Polynesia and Melanesia, Sugarcane belongs to the grass family Poaceae. The plant is two to six meters (six to twenty feet) tall. It has stout, jointed, fibrous stalks that are rich in the sugar sucrose, which accumulates in the stalk internodes. Sugarcane is the world's largest crop by production quantity, with 1.9 billion tones produced in 2016, and Brazil accounting for 41% of the world total. In 2012, The Food and Agriculture Organization estimated that, it was cultivated on about 26×10^6 hectares (6.4×10^7 acres), in more than 90 countries. The global demand for sugar is the primary driver of sugarcane agriculture. Cane accounts for 80% of sugar produced; most of the rest is made from sugar beets. Sugarcane predominantly grows in the tropical and subtropical regions (sugar beets grow in colder temperate regions). Other than sugar, products derived from sugarcane include falernum, molasses, rum, *cachaça* (a traditional spirit from Brazil), bagasse, and ethanol. In some regions, people use sugarcane reeds to make pens, mats, screens, and thatch. The young, unexpanded inflorescence of *Saccharum edule* (*duruka* or *tebu telor*) is eaten raw, steamed, or toasted, and prepared in various ways in Southeast Asia, including Fiji and certain island communities of Indonesia.

2.2 Nutritional value of sugarcane juice

Consuming sugarcane can be very healthy as it contains a lot of vital ingredients which is ideal for forming a strong immune system. Swaminathan, (1995) reported that sugarcane juice is rich in enzyme and has many medicinal properties. It contains water (75%-85%), reducing sugar (0.3-3.0%), non reducing sugar (10-21%). Sugarcane juice contains 30 grams of natural sugar along with 250 calories. The fat, protein, fiber and cholesterol content is generally less but it contains copious amounts of iron, magnesium, calcium, potassium and sodium. Health Benefits of Sugarcane juice:

Mentioned below are the best health benefits of Sugarcane juice.

1 Can be consumed by diabetic patients too

Even though diabetic people are not allowed to consume sugar products due to the high sugar content in their blood, sugarcane is relatively safer to consume. Sugarcane contains natural sugar which has extremely low glycemic index, this prevents sudden rise in blood glucose levels in diabetic patients. Sugarcane also acts as a substitute for soft drinks or aerated drinks for these people. However, it should still be consumed in moderation.

2 Helps to prevent cancer from occurring

Sugarcane helps to prevent cancer to some extent. This is because; it is high in alkaline content due to concentration of manganese, iron, potassium, magnesium and calcium. It is difficult for cancer cells to survive in an alkaline environment which is why breast cancer and prostate cancer can be kept at bay by consuming sugarcane juice regularly.

3 Helps in the normal functioning of the kidneys

Sugarcane juice generally boosts the protein levels in the body, thus maintaining the functioning of the kidneys. If it is taken in diluted form along with coconut water and lime juice, it can help to reduce the burning sensation which is associated with prostatitis, kidney stones, sexually

transmitted diseases and urinary tract infection (UTI). There are antioxidants present in sugarcane juice which helps to boost the immune system and fight infections in the body. The digestive system also functions smoothly by consuming sugarcane juice.

4 Sugarcane also helps to combat skin conditions

Sugarcane juice has an extremely important component called hydroxy acid and glycolic acid which has plenty of skin benefits. Sugarcane juice can keep the skin hydrated, prevents ageing, reduces blemishes and fights acne. The presence of phenolic compounds, flavonoids and antioxidants can also delay the signs of aging and reduce the appearance of wrinkles. Applying this juice on the face regularly can keep it radiant, soft and supple.

5 Acts as an energy booster

Since sugarcane is a rich source of sucrose (simple sugar), it can be easily absorbed by the body. This sugar can be used to replenish low sugar levels in the body. Therefore, consuming sugarcane juice can make you feeling refreshed, charged and energetic.

6 Helpful during pregnancy

Sugarcane juice is said to be an useful addition to the diet of an expecting mother. It facilitates safer pregnancy and quicker conceptions. Sugarcane has copious amounts of vitamin B9 and folic acid which protects the unborn child from birth defectssuch as Spina bifida. According to studies, sugarcane juice also helps women with ovulating problems, therefore increasing the probability of a safer conception.

7 Prevents tooth decay

Bad breath can be very embarrassing and can also feel very uncomfortable. Sugarcane can be used as a home remedy to combat this problem. Sugarcane contains important minerals like phosphorous and

calcium which helps to build up the teeth enamel, thus keeping tooth decay at bay.

8 Treats febrile disorders

Febrile disorders are very common in kids or early teens. It can cause regular fevers with extremely high temperature. During the fever, a lot of protein content is lost. Sugarcane juice helps to replenish the lost proteins and helps in speedy recovery from the fever.

9 Helps lose weight

Sugarcane is ideal for people who want to reduce weight, since it has natural sugar, Sugarcane juice can also reduce unhealthy cholesterol levels. It is high in soluble fibre which is why it can reduce weight drastically. Sugarcane is ideal for a sore throat and increases muscle power as it contains natural glucose which is required by the body to maintain stamina. In many cases, jaundice or yellow fever can also be cured by consuming sugarcane juice. Sugarcane can also be used to manufacture recycled paper, this helps to reduce deforestation to a great extent. The side effects of consuming sugarcane juice is extremely low, but however it can lead to some health issues. An ingredient called policosanol present in sugarcane can cause insomnia, upset stomach, dizziness, headaches and weight loss if consumed excessively. It can also cause blood thinning and can affect the cholesterol levels in the blood.

2.3 Preservation of sugarcane juice

Sugarcane juice is one of the delicious drinks that enjoy a wide popularity in view of its pleasing taste, refreshing tingle and availability during the greater part of the year throughout the country. The main problem associated with fresh sugarcane juice is its short life and heat sensitivity of its flavor. Therefore the drink is mostly sold – fresh by roadsides and small eateries. Therefore, most of

the attempts to preserve the sugarcane juice have been focusing on the use of refrigeration, heat treatment and preservatives.

Kapur et al. (1978) reported, that the freshly crushed cane juice heated to 80°C, mixed with citric acid (0.2%) alone or in combination with ascorbic acid (0.05%) Filtered, filled hot in to sterilized bottles and sealed immediately, maintained the original taste and flavor of sugarcane juice.

Bobadilla and Preston, (1981).reported that sodium benzoate at a concentration of 0.05%was sufficient to stop fermentation sugarcane juice for 2 – 3 days and 0.1% was enough to preserve the juice for 6days. On the other hand aqueous ammonia at a concentration of 0.32% preserved the juice for 2 days and 1.28% upto 6 days

Attempts made to preserve the sugarcane juice involved blanching of stems, juice extraction, pasteurization followed by addition of ascorbic or citric acid and preservatives such as potassium meta bisulphate, sulphur dioxide and bromo nitro propanediol. Ginger or lemon extract were added to improve the taste. The juice thus prepared was bottled in sterilized glass bottles to be used as an acceptable beverage with a low shelf life under room temperature at refrigerated condition (Anonymous 1991; Mann and Singh 1988)

Bhupinder et al. (1991) developed a process for preservation of sugarcane juice into acceptable ready to serve beverage by pasteurization of sugarcane juice for 10 min at 80°C with addition of preservative 140 ppm potassium meta bi sulphite.

Nicolas et al, (1994) reported the variations in the effect of different acids on enzymes activity; as an example, malic acid and citric acid has been reported to be more efficient in the inhibitor of enzyme activity in sugarcane juice.

Whitaker 1994 and Lin et al.,(2007). Reported that ascorbic acid is the most widely used inhibitors of enzymes activity in addition to its reducing properties and lowering pH

Kaur et al. (1995) reported that, addition of preservatives and sterilization increased the shelf life of sugarcane juice up to 24 weeks.

Kaur *et al.* (1995) reported the scheme for producing bottled cane juice by pasteurizing the extracted juice for 10 min at 80⁰ C, adding K₂S₂O₅ equivalent to 70 ppm. SO₂ hot bottling, sterilizing for 30 min at 100⁰C and cooling for storage and reported that *Leuconostoc mesenteroides* activity was absent from both plain and treated juices.

Khanna (1995) reported the use of Deola (*Hibiscus esculentus*) extract for cane juice clarification. The extract was added to the boiling pan in three installments (10:10:5) with an interval of 5-10 minutes result in clear, transparent and light brownish yellow colour juice.

Yusof *et al.* (1999) studied the quality of sugar-cane juice extracted from stored canes, as well as changes in quality of fresh juice stored at different temperatures and reported that low temperature storage of canes was able to maintain the quality of juice for up to 9 days .During storage, sucrose contents decreased while fructose, glucose and titratable acidity increased in the samples.

Chauhan et al., (2002) reported that during storage of sugarcane juice the pH, total soluble solids and total sugars decreased, whereas, titratable acidity and reducing sugars increased significantly ($P < 0.01$).

Hanan *et al.* (2002) studied the effect of sugarcane maturation on the contents of chlorophyll, tannin, and polyphenol oxidase (PPO) activity and on color change of sugarcane juice and reported that there were significant decreases in total chlorophyll a and b and tannin contents during maturity followed by slower rates of decrease of both parameters at the end of maturity stages. There was a highly significant difference in PPO activity of cane juice during maturity. PPO activity was high at the early development stage, decreased during maturation, and then remained relatively constant at the end of maturity.

Prasad and Nath, (2002) reported the quality of sugarcane juice extracted from stored canes, as well as changes in quality of fresh juice stored at different temperatures.

Pushpa et al. (2002) reported the process for the preparation of sugarcane juice concentrate using citric acid alone and in combination with sodium benzoate. Treated samples were packed in glass bottles shows that concentrate when treated with citric acid (0.5%) and sodium benzoate (500 ppm) in combination gives the best result in terms of appearance, clarity, colour, flavour, taste and shelf life was increased upto 8 months.

Bosse et al. (2006) preserved sugarcane juice by heating it to a different temperature (75, 85 and 95⁰C) with addition of potassium meta bi sulphite and citric acid and observed decrease in pH at elevated time and temperature of pasteurization. The reducing sugars showed an upward trend as the temperature of pasteurization increased and sucrose levels decreased.

Lin Chun Mao et al. (2007) reported that the combination of blanching of sugarcane stems and addition of ascorbic acid produce the best quality of sugarcane juice.

Mao et al. (2007) reported the physiochemical changes in fresh sugarcane juice stored at 10°C. It was seen that blanching of stems before squeezing the juice and use of ascorbic acid (0.1%) improved the quality of sugarcane juice by preventing degreening and/or browning, and reduced activities of PPO (polyphenol oxidase) and SNI (sugarcane neutral invertase) in fresh sugarcane juice. Addition of ascorbic acid appeared to be more effective than blanching.

Azra Yasmin et al. (2010) reported that sugarcane juice pasteurized just after extraction and its pH was maintained at 4.3 by the addition of citric acid, hot filled in sterilized glass bottles, stored at room temperature for four months. The developed product had attractive color, flavor and refreshing with uniform consistency. Bottled juice samples were merely accepted up to a storage period of 120 days

Karmakar *et al.* (2011) prepared sugarcane juice beverage by pasteurizing 100 ml of sugarcane juice at 80, 85, 90, 95 °C for 2 min and stored at 25 and 40 °C in glass bottles and reported that an acceptable quality beverage of sugarcane juice with satisfactory storage stability for 25 days at 40 °C could be prepared.

Mishra *et al.* (2011) analyzed that a combination of gamma radiation (5 kGy) with permitted preservatives and low temperature storage (10°C) could preserve raw sugarcane juice for more than a month. The preservatives used were citric acid (0.3%), sodium benzoate (0.015%), potassium sorbate (0.025%), and sucrose (10%). The treatment helped in extending the shelf life to 15 days at ambient temperature (26± 2°C) and 35 days at 10°C.

Krishna *et al.* (2013) determined the quality of sugarcane juice extracted from stored canes, as well as changes in quality of fresh juice stored at different temperatures and reported that low temperature storage (10°C) of canes was able to maintain the quality of juice for 10 days, while low temperature storage (5°C) of juice could last for only 4 days. Spoilage of cane stored at 30°C occurred faster than that stored at 10°C. Fresh sugarcane juice became spoil within a day when stored at 30° C. Microbial count (bacteria, yeast, fungi) especially lactic acid bacteria count increased, during storage of cane juice.

2.4 Processing and sensory quality of juices:

Lionnet, (1986) reported that rate at which harvested cane deteriorates is influenced primarily by temperature, humidity, cane variety and the state of the stalk.

Goyle and Ojha, (1998).reported the sensory quality of preserved orange juice with potassium metabisulphite and stored at refrigerated temperature (1°C) and at room temperature for a period of 4 weeks. On sensory evaluation scores for flavour, acceptability and taste of experimental juice and a control were comparable during first three weeks. In 4th week the sweet and after taste was found to be considerably lower.

Orange juice was stored at refrigerated temperature without preservative and at room temperature with preservative for a period of 4 weeks. At refrigerated temperature there was decrease in pH and titrable acidity. But there was no change in pH and titrable acidity (TA) at room temperature. The storage of orange juice brought about a loss of 5-8% vitamin C at refrigerated and room temperatures after 4 weeks (Goyle and Ojha, 1998).

Litchi juice was extracted from ripe litchi fruits. The extracted juice was subjected to different treatments: T1 = heating at 85°C + citric acid (1%) + KMS (500 ppm); T2 = citric acid (1%) + KMS (500 ppm); T3 = heating at 85 degrees C + citric acid (1%) + ascorbic acid (0.01%) + KMS (500 ppm); and T4 = citric acid (1%) + ascorbic acid (0.01%) + KMS (500 ppm). The processed juice samples were kept in glass bottles sealed with crown corks and stored for 12 months at 11-36°C (room temperature) and at 5-8°C (low temperature). Results showed that bottled litchi juice maintained acceptable colour and quality for up to 12 months when stored at low temperature and only up to 6 months for T1 and T3 samples stored at room temperature (Alex et al., 2003).

The chemical, enzymatic changes and shelf life of pasteurized orange juice packed in polyethylene terephthalate bottles stored at 4°C were evaluated. Two batches of orange juice samples were pasteurized at 72°C for 16 seconds and at 90°C for 40 seconds. The orange juice samples were evaluated periodically for relative density, soluble solids (degrees Brix), titrable acidity, pH, suspended solids (pulp), ascorbic acid, colour, and pectin esterase activity. The juice samples exhibited good stability and no statistical difference was observed among the treatments, except for colour (turbidity) that increased with longer storage time (Correo et al., 2003).

Unpasteurized and pasteurized sweet orange juices were filled in sterilized and unsterilized bottles, stored at room temperature and refrigerated temperature. Visual appearance of all the juice samples was maintained within highly acceptable range during the entire period of storage but with the

advancement of time, deterioration in taste and flavor occurred which adversely affected the overall acceptability of juice samples (Jain et al., 2003).

A study was done to develop a low cost method of lime juice preservation. Pasteurized (77°C for 60 seconds) and non-pasteurized juice were treated with different levels of potassium metabisulfite (500 ppm, 1000 ppm, 1500 ppm and 2000 ppm) and stored in glass bottles at room temperature for 3 months. Results obtained showed that total soluble solids increased during storage while sulfur dioxide (SO₂) and ascorbic acid contents gradually decreased. Non-pasteurized juice was found to be the most acceptable in terms of organoleptic evaluation. It was found that lime juice could be preserved for a period of about 10 weeks without any undesirable changes by using SO₂ at an initial level of 500 ppm (Ekanayake et al., 2004).

Falade et al., (2004) reported that, sweetened mango juices processed, bottled and stored at 25 °C for 12 weeks. The titrable acidity, pH, total solids, ash, soluble solids and ascorbic acid contents were evaluated immediately after processing and subsequently at 2-week intervals. The quality attributes of the sweetened mango juices decreased during storage. After 12 weeks of storage, the percentage ascorbic acid loss of sweetened Julie and Ogbomoso mango juices was 16.25 and 16.67%, respectively. Browning index increased during storage.

Shelf life of Passion Fruit Juice was increased using sugar, benzoic acid, citric acid and a combination of citric and benzoic acid. The result showed that 30% benzoic acid was able to preserve the juice for one month and was found to be the best method. The juice with 4% sugar was spoiled after three days, while that of 4% citric acid could be stored for one week and some days. The combination of 3% benzoic acid and 4% citric acid increased the shelf life upto two to three weeks (Akpan and Kovo, 2005).

Madsen et al. (2005) reported work on preservation of sugarcane juice using mixed dithiocarbamates. The study focused on use of mixtures containing methyldithiocarbamic acid sodium (Vapam) and ethylenebis (dithiocarbamic acid)

disodium salt (Nabam) on sugarcane juice. Biocides were applied at levels ranging from 5 to 20 mg/kg. It was shown that biocide applied at levels greater than 5 mg/kg could preserve sugar cane juice.

A study was done by Zaroni, et al (2005) to examine the quality and shelf-life of freshly squeezed, unpasteurized blonde and blood orange juice from organic farming methods. Thermal treatment had an effect on nutritional and sensory quality of blonde orange juice. Shelf-life studies revealed that unpasteurized juices could be stored at approximately 10°C for 10 days without significant growth of spoilage microorganisms. However decrease in ascorbic acid and anthocyanin content was observed during storage.

A study was done to determine the most suitable combination of preservative (sodium benzoate) and pasteurization levels for the preservation of pomegranate juice at room temperature. After extraction, the juice was submitted to various levels of pasteurization (i.e., no pasteurization, 60°C, 70°C and 80°C) and sodium benzoate treatment (i.e., 0, 400, 500 and 600 ppm). In juice samples treated with both pasteurization at 70°C and sodium benzoate at 500 ppm minimum changes were observed in all physicochemical parameters (Suryawanshi et al., 2008).

Grape juice was studied for one month at room temperature, after adding sodium benzoate and potassium sorbate (treatment T1: 0.1% sodium benzoate, T2: 0.2% potassium sorbate, T3: 0.1% sodium benzoate + 0.2% potassium sorbate, T4: control). Stability of ascorbic acid was found to be highest in T3, followed by T1, T2, and T4. Acidity and total soluble solids increased during storage. T1 obtained maximum score (8.0) for overall acceptability. T1 was found to be better for preserving nutritive and sensory values of the grape juice (Alam et al., 2009).

2.5 Microbiological and storage stability of juices

Hassandar et al. (1992) reported that heat treatment reduced the overall microbial load by 96.0 – 93.3% without any detrimental effect on physico-chemical characteristics of the juice.

Raw sugarcane juice was found to have 2.7×10^6 bacterial colonies per ml and 4.8×10^5 yeast and mould count per ml of sample. E coli count was found to be 4.99×10^4 cfu/ml (Nagalakshmi, 1995).

Chauhan et al. (2002) reported that microbiological population (total plate counts, and yeast and mold counts) increased during storage of sugarcane juice. The extent of increase in microbial population was also higher at room temperature as compared to refrigeration temperature. Highest counts were obtained during storage of pasteurized juice followed by pasteurization with addition of ascorbic acid followed by pasteurization with addition of citric acid. Addition of potassium metabisulphite to juice further reduced the counts appreciably. Raw sugarcane juice had 10 to 20 colonies of coliforms per 10 ml. The coliforms disappeared after pasteurization and no coliform could be detected throughout the storage.

Foley et al., (2002) reported that irradiation was less effective in reducing the yeast and mould population. The lower dose of 0.7 and 1.4 kGy had a minimal effect: while 3.5 and 4.0 kGy reduced the population by over 1 logs in fresh orange juice.

Oliveira et al., (2006) reported that fresh sugarcane juice sold by street vendors without any heat treatment showed thermo tolerant coliform levels were higher than allowed by Brazilian standards. Salmonella spp. and parasites were absent in all samples.

Solomon *et al.*, (2006), reported the sugarcane juice deterioration products including high invert sugars, polysaccharides (e.g. Dextran) and microbial contamination (e.g., ethanol and lactic acid formation) at the factory level.

Frazier and Westhoff, (2007) reported that sugarcane juice is widely spoiled by bacteria such as, Leuconostoc, Enterobacter, Flavobacterium, Micrococcus, Lactobacillus, Actinomyces. Among yeast and molds, Aspergillus,

Cladosporium, Monilla, Penicillium, Saccharomyces, Candidia, Pichia, Torulopsis are responsible for spoilage.

Subbannayya et al. (2007) reported that heat treatment reduces bacterial count in juice ranged from 10⁵ to 10⁷cfu/ml. Salmonella, Shigella and Vibrios were not present. However presence of E.coli, other coliforms and Enterococci indicated faecal contamination of juice.

Solomon, (2009) reported that sugarcane juice is highly fermentable, it contains about 15-18% sucrose, 0.5% reducing sugars and adequate amount of organic nitrogen and mineral salts for microbial growth. Its pH ranges from 5.0-5.5 making it selective for acidophilic microorganism especially yeast and lactic acid bacteria. Large population of yeast favors the ethanol production at the expense of sucrose. The microbial contamination of the juice is usually extremely high, typical viable counts being 10⁸-10⁹ cells/ml of juice. The major loss of sugar occurs due to inversion of sucrose in raw sugar cane juice and other types of degradation of the juice caused by bacterial activities, enzymes and other biological factors moreover spoilage depends on the type of cane used for juice extraction.

Tambekar et al. (2009) reported that out of 52 fruit juice samples analyzed in Amravati city, India, for presence of enteric bacterial pathogens. The dominant bacterial pathogen present in juices were *Escherichia coli* (40%), followed by *Pseudomonas aeruginosa* (25%), *Salmonella spp.* (16%), *Proteus spp.* (9%), *Staphylococcus aureus* (6%), *Klebsiella spp.* (3%) and *Enterobacter spp.* (1%). The highest bacterial contamination was observed in sweet lemon juice (35%), pineapple (29%), pomegranate, apple and orange (12%) each.

Similar study was done in Nagpur city where fruit and vegetable juice samples were analyzed for their microbiological quality. A total of 38 samples were analyzed for total viable count, total and fecal coliforms, staphylococci on mannitol salt agar and salmonella. The total viable count in all the fruit and vegetable samples were in the range of 2.0x10⁴– 4.6x10⁶. Almost 50% of the fruit

and vegetable juices also showed the presence of Salmonella (Titarmare et al., 2009).

From the preceding paragraphs, it can be reduced that the systematic information on the object of the present proposed programme of study is still quite meagre. However, there are tremendous opportunities to work for diversified uses at home and commercial scale. Therefore, the present investigation has been undertaken to provide the relevant knowledge in the selected field of study.

MATERIALS AND METHODS

This chapter deals with the various experimental techniques, materials and methodologies used to conduct studies on shelf life extension of sugarcane juice. The present research work was carried out in the Department of Food Science and Technology, College of Agriculture, Jawaharlal Nehru Krishi, Vishwa Vidyalaya, Jabalpur (M.P.) during the year 2017-2018.

3.1 Experimental materials

The freshly harvested, sugarcanes were procured from Jawaharlal Nehru Krishi Vishwa Vidyalaya research farm. The glass bottles were purchased from Priya Darshani Co-operative Store, Civic Centre, Jabalpur. All the chemicals used in present investigation were of standard analytical grades from BDH (India), E-Merck and Sarabhai, M. Company. The glassware's used in the present investigation were from Qualigens and Borosil Company respectively.

3.1.3 Equipments and machines

Following equipments and machines were used for conducting experiment.

S. No.	Name of equipment/machine	Source of supply
1.	Sugarcane juice extractor/ crusher	M/s D.K. Berry & Co. Pvt. Ltd. 11/35, West Punjabi Bagh, New Delhi, India
2.	Autoclave	M/S Ambica Sons Near marbel city Hospital Jabalpur
3	Referagerator	M/s Blue Star Ltd. Block-2A DLF corporate park DLF Qutab Enclave, Phase 3, Gurgaon, Haryana
4	juice and water analysis testing kit, model 191 E Delux	M/s Environmental and Scientific Instruments Corporation, Panchkula, Haryana, India
5	Hand refracto meter (ERMA)	M/s D.K. Berry & Co. (P) Ltd. 11/35 West Punjabi Bagh, New Delhi, India
6	Electronic balance TS 2005	M/s DHAUS Corporation, Made in USA

7	Citizen electronic balance	M/s Citizen scale (I) Pvt. Ltd. Daora (UT & DNN)
8	Sensory Evaluation Lab	Godrej Industries Limited. Pirojshanagar, Eastern Express Highway, Vikhroli, Mumbai-400079, India
9	Colorimeter spectronic-20	Bausch and Lomb (Pvt.) Ltd.
10	Utensils for handling and storage S.S. trays, glass jar, plastic Basket and bucket	M/s D.K. Berry & Co. (P) Ltd. 11/35 West Punjabi Bagh New Delhi. M/s J.K. Sales & promoters Gole Bazar, Write Town, Jabalpur (M.P.)

3.2 Experimental methods

3.2.1 Experimental plan and design

In accordance with the objectives of investigation the experiment was conducted in Complete Randomized Design (CRD) with three replications the details of treatment combinations are presented in Table 3.1

Table 3.1: Treatment combination of experiments to study effect of preservatives and pasteurization temperature for extension of shelf life of sugarcane juice

SNo	Treatment	Description
1	CA ₁ T ₁	0.02 % Citric acid and 60 °C Pasteurization
2	CA ₂ T ₁	0.04 % Citric acid and 60 °C Pasteurization
3	CA ₃ T ₁	0.06 % Citric acid and 60 °C Pasteurization
4	CA ₁ T ₂	0.02 % Citric acid and 70 °C Pasteurization
5	CA ₂ T ₂	0.04 % Citric acid and 70 °C Pasteurization
6	CA ₃ T ₂	0.06 % Citric acid and 70 °C Pasteurization
7	CA ₁ T ₃	0.02 % Citric acid and 80 °C Pasteurization
8	CA ₂ T ₃	0.04 % Citric acid and 80 °C Pasteurization
9	CA ₃ T ₃	0.06 % Citric acid and 80 °C Pasteurization
10	KMS ₁ T ₁	0.25 ppm Potassium metabisulphite and 60 °C Pasteurization

11	KMS ₂ T ₁	0.5 ppm Potassium meta bi sulphite and 60 °C Pasteurization
12	KMS ₃ T ₁	0.75 ppm Potassium meta bi sulphite and 60 °C Pasteurization
13	KMS ₁ T ₂	0.25 ppm Potassium meta bi sulphite and 70 °C Pasteurization
14	KMS ₂ T ₂	0.5 ppm Potassium meta bi sulphite and 70 °C Pasteurization
15	KMS ₃ T ₂	0.75 ppm Potassium meta bi sulphite and 70 °C Pasteurization
16	KMS ₁ T ₃	0.25 ppm Potassium meta bi sulphite and 80 °C Pasteurization
17	KMS ₂ T ₃	0.5 ppm Potassium meta bi sulphite and 80 °C Pasteurization
18	KMS ₃ T ₃	0.75 ppm Potassium meta bi sulphite and 80 °C Pasteurization
19	PS ₁ T ₁	0.15 % Potassium sorbate and 60 °C Pasteurization
20	PS ₂ T ₁	0.25 % Potassium sorbate and 60 °C Pasteurization
21	PS ₃ T ₁	0.5 % Potassium sorbate and 60 °C Pasteurization
22	PS ₁ T ₂	0.15 % Potassium sorbate and 70 °C Pasteurization
23	PS ₂ T ₂	0.25 % Potassium sorbate and 70 °C Pasteurization
24	PS ₃ T ₂	0.5 % Potassium sorbate and 70 °C Pasteurization
25	PS ₁ T ₃	0.15 % Potassium sorbate and 80 °C Pasteurization
26	PS ₂ T ₃	0.25 % Potassium sorbate and 80 °C Pasteurization
27	PS ₃ T ₃	0.5 % Potassium sorbate and 80 °C Pasteurization
28	AA ₁ T ₁	0.02 % Ascorbic acid and 60 °C Pasteurization
29	AA ₂ T ₁	0.04 % Ascorbic acid and 60 °C Pasteurization
30	AA ₃ T ₁	0.06 % Ascorbic acid and 60 °C Pasteurization
31	AA ₁ T ₂	0.02 % Ascorbic acid and 70 °C Pasteurization
32	AA ₂ T ₂	0.04 % Ascorbic acid and 70 °C Pasteurization
33	AA ₃ T ₂	0.06 % Ascorbic acid and 70 °C Pasteurization
34	AA ₁ T ₃	0.02 % Ascorbic acid and 80 °C Pasteurization
35	AA ₂ T ₃	0.04 % Ascorbic acid and 80 °C Pasteurization
36	AA ₃ T ₃	0.06 % Ascorbic acid and 80 °C Pasteurization

3.2.2 Processing of sugarcane juice for storage study

The fresh sugarcane juice was obtained after crushing sugarcanes using commercial sugarcane crusher. The juice was filtered and immediately bulk

packed in 10 lit capacity glass bottles for further use in experiments as per experimental plan (Table 3.1). The glass bottles used for storage study were thoroughly cleaned with warm water to remove adhering dust and dirt etc. The bottles were sterilized in autoclave at 15 psi (121⁰C) for 15 min according to procedure as described by Ranganna (1986). The sugarcane juice was also pasteurized separately and treated with different concentration of preservatives as per experiment plan and filled in 200 ml capacity pre sterilized bottle, corked, stored at refrigerated condition (5⁰C) and at ambient room temperature (31⁰C) for 07, 14, and 21 days.

3.3 Analytical Methods

The following methods were used for various determinations.

3.3.1 Chemical composition

3.3.1.1 Moisture

The moisture content in the sample was estimated according to the method of AOAC (1984).

The sample (5g) was taken in pre-weighed moisture box, dried at 105⁰C for 24hr in hot air oven, cooled in dessicator and weighed. The difference in weight of moisture box represents the moisture content of the sample.

Calculation

$$\text{Moisture (\%)} = \frac{\text{Difference in weight}}{\text{Weight of sample}} \times 100$$

3.3.1.2 Carbohydrates

Total carbohydrates in the samples were estimated by hydrolysis method as described in AOAC (1984).

Reagents:

1. Conc. HCl (AR sp.gr.1.25).

2. Fehling's solution:

Fehling's solution A: 34.64g of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ was dissolved in 500ml of distilled water.

Fehling's solution B: 173g of sodium potassium tartarate and 50g of sodium hydroxide were dissolved in 500ml of distilled water. The Fehling's solution was prepared by mixing the equal volume of solution A and solution B. It was prepared fresh daily.

3. Sodium hydroxide solution: 40% (w/v).

4. Methyl blue indicator: 0.1% (w/v) in 95% alcohol.

5. 3N HCl-68.18ml concentrated HCl was made up to 250ml with distilled water.

6. Dextrose 1% - 1g of dextrose was dissolved in 100ml distilled water.

Procedure

The sample (2.5g) was taken in the flask and suspended in 200ml of distilled water. Twenty ml of 3NHCl was added and refluxed in an air condenser for 3hr. On cooling, it was neutralized with alkali to pH 7.0, filtered and volume was made to 250ml with distilled water.

The total carbohydrates in the filtrate were determined by titrating it with Fehling's solution (A & B, 5ml each) using 1ml of methyl blue indicator. Factor was worked out by titrating 1% dextrose with Fehling's solution. In each titration Fehling's solution in the conical flask was heated with a constant flame and titration was done with filtrate in the burette until the end point (brick-red colour) was obtained. The total carbohydrate content was calculated as under.

Calculation

$$\text{Factor} = \frac{\text{Titre value} \times 2.5}{1000}$$

$$\text{Dextrose \%} = \frac{\text{Factor x 250}}{\text{Titrated value x Weight of sample}} \times 100$$

$$\text{Total carbohydrates (\%)} = \text{Dextrose \%} \times 0.9$$

3.3.1.3 Ash

The ash content in the sample was determined by procedure as describe in AOAC (1984). The sample (5g) was placed in pre-weighed crucible. It was burned on gas flame until it was completely charred. The samples were then placed in muffle furnace for combustion at 520°C for 5hr; their-after samples were cooled in desiccators and weighed. The heating in muffle furnace was repeated until constant weight was obtained. The ash content was calculated as under.

Calculation

$$\text{Ash (\%)} = \frac{\text{Initial weight of empty crucible and sample} - \text{Final weight of crucible with ash}}{\text{Weight of sample}} \times 100$$

3.3.1.4 Sugar

The sugars content of the sample was determined by the procedure as described by Ranganna (1991).

Reagents:

1. Fehling's solution A.
2. Fehling's solution B.
3. Methylene blue indicator.
4. 45% Neutral Lead Acetate solution: 225gm of neutral lead acetate were dissolved in distilled water and diluted to 500ml.

5. 22% Potassium Oxalate solution: 110 gm of potassium oxalate ($K_2C_2O_4 \cdot H_2O$) were dissolved in distilled water and make up to 500ml.

6. Standard invert sugar solution:

Weighed 9.5g of AR grade sucrose into 1 litre volumetric flask. Then added 100ml water and 5ml concentrated HCl. Allowed it to stand for 3 days at 20-25°C for inversion to take place and then make up to mark with water and its stable for several days. Pipette 25ml of standard invert sugar solution into 100ml volumetric flask and added 50ml water. Added few drops of phenolphthalein indicator and neutralized with 20% NaOH until pink. Then it was acidified with 1N HCl by adding it drop wise until 1 drop causes the pink to disappear. Make up to mark with water (1ml=2.5mg of invert sugar).

7. Standardization of Fehling's solution:

Mix 50ml solutions of Fehling A and 50ml of Fehling B. Pipette 10ml of mixed solution into 250ml conical flask and added 25ml water. Take invert sugar solution prepared by inversion of sucrose in 50ml burette. Add to the mixed Fehling solution almost the whole of the standard invert sugar solution (18-19ml) required to effect the reduction of copper, so that not more than 1ml will be required to complete titration. The flask was heated over a hot plate covered with asbestos filled wire gauze. When liquid began to boil, keep in moderate ebullition for 2min. On flame, added three drops of methylene blue and completed titration in 1min, over all 3 min and no interruption. End point was noted which was indicated by the discoloration of indicator (i.e. brick red).

$$\text{Factor for Fehling's solution (g of invert sugar)} = \frac{\text{Titre value} \times 2.5}{1000}$$

Preparation of sample:

25g of sample powder was taken and transferred to 250ml flask. Added 100ml water and neutralized with 1N NaOH and then filtered. 2ml of lead acetate solution was added, shaken well and allowed it to stand for 10 min. Added 2ml of

potassium oxalate solution to remove excess lead and make up the volume with water and filtered.

(A) Reducing sugar

Pipette 10ml of mixed Fehling solution into 250ml conical flask (5A and 5B). The burette was filled with the sample solution prepared. Then run into the flask almost the whole volume(15-50ml) of solution required to reduce the Fehling solution so that 0.5-1.0ml is require to later complete titration. Mixed the content and was heated to boiling and boiled moderately for 2min. Then added 3 drops of methylene blue and by not touching the sides. Titration was completed within 1min by adding 2-3 drops of sugar solution at 5-10sec intervals, until the indicator is completely decolourized from blue to brick red of cuprous oxide. Noted the volume of the solution required.

Note: End point was determined within 1 drop of sugar and not interrupting the boiling more than few seconds as the indicator undergoes back oxidation rapidly when air has free excess into the flask.

Calculation

$$\text{Reducing sugar \%} = \frac{\text{Factor} \times \text{Dilution} \times 100}{\text{Titer value} \times \text{Weight of sample}}$$

(B) Total sugar

50ml of the clarified solution was pipette into 250ml flask and added 5g of citric acid and 50ml of water. It was boiled gently for 10 min to complete inversion of sucrose, and then cooled. Transferred it to 250ml flask and neutralized with 1N NaOH using phenolphthalein and make up volume and was titrated with Fehling solution.

Calculation:

$$\text{Total sugar \%} = \frac{\text{Factor} \times \text{Dilution} \times 100}{\text{Titer value} \times \text{Weight of sample}}$$

Non reducing sugar % = Total sugar - Reducing sugar

3.3.1.5 Total soluble solids

The Total soluble solids (TSS) were measured with the standard methods of A.O.A.C. (2005), by Atago digital refractometer at room temperature. To calibrate the instrument, distilled water and test solution of known sucrose concentration was used. After each reading, a clean soft lint free material was used to clean the refractometer. A drop of sugarcane juice (Treated with chemical and fresh) sample kept onto the dry prism of refractometer, and readings were taken in ° brix, factors related to temperature was also added or removal of errors.

3.3.1.6 Acidity

Titration acidity content of samples was determined by simple acid-alkaline titration method as described in AOAC (1984). The samples were homogenized with 100ml distilled water. Now 25ml of diluted sample was transferred into a 250ml beaker. Three drops of phenolphthalein indicator was added to the solution. N/10 NaOH solution was taken in burette and titrated with alkali solution drop by drop with constant stirring till the pink end point is reached. End point readings were recorded and the percentage acidity was calculated in terms of citric acid by the following formula.

Calculation

$$\text{T.A. \%} = \frac{\text{Reading} \times \text{Normality of NaOH} \times \text{Equivalent weight of Citric acid}}{\text{Volume of sample taken} \times 1000} \times 100$$

3.3.1.7 Total Phenol

The total phenol content of samples was determined by procedure as described by Sadasivam and Manickam (1991).

Reagents:

1. 80% ethanol: (80ml ethanol + 20ml distilled water).

2. Folin - ciocalteau reagent.
3. 20% sodium carbonate (Na₂CO₃).
4. Standard - (100mg catechol in 100ml distilled water), dilute 10 times for a working standard.
5. Blank - (3ml distilled water + 0.5ml Folin reagent + 2ml Na₂CO₃).

Procedure

The sample (0.5 to 1.0g) was grind in pestle and mortar in 10 time volume of 80% ethanol. The aliquot was centrifuge and homogenized at 10,000 RPM for 20 min. The supernatant collected and residue was re-extracted five times with 80% ethanol. The supernatant was pooled and evaporated to dryness. The dried residue was further dissolved in a known volume of distilled water (5 to 10 ml). Form the above solution pipette out aliquots, 0.2ml into 20ml test tube. Now make up the volume in each test tube upto 3ml with distilled water (add 2.8ml distilled water). 3ml of folin reagent was taken and its volume was makeup to 9ml. From this solution add 0.5ml of folin reagent in to the test tube. After a period of 3 min, add 2ml of 20% Na₂CO₃ solution to each tube. Mix it thoroughly and place the tubes in boiling water for exactly 1 min. Cool it and measure the absorbance at 650 nm against a reagent blank. Also prepare a standard curve simultaneously using different concentration of catechol.

The phenol content was calculated from the standard curve prepared by catechol and express as mg of phenol/100g sample.

Calculation

$$\frac{\text{Concentration of std.}}{\text{Absorbance of std.}} \times \frac{\text{Abs.of sample}}{\text{Vol.of assay}} \times \frac{\text{Vol.make up}}{\text{Wt.of sample}} \times 100$$

3.3.1.8 Polyphenol oxidase (PPO) activity

Polyphenol oxidase (PPO) was determination according to the methods described by Zugustin M.A. and Ghazil H.M. (1985).

Reagents

1. Chilled acetone
2. 0.2M Potassium phosphate buffer, pH 6.8
3. 0.05M Catechol : Prepare in 0.2M phosphate buffer, pH 6.8

Method

For the PPO activity, the phosphate buffer (pH 6.0) and catechol were added to an aliquot of the sample and incubated at 30⁰C for 30 min. The reaction was then terminated with perchloric acid the absorbance was read in a spectrophotometer at 395 nm. A blank absorbance of sugarcane juice added to the phosphate buffer solution (Substitute for the reagents) was determined

Enzyme activity was expressed In U/mL with one unit equivalent to variation of 0.001 absorbance per min. per mL of sample. The eq. 1st was applied to calculate the enzyme activity.

$$\text{Activity (U/ml)} = \frac{(\text{Ab}_{\text{sample}} - \text{Ab}_{\text{blank}})}{0.001 \times t}$$

Where Ab sample is the sample absorbance; AB blank is the blank absorbance; and t the incubation time of sample with reagents (min.)

3.3.2 Microbiological analysis of sugarcane juice

3.3.2.1 Total Plate Count

The total plate count in stored sugarcane juice was determined by following the standard procedure as described by Ranganna (1986)

3.4.3 Sensory evaluation of stored sugarcane juice

A semi trained panel of 10 members was given samples of fresh and stored sugarcane juice to evaluate the sensory quality attributes viz.; colour/appearance, taste, flavor and overall acceptability. The evaluation of

sensory attributes was carried out according to the method of Amerine *et al.* (1965) on 9 point hedonic scale as given in Appendix- I.

3.5 Statistical analysis of data

The experiment was conducted in complete randomized design (CRD) using WASP, Version 1.0 (<http://icargoa.res.in/wasp>). The data was subjected to one-way analysis of variance (ANOVA) at 5% level of significance. The skeleton of ANOVA for complete randomized design (CRD) is presented in table given below:

Skeleton of Analysis of variance

Source of variation	d.f.	S.S.	M.S.S.	F cal.	F tabulated	Value at
					5%	1%
Treatments	t-1 (24)	SSt	MSSt	MSSt/MSSe		
Error	n-t (25)	SSe	MSSe			
Total	n-1 (49)					

n = Total number of observations

t = Number of treatments

The 'F' test was applied to check the overall significance of various treatments in general and comparison of individual treatment was made with the help of critical difference at 5% level of significance, which was calculated as given below:

$$\text{SEm} \pm \text{ for treatment } t = \sqrt{\frac{\text{MSSe}}{\text{No. of replication}} \times 100}$$

$$\text{SEd for treatments} = \text{SEm} \times \sqrt{2}$$

CD for treatments = SEd x 't' value at 5% error degree of freedom.

Where,

SEm ± = Standard Error of treatment means

SEd = Standard Error of difference between two treatments

CD = Critical difference

The results are also presented through bar diagram to show the effect of variables on selected response during storage. The findings of experiments are presented in subsequent chapter.

RESULT

The present investigation was planned to enhance the shelf life of sugarcane juice using different chemical preservatives (citric acid, potassium metabisulfite, potassium sorbate, ascorbic acid) and pasteurization temperature. The storage study was conducted for 7, 14 and 21 days at refrigerated and ambient condition. The findings of various experiments are presented through table 1 to 32 and the results are explained as under:

4.1 Chemical Composition of sugarcane juice

The fresh sugarcane cultivar, CO 99004 was procured from JNKVV research farm. The sugarcane was extracted and evaluated for various quality attributes. The findings are presented in Table 1. A perusal of table indicates that juice yield of sugarcane variety was 59 percent. The proximate composition shows that sugarcane juice contain moisture (80%) Protein (.55%) fat (0.19%) crude fibre (13.32%) and ash (0.30%).The cultivar is rich source of sugars. The total soluble sugar TSS of juice was 19 % and 20 Brix. On the basis of quality attributes variety CO99004 grown in Madhya Pradesh was selected for experimental purpose.

4.2 Effect of citric acid and pasteurization temperature on chemical composition of stored sugarcane juice

4.2.1 Moisture, Carbohydrate and Total ash content stored sugarcane juice at refrigerated condition.

A cursory view of table 2 reveals the results of effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition. The moisture content in different treatment combinations of stored sugarcane juice for 7,14 and 21 days was found to be 75.4% (C₂T₁, C₃T₁) to 80.88 (C₃T₃); 76.75(C₃T₂) to 85.6 (C₃T₃); 76.4 (C₂T₁) to 80.79(C₃T₃) respectively. Similarly the lowest and highest values of carbohydrate were found to be 13.68% (C₃T₂) to 16.20% (C₂T₁) for 7 days, 13.73% (C₁T₃) to 16.27% (C₂T₃) for 14 days and 13.82% (C₃T₁) to 17.23% (C₃T₃) for 7,14 and 21 days of storage period respectively. The highest and lowest values of ash content in stored sugarcane juice were

Table 1 Chemical Composition of sugarcane juice.

S. No	Characteristics	Variety: CO 99004
1.	Juice yield%	59
2.	Moisture%	80.00
3.	Protein%	0.55
4.	Fat %	0.19
5.	Ash %	0.30
6.	Crude Fibre%	13.32
7.	Total soluble solid	20.00
8.	Total Sugar%	19.00
9.	Acidity %	0.30

Table 2 Effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (5° C)

Treatments	Storage period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
CA₁T₁	75.8	14.89	0.2	80.5	14.41	0.6	76.72	16.01	0.4
CA₂T₁	75.4	16.20	0.4	80.5	16.20	0.2	76.4	14.21	0.2
CA₃T₁	75.4	14.78	0.4	80.12	14.84	0.3	78.58	13.82	0.2
CA₁T₂	76	14.36	0.2	81.43	14.78	0.4	80.12	17.16	0.4
CA₂T₂	79	13.97	0.4	81.43	14.84	0.4	80.4	14.01	0.4
CA₃T₂	76.4	13.68	0.2	76.75	15	0.3	76.72	16.46	0.6
CA₁T₃	76.7	14.78	0.6	80.12	13.73	0.6	80.5	13.87	0.4
CA₂T₃	78.8	14.94	0.4	82.81	16.27	0.4	80.12	13.92	0.2
CA₃T₃	80.88	14.67	0.2	85.6	13.78	0.2	80.79	17.23	0.4
CD@5%	0.05	0.08	0.28	0.10	0.11	0.25	0.01	0.03	0.26
SEM±	0.02	0.23	0.10	0.30	0.04	0.09	0.03	0.01	0.09

observed in treatment combinations as follows: C₁T₃ (0.6%) & C₁T₁, C₁T₂, C₃T₂, C₃T₃ (0.2%) for 7 days; C₁T₁, C₁T₃ (0.6%) & C₂T₁, C₃T₃ (0.2%) for 14 days; C₃T₂ (0.6%) & C₂T₁, C₃T₁, C₂T₃ (0.2%) for 21 days.

The statistical analysis data indicates that treatments have significant effect on moisture carbohydrate and ash content of stored sugarcane juice. Treatment CA₃T₃ was at par to CA₂T₂ followed by CA₂T₃ as regards for moisture content of juice. Similarly in carbohydrate and ash content treatment CA₂T₁ was at par CA₂T₃ followed by CA₁T₁, CA₁T₃ and CA₂T₁, followed by CA₁T₁ respectively, in 7 days storage of period.

On 14 days of storage period for moisture content treatment CA₃T₃ was at par to CA₂T₃ followed by CA₁T₂. Similarly in carbohydrate and ash content treatment CA₂T₃ was at par with CA₂T₁ and CA₃T₁; treatment CA₁T₁ was at par to CA₁T₂ followed by CA₃T₁ respectively.

Finally on 21 days of storage it was seen that treatment CA₃T₃ was at par to CA₁T₂ and CA₁T₃ for moisture; treatment CA₃T₃ was at par to CA₁T₂ followed by CA₃T₂ for carbohydrate; treatment CA₃T₂ was at par with CA₁T₁ followed by CA₂T₁ for ash content in stored sugarcane juice.

4.2.2 Moisture, Carbohydrate and Total ash content stored sugarcane juice at ambient condition.

Table 3 comprises the results of effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at ambient condition. The minimum and maximum value of moisture content was seen at treatments CA₁T₃ (76.30), & CA₃T₃ (81.70); CA₃T₂ (76.77) & CA₃T₃ (84.35); CA₁T₁ (76.10) & CA₂T₃ (80.79) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of carbohydrate were found in treatments CA₃T₂ (13.73) & CA₃T₃ (15.76) for 7 days, CA₁T₁ (13.64) & CA₂T₂ (17.16) for 14 days and CA₂T₁ (17.23) & CA₁T₃ (18.16) for 21 days. Finally, for ash content the higher and lower values were as follows: CA₂T₁, CA₂T₂ and CA₃T₂ (0.4) & CA₃T₁, CA₁T₂, CA₃T₂, (0.2) for 7 days; CA₁T₁, CA₂T₁, CA₂T₂, CA₃T₂ CA₁T₃ & CA₂T₃ (0.4) & CA₃T₁, CA₁T₂ CA₃T₃ (0.2) for 14 days; CA₁T₃, CA₂T₃, (0.4) & CA₁T₁, CA₂T₁, CA₁T₂, CA₂T₂, CA₃T₂, CA₃T₃ (0.2) for 21 days.

Table 3 Effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at ambient condition (30° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
CA₁T₁	76.91	13.97	0.3	80.80	13.64	0.4	76.10	17.23	0.2
CA₂T₁	76.80	14.16	0.4	80.54	13.78	0.4	76.50	13.78	0.2
CA₃T₁	76.62	13.78	0.2	81.80	14.01	0.2	80.50	15.76	0.3
CA₁T₂	77.20	13.92	0.2	80.87	13.68	0.2	80.61	14.26	0.2
CA₂T₂	77.70	14.52	0.4	81.57	17.16	0.4	80.50	17.84	0.2
CA₃T₂	76.50	13.73	0.4	76.77	14.11	0.4	80.50	13.78	0.2
CA₁T₃	76.30	14.31	0.3	81.12	14.41	0.4	80.40	18.16	0.4
CA₂T₃	78.41	13.82	0.3	82.54	14.16	0.4	81.40	14.01	0.4
CA₃T₃	81.70	15.76	0.2	84.35	16.40	0.2	80.50	16.40	0.2
CD@5%	2.80	2.56	0.05	2.660	2.41	0.05	2.36	2.23	0.04
SEM±	0.96	0.88	0.02	0.92	0.83	0.02	0.81	0.77	0.01

The table 3 indicates that treatment CA₃T₃ was at par to CA₂T₃ followed by CA₂T₂ in moisture, similarly in carbohydrate and Ash content treatment CA₃T₃ was at par CA₂T₂ followed by CA₁T₃; CA₂T₁ was at par with CA₁T₁, followed by CA₃T₁ respectively, in 7 days of storage period.

On 14 days of storage period for moisture treatment CA₃T₃ was at par to CA₂T₃ followed by CA₃T₁; in carbohydrate treatment CA₂T₂ was at par with CA₃T₃ and CA₁T₃; treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁ in ash content.

Finally on 21 days of storage it was seen that treatment CA₂T₃ was at par to CA₁T₂ and CA₃T₁ for moisture; for carbohydrate CA₁T₃ was at par to CA₂T₂ followed by CA₁T₁; in ash treatment CA₁T₃ was at par with CA₃T₁ followed by CA₁T₁.

4.2.3 Total sugar, Reducing sugar and Titrable acidity of stored sugarcane juice at refrigerated condition.

A perusal of table 4 shows the effect of citric acid and pasteurization temperature was seen on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition. The minimum and maximum value of total sugar content was seen in treatments 19.91 (CA₁T₂, CA₃T₂), & 16.78 (CA₁T₃, CA₃T₃); 19.86 (CA₁T₂, CA₃T₂) & 16.71 (CA₃T₃); 19.81 (CA₁T₂, CA₃T₂) & 16.65 (CA₃T₃) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of reducing sugar were found in treatments CA₂T₂ (0.242) & CA₁T₂ (0.204) for 7 days, CA₃T₁ (0.364) & CA₁T₂ (0.258) for 14 days and CA₂T₁ (0.679) & CA₃T₁ (0.399) for 21 days. The higher and lower values of titrable acidity were found in treatments to be CA₁T₁ (0.26) & CA₂T₂ (0.53) and for 7 days; CA₁T₁, (0.35) & CA₂T₂, (0.62) for 14 days; CA₂T₂ CA₂T₃ (0.70) & CA₁T₁ (0.42) for 21 days.

The findings also indicates that treatment CA₁T₂ was at par to CA₂T₂ followed by CA₂T₃ in total sugar, similarly in case of reducing sugar and titrable acidity, treatment CA₂T₂ was at par CA₂T₃ followed by CA₃T₂; CA₂T₁ was at par with CA₁T₁, followed by CA₃T₁ respectively, in 7 days storage of period.

Table 4 Effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
CA₁T₁	18.11	0.240	0.26	18.06	0.334	0.35	18.01	0.626	0.42
CA₂T₁	18.11	0.220	0.46	18.04	0.279	0.55	18.00	0.679	0.63
CA₃T₁	17.53	0.215	0.47	17.47	0.364	0.56	17.42	0.399	0.63
CA₁T₂	19.91	0.204	0.44	19.86	0.258	0.53	19.81	0.543	0.61
CA₂T₂	19.49	0.264	0.53	19.43	0.318	0.62	19.38	0.599	0.70
CA₃T₂	19.91	0.242	0.49	19.86	0.351	0.58	19.81	0.443	0.66
CA₁T₃	16.78	0.211	0.37	16.72	0.283	0.46	16.67	0.509	0.61
CA₂T₃	18.17	0.247	0.49	18.11	0.291	0.58	18.05	0.485	0.70
CA₃T₃	16.78	0.227	0.40	16.71	0.275	0.49	16.65	0.438	0.66
CD@5%	2.46	0.16	21.15	2.20	0.24	0.23	2.33	0.25	0.34
SEM±	0.85	0.06	7.27	0.76	0.08	0.08	0.80	0.09	0.12

On 14 days of storage period as regards for total sugar, treatment combination CA₂T₂ was at par to CA₃T₂ followed by CA₃T₁; treatment CA₃T₁ was at par with CA₃T₂ and CA₁T₁; treatment CA₂T₂ was at par to CA₃T₂ followed by CA₃T₁ in titrable acidity.

On 21 days of storage period it was seen that treatment CA₁T₂ was at par to CA₂T₂ and CA₂T₃ for total sugar); CA₂T₁ was at par to CA₁T₁ followed by CA₁T₂ for reducing sugar; CA₂T₂ was at par with CA₃T₂ followed by CA₂T₁ for titrable acidity respectively.

4.2.4 Total sugar, Reducing sugar and Titrable acidity of stored sugarcane juice at ambient condition.

Table 5 comprises the results of effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition. The minimum and maximum value of total sugar content was seen at treatments CA₂T₁ (22.62) & CA₁T₁ (17.86); CA₁T₁ (22.56) & CA₁T₁ (17.81); CA₂T₁ (22.52) & CA₁T₁ (17.77) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of reducing sugar were found in treatments CA₁T₁ (0.271) & CA₂T₂ (0.204) for 7 days, CA₂T₂ (0.370) & CA₁T₃ (0.273) for 14 days and CA₁T₃ (0.679) & CA₂T₃ (0.424) for 21 days storage period. The higher and lower values titrable acidity were as follows: CA₃T₃ (0.53) & CA₁T₁ (0.28) and for 7 days; CA₃T₃, (0.62) & CA₁T₁, (0.37) for 14 days; CA₃T₃ (0.70) & CA₁T₁ (0.45) for 21 days.

The statistical analysis of reveals that treatment CA₂T₁ was at par to CA₃T₁ followed by CA₃T₃ in case of total sugar, similarly in reducing sugar and titrable acidity treatment CA₁T₁ was at par CA₃T₁ followed by CA₃T₃; CA₃T₃ was at par with CA₁T₃, followed by CA₁T₂ respectively, in 7 days storage of period.

On 14 days of storage period for total sugar, treatment CA₂T₁ was at par to CA₃T₁ followed by CA₃T₃; in case of reducing sugar CA₂T₂ was at par with CA₃T₂ and CA₁T₁ and treatment CA₃T₃ was at par to CA₁T₃ followed by CA₁T₂ in titrable acidity.

On 21 days of storage period, it was seen that treatment CA₂T₁ was at par to CA₁T₃ and CA₃T₁ for total sugar; for reducing sugar CA₂T₁ was at par to

Table 5 Effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
CA₁T₁	17.86	0.271	0.28	17.81	0.318	0.37	17.77	0.590	0.45
CA₂T₁	22.62	0.219	0.45	22.56	0.287	0.54	22.52	0.657	0.62
CA₃T₁	20.49	0.264	0.43	20.44	0.313	0.52	20.39	0.574	0.60
CA₁T₂	18.75	0.242	0.50	18.69	0.302	0.59	18.63	0.485	0.67
CA₂T₂	18.75	0.204	0.49	18.7	0.370	0.58	18.65	0.515	0.66
CA₃T₂	18.11	0.211	0.39	18.05	0.351	0.48	18	0.668	0.54
CA₁T₃	20.49	0.235	0.51	20.44	0.273	0.60	20.4	0.679	0.68
CA₂T₃	19.84	0.226	0.42	19.79	0.291	0.51	19.73	0.424	0.59
CA₃T₃	20.24	0.255	0.53	20.2	0.299	0.62	20.16	0.468	0.70
CD@5%	2.13	0.10	0.27	2.51	0.20	24.61	1.87	0.23	0.21
SEM±	0.73	0.04	0.09	0.86	0.07	8.46	0.64	0.08	0.07

CA₁T₁ followed by CA₁T₂ and in titrable acidity treatment CA₂T₂ was at par with CA₃T₂ followed by CA₂T₁.

4.2.5 Poly phenol oxidase, Total soluble solid and Phenol content stored sugarcane juice at refrigerated condition

Table 6 presents the results of effect of citric acid and pasteurization temperature on poly phenol oxidase, total soluble solid and phenol content of stored sugarcane juice at refrigerated condition. The minimum and maximum value of poly phenol oxidase activity was seen in treatments CA₁T₁, CA₁T₂ (0.035) & CA₃T₃ (0.027); CA₂T₁, CA₁T₂, CA₂T₂ (0.023) & CA₂T₃ (0.018); CA₁T₁ (0.024) & CA₂T₃ (0.015) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of total soluble solid were found in treatments CA₃T₃ (20.5) & CA₁T₁ (19.1) for 7 days, CA₂T₃ (19.8) & CA₁T₁ (18.8) for 14 days and CA₂T₃ (19.6) & CA₁T₁ (18.5) for 21 days. The higher and lower values phenol content were found to be: CA₁T₂ (14.76) & CA₂T₁ (14.55) and for 7 days; CA₁T₂, (15.16) & CA₂T₁, (14.95) for 14 days and CA₁T₂ (15.56) & CA₁T₁ (15.4) for 21 days.

The statistical analysis shows that treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁ in case of poly phenol oxidase activity, similarly in case of total soluble solid and phenol content treatment, CA₃T₃ was at par CA₂T₃ followed by CA₃T₁ and CA₁T₂ was at par with CA₃T₁, followed by CA₂T₂ respectively, in 7 days storage of period.

On 14 days of storage period for poly phenol oxidase activity treatment, CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁; in CA₂T₃ was at par with CA₃T₁ and CA₃T₂ and treatment CA₁T₂ was at par to CA₂T₃ followed by CA₁T₃ in phenol content.

Finally on 21 days of storage it was seen that treatment CA₁T₁ was at par to CA₂T₁ and CA₃T₁ for poly phenol oxidase activity; for total soluble solid CA₂T₃ was at par to CA₃T₁ followed by CA₃T₂ and in phenol content treatment CA₁T₂ was at par with CA₁T₃ followed by CA₃T₁.

Table 6 Effect of citric acid and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
CA₁T₁	0.035	19.1	14.60	0.025	18.8	15	0.024	18.5	15.4
CA₂T₁	0.034	19.5	14.55	0.023	19.2	14.95	0.022	18.9	15.35
CA₃T₁	0.033	19.8	14.65	0.022	19.5	15.05	0.021	19.2	15.45
CA₁T₂	0.035	19.7	14.76	0.023	19.5	15.16	0.022	19.2	15.56
CA₂T₂	0.031	19.5	14.63	0.023	19.2	15.03	0.021	18.8	15.43
CA₃T₂	0.030	19.5	14.67	0.022	19.3	15.07	0.020	19	15.47
CA₁T₃	0.034	19.8	14.74	0.019	19.5	15.08	0.017	19.3	15.54
CA₂T₃	0.030	20.2	14.63	0.018	19.8	15.14	0.015	19.6	15.43
CA₃T₃	0.027	20.5	14.65	0.019	19.2	15.03	0.017	18.9	15.45
CD@5%	0.02	2.51	2.35	0.04	2.42	2.23	0.05	1.84	2.15
SEM±	0.01	0.86	0.81	0.01	0.83	0.77	0.02	0.63	0.74

4.2.6 Poly phenol oxidase, Total soluble solid and Phenol content stored sugarcane juice at ambient condition

The results of effect of citric acid and pasteurization temperature on poly phenol oxidase activity, total soluble solid and phenol content of stored sugarcane juice at ambient condition are presented in table 7. The minimum and maximum value of poly phenol oxidase activity was seen in treatments CA₁T₁, CA₁T₂ (0.055) & CA₃T₃ (0.047); CA₁T₁ (0.045), & CA₃T₃ (0.031); CA₁T₁ (0.039) & CA₃T₃ (0.026) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of total soluble solid were found in treatments CA₂T₃ (19.8) & CA₁T₁ (18.8) for 7 days, CA₂T₃ (19.6) & CA₁T₁ (18.6) for 14 days and CA₁T₂, CA₂T₃ (19.3) & CA₁T₁ (18.3) for 21 days. The highest and lowest content of phenol were found in treatment combination to be: CA₁T₂ (14.36) & CA₂T₁ (14.15) and for 7 days; CA₁T₂, (14.76) & CA₂T₁, (14.55) for 14 days; CA₁T₂ (15.16) & CA₂T₁ (14.95) for 21 day.

The treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁ in Poly phenol oxidase, similarly in total soluble solid and phenol content treatment CA₃T₃ was at par CA₂T₃ followed by CA₃T₁; CA₁T₂ was at par with CA₃T₁, followed by CA₂T₂ respectively, in 7 days storage of period (Table 7).

On 14 days of storage period for poly phenol oxidase activity treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁; in CA₂T₃ was at par with CA₃T₁ and CA₃T₂ and treatment CA₁T₂ was at par to CA₂T₃ followed by CA₁T₃ in phenol content.

Finally on 21 days of storage period, it was seen that treatment CA₁T₁ was at par to CA₂T₁ and CA₃T₁ for poly phenol oxidase activity; for total soluble solid CA₂T₃ was at par to CA₃T₁ followed by CA₃T₂; in phenol content treatment CA₁T₂ was at par with CA₁T₃ followed by CA₃T₁.

Table 7 Effect of citric acid and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30°C)

Treatment	Storage period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
CA₁T₁	0.055	18.8	14.20	0.045	18.6	14.60	0.039	18.3	15
CA₂T₁	0.054	19.2	14.15	0.043	19	14.55	0.034	18.7	14.95
CA₃T₁	0.053	19.5	14.25	0.042	19.3	14.65	0.034	19	15.05
CA₁T₂	0.055	19.7	14.36	0.043	19.5	14.76	0.031	19.3	15.16
CA₂T₂	0.051	19.5	14.23	0.043	19.2	14.63	0.030	18.9	15.03
CA₃T₂	0.05	19.5	14.27	0.042	19.3	14.67	0.028	19	15.07
CA₁T₃	0.054	19.5	14.34	0.038	19.2	14.74	0.029	18.8	15.14
CA₂T₃	0.05	19.8	14.23	0.035	19.6	14.63	0.029	19.3	15.03
CA₃T₃	0.047	19.2	14.25	0.031	19	14.65	0.026	18.6	15.05
CD@5%	0.03	1.05	0.44	0.02	1.05	0.52	0.02	1.07	0.41
SEM±	0.01	0.36	0.15	0.01	0.36	0.18	0.01	0.37	0.14

4.3 Effect of Potassium metabisulfite and pasteurization temperature on chemical composition of stored sugarcane juice

4.3.1 Moisture, Carbohydrate and Total ash content: of stored sugarcane juice at refrigerated condition

The effect of potassium meta bisulphite and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition is presented in table 8. The maximum and minimum value of moisture content was seen at treatments KMS_3T_3 (81.43) & KMS_1T_1 (75.2); KMS_3T_3 (85.4), & KMS_3T_2 (77.5); KMS_2T_3 (82.2) & KMS_1T_1 (76.4) for 7, 14 and 21 days storage respectively. Similarly the highest and lowest values of carbohydrate were found in treatments KMS_1T_2 (16.74) & KMS_2T_2 (13.73) for 7 days, KMS_3T_3 (17.02) & KMS_3T_2 (13.59) for 14 days and KMS_3T_1 , (16.40) & KMS_3T_2 (13.92) for 21 days. The total ash content were higher and lower in the following treatment combinations: KMS_2T_1 , KMS_1T_2 KMS_3T_2 KMS_1T_3 & KMS_3T_3 (0.4) and KMS_1T_1 , KMS_2T_3 (0.2) for 7 days; KMS_1T_2 , (0.6) & KMS_3T_1 , KMS_3T_3 (0.2) for 14 days; KMS_1T_1 , KMS_3T_1 , KMS_1T_2 , KMS_2T_2 , KMS_1T_3 & KMS_2T_3 (0.4) & KMS_2T_1 KMS_3T_3 (0.2) for 21 day.

The statistical analysis data indicates that treatments have got significant effect on moisture carbohydrate and ash content of stored sugarcane juice. Treatment KMS_3T_3 was at par to KMS_2T_3 followed by KMS_2T_3 as regards for moisture content of juice. Similarly in carbohydrate and ash content treatment KMS_1T_2 was at par KMS_3T_3 followed by KMS_2T_1 , KMS_2T_1 and KMS_1T_2 , followed by KMS_1T_1 respectively, in 7 days storage of period.

On 14 days of storage period for moisture content treatment KMS_3T_3 was at par to KMS_2T_3 followed by KMS_2T_2 . Similarly in carbohydrate and total ash content treatment KMS_3T_3 was at par with KMS_2T_3 and KMS_1T_2 ; treatment KMS_1T_2 was at par to KMS_1T_1 followed by KMS_2T_1 respectively.

On 21 days of storage it was seen that treatment combination KMS_2T_3 was at par to KMS_2T_2 and KMS_3T_2 for moisture; treatment combination KMS_3T_1 was at par to KMS_1T_3 followed by KMS_1T_2 for carbohydrate and

Table 8 Effect of Potassium metabisulfite and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
KMS₁T₁	75.2	14.06	0.2	81.43	14.26	0.4	76.4	14.36	0.4
KMS₂T₁	75.82	14.46	0.4	78.58	14.36	0.3	76.47	14.46	0.2
KMS₃T₁	75.61	14.21	0.3	80.2	14.41	0.2	79	16.40	0.3
KMS₁T₂	76.7	16.74	0.4	80.5	16.27	0.6	80.01	15.06	0.4
KMS₂T₂	78.52	13.73	0.3	81.66	13.97	0.4	80.79	15.00	0.4
KMS₃T₂	76.2	13.97	0.4	77.5	13.59	0.3	80.12	13.92	0.3
KMS₁T₃	76.54	13.97	0.4	80.79	15.06	0.4	80.15	16.01	0.4
KMS₂T₃	78.4	14.31	0.2	82.7	16.88	0.4	82.2	14.21	0.4
KMS₃T₃	81.43	15.58	0.4	85.4	17.02	0.2	80.12	14.41	0.2
CD@5%	2.83	2.10	0.20	2.63	2.28	0.06	0.06	0.09	0.04
SEM±	0.97	0.72	0.07	0.91	0.78	0.02	0.02	0.03	0.01

treatment combination KMS_1T_1 was at par with KMS_3T_2 followed by KMS_2T_1 for ash content in stored sugarcane juice.

4.3.2 Moisture, Carbohydrate and Total ash content: of stored sugarcane juice at ambient condition

The effect of potassium meta bisulphite and pasteurization temperature treatment combination on moisture, carbohydrate and ash content of stored sugarcane juice at ambient condition is presented in table 9. The maximum and minimum value of moisture content was seen in treatments KMS_3T_3 (81.74) & KMS_1T_2 (76.32); KMS_3T_3 (85.10) & KMS_3T_2 (77.40); KMS_2T_3 (81.75) & KMS_1T_1 (76.50) for 7, 14 and 21 days respectively. Similarly the highest and lowest values of carbohydrate were found in treatments KMS_1T_2 (17.23) & KMS_2T_1 , KMS_2T_2 , KMS_2T_3 (13.78) for 7 days, KMS_2T_1 (16.60) & KMS_1T_1 , KMS_2T_3 , (13.78) for 14 days and KMS_3T_2 , (18.33) & KMS_1T_1 (13.73) for 21 days. Finally, for ash content the higher and lower values were as follows: KMS_1T_2 , KMS_2T_2 , KMS_1T_3 (0.4) and KMS_2T_1 , KMS_3T_1 , KMS_3T_2 and KMS_3T_3 (0.2) for 7 days; KMS_2T_2 , KMS_3T_2 , KMS_1T_3 (0.4) & KMS_1T_1 , KMS_3T_1 , KMS_1T_2 (0.2) for 14 days; KMS_3T_2 , (0.6) & KMS_2T_1 , KMS_3T_1 , KMS_2T_2 , KMS_1T_3 , KMS_3T_3 (0.2) for 21 day.

The statistical analysis data indicates that significant difference exist among the treatment combination for moisture, carbohydrate and ash content of stored sugarcane juice. Treatment combination KMS_3T_3 was at par to KMS_2T_3 followed by KMS_2T_2 as regards for moisture content of juice. Similarly in carbohydrate and ash content treatment KMS_1T_2 was at par KMS_1T_3 followed by KMS_1T_1 , KMS_1T_2 and KMS_2T_2 followed by KMS_1T_1 respectively, in 7 days storage of period.

On 14 days of storage period treatment combination KMS_3T_3 was at par to KMS_2T_3 followed by KMS_1T_3 for moisture content. Similarly in carbohydrate and ash content of juice treatment combination KMS_2T_1 was at par with KMS_1T_3 and KMS_2T_2 ; treatment combination KMS_2T_2 was at par to KMS_2T_1 followed by KMS_1T_1 respectively.

Upon 21 days of storage it was seen that treatment combination KMS_2T_3 was at par to KMS_1T_3 and KMS_3T_2 for moisture; treatment KMS_3T_2

Table 9 Effect of Potassium metabisulfite and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (30 °C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
KMS₁T₁	76.87	14.57	0.3	80.40	13.78	0.2	76.50	13.73	0.4
KMS₂T₁	76.54	13.78	0.2	81.20	16.60	0.3	77.11	17.61	0.2
KMS₃T₁	76.50	14.16	0.2	81.24	13.97	0.2	80.37	14.16	0.2
KMS₁T₂	76.32	17.23	0.4	81.21	14.16	0.2	80.50	16.33	0.4
KMS₂T₂	77.81	13.78	0.4	82.00	14.26	0.4	80.10	14.31	0.2
KMS₃T₂	76.48	14.52	0.2	77.40	14.11	0.4	81.21	18.33	0.6
KMS₁T₃	76.54	16.67	0.4	81.41	14.67	0.4	81.40	14.11	0.2
KMS₂T₃	78.90	13.78	0.3	82.61	13.78	0.3	81.75	14.67	0.4
KMS₃T₃	81.74	14.06	0.2	85.10	13.92	0.3	81.21	14.52	0.2
CD@5%	1.87	1.08	0.04	1.87	1.87	0.04	2.63	2.61	0.05
SEM±	0.64	0.37	0.01	0.64	0.64	0.01	0.91	0.09	0.02

was at par to KMS_2T_1 followed by KMS_1T_2 for carbohydrate and for ash content in stored sugarcane juice treatment KMS_3T_2 was at par with KMS_1T_1 , KMS_1T_2 followed by KMS_2T_1 .

4.3.3 Total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition:

Table 10, comprises the results of effect of potassium meta bisulphite and pasteurization temperature treatment combination on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition. The minimum and maximum value of total sugar content for 7, 14 and 21 days were found for treatments KMS_1T_3 (16.78) & KMS_2T_1 (22.62); KMS_1T_3 (16.73) & KMS_2T_1 (22.56); KMS_1T_3 (16.69) & KMS_2T_1 (22.5) respectively. Similarly the lowest and highest values of reducing sugar were found in treatments KMS_2T_3 (0.209) & KMS_1T_2 (0.258) for 7 days, KMS_3T_1 (0.266) & KMS_3T_2 (0.370) for 14 days and KMS_1T_1 (0.374) & KMS_2T_2 (0.657) for 21 days storage period. The higher and lower values of titrable acidity were as follows: KMS_1T_3 (0.51) & KMS_1T_1 (0.27) for 7 days; KMS_1T_3 , KMS_3T_3 (0.60) & KMS_1T_1 (0.36) for 14 days; KMS_1T_3 (0.68) & KMS_1T_2 (0.43) for 21 days.

The statistical analysis data indicates that significant difference exist among the treatment combination. The treatment KMS_2T_1 was at par to KMS_1T_1 , KMS_3T_2 followed by KMS_2T_2 in case of total sugar, similarly in reducing sugar and titrable acidity, treatment KMS_1T_2 was at par to KMS_2T_2 followed by KMS_2T_1 ; KMS_1T_3 , KMS_3T_3 collectively were at par with KMS_2T_2 , followed by KMS_1T_2 respectively, in 7 days storage of period.

On 14 days of storage period for total sugar, treatment KMS_2T_1 was at par to KMS_3T_2 followed by KMS_1T_1 ; in case of reducing sugar KMS_3T_2 was at par with KMS_2T_2 followed by KMS_1T_2 and the treatments KMS_1T_3 and KMS_3T_3 were collectively at par to KMS_2T_2 followed by KMS_1T_2 in case of titrable acidity.

On 21 days of storage period, it was found that treatment KMS_2T_1 was at par to KMS_3T_2 and KMS_1T_1 for total sugar; for reducing sugar KMS_2T_2 was

Table 10 Effect of Potassium metabisulfite and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
KMS₁T₁	20.24	0.227	0.27	20.17	0.295	0.36	20.12	0.374	0.43
KMS₂T₁	22.62	0.234	0.46	22.56	0.277	0.55	22.5	0.392	0.63
KMS₃T₁	18.52	0.226	0.41	18.45	0.266	0.50	18.4	0.608	0.59
KMS₁T₂	18.11	0.258	0.48	18.05	0.339	0.57	18	0.626	0.65
KMS₂T₂	18.92	0.248	0.50	18.86	0.345	0.59	18.81	0.657	0.67
KMS₃T₂	20.24	0.214	0.40	20.38	0.370	0.49	20.33	0.468	0.57
KMS₁T₃	16.78	0.210	0.51	16.73	0.299	0.60	16.69	0.522	0.68
KMS₂T₃	18.17	0.209	0.41	18.11	0.297	0.50	18.05	0.509	0.59
KMS₃T₃	18.85	0.233	0.51	18.78	0.316	0.60	18.74	0.566	0.67
CD@5%	2.26	0.05	0.32	1.78	0.24	0.43	2.53	0.05	0.47
SEM±	0.78	0.02	0.11	0.61	0.08	0.15	0.87	0.02	0.16

at par to KMS_1T_2 followed by KMS_3T_1 and in case of titrable acidity the treatment KMS_1T_3 was at par with KMS_3T_3 , KMS_2T_2 followed by KMS_1T_2 .

4.3.4 Total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition:

Table 11 indicates the effect of potassium meta bisuphite and pasteurization temperature on total sugar, reducing sugar, titrable acidity of stored sugarcane juice at ambient condition. The minimum and maximum range of total sugar content was seen in treatments KMS_3T_1 (15.05) & KMS_2T_1 (23.61); KMS_3T_1 (15.00) & KMS_2T_1 (23.56); KMS_3T_1 (14.94) & KMS_2T_1 (23.51) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of reducing sugar were found in treatments KMS_3T_2 (0.212) & KMS_2T_2 (0.281) for 7 days, KMS_1T_2 (0.271) & KMS_1T_3 (0.370) for 14 days and KMS_3T_2 (0.407) & KMS_1T_3 (0.679) for 21 days. In case of titrable acidity content the minimum and maximum values were as follows: KMS_3T_1 (0.28) & KMS_2T_2 (0.53) for 7 days; KMS_3T_1 (0.37) & KMS_2T_2 (0.62) for 14 days; KMS_3T_1 (0.45) & KMS_2T_2 (0.70) for 21 days.

The statistical analysis of data indicates that treatments have got significant effect on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition. The treatment combination KMS_2T_1 was at par to KMS_2T_2 followed by KMS_1T_1 for total sugar, similarly in case of reducing sugar and titrable acidity, treatment KMS_2T_2 was at par with KMS_3T_1 followed by KMS_3T_3 , and KMS_2T_2 was at par with KMS_3T_3 , followed by KMS_2T_1 respectively, on 7 days storage of period.

On 14 days of storage period for total sugar, treatment KMS_2T_1 was at par to KMS_2T_2 followed by KMS_1T_1 ; in reducing sugar the treatment KMS_1T_3 was at par with KMS_2T_1 and KMS_3T_2 ; treatment KMS_2T_2 was at par to KMS_3T_3 followed by KMS_2T_1 in titrable acidity.

Upon 21 days of storage, it was seen that treatment combination KMS_2T_1 was at par to KMS_2T_2 and KMS_1T_1 for total sugar; for reducing sugar treatment KMS_1T_3 was at par to KMS_2T_3 followed by KMS_1T_2 and in case of titrable acidity, treatment KMS_2T_2 was at par with KMS_3T_3 followed by KMS_2T_3 .

Table 11 Effect of Potassium metabisulfate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
KMS₁T₁	19.49	0.248	0.37	19.43	0.291	0.46	19.38	0.558	0.54
KMS₂T₁	23.61	0.221	0.50	23.56	0.345	0.59	23.51	0.599	0.56
KMS₃T₁	15.05	0.255	0.28	15	0.283	0.37	14.94	0.452	0.45
KMS₁T₂	16.37	0.237	0.44	16.32	0.271	0.53	16.27	0.599	0.61
KMS₂T₂	21.23	0.281	0.53	21.18	0.321	0.62	21.14	0.458	0.70
KMS₃T₂	19.1	0.212	0.34	19.04	0.342	0.43	19	0.407	0.51
KMS₁T₃	17.59	0.220	0.45	17.53	0.370	0.54	17.48	0.679	0.62
KMS₂T₃	17.86	0.237	0.48	17.81	0.323	0.57	17.76	0.657	0.65
KMS₃T₃	18.92	0.250	0.51	18.87	0.313	0.60	18.84	0.447	0.68
CD@5%	2.35	0.22	0.26	3.00	0.21	0.08	1.21	0.43	0.14
SEM±	0.81	0.08	0.09	1.03	0.07	0.03	0.42	0.15	0.05

4.3.5 Poly phenol oxidase, Total Soluble Solids and Phenol content of stored sugarcane juice at refrigerated condition

The table 12 show result of effect of KMS and pasteurization temperature on PPO, TSS and phenol content of stored sugarcane juice at refrigerated condition. The lowest and highest values of PPO was found in treatments KMS_3T_3 (0.021) & KMS_1T_1 (0.028); KMS_3T_3 (0.015) & KMS_1T_1 (0.025); KMS_3T_3 (0.013) & KMS_1T_1 (0.024) for 7, 14 and 21 days. Similarly, the lowest and highest values of TSS was seen in treatments KMS_3T_2 (18.2) & KMS_1T_1 , KMS_2T_3 (19.8) for 7 days, KMS_2T_3 (18.00) & KMS_1T_1 , KMS_2T_3 (19.5) for 14 days and KMS_3T_2 (17.70) & KMS_2T_3 (19.30) for 21 days. For phenol content the lowest and highest values was seen in treatments: KMS_3T_2 , (14.54) & KMS_2T_2 , KMS_2T_3 (14.85) for 7 days; KMS_3T_2 (14.94) & KMS_2T_3 (15.26) for 14 days; KMS_3T_2 (15.34) & KMS_2T_2 , KMS_2T_3 (15.65) for 21 days treatments.

The statistical analysis of data indicates that treatments exhibits significant effect on PPO, TSS and phenol content of stored sugarcane juice. Table 12 shows that treatment combination KMS_1T_1 was at par to KMS_2T_1 , KMS_3T_1 followed by KMS_1T_2 in poly phenol oxidase, similarly in total soluble solid and phenol content, treatment KMS_1T_1 , KMS_2T_3 were at par with KMS_1T_2 and KMS_3T_3 followed by KMS_3T_1 ; KMS_2T_2 was at par with KMS_3T_1 & KMS_1T_3 , followed by KMS_1T_2 respectively, on 7 days storage of period.

On 14 days of storage period for poly phenol oxidase, treatment KMS_1T_1 was at par to KMS_1T_2 followed by KMS_2T_1 ; similarly for TSS treatment KMS_1T_1 & KMS_2T_3 collectively were at par with KMS_3T_3 and KMS_1T_2 ; KMS_2T_3 was at par to KMS_2T_2 followed by KMS_1T_3 in phenol content.

Upon 21 days of storage it was seen that for poly phenol oxidase, treatment KMS_1T_1 was at par to KMS_1T_2 followed by KMS_2T_1 ; for total soluble solid the treatment KMS_2T_3 was at par to KMS_1T_1 followed by KMS_3T_1 , for phenol content treatment KMS_2T_2 & KMS_2T_3 were at par with KMS_1T_3 , KMS_3T_1 followed by KMS_1T_2 .

Table 12 Effect of Potassium metabisulfite and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5⁰C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
KMS₁T₁	0.028	19.8	14.62	0.025	19.5	15.02	0.024	19.2	15.42
KMS₂T₁	0.026	18.5	14.55	0.023	18.3	14.95	0.021	18	15.35
KMS₃T₁	0.026	19.4	14.77	0.021	19.1	15.17	0.018	18.9	15.57
KMS₁T₂	0.025	19.5	14.76	0.024	19.2	15.16	0.022	18.8	15.56
KMS₂T₂	0.023	18.7	14.85	0.021	18.5	15.25	0.020	18.2	15.65
KMS₃T₂	0.023	18.2	14.54	0.019	18	14.94	0.018	17.7	15.34
KMS₁T₃	0.024	18.8	14.77	0.022	18.5	15.22	0.020	18.3	15.57
KMS₂T₃	0.022	19.8	14.85	0.019	19.5	15.26	0.020	19.3	15.65
KMS₃T₃	0.021	19.5	14.55	0.015	19.2	15.12	0.013	18.7	15.35
CD@5%	0.02	0.16	1.44	0.02	0.07	0.05	0.02	0.05	0.05
SEM±	0.01	0.05	0.50	0.01	0.03	0.02	0.01	0.02	0.02

4.3.6 Poly phenol oxidase, Total Soluble Solids and Phenol content of stored sugarcane juice at ambient condition

Table 13 comprises the results of effect of potassium meta bisulphite and pasteurization temperature on PPO, TSS and phenol content of stored sugarcane juice at ambient condition. The lowest and highest values of PPO was found in treatments KMS₃T₃ (0.034) & KMS₁T₁ (0.048); KMS₃T₃ (0.025) & KMS₁T₁ (0.045); KMS₃T₃ (0.027) & KMS₁T₁ (0.041) for 7, 14 and 21 days. Similarly TSS was found the lowest and highest ranged in treatments KMS₃T₂ (18.20) & KMS₂T₃ (19.90) on 7 days of storage period, KMS₃T₂ (18.00) & KMS₂T₃ (19.70) for 14 days and KMS₃T₂ (17.70) & KMS₂T₃ (19.50) for 21 days. Phenol content was found to be lowest and highest in treatments: KMS₃T₂, (14.14) & KMS₂T₂, KMS₂T₃ (14.45) for 7 days; KMS₃T₂ (14.54) & KMS₂T₂, KMS₂T₃ (14.85) for 14 days; KMS₃T₂, for 21 days of storage period KMS₃T₂ (14.95) & KMS₂T₂ (15.25).

The statistical analysis of data indicates that significant difference exists among the treatments combinations. The treatment KMS₁T₁ was at par to KMS₂T₁ followed by KMS₁T₂ in poly phenol oxidase, similarly in total soluble solid and phenol content treatment KMS₂T₁ was at par with KMS₂T₁ and followed by KMS₃T₃; KMS₂T₂ was at par with KMS₂T₃, followed by KMS₃T₁ respectively, in 7 days storage of period. On 14 days of storage period for poly phenol oxidase, treatment KMS₁T₁ was at par to KMS₂T₁ followed by KMS₂T₂; in KMS₂T₃ was at par with KMS₁T₁ & KMS₁T₂; treatment KMS₂T₂ was at par to KMS₂T₃ followed by KMS₃T₁ in phenol content. Finally on 21 days of storage it was seen that treatment KMS₁T₁ was at par to KMS₂T₁ and KMS₃T₁ for poly phenol oxidase; for total soluble solid KMS₂T₃ was at par to KMS₁T₁ & KMS₁T₂ followed by KMS₃T₁; as regards for phenol content treatment combination KMS₂T₂ was at par with KMS₂T₃ followed by KMS₃T₁ & KMS₃T₂.

Table 13 Effect of Potassium metabisulfite and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
KMS₁T₁	0.048	19.5	14.22	0.045	19.3	14.62	0.041	19	15.02
KMS₂T₁	0.046	18.3	14.15	0.043	18.1	14.55	0.040	17.8	14.95
KMS₃T₁	0.046	19.1	14.37	0.041	18.9	14.77	0.038	18.5	15.17
KMS₁T₂	0.040	19.5	14.36	0.040	19.3	14.76	0.038	19	15.16
KMS₂T₂	0.041	18.7	14.45	0.041	18.5	14.85	0.035	18.2	15.25
KMS₃T₂	0.039	18.2	14.14	0.035	18	14.54	0.031	17.7	14.94
KMS₁T₃	0.037	18.5	14.37	0.032	18.3	14.77	0.031	18	15.17
KMS₂T₃	0.035	19.9	14.45	0.029	19.7	14.85	0.030	19.5	15.25
KMS₃T₃	0.034	19.2	14.15	0.025	19	14.55	0.027	18.7	14.95
CD@5%	0.02	0.05	0.05	0.02	0.47	0.04	0.03	0.07	0.56
SEM±	0.01	0.02	0.01	0.01	0.16	0.01	0.01	0.02	0.19

4.4 Effect of effect of potassium sorbate and pasteurization temperature on chemical composition of stored sugarcane juice

4.4.1 Moisture, Carbohydrate and Total ash content stored sugarcane juice at refrigerated condition.

The result presented in table 14 shows the effect of potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition. The lowest and highest values of moisture content was found in treatments PS₃T₁ (75.90) & PS₃T₃ (81.20); PS₃T₂ (77.20) & PS₁T₃ (84.63); PS₂T₁ (76.79) & PS₁T₃ (80.79) for 7, 14 and 21 days of storage of sugarcane juice. Similarly it was observed that carbohydrate content was found to be lowest and highest in treatments combinations PS₁T₁ (13.50) & PS₃T₁ (16.74) for 7 days, PS₁T₂ (13.06) & PS₁T₃ (16.88) for 14 days and PS₂T₂ (13.92) & PS₁T₃ (17.23) for 21 days storage of sugarcane juice. As regards for total ash content the lowest and highest values were observed in treatments: PS₃T₁, PS₃T₂, PS₃T₃ (0.20) & PS₂T₁, PS₁T₂, PS₂T₂, PS₂T₃ (0.40) for 7 days; PS₁T₁, PS₁T₃, PS₃T₃ (0.20) & PS₂T₁, PS₃T₁, PS₃T₂, PS₂T₃ (0.40) for 14 days; PS₁T₁, PS₃T₁, PS₁T₂, PS₁T₃, PS₃T₃ (0.20) & PS₂T₁, PS₂T₂, PS₃T₂ (0.40) for 21 days storage of juice.

The statistical analysis shows that treatment combination has got significant effect on chemical constituents of stored sugarcane juice. Treatment combination PS₃T₃ was at par to PS₂T₃ followed by PS₁T₂ in moisture content, similarly in carbohydrate and ash content treatment, PS₃T₁ was at par with PS₂T₁ and followed by PS₃T₂; PS₂T₁ was at par with PS₁T₂, followed by PS₁T₁ respectively, in 7 days storage of period.

On 14 days of storage period, for moisture content, treatment PS₃T₃ was at par to PS₂T₃ followed by PS₃T₁; PS₁T₃ was at par with PS₃T₃ & followed by PS₃T₁; treatment PS₂T₁ was at par to PS₃T₁ followed by PS₁T₂ in ash content.

Finally on 21 days of storage study, it was seen that treatment PS₁T₃ was at par to PS₁T₂ and PS₂T₃ for moisture; for carbohydrate PS₁T₃ was at par to PS₁T₂ & followed by PS₁T₁; in ash content treatment combination PS₂T₁ was at par with PS₂T₂ followed by PS₂T₃.

Table 14 Effect of Potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
PS₁T₁	76.1	13.5	0.3	80.88	13.41	0.2	76.82	16.01	0.2
PS₂T₁	76.6	14.46	0.4	80.5	13.73	0.4	76.4	14.52	0.4
PS₃T₁	75.9	16.74	0.2	81.43	15.82	0.4	79.22	14.46	0.2
PS₁T₂	76.75	13.97	0.4	81.23	13.06	0.3	80.78	17.16	0.2
PS₂T₂	76.1	13.87	0.4	80.9	14.78	0.3	80.1	13.92	0.4
PS₃T₂	76.1	14.21	0.2	77.2	14.94	0.4	79.22	15.06	0.4
PS₁T₃	76.55	13.97	0.3	80.5	16.88	0.2	80.79	17.23	0.2
PS₂T₃	78.35	13.87	0.4	82.25	14.78	0.4	80.6	14.41	0.3
PS₃T₃	81.2	14.21	0.2	84.63	16.40	0.2	80.25	16.01	0.2
CD@5%	0.60	0.07	0.05	0.51	0.06	0.03	0.36	0.29	0.06
SEM±	0.21	0.02	0.02	0.18	0.02	0.01	0.12	0.10	0.02

4.4.2 Moisture, Carbohydrate and Total ash content stored sugarcane juice at ambient condition.

The effect of potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash of stored sugarcane juice at ambient condition is presented in table 15. The lowest and highest values of moisture content was found in treatments combination PS_3T_2 (76.58) & PS_3T_3 (80.57); PS_3T_2 (77.85) & PS_3T_3 (83.47); PS_1T_1 (76.72) & PS_3T_2 (81.75) for 7, 14 and 21 days of storage study of sugarcane juice. Similarly it was also observed that, carbohydrate content was found to be lowest and highest range in treatment combination PS_3T_1 (13.78) & PS_3T_3 (15.76) for 7 days, PS_2T_1 (13.73) & PS_2T_2 (17.16) for 14 days and PS_2T_3 (13.92) & PS_2T_2 (17.84) for 21 days of storage of sugarcane juice. The ash content was found to be lowest and highest in treatment combinations PS_2T_1 , PS_2T_2 , PS_1T_3 , PS_2T_3 (0.20) & PS_3T_1 , PS_1T_2 , PS_3T_2 , PS_3T_3 (0.40) for 7 days; PS_1T_1 , PS_2T_2 , PS_1T_3 (0.20) & PS_3T_1 , PS_1T_2 , PS_3T_2 , PS_2T_3 , PS_3T_3 (0.40) for 14 days; PS_1T_1 , PS_3T_1 , PS_2T_2 , PS_2T_3 , PS_3T_3 (0.20) & PS_1T_1 , PS_3T_2 , PS_1T_3 (0.40) for 21 days storage study.

The statistical analysis shows that treatment combination PS_3T_3 was at par to PS_1T_1 followed by PS_2T_3 in moisture, similarly in carbohydrate and ash content treatment PS_3T_3 was at par with PS_3T_2 and followed by PS_1T_2 ; PS_1T_2 was at par with PS_1T_1 , followed by PS_2T_1 respectively, in 7 days storage of period.

On 14 days of storage period for moisture content, treatment PS_3T_3 was at par to PS_2T_3 followed by PS_2T_2 ; in PS_2T_2 was at par with PS_1T_1 & followed by PS_3T_2 ; treatment PS_3T_1 was at par to PS_2T_1 followed by PS_1T_1 in case of ash content.

Upon 21 days of storage study, it was seen that treatment PS_3T_2 was at par to PS_1T_2 and followed by PS_2T_3 for moisture; for carbohydrate PS_3T_2 was at par to PS_1T_3 & followed by PS_2T_2 ; for ash content treatment combination PS_1T_1 was at par with PS_3T_1 followed by PS_2T_2 .

Table 15 Effect of Potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash of stored sugarcane juice at ambient condition (30°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
PS₁T₁	78.01	14.16	0.3	80.72	14.57	0.2	76.72	14.16	0.4
PS₂T₁	76.75	13.92	0.2	80.50	13.73	0.3	77.54	12.19	0.3
PS₃T₁	76.80	13.78	0.4	81.78	13.78	0.4	81.21	15.76	0.3
PS₁T₂	77.10	14.46	0.4	81.75	13.82	0.4	81.52	17.59	0.4
PS₂T₂	78.00	14.11	0.2	82.40	17.16	0.2	80.00	17.84	0.2
PS₃T₂	76.58	15.46	0.4	77.85	14.26	0.4	81.75	19.1	0.4
PS₁T₃	76.70	13.82	0.2	81.12	14.21	0.2	80.50	18.16	0.4
PS₂T₃	78.00	13.92	0.2	82.76	17.16	0.4	81.42	13.92	0.2
PS₃T₃	80.57	15.76	0.4	83.47	13.78	0.4	80.50	14.84	0.2
CD@5%	0.47	0.10	0.06	0.63	0.28	0.48	0.43	0.39	0.05
SEM±	0.16	0.04	0.02	0.22	0.81	0.17	0.15	0.13	0.02

4.4.3 Total sugar, reducing sugar and titrable acidity stored sugarcane juice at refrigerated condition

Table 16 shows the effect of potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition. The results indicates that lowest and highest values of total sugar was found in treatments PS₁T₁ (76.58) & PS₁T₂ (80.57); PS₁T₁ (77.85) & PS₁T₂ (83.47); PS₁T₁ (76.72) & PS₁T₂ (81.75) for 7, 14 and 21 days. Similarly observed the reducing sugar was found the lowest and highest ranged in treatments PS₃T₁ (13.78) & PS₃T₃ (15.76) for 7 days, PS₃T₁ (13.73) & PS₃T₂ (17.16) for 14 days and PS₂T₁ (13.92) & PS₃T₁ (17.84) for 21 days. Finally, for reducing sugar was found the lowest and highest values in treatments: PS₂T₁ (0.20) & PS₃T₁ (0.40) for 7 days; PS₂T₁ (0.20) & PS₃T₁ (0.40) for 14 days; PS₂T₁ (0.20) & PS₃T₁ (0.40) for 21 days treatments.

On CRD analysis all the treatments were found significant. Treatment PS₁T₂ was at par to PS₂T₂ followed by PS₃T₂ in total sugar, similarly in reducing sugar and titrable acidity treatment PS₃T₃ was at par with PS₁T₂ and followed by PS₂T₂; PS₂T₁ was at par with PS₂T₂, followed by PS₂T₃ respectively, in 7 days storage of period.

On 14 days of storage period for total sugar, treatment PS₁T₂ was at par to PS₂T₂ followed by PS₃T₂; in PS₂T₁ was at par with PS₂T₂ & followed by PS₁T₂; treatment PS₂T was at par to PS₂T₂ followed by PS₁T₂ in titrable acidity.

Finally on 21 days of storage it was seen that treatment PS₁T₂ was at par to PS₂T₂ and followed by PS₃T₂ for total sugar; Reducing PS₂T₂ was at par to PS₂T₁ & followed by PS₃T₁; In titrable acidity treatment PS₂T₁ was at par with PS₃T₃ followed by PS₂T₂.

4.4.4 Total sugar, reducing sugar and titrable acidity stored sugarcane juice at ambient condition.

Table 17 indicates the effect of potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition. The minimum and maximum value of total sugar was seen in treatments PS₁T₁ (15.21) & PS₁T₂ (21.88); PS₁T₁

Table 16 Effect of Potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
PS₁T₁	14.47	0.209	0.36	14.36	0.313	0.45	14.31	0.407	0.53
PS₂T₁	19.49	0.219	0.52	19.36	0.299	0.61	19.32	0.599	0.69
PS₃T₁	19.1	0.204	0.27	19.04	0.271	0.36	19	0.582	0.43
PS₁T₂	22.62	0.256	0.42	22.56	0.318	0.51	22.5	0.550	0.48
PS₂T₂	20.49	0.251	0.51	20.36	0.339	0.60	20.31	0.657	0.67
PS₃T₂	19.91	0.214	0.32	19.86	0.342	0.41	19.82	0.468	0.49
PS₁T₃	15.79	0.205	0.44	15.74	0.318	0.53	15.7	0.509	0.61
PS₂T₃	17.53	0.247	0.49	17.47	0.295	0.58	17.42	0.452	0.66
PS₃T₃	18.11	0.263	0.52	18.04	0.279	0.61	17.99	0.503	0.69
CD@5%	0.49	0.05	0.32	0.79	0.08	0.38	0.53	0.05	0.06
SEM±	0.17	0.02	0.11	0.27	0.03	0.13	0.18	0.02	0.02

Table 17 Effect of Potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
PS₁T₁	15.21	0.212	0.53	15.15	0.275	0.62	15.11	0.558	0.70
PS₂T₁	18.17	0.234	0.46	18.13	0.299	0.55	18.08	0.590	0.63
PS₃T₁	16.53	0.255	0.49	16.48	0.326	0.58	16.43	0.582	0.66
PS₁T₂	21.88	0.204	0.40	21.84	0.283	0.49	21.8	0.599	0.57
PS₂T₂	15.79	0.206	0.46	15.74	0.374	0.55	15.7	0.458	0.63
PS₃T₂	16.53	0.231	0.49	16.47	0.370	0.58	16.42	0.407	0.66
PS₁T₃	17.53	0.248	0.28	17.48	0.258	0.27	17.43	0.497	0.35
PS₂T₃	21.88	0.220	0.47	21.83	0.291	0.56	21.78	0.424	0.64
PS₃T₃	18.17	0.226	0.41	18.13	0.357	0.50	18.06	0.452	0.58
CD@5%	0.58	0.06	0.06	0.97	0.06	0.05	1.09	0.08	0.07
SEM±	0.20	0.02	0.02	0.33	0.02	0.02	0.37	0.03	0.02

(15.15) & PS₁T₂ (21.84); PS₁T₂ (21.8) & PS₂T₂ (15.7) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of reducing sugar were found in treatments PS₃T₁ (0.255) & PS₁T₂ (0.204), for 7 days, PS₁T₂ (21.84) & PS₁T₁, (15.15) for 14 days and PS₃T₂ (0.407) & PS₁T₂ (0.599) for 21 days.

Finally, for the titrable acidity higher and lower values were as follows: PS₁T₃ (0.28), PS₁T₁ (0.53); PS₁T₃ (0.27) & PS₁T₃; PS₁T₃ (0.35) and PS₁T₁ (0.70) for 7 days

On CRD analysis all the treatments were found significant. Treatment PS₁T₂ was at par to PS₂T₃ followed by PS₂T₁ in total sugar, similarly in reducing sugar and titrable acidity treatment PS₃T₁ was at par with PS₁T₃ and followed by PS₂T₂; PS₁T₁ was at par with PS₃T₁ followed by PS₃T₂ respectively, in 7 days storage of period.

On 14 days of storage period for total sugar, treatment PS₁T₂ was at par to PS₂T₃ followed by PS₂T₁; in PS₂T₂ was at par with PS₃T₃ & followed by PS₁T₁; treatment PS₁T₁ was at par to PS₃T₁ followed by PS₃T₂ in titrable acidity.

Finally on 21 days of storage it was seen that treatment PS₁T₂ was at par to PS₂T₃ and followed by PS₂T₁ for total sugar; Reducing PS₁T₂ was at par to PS₂T₁ & followed by PS₃T₁; in titrable acidity treatment PS₁T₁ was at par with PS₃T₁ followed by PS₃T₂.

4.4.5 Poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition

Table 18 reveals that minimum and maximum value of poly phenol oxidase, in treatments PS₃T₃ (0.031) & PS₁T₁ (0.035); PS₃T₃ (0.025) & PS₁T₁ (0.029); PS₃T₃ (0.023) & PS₁T₁ (0.028) for 7, 14 and 21 days of storage period of sugarcane juice respectively. Similarly the lowest and highest values of total soluble solid were found in treatments PS₁T₁ (18.3) & PS₂T₃ (20.7), for 7 days, PS₁T₁ (18.0) & PS₃T₃, (20.5) for 14 days and PS₁T₁ (17.8) & PS₁T₂ (20.2) for 21 days of storage period. The phenol content was higher and lower in treatment combination: PS₁T₁ (18.3) & PS₂T₃ (20.7); PS₁T₁ (14.92) & PS₃T₁ (15.27); PS₁T₁ (15.32) PS₃T₁ (15.67) for 7, 14 and 21 days respectively

Table 18 Effect of Potassium sorbate and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
PS₁T₁	0.035	18.3	14.52	0.029	18	14.92	0.028	17.8	15.32
PS₂T₁	0.033	18.7	14.74	0.028	18.5	15.14	0.027	18.3	15.54
PS₃T₁	0.032	19.7	14.87	0.028	19.5	15.27	0.027	19.3	15.67
PS₁T₂	0.032	19.1	14.82	0.027	18.8	15.22	0.027	18.5	15.62
PS₂T₂	0.031	19.8	14.64	0.027	19.5	15.04	0.025	19.2	15.44
PS₃T₂	0.033	18.5	14.78	0.026	18.3	15.18	0.026	18	15.58
PS₁T₃	0.031	20	14.64	0.026	19.8	15.05	0.024	19.6	15.44
PS₂T₃	0.031	20.7	14.54	0.025	20.5	15.03	0.024	20.2	15.34
PS₃T₃	0.031	18.4	14.59	0.025	18.2	15.15	0.023	17.9	15.39
CD@5%	0.03	0.40	0.08	0.02	0.88	0.70	0.03	0.72	0.04
SEM±	0.01	0.14	0.03	0.01	0.30	0.24	0.01	0.25	0.01

Table 18 indicate that treatment combination PS₁T₁ was at par to PS₂T₁ followed by PS₃T₁ in poly phenol oxidase activity similarly in total soluble solid and phenol content treatment PS₂T₃ was at par with PS₁T₃ and followed by PS₂T₂; PS₃T₁ was at par with PS₁T₂ followed by PS₃T₂ respectively, in 7 days storage of period.

On 14 days of storage period for poly phenol oxidase, treatment PS₁T₁ was at par to PS₂T₁ followed by PS₁T₂; in PS₂T₃ was at par with PS₁T₃ & followed by PS₃T₁; treatment PS₁T₂ was at par to PS₃T₁ followed by PS₂T₁ in phenol content.

Finally on 21 days of storage of sugarcane juice, it was seen that treatment PS₁T₁ was at par to PS₂T₁ and followed by PS₃T₂ for poly phenol oxidase; total soluble solid PS₂T₃ was at par to PS₁T₃ & followed by PS₃T₁; in phenol content treatment combination PS₃T₁ was at par with PS₁T₂ followed by treatment PS₃T₂.

4.4.6 Poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition

Table 19 indicates that minimum and maximum value of poly phenol oxidase, was seen in treatments PS₃T₃ (0.029) & PS₁T₁ (0.035); PS₃T₃ (0.025) & PS₁T₁ (0.032); PS₃T₃ (0.025) & PS₁T₁ (0.032) for 7, 14 and 21 days respectively. The lowest and highest range of the total soluble solid were seen in treatments PS₃T₃ (18.2) & PS₂T₃ (20.5), for 7 days, PS₃T₃ (18.0) & PS₂T₃ (20.2) for 14 days and PS₂T₁ (18.0) & PS₂T₃ (19.5) for 21 days. The phenol content was higher and lower in treatment combination PS₃T₁ (14.47) & PS₁T₁ (14.12); PS₃T₁ (14.87), & PS₁T₁ (14.52); PS₃T₁ (15.27) PS₁T₁ (14.92) for 7, 14 and 21 days respectively.

Table 19 also indicates that treatment combination PS₁T₁ was at par to PS₃T₁ followed by PS₁T₂ in poly phenol oxidase, similarly in total soluble solid and phenol content, treatment PS₂T₃ was at par with PS₂T₂ and followed by PS₁T₃; PS₃T₁ was at par with PS₁T₂ followed by PS₃T₂ respectively, in 7 days storage of period.

On 14 days of storage period for poly phenol oxidase, treatment PS₁T₁ was at par to PS₁T₂ followed by PS₂T₁; in PS₂T₃ was at par with PS₂T₂

Table 19 Effect of Potassium sorbate and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30° C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
PS₁T₁	0.0350	18.7	14.12	0.032	18.5	14.52	0.032	18.2	14.92
PS₂T₁	0.0350	18.5	14.34	0.030	18.3	14.74	0.030	18	15.14
PS₃T₁	0.0348	19.5	14.47	0.030	19.3	14.87	0.030	19	15.27
PS₁T₂	0.033	19.7	14.42	0.031	19.5	14.82	0.031	19.2	15.22
PS₂T₂	0.034	19.9	14.24	0.030	19.7	14.64	0.030	19.3	15.04
PS₃T₂	0.034	18.5	14.38	0.029	18.2	14.78	0.029	17.9	15.18
PS₁T₃	0.031	19.8	14.24	0.027	19.4	14.64	0.027	18	15.04
PS₂T₃	0.03	20.5	14.14	0.025	20.2	14.54	0.025	19.5	14.94
PS₃T₃	0.029	18.2	14.19	0.025	18	14.59	0.025	18.7	14.99
CD@5%	0.03	0.86	0.66	0.03	0.05	1.11	0.03	0.88	0.87
SEM±	0.01	0.30	0.23	0.01	0.02	0.38	0.01	0.30	0.30

& followed by PS₁T₂; treatment PS₃T₁ was at par to PS₁T₂ followed by PS₃T₂ in phenol content.

Finally on 21 days of storage it was seen that treatment PS₁T₁ was at par to PS₁T₂ and followed by PS₂T₁ for poly phenol oxidase; Total soluble solid PS₂T₃ was at par to PS₂T₂ & followed by PS₁T₂; in phenol content treatment PS₃T₁ was at par with PS₁T₂ followed by PS₃T₂.

4.5. Effect of different combinations of ascorbic acid and pasteurization temperature on various chemical constituents of stored sugar cane juice

4.5.1 Moisture, Carbohydrate and Ash content of stored sugar cane juice at refrigerated condition

Table 20 comprises the results of effect of different combinations of ascorbic acid and pasteurization temperature on various chemical constituents of stored sugar cane juice. The minimum and maximum value of moisture, carbohydrate and ash content under refrigerated condition (Table 20) was seen in treatments AA₂T₁ (75.4) & AA₂T₃ (81.2); AA₁T₂ (76.54) & AA₃T₃ (85.6); AA₂T₁ (76.24) & AA₁T₃ (82.25) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of carbohydrate were found in treatments AA₁T₃ (13.73) & AA₃T₃ (16.27), for 7 days, AA₂T₁ (13.59) & AA₂T₂, (15.70) for 14 days and AA₃T₂ (17.23) & AA₁T₃ (13.73) for 21 days. Finally, for the Ash content higher and lower values were as follows: AA₁T₂, AA₃T₃, (0.4) and AA₁T₁, AA₂T₂ AA₃T₂ AA₂T₃ (0.2) for 7 days; AA₂T₂ (0.6) & AA₁T₁, AA₃T₂, AA₁T₃ (0.2) for 14 days; AA₁T₁ & AA₂T₂ (0.6) & AA₃T₂, AA₁T₃, AA₂T₃, (0.2) for 21 day.

The statistical analysis shows that treatment AA₂T₃ was at par to AA₃T₃ followed by AA₁T₂ in moisture, similarly in carbohydrate and ash content treatment AA₃T₃ was at par with AA₂T₂ and followed by AA₂T₃; AA₁T₂ was at par with AA₃T₃, followed by AA₁T₃ respectively, in 7 days storage of period.

On 14 days of storage period for moisture, treatment AA₃T₃ was at par to AA₃T₂ followed by AA₂T₃; in AA₂T₂ was at par with AA₃T₁ & followed by AA₁T₂; treatment AA₂T₂ was at par to AA₃T₁ followed by AA₂T₃ in ash content.

Table 20 Effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
AA₁T₁	76.2	13.82	0.2	80.12	14.26	0.2	76.7	14.94	0.6
AA₂T₁	75.4	14.41	0.3	80.55	13.59	0.3	76.24	14.78	0.4
AA₃T₁	75.61	14.31	0.3	82.2	13.78	0.4	80.12	16.40	0.3
AA₁T₂	76.75	13.78	0.4	76.54	14.73	0.3	80.5	13.92	0.4
AA₂T₂	79.1	15.82	0.2	82.81	15.70	0.6	81.66	13.97	0.6
AA₃T₂	76.4	13.92	0.2	84.6	13.78	0.2	80.55	17.23	0.2
AA₁T₃	76.54	13.73	0.3	76.75	13.73	0.2	82.25	13.73	0.2
AA₂T₃	81.2	14.57	0.2	84.6	14.84	0.4	80.9	14.57	0.2
AA₃T₃	78.35	16.27	0.4	85.6	15	0.3	81.66	14.36	0.3
CD@5%	0.85	0.03	0.04	0.65	0.06	0.04	0.99	0.88	0.05
SEM±	0.29	0.01	0.01	0.22	0.02	0.01	0.34	0.30	0.02

Finally on 21 days of storage it was seen that treatment AA₁T₃ was at par to AA₂T₂ and followed by AA₂T₃ for moisture; for carbohydrate AA₃T₂ was at par to AA₃T₁ & followed by AA₁T₁; in ash content treatment AA₁T₁ was at par with AA₂T₁ followed by AA₃T₁.

4.5.2 Moisture, Carbohydrate and Ash content of stored sugar cane juice at ambient condition

Table 21 shows that highest and lowest moisture content were found in treatments AA₂T₃ (76.32) & AA₂T₃ (78.00); AA₃T₂ (78.00) & AA₃T₃ (84.60); AA₂T₁ (75.50) & AA₃T₂, AA₃T₃ (81.40) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of carbohydrate were found in treatments AA₂T₂ (13.64) & AA₁T₃ (16.67), for 7 days, AA₃T₂ (13.59) & AA₃T₁, (16.53) for 14 days and AA₃T₁ (13.92) & AA₂T₃ (18.41) for 21 days storage of sugarcane juice. The ash content was higher and lower in treatments viz AA₁T₁, AA₂T₂, AA₁T₃ (0.4) and AA₂T₁, AA₃T₁ AA₁T₂ AA₃T₂ AA₂T₃ AA₃T₃ (0.2) for 7 days; AA₂T₁, AA₃T₁, AA₃T₂, AA₂T₃ (0.4) & AA₁T₂, AA₁T₃, AA₃T₃ (0.2) for 14 days; AA₁T₂ & AA₂T₂, AA₁T₃ (0.4) & AA₁T₁, AA₃T₂, AA₁T₃, AA₂T₃, AA₃T₃ (0.2) for 21 day respectively.

Findings exhibits that treatments have got significant effect on chemical composition of juice. Treatment AA₂T₃ was at par to AA₂T₂ followed by AA₃T₃ for moisture, similarly in carbohydrate and ash content treatment AA₁T₃ was at par with AA₁T₁ and followed by AA₃T₃; AA₁T₁ was at par with AA₂T₂, followed by AA₂T₁ respectively, in 7 days storage of period.

On 14 days of storage period, for moisture content treatment combination AA₁T₃ was at par to AA₂T₂ followed by AA₁T₂; in AA₃T₁ was at par with AA₃T₃ & followed by AA₁T₁; treatment AA₂T₁ was at par to AA₃T₁ followed by AA₁T₁ in ash content.

On 21 days of storage study, it was seen that treatment AA₃T₂ was at par to AA₃T₃ and followed by AA₂T₂ for moisture; for carbohydrate AA₂T₃ was at par to AA₃T₂ & followed by AA₁T₁; in ash content treatment AA₁T₂ was at par with AA₂T₁ followed by AA₁T₁.

Table 21 Effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at ambient condition (30°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash	Moisture	Carbohydrate	Ash
AA₁T₁	76.50	14.67	0.4	80.12	14.52	0.3	76.80	17.23	0.2
AA₂T₁	76.82	13.82	0.2	80.40	14.21	0.4	76.50	14.52	0.3
AA₃T₁	76.80	13.97	0.2	81.30	16.53	0.4	80.50	13.92	0.2
AA₁T₂	76.50	13.73	0.2	81.12	14.01	0.2	80.70	16.33	0.4
AA₂T₂	77.10	13.64	0.4	81.70	13.92	0.3	81.21	14.21	0.4
AA₃T₂	76.62	14.41	0.2	78.00	13.59	0.4	81.40	18.33	0.3
AA₁T₃	76.32	16.67	0.4	84.60	13.82	0.2	80.75	14.26	0.4
AA₂T₃	78.00	14.52	0.2	81.30	13.78	0.4	80.14	18.41	0.2
AA₃T₃	76.91	14.41	0.2	84.60	14.89	0.2	81.40	16.40	0.2
CD@5%	0.65	0.06	0.06	1.09	0.04	0.05	0.96	0.47	0.06
SEM±	0.22	0.02	0.02	0.37	0.01	0.02	0.33	0.16	0.02

4.5.3 Total sugar, reducing sugar and Titrable acidity of stored sugarcane juice at refrigerated condition

Table 22 reveals the effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition. The minimum and maximum value of total sugar was seen in treatments AA₃T₁, (22.62) & AA₁T₃ (15.63); AA₃T₁, (22.57) & AA₁T₃ (15.18); AA₂T₁ (22.53) & AA₁T₃ (15.47) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of reducing sugar were found in treatments AA₂T₃ (0.247) & AA₁T₂ (0.204) for 7 days, AA₃T₂ (0.370) & AA₃T₁ (0.258) for 14 days and AA₃T₁ (22.53) & AA₁T₃ (15.47) for 21 days. Finally, for titrable acidity the higher and lower values were as follows: AA₁T₁ (0.51) & AA₃T₃ (0.26), for 7 days; AA₁T₁ (0.60) & AA₃T₃ (0.35) for 14 days; AA₃T₁ (0.67) & AA₃T₃ (0.45) for 21 days.

The statistical analysis of data shows significant effect of treatments on quality of juice. The treatment combination AA₃T₁ was at par to AA₃T₃ followed by AA₁T₂ for total sugar, similarly in case of reducing sugar and titrable acidity treatment AA₂T₃ was at par with AA₂T₂ and followed by AA₃T₁; AA₁T₁ was at par with AA₃T₁, followed by AA₃T₂ respectively, in 7 days storage of period.

On 14 days of storage period, for total sugar, treatment AA₃T₁ was at par to AA₂T₁ followed by AA₁T₂; in AA₃T₂ was at par with AA₂T₂ & followed by AA₁T₂; treatment AA₁T₁ was at par to AA₃T₁ followed by AA₃T₂ in titrable acidity. Finally on 21 days of storage it was seen that treatment AA₃T₁ was at par to AA₂T₁ and followed by AA₂T₂ for total sugar; for reducing sugar AA₁T₁ was at par to AA₂T₂ & followed by AA₃T₁; in titrable acidity treatment AA₃T₁ was at par with AA₁T₁ followed by AA₃T₂.

4.5.4 Total sugar, reducing sugar and Titrable acidity of stored sugarcane juice at ambient condition

Table 23 reveals the effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition. The maximum and minimum value of total sugar was seen in treatments AA₂T₃ (21.3) & AA₁T₁, (15.79); AA₂T₃,

Table 22 Effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
AA₁T₁	18.75	0.214	0.51	18.69	0.318	0.60	18.63	0.679	0.66
AA₂T₁	21.88	0.229	0.46	21.82	0.287	0.55	21.76	0.458	0.62
AA₃T₁	22.62	0.240	0.50	22.57	0.258	0.59	22.53	0.582	0.67
AA₁T₂	20.49	0.204	0.38	20.43	0.339	0.47	19.39	0.384	0.54
AA₂T₂	20.24	0.244	0.45	20.19	0.360	0.54	20.15	0.590	0.62
AA₃T₂	17.11	0.209	0.49	17.06	0.370	0.58	17.01	0.443	0.66
AA₁T₃	15.63	0.205	0.27	15.58	0.321	0.36	15.47	0.452	0.45
AA₂T₃	19.49	0.247	0.49	19.36	0.281	0.58	19.31	0.529	0.66
AA₃T₃	21.23	0.218	0.26	21.17	0.271	0.35	21.12	0.497	0.43
CD@5%	1.95	0.03	0.03	2.20	0.05	0.04	1.91	0.06	0.04
SEM±	0.67	0.01	0.01	0.76	0.02	0.01	0.66	0.02	0.01

Table 23 Effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity sugar of stored sugarcane juice at ambient condition (30°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity	Total sugar	Reducing sugar	Titrable acidity
AA₁T₁	15.79	0.218	0.52	15.74	0.313	0.61	15.7	0.407	0.69
AA₂T₁	21.23	0.224	0.41	21.18	0.295	0.50	21.15	0.608	0.58
AA₃T₁	18.85	0.240	0.28	18.8	0.275	0.27	18.74	0.558	0.35
AA₁T₂	18.17	0.256	0.50	18.12	0.342	0.49	18.07	0.679	0.57
AA₂T₂	17.36	0.204	0.30	17.32	0.345	0.39	17.28	0.479	0.47
AA₃T₂	16.53	0.242	0.47	16.5	0.351	0.56	16.45	0.438	0.64
AA₁T₃	19.1	0.226	0.53	19.04	0.331	0.62	19	0.485	0.70
AA₂T₃	21.3	0.251	0.45	21.25	0.339	0.54	21.2	0.509	0.62
AA₃T₃	17.53	0.227	0.51	17.49	0.281	0.60	17.44	0.443	0.68
CD@5%	1.82	0.03	0.05	1.81	0.03	0.03	1.96	0.03	0.04
SEM±	0.63	0.01	0.02	0.62	0.01	0.01	0.67	0.01	0.01

(21.25) & AA₁T₁ (15.74); AA₂T₃ (21.2) & AA₁T₁ (15.7) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of reducing sugar were found in treatments AA₂T₂ (0.204) & AA₁T₂ (0.256) for 7 days, AA₃T₁ (0.275) & AA₃T₂ (0.351) for 14 days and AA₁T₂ (0.407) & AA₂T₁ (0.608) for 21 days storage of sugarcane juice. The higher and lower values of titrable acidity were in treatment combination, AA₁T₃ (0.53) & AA₃T₁ (0.28), for 7 days; AA₁T₃ (0.62) & AA₃T₁ (0.27) for 14 days; AA₁T₃ (0.70) & AA₃T₁ (0.35) for 21 days respectively.

The statistical analysis all the treatments were found significant. Treatment combination AA₂T₃ was at par to AA₂T₁ followed by AA₁T₃ in total sugar, similarly in case of reducing sugar and titrable acidity treatment AA₁T₂ was at par with AA₂T₃ and followed by AA₃T₂; AA₁T₃ was at par with AA₁T₁, followed by AA₃T₃ respectively, in 7 days storage of period.

On 14 days of storage period, as regards for total sugar, treatment AA₂T₃ was at par to AA₂T₁ followed by AA₁T₃; in AA₃T₂ was at par with AA₂T₂ & followed by AA₁T₂; treatment AA₃T₂ was at par to AA₂T₂ followed by AA₁T₂ in titrable acidity respectively.

Finally on 21 days of storage, it was seen that treatment AA₂T₃ was at par to AA₂T₁ and followed by AA₁T₃ for total sugar; for reducing sugar AA₁T₂ was at par to AA₂T₁ & followed by AA₃T₁; in titrable acidity treatment AA₁T₃ was at par with AA₁T₁ followed by treatment AA₃T₃ respectively.

4.5.5 Poly phenol oxidase, TSS and phenol content of stored sugarcane juice at refrigerated condition

Table 24 reveals the results of effect of ascorbic acid and pasteurization temperature on poly phenol oxidase, TSS and phenol content of stored sugarcane juice at refrigerated condition. The maximum and minimum value of PPO content was seen at treatments AA₁T₂, AA₁T₃ (0.034) AA₃T₃ (0.029); & AA₁T₁ (0.031) AA₁T₂, AA₂T₂ (0.023); AA₁T₁ (0.029) & AA₁T₂, AA₂T₂, AA₃T₃ (0.022) for 7, 14 and 21 days storage respectively. Similarly the lowest and highest values of total soluble solid were found in treatments AA₃T₂ (20.4) & AA₃T₃ (18.3), for 7 days, AA₃T₂ (19.8) & AA₃T₃, (18) for 14 days and AA₃T₂ (19.5) & AA₃T₃ (17) for 21 days of storage. The highest and lowest

Table 24 Effect of ascorbic acid and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
AA₁T₁	0.032	18.5	14.65	0.031	18.3	15.05	0.029	18	15.45
AA₂T₁	0.031	19.7	14.76	0.030	19.5	15.16	0.028	19.3	15.56
AA₃T₁	0.033	18.6	14.62	0.029	18.4	15.02	0.027	18.1	15.42
AA₁T₂	0.034	19.1	14.72	0.023	18.8	15.12	0.022	18.5	15.52
AA₂T₂	0.031	19.7	14.74	0.023	19.5	15.14	0.022	19.2	15.54
AA₃T₂	0.032	20.4	14.66	0.025	19.8	15.06	0.023	19.5	15.46
AA₁T₃	0.034	18.5	14.62	0.030	18.3	15.08	0.028	18	15.42
AA₂T₃	0.030	19.7	14.65	0.030	19.5	15.11	0.028	19.2	15.45
AA₃T₃	0.029	18.3	14.74	0.025	18	15.13	0.022	17	15.54
CD@5%	0.03	0.05	0.61	0.03	0.58	0.83	0.06	0.75	1.03
SEM±	0.01	0.02	0.21	0.01	0.20	0.29	0.03	0.26	0.35

value of phenol content were in treatments, AA₂T₁, (14.76) and AA₃T₁, AA₁T₃, (14.62) for 7 days; AA₂T₁ (15.16) & AA₃T₁, (15.02) for 14 days; AA₂T₁ (15.56) & AA₃T₁, AA₁T₃, (15.42) for 21 day storage of sugarcane juice.

The finding presented in the table 24 also shows that treatments have got significant effect on poly phenol oxidase, TSS and phenol content of stored sugarcane juice. The treatment AA₁T₂ and AA₃T₂ was at par to AA₃T₁ followed by AA₃T₂ in poly phenol oxidase, similarly in total soluble sugar and phenol content treatment AA₃T₂ was at par with AA₂T₁ and followed by AA₁T₂; AA₂T₁ was at par with AA₂T₂, followed by AA₁T₂ respectively, in 7 days storage of period.

On 14 days of storage period for total sugar, treatment combination AA₁T₁ was at par to AA₂T₁ followed by AA₃T₁; in AA₃T₂ was at par with AA₂T₁ & followed by AA₁T₂; treatment AA₂T₁ was at par to AA₂T₂ followed by AA₃T₃ in phenol content.

Finally on 21 days of storage it was seen that treatment AA₁T₁ was at par to AA₂T₁ and followed by AA₃T₁ for poly phenol oxidase; for total soluble sugar AA₃T₂ was at par to AA₂T₁ & followed by AA₂T₂; in phenol content treatment AA₂T₁ was at par with AA₂T₂ followed by AA₁T₂.

4.5.6 Poly phenol oxidase, TSS and phenol content of stored sugarcane juice at ambient condition

Table 25 indicates the effect of ascorbic acid and pasteurization temperature on poly phenol oxidase, TSS and phenol content of stored sugarcane juice at ambient condition. The minimum and maximum value of poly phenol oxidase activity was seen in treatments AA₁T₂, AA₁T₃ (0.054) & AA₃T₃ (0.049); AA₁T₁, AA₁T₂, AA₂T₂ (0.043) & AA₁T₃ (0.030); AA₂T₁ (0.031) & AA₃T₃ (0.026) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of total soluble solid were found in treatments AA₃T₂ (20.4) & AA₃T₃ (18) for 7 days, AA₃T₂ (20.1) & AA₃T₃ (17.7) for 14 days and AA₂T₁ (19.2) & AA₃T₃ (17.3) for 21 days. The phenol content, higher and lower values were as follows: AA₂T₁ (14.36) & AA₃T₁, AA₁T₃ (14.22) and for 7 days; AA₂T₁ (14.76), AA₃T₁, AA₃T₁, (14.62) for 14 days; AA₂T₁ (15.16) & AA₃T₁, AA₁T₃, (15.02) for 21 days respectively.

Table 25 Effect of ascorbic acid and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30°C)

Treatment	Storage Period								
	7 days			14 days			21 days		
	PPO	TSS	Phenol	PPO	TSS	Phenol	PPO	TSS	Phenol
AA₁T₁	0.052	18.3	14.25	0.043	18.1	14.65	0.030	17.8	15.05
AA₂T₁	0.051	19.8	14.36	0.042	19.6	14.76	0.031	19.2	15.16
AA₃T₁	0.053	18.4	14.22	0.040	18.2	14.62	0.029	17.9	15.02
AA₁T₂	0.054	19.1	14.32	0.043	18.9	14.72	0.029	18.5	15.12
AA₂T₂	0.051	19.7	14.34	0.043	19.5	14.74	0.030	19.1	15.14
AA₃T₂	0.052	20.4	14.26	0.042	20.1	14.66	0.028	19.7	15.06
AA₁T₃	0.054	18.3	14.22	0.030	18.1	14.62	0.029	17.7	15.02
AA₂T₃	0.050	19.5	14.25	0.035	19.3	14.65	0.029	19	15.05
AA₃T₃	0.049	18	14.34	0.031	17.7	14.74	0.026	17.3	15.14
CD@5%	0.08	0.72	0.62	0.02	0.99	0.50	0.05	1.05	0.65
SEM±	0.03	0.25	0.21	0.01	0.34	0.17	0.02	0.36	0.22

Table 25 shows that treatment AA₁T₂ was at par to AA₃T₁ followed by AA₁T₁ in poly phenol oxidase, similarly in total soluble solid and phenol content treatment AA₃T₂ was at par with AA₂T₁ and followed by AA₂T₂; AA₂T₁ was at par with AA₂T₂, followed by AAT₂ respectively, in 7 days storage of period.

On 14 days of storage period for poly phenol oxidase, treatment AA₁T₁ was at par to AA₂T₁ followed by AA₃T₂; in AA₃T₂ was at par with AA₂T₁ & followed by AA₂T₂; treatment AA₂T₁ was at par to AA₂T₂ followed by AA₁T₂ in phenol content.

Finally on 21 days of storage it was seen that treatment AA₂T₁ was at par to AA₁T₁ and followed by AA₃T₁ for poly phenol oxidase; for total soluble solid treatment AA₃T₂ was at par to AA₂T₁ & followed by AA₂T₂; In case of phenol content, treatment combination AA₂T₁ was at par with AA₂T₂ followed by AA₃T₁.

4.6 Effect of Citric acid and pasteurization temperature on total plate count of stored sugar cane juice at ambient and refrigerated condition.

Table 26 indicates the effect of citric acid and pasteurization temperature on total plate count of stored sugarcane juice at ambient and refrigerated condition. The minimum and maximum value of total plate count was seen in treatments CA₃T₃ (1.66) & CA₁T₁ (1.86); CA₃T₃ (1.76) & CA₁T₁ (1.91); CA₃T₃ (1.86) & CA₁T₁ (2.06) for 7, 14 and 21 days respectively in refrigerated condition. Similarly the lowest and highest values of total plate count in ambient condition for 7, 14 and 21 days were recorded in treatments CA₃T₃ (1.79) & CA₁T₁ (1.96); CA₃T₃ (1.89) & CA₁T₁ (2.06); CA₃T₃ (1.99) & CA₁T₁ (2.16) respectively.

Table 26 shows that treatment CA₁T₁ at par to CA₂T₁ followed by CA₃T₁ in refrigerated condition, similarly, in ambient condition treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁ on 7 days storage of period.

On 14 days of storage period in refrigerated condition, treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁ and in same manner in ambient condition treatment CA₁T₁ was at par with CA₂T₁ followed by CA₃T₁.

Table 26 Effect of citric acid and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
CA₁T₁	1.86	1.96	2.06	1.96	2.06	2.16
CA₂T₁	1.82	1.92	2.02	1.92	2.02	2.12
CA₃T₁	1.80	1.9	2	1.9	2	2.1
CA₁T₂	1.78	1.88	1.98	1.88	1.98	2.08
CA₂T₂	1.75	1.85	1.95	1.85	1.95	2.05
CA₃T₂	1.72	1.82	1.92	1.82	1.92	2.02
CA₁T₃	1.71	1.81	1.91	1.86	1.96	2.06
CA₂T₃	1.69	1.79	1.89	1.82	1.92	2.02
CA₃T₃	1.66	1.76	1.86	1.79	1.89	1.99
CD@5%	0.92	0.87	0.28	0.97	1.04	0.95
SEM±	0.28	0.27	0.41	0.30	0.32	0.29

Finally on 21 days of storage in refrigerated condition, it was seen that treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁. Similarly, in ambient condition treatment CA₁T₁ was at par to CA₂T₁ followed by CA₃T₁ for total plate count.

4.7 Effect of potassium meta bisulphite and pasteurization temperature on total plate count of stored sugar cane juice at ambient and refrigerated condition.

Table 27 indicates the effect of potassium meta bisulphite and pasteurization temperature on total plate count of stored sugarcane juice at ambient and refrigerated condition. The minimum and maximum value of total plate count was seen in treatments KMS₃T₁ (1.70) & KMS₁T₁ (1.78); KMS₃T₁ (1.80) & KMS₁T₁ (1.88); KMS₃T₁ (1.90) & KMS₁T₁ (1.98) for 7, 14 and 21 days respectively in refrigerated condition. Similarly the lowest and highest values of total plate count in ambient condition for 7, 14 and 21 days were recorded in treatments KMS₂T₁, KMS₃T₃ (14.15) & KMS₂T₂, KMS₂T₃ (14.45); KMS₂T₁, KMS₃T₃ (14.55) & KMS₂T₂, KMS₂T₃ (14.85); KMS₃T₂ (14.94) & KMS₂T₂, KMS₂T₃ (15.25) respectively.

Table 27 shows that treatment KMS₁T₁ at par to KMS₂T₁ followed by KMS₃T₁ in refrigerated condition, similarly, in ambient condition treatment KMS₂T₂ was at par to KMS₂T₃, KMS₃T₁ followed by KMS₁T₃ on 7 days storage of period.

On 14 days of storage period in refrigerated condition, treatment KMS₁T₁ was at par to KMS₂T₁ followed by KMS₂T₂ and in same manner in ambient condition treatment KMS₂T₂ & KMS₂T₃ was at par with KMS₃T₁, KMS₁T₃ followed by KMS₁T₂.

Finally on 21 days of storage in refrigerated condition, it was seen that treatment KMS₁T₁ was at par to KMS₂T₁ followed by KMS₂T₂. Similarly, in ambient condition treatment KMS₂T₂, KMS₂T₃ was at par to KMS₃T₁ & KMS₁T₃ followed by KMS₁T₂ for total plate count.

Table 27 Effect of Potassium metabisuphite and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
KMS₁T₁	1.78	1.88	1.98	2.10	2.12	2.15
KMS₂T₁	1.76	1.86	1.96	2.07	2.1	2.1
KMS₃T₁	1.70	1.8	1.9	2.02	2.04	2.07
KMS₁T₂	1.76	1.86	1.96	1.97	1.99	2.01
KMS₂T₂	1.75	1.85	1.95	1.93	1.96	1.99
KMS₃T₂	1.73	1.83	1.93	1.88	1.9	1.93
KMS₁T₃	1.73	1.83	1.93	1.85	1.88	1.89
KMS₂T₃	1.73	1.83	1.93	1.80	1.82	1.82
KMS₃T₃	1.72	1.82	1.92	1.76	1.78	1.8
CD@5%	0.60	0.76	0.71	0.81	0.55	0.92
SEM±	0.19	0.24	0.22	0.25	0.17	0.28

4.8 Effect of potassium sorbate and pasteurization temperature on total plate count of stored sugar cane juice at ambient and refrigerated condition.

Table 28 indicates the effect of potassium sorbate and pasteurization temperature on total plate count of stored sugarcane juice at ambient and refrigerated condition. In refrigerated condition the minimum and maximum value of total plate count was seen in treatments PS_3T_1 (1.60) & PS_1T_1 (1.68); PS_3T_1 (1.70) & PS_1T_1 (1.88); PS_3T_1 (1.80) & PS_1T_1 (1.88) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of total plate count in ambient condition for 7, 14 and 21 days were recorded in treatments PS_2T_1 , PS_3T_3 (13.15) & PS_2T_2 , PS_2T_3 (13.45); KMS_2T_1 , PS_3T_3 (13.55) & PS_2T_2 , PS_2T_3 (13.85); PS_3T_2 (13.94) & PS_2T_2 , PS_2T_3 (14.25) respectively.

Table 27 shows that treatment PS_1T_1 at par to PS_2T_1 followed by PS_3T_1 in refrigerated condition, similarly, in ambient condition treatment PS_2T_2 was at par to PS_2T_3 , PS_3T_1 followed by PS_1T_3 on 7 days storage of period.

On 14 days of storage period in refrigerated condition, treatment PS_1T_1 was at par to PS_2T_1 followed by PS_2T_2 and in same manner in ambient condition treatment PS_2T_2 & PS_2T_3 was at par with PS_3T_1 , PS_1T_3 followed by PS_1T_2 .

Finally on 21 days of storage in refrigerated condition, it was seen that treatment PS_1T_1 was at par to PS_2T_1 followed by PS_2T_2 . Similarly, in ambient condition treatment PS_2T_2 , PS_2T_3 was at par to PS_3T_1 & PS_1T_3 followed by PS_1T_2 for total plate count.

4.9 Effect of ascorbic acid and pasteurization temperature on total plate count of stored sugarcane juice at ambient and refrigerated condition

Table 29 indicates the effect of potassium sorbate and pasteurization temperature on total plate count of stored sugarcane juice at ambient and refrigerated condition. In refrigerated condition the minimum and maximum value of total plate count was seen in treatments PS_3T_1 (1.60) & PS_1T_1 (1.68); PS_3T_1 (1.70) & PS_1T_1 (1.88); PS_3T_1 (1.80) & PS_1T_1 (1.88) for 7, 14 and 21 days respectively. Similarly the lowest and highest values of total plate count

Table 28 Effect of Potassium sorbate and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
PS₁T₁	1.76	1.83	1.9	1.88	1.98	2.08
PS₂T₁	1.76	1.88	1.96	1.89	1.97	2.04
PS₃T₁	1.71	1.82	1.91	1.87	1.97	2.06
PS₁T₂	1.78	1.85	1.95	1.76	1.89	1.98
PS₂T₂	1.75	1.85	1.92	1.78	1.89	1.97
PS₃T₂	1.75	1.82	1.93	1.89	1.98	2.14
PS₁T₃	1.69	1.75	1.83	1.98	2.12	2.19
PS₂T₃	1.71	1.86	1.91	1.67	1.79	1.94
PS₃T₃	1.75	1.80	1.89	1.76	1.89	1.97
CD@5%	0.58	0.99	0.77	0.50	0.61	0.56
SEM±	0.18	0.30	0.24	0.26	0.19	0.17

Table 29 Effect of Ascorbic acid and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
AA₁T₁	1.72	1.75	1.78	1.73	1.76	1.80
AA₂T₁	1.75	1.78	1.80	1.76	1.78	1.81
AA₃T₁	1.78	1.82	1.83	1.79	1.84	1.85
AA₁T₂	1.79	1.81	1.84	1.82	1.86	1.88
AA₂T₂	1.82	1.84	1.85	1.83	1.86	1.87
AA₃T₂	1.85	1.87	1.89	1.97	2.01	2.2
AA₁T₃	1.88	1.93	1.94	1.94	1.99	2.
AA₂T₃	1.98	2.01	2.2	1.98	2.07	2.12
AA₃T₃	1.99	2.05	2.3	1.94	2.2	2.4
CD@5%	0.63	2.82	0.77	1.61	0.77	2.81
SEM±	0.19	0.87	0.24	0.80	0.24	0.86

in ambient condition for 7, 14 and 21 days were recorded in treatments PS₂T₁, PS₃T₃ (13.15) & PS₂T₂, PS₂T₃ (13.45); KMS₂T₁, PS₃T₃ (13.55) & PS₂T₂, PS₂T₃ (13.85); PS₃T₂ (13.94) & PS₂T₂, PS₂T₃ (14.25) respectively.

Table 27 shows that treatment PS₁T₁ at par to PS₂T₁ followed by PS₃T₁ in refrigerated condition, similarly in ambient condition treatment PS₂T₂ was at par to PS₂T₃, PS₃T₁ followed by PS₁T₃ on 7 days storage of period.

On 14 days of storage period in refrigerated condition, treatment PS₁T₁ was at par to PS₂T₁ followed by PS₂T₂ and in same manner in ambient condition treatment PS₂T₂ & PS₂T₃ was at par with PS₃T₁, PS₁T₃ followed by PS₁T₂.

Finally on 21 days of storage in refrigerated condition, it was seen that treatment PS₁T₁ was at par to PS₂T₁ followed by PS₂T₂. Similarly, in ambient condition treatment PS₂T₂, PS₂T₃ was at par to PS₃T₁ & PS₁T₃ followed by PS₁T₂ for total plate count.

4.10 Effect of Citric acid and pasteurization temperature on Overall acceptability sensory score of stored sugarcane juice

Table 30 indicates the results of effect of citric acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition. It varies from 8.46 to 8.86, 8.35 to 8.78 and 8.29 to 8.67 at 7, 14 and 21 days store sugarcane juice under refrigerated conditions. The mean value of sensory score at ambient condition of storage were found to be 8.35 to 8.74, 8.26 to 8.6 and 8.17 to 8.45 at 7, 14 and 21 days store sugarcane juice. The also reveals that stored sugarcane juice under both conditions of storage have received the sensory score above 8.0 which comes under category of liked very much.

4.11 Effect of potassium metabisulfite and pasteurization temperature on Overall acceptability sensory score of stored sugarcane juice

Table 31 shows the results of effect of potassium metabisulfite and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition. It varies from 8.85 to 8.90, 8.61 to 8.75 and 8.34 to 55 at 7, 14 and 21 days store sugarcane juice

Table 30 Effect of citric acid and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
CA₁T₁	8.46	8.35	8.29	8.35	8.26	8.18
CA₂T₁	8.86	8.76	8.65	8.56	8.45	8.34
CA₃T₁	8.67	8.56	8.48	8.38	8.28	8.17
CA₁T₂	8.68	8.54	8.45	8.49	8.37	8.26
CA₂T₂	8.85	8.78	8.67	8.58	8.46	8.35
CA₃T₂	8.76	8.67	8.56	8.64	8.56	8.45
CA₁T₃	8.67	8.56	8.45	8.74	8.62	8.45
CA₂T₃	8.56	8.46	8.35	8.38	8.29	8.23
CA₃T₃	8.65	8.56	8.45	8.49	8.34	8.26

Table 31 Effect of potassium metabisulfite and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
KMS₁T₁	8.98	8.71	8.55	8.84	8.57	8.25
KMS₂T₁	8.95	8.72	8.53	8.75	8.56	8.39
KMS₃T₁	8.94	8.74	8.53	8.75	8.64	8.28
KMS₁T₂	8.93	8.69	8.34	8.53	8.26	8.49
KMS₂T₂	8.97	8.75	8.51	8.63	8.34	8.57
KMS₃T₂	8.85	8.61	8.45	8.73	8.52	8.38
KMS₁T₃	8.96	8.74	8.53	8.85	8.73	8.40
KMS₂T₃	8.94	8.68	8.43	8.75	8.62	8.58
KMS₃T₃	8.93	8.74	8.52	8.73	8.61	8.56

under refrigerated conditions. The mean value of sensory score at ambient condition of storage were found to be 8.53 to 8.85, 8.34 to 8.73 and 8.25 to 8.58 at 7, 14 and 21 days stored sugarcane juice. The table also reveals that stored sugarcane juice under both conditions of storage have again received the sensory score above 8.0 which comes under category of liked very much.

4.12 Effect of potassium sorbate and pasteurization temperature on Overall acceptability sensory score of stored sugarcane juice

The effect of potassium sorbate and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition is presented in table 32. It varies from 8.57 to 8.84, 8.34 to 8.71 and 8.20 to 8.52 at 7, 14 and 21 days store sugarcane juice under refrigerated conditions. The mean value of sensory score at ambient condition of storage were found to be 8.54 to 8.76, 8.43 to 8.67 and 8.34 to 8.57 at 7, 14 and 21 days stored sugarcane juice. The table also reveals that stored sugarcane juice under both conditions of storage have again received the sensory score above 8.0 which comes under category of liked very much.

4.13 Effect of ascorbic acid and pasteurization temperature on Overall acceptability sensory score of stored sugarcane juice

The effect of ascorbic acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition is presented in table 33. It varies from 8.68 to 8.94, 8.67 to 8.85 and 8.46 to 8.76 at 7, 14 and 21 days store sugarcane juice under refrigerated conditions. The mean value of sensory score at ambient condition of storage were found to be 8.36 to 8.65, 8.25 to 8.57 and 8.16 to 8.48 at 7, 14 and 21 days stored sugarcane juice. The table also reveals that stored sugarcane juice under both conditions of storage have again received the sensory score above 8.0 which comes under category of liked very much.

The finding presented indicates the possibilities of preservation of sugarcane juice by using chemical preservatives. The justification of findings with the results of other research are given in next chapter.

Table 32 Effect of potassium sorbate and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
PS₁T₁	8.65	8.43	8.24	8.65	8.50	8.45
PS₂T₁	8.57	8.34	8.20	8.54	8.43	8.37
PS₃T₁	8.57	8.37	8.20	8.76	8.65	8.57
PS₁T₂	8.58	8.71	8.52	8.74	8.65	8.55
PS₂T₂	8.69	8.41	8.20	8.64	8.56	8.45
PS₃T₂	8.84	8.65	8.43	8.73	8.46	8.34
PS₁T₃	8.75	8.54	8.35	8.64	8.56	8.45
PS₂T₃	8.74	8.51	8.37	8.64	8.57	8.45
PS₃T₃	8.82	8.64	8.40	8.75	8.67	8.56

Table 33 Effect of ascorbic acid and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

Treatment	Storage Period					
	(5°C)			(30°C)		
	7 days	14 days	21 days	7 days	14 days	21 days
AA₁T₁	8.78	8.67	8.56	8.45	8.35	8.23
AA₂T₁	8.87	8.79	8.65	8.36	8.25	8.16
AA₃T₁	8.79	8.68	8.56	8.36	8.26	8.18
AA₁T₂	8.68	8.57	8.46	8.47	8.36	8.25
AA₂T₂	8.75	8.69	8.57	8.58	8.49	8.35
AA₃T₂	8.94	8.85	8.76	8.65	8.57	8.48
AA₁T₃	8.73	8.68	8.76	8.54	8.47	8.36
AA₂T₃	8.85	8.67	8.57	8.47	8.36	8.27
AA₃T₃	8.85	8.76	8.68	8.53	8.45	8.39

Summary, conclusion, Suggestion and Further Work

Summary

Sugarcane is one of the most important agro-industrial crops in our country. The total production of sugarcane in India during the year 2017-18 was 330 million tones. Uttar Pradesh and Maharashtra are the two largest sugarcane producing states in the country, accounting for more than 80 percent of the annual crop production (Anonymous 2011). India stands second among other sugarcane growing countries, contributing nearly 20.40% area and 18.60% production. Fresh Sugarcane juice is popular beverage in Southeast Asia, South Asia, Latin America, and also in other countries where sugarcane is grown commercially. Sugarcane juice provide vitamins A, C, B1, B2, B3, B5, B6 and iron with a high concentration of phytonutrients (including chlorophyll), antioxidants, proteins, soluble fibre and numerous other health supportive compounds (Karthikeyan and Samipillai, 2010). As sugarcane juice is rich in sugar, it starts fermenting immediately after extraction and also turns brown due to poly phenol oxidease activity (Qudsieh *et al* 2002). Microbial fermentation of the juice turns its taste to sour within few hours of extraction (Yusof *et al* 2000). All these negative changes limit the processing and marketing of sugarcane juice (Eggleston 2002). Development of a preservation method, which can control microbial growth in raw sugarcane juice and maintain its freshness, is a challenging problem for ensuring its wider distribution and availability. In this context the present investigation entitle “studies of shelf life extension of sugarcane juice” has been planned. The find are summarized as under

The effect of levels of citric acid and pasteurization temperature shows significant effect of treatment combination on moisture, carbohydrate ash, total sugar, reducing sugar, titrable acidity, poly phenol oxidease (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice under refrigerated and ambient condition.

The effect of levels of potassium metabisulfite and pasteurization temperature shows significant effect of treatment combination on moisture,

carbohydrate ash, total sugar, reducing sugar, titrable acidity, poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice under refrigerated and ambient condition. .

The effect of levels of potassium sorbate and pasteurization temperature exhibited significant effect of treatment combination on moisture, carbohydrate ash, total sugar, reducing sugar, titrable acidity, poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice under refrigerated and ambient condition. .

The effect of levels of ascorbic acid and pasteurization temperature shows significant effect of treatment combination on moisture, carbohydrate ash, total sugar, reducing sugar, titrable acidity, poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice under refrigerated and ambient condition.

The levels of citric acid, potassium metabisulphite, potassium sorbate, ascorbic acid and pasteurization temperature exhibits significant effect on total plate count at refrigerated and ambient condition.

The effect of citric acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition varies from 8.46 to 8.86, 8.35 to 8.78, 8.29 to 8.67 and 8.35 to 8.74, 8.26 to 8.6 and 8.17 to 8.45 at 7, 14 and 21 days stored sugarcane juice under refrigerated and ambient condition respectively.

The effect of levels of potassium metabisulfite and pasteurization on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition varies from 8.85 to 8.90, 8.61 to 8.75 and 8.34 to 55 and 8.53 to 8.85, 8.34 to 8.73, 8.25 to 8.58 at 7, 14 and 21 days storage sugarcane juice.

The results of effect of potassium sorbate and pasteurization temperature on varies from 8.57 to 8.84, 8.34 to 8.71 and 8.20 to 8.52 at 7, 14 and 21 days store sugarcane juice under refrigerated conditions. While the mean value of sensory score at ambient condition of storage were found to be 8.54 to 8.76, 8.43 to 8.67 and 8.34 to 8.57 at 7, 14 and 21 days stored sugarcane juice

The effect of ascorbic acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition varies from 8.68 to 8.94, 8.67 to 8.85, 8.46 to 8.76 and 8.36 to 8.65, 8.25 to 8.57, 8.16 to 8.48 at 7, 14 and 21 days storage of sugarcane juice respectively.

Conclusion and Suggestion for Further Work

Conclusion

In a nut shell it is reduced from the findings of present investigation and their justification that different treatment combination of preservatives with pasteurization temperature can be successfully utilized for development of preserved bottled sugarcane juice without sacrificing the quality of product. The shelf life of sugarcane juice can be enhanced for 14 days by using treatment combination potassium metabisulfite and pasteurization temperature of at concentration of 0.15ppm and 70 °C. respectively. The preservation technology has got great potential for value addition in food processing industries. The standardized optimum condition of can be successfully translated at commercial scale for domestic and commercial application.

Suggestion for Further Work

1. Further research work is needed to explore use other preservative for long term storage of sugarcane juice.
2. Research work is needed to study the effect of preservatives and processing technique on quality of product and health aspects of consumer
3. Further research work is required to study on ultra filtration of sugarcane juice for long term storage.

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ABSTRACT

The present investigations entitled "**Study on shelf life extension of sugarcane juice**". Sugarcane juice was subjected to evaluation with treatment (Citric acid 0.2, 0.4, 0.6, potassium meta-bisulphite 0.15, 0.20, 0.25 ppm, potassium sorbate 0.25, 0.50, 0.75 %, ascorbic acid 0.2, 0.4, 0.6 %, pasteurization with 60, 70, 80 °C and storage at 5 and 30 °C temperature). The total number of treatments were subjected to various chemical evaluation (Moisture, carbohydrates, ash, total sugar, reducing sugar, TSS, titrable acidity, polyphenol oxidase activity, phenol content and microbial count) as well as sensory evaluation with the storage study for 7, 14 and 21 days in glass bottle. The best treatment combination was 70 °C pasteurization temperature, 0.15 ppm potassium metabisulphate, storage time 14 days at 5 °C temperature. The microbial count was minimum in the 14 days stored at 5 °C with the treatment combination of 0.15 % potassium metabisulphite and 70 °C pasteurization temperature. The stored sugarcane juice was subjected to sensory evaluation and the highest sensory score was 14 days stored sugarcane juice at 5 °C temperature.

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DISCUSSION

Extension of shelf life of sugarcane juice is an area of current research interest because of unavailability of viable technology of sugarcane juice processing and preservation at domestic and commercial level. Very meager information is available on the same; therefore the proposed research work has been planned. This chapter deals with the justification of the findings obtained in the experiments related to studies on effect of different chemical preservatives on chemical composition, microbial counts and sensory quality attributes of sugarcane juice for 7, 14 and 21 days storage period. The findings obtained in various experiments of study have been explained with the help of figures 2 to 33 and with the reported values of various researchers and discussed as under:

5.1 Effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice.

The finding presented through figure 2 to 3 exhibits that after 7 days of storage period of sugarcane juice under refrigerated condition, the moisture, carbohydrate and ash varied from 75.40 to 80.88%; 13.68 to 16.20%; 0.2 to 0.6 % respectively. On 14 days of storage they varied from 76.75 to 82.81%; 13.73 to 16.27%; 0.2 to 0.6% respectively. Similarly upon 21 days of storage period it varied from 76.4 to 80.79%; 13.82 to 17.23%; 0.2 to 0.6% respectively.

Under ambient condition of storage of sugarcane juice the moisture, carbohydrate and ash content showed variation from 76.30 to 81.70%; 13.73 to 15.76%; 0.2 to 0.4% (7 days) 76.77 to 84.35%; 13.64 to 17.16%; 0.2 to 0.4% (14 days) and 76.10 to 81.40%; 13.78 to 18.16%; 0.2 to 0.4% (21 days) of storage period respectively. The similar findings of moisture, carbohydrate and ash content were also reported by Chouhan (2002).

5.2. Effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice.

The finding presented in figure 4 to 5 indicates that on 7 days of storage period of sugarcane juice the effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity content varied from 16.78 to 19.91%; 0.204 to 0.264%; 0.26 to 0.53% respectively. Similarly for 14 days of storage the same varied from 16.71 to 19.86%; 0.258 to 0.364%; 0.35 to 0.62% respectively. During 21 days of storage period the values of total sugar, reducing sugar and titrable acidity were found to be 16.65 to 19.81%; 0.399 to 0.679%; 0.42 to 0.70% respectively. The titrable acidity of sugarcane juice increased during storage. This might be due to the fact that acetic and lactic acid production took place during storage.

A cursory view of findings informs that at ambient condition during 7 days of storage period of sugarcane juice the effect of citric acid and pasteurization temperature on the total sugar, reducing sugar and titrable acidity, indicates variability of 17.86 to 22.62%; 0.204 to 0.271%; 0.28 to 0.53% respectively. Similarly on 14 days of storage the same varied from 17.81 to 22.56%; 0.273 to 0.370%; 0.37 to 0.62% respectively. Upon 21 days of storage period the same varied from 17.77 to 19.81%; 0.424 to 0.679%; 0.45 to 0.70 %respectively. It may be due to total sugars content was decreased during storage. This could be due to action of microorganism present in the juice, converted the total sugars into reducing sugars (glucose and fructose). Sugarcane juice stored at 30⁰ C recorded higher acidity range compared to those stored at 5⁰ C similar findings were also reported by Chauhan et al. (1997) and Yusuf et al. (2002).

There was an increase in reducing sugars during storage. This might be due to the fact that the hydrolysis of non-reducing sugars (sucrose) into reducing sugars (glucose and fructose) by the action of microorganism. The results are in agreement with findings of Singh et al. (2002).

5.3. Effect of citric acid and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice.

The effect of citric acid and pasteurization temperature on poly phenol oxidase, total soluble solid and phenol content of stored sugarcane juice are depicted through figure 6 to 7. The concentration of these chemical constituents under refrigerated condition varied from 0.027 to 0.035, 19.1 to 20.5, 14.55 to 14.76%; 0.018 to 0.023, 18.8 to 19.8, 14.95 to 15.16% and 0.015 to 0.024, 18.5 to 19.6, 15.4 to 15.56% at 7, 14 and 21 days of storage respectively. Total soluble solids were decreased with increase in storage periods. This might be due to the hydrolysis of sugars present in the juice. Total soluble solids decrease was less in glass bottle during storage. Similar findings were also reported by Chauhan et al. (1997) and Yusuf et al. (2002).

Under ambient condition upon 7 days of storage period of sugarcane juice the polyphenol oxidase, total soluble solid and phenol content values varied from 0.047 to 0.055, 18.8 to 19.8, 14.15 to 14.36% respectively. Similarly on 14 days of storage the same varied from 0.031 to 0.031, 18.6 to 19.6; 14.55 to 14.76% respectively. Similarly on 21 days of storage the values of same varied from 0.026 to 0.039, 18.3 to 19.3, 14.95 to 15.16% respectively. The decrease in PPO was due to pasteurization temperatures (85, 90 and 95°C) since no residual activity was detected in processed samples. Mao (2007) and others reported that PPO activity converts phenolic compounds into brown colored polymers, which causes darkening when sugarcane is crushed. Despite inactivation of PPO during pasteurization alterations in color were still observed. Similar findings were also reported by Chauhan et al. (1997) and Yusuf et al. (2002).

5.4 Effect of potassium metabisulfite and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice.

A cursory view of Figure 8 to 9 exhibits that after 7 days of storage period of sugarcane juice under refrigerated condition. The moisture, carbohydrate and ash varied from 75.20 to 81.43%, 13.73 to 16.74%, 0.2 to

0.4% respectively. On 14 days of storage it varied from 77.50 to 85.40%, 13.59 to 17.02%, and 0.2 to 0.6% respectively. Similarly upon 21 days of storage period it varied from 76.40 to 80.12%, 13.92 to 16.40%, and 0.2 to 0.4% respectively.

Under ambient condition of storage of sugarcane juice the moisture, carbohydrate and ash content showed variation from 77.81 to 81.74%, 13.78 to 17.23% and 0.2 to 0.4% (7 days) 77.40 to 85.10%, 13.78 to 16.60%, 0.2 to 0.4% (14 days) and 76.50 to 81.75%, 13.38 to 18.33% and 0.2 to 0.6% (21 days) of storage period respectively. The similar findings were also reported by Chauhan et al. (1997) and Yusuf et al. (2002).

5.5 Effect of potassium metabisulfite and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice.

Findings of experiments are depicted through figure 10 to 11. Under refrigerated condition at 7 days of storage period of sugarcane juice, the effect of potassium metabisulfite and pasteurization temperature on total sugar, reducing sugar and titrable acidity values found to be 16.78 to 22.62%; 0.209 to 0.258%; 0.27 to 0.51% respectively. Similarly on 14 days of storage the values of same varied from 16.73 to 22.56%; 0.266 to 0.370%; 0.36 to 0.60% respectively. Finally upon 21 days of storage period it varied from 16.69 to 22.50%; 0.374 to 0.657%; 0.43 to 0.68 %respectively.

The effect of potassium metabisulfite and pasteurization temperature on total sugar, reducing sugar and titrable acidity of sugarcane juice under ambient condition during 7 days of storage period varied from 15.50 to 23.61%; 0.212 to 0.255%; 0.28 to 0.53% respectively. Similarly on 14 days of storage it varies from 15.00 to 23.56%; 0.271 to 0.370%; 0.37 to 0.62% respectively. Upon 21 days of storage period the values of total sugar, reducing sugar and titrable acidity varied from 14.49 to 23.51%; 0.407 to 0.679%; 0.45 to 0.70% respectively. The present findings of sugar content are in conformity with reported results of Chauhan et al. (1997) and Hesham et al. (2010).

5.6 Effect of potassium metabisulfite and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice.

The results reveals that (Figure 12 to 13) under refrigerated condition at 7 days of storage period of sugarcane juice the effect of potassium metabisulfite and pasteurization temperature on polyphenol oxidase, total soluble solid and phenol values varied from 0.034 to 0.048; 18.20 to 19.90; 14.14 to 14.45 %respectively. Upon 14 days of storage the same varied from 0.025 to 0.045; 18.00 to 19.70; 14.54 to 14.85% respectively. Similarly on 21 days of storage period it varied from 0.027 to 0.041; 17.70 to 19.50; 14.94 to 15.25 %respectively.

On 7 days (ambient condition) of storage period of sugarcane juice effect of potassium metabisulfite and pasteurization temperature on polyphenol oxidase, total soluble solid and phenol values shows that it varies from 0.047 to 0.055; 18.8 to 19.8; 14.15 to 14.36% respectively. Upon 14 days of storage it varied from 0.031 to 0.031; 18.6 to 19.6; 14.55 to 14.76 % respectively. And on 21 days of storage period the same varied from 0.026 to 0.039; 18.3 to 19.3; 14.95 to 15.16% respectively. The present findings are in conformity with reported results of Chauhan et al. (1997) and Hesham et al. (2010)

5.7 Effect of Potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice.

Findings depicted through figure (14 to 15) indicates that under refrigerated condition. The moisture, carbohydrate and ash content, values varied from 75.90 to 81.20%,13.50 to 16.74%, 0.2 to 0.4%;77.20 to 84.63%,13.06 to 16.88%,0.2 to 0.4%; 76.40 to 80.79%, 13.92 to 17.23% and 0.2 to 0.4 % at 7, 14 and 21 days storage respectively.

Under ambient condition the effect of potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash values shows that it varies from 76.58 to 80.57%; 13.78 to 15.76%; 0.2 to 0.4% respectively at 7 days of storage period of sugarcane juice. Similarly on 14 days of

storage, the same varied from 77.85 to 83.47%; 13.73 to 17.16% and 0.2 to 0.4% respectively. And upon 21 days of storage period it varied from 76.72 to 81.75%; 12.19 to 19.10%; 0.2 to 0.4 % respectively.

5.8 Effect of potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice.

The effect of potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity are shown through figure 16 to 17. They varied from 16.78 to 22.62%, 0.209 to 0.258%, 0.27 to 0.51%; 16.73 to 22.56%, 0.266 to 0.370%, 0.36 to 0.60 %and 16.69 to 22.50%, 0.374 to 0.657%, 0.43 to 0.68% respectively on 7,14 and 21 days of storage period respectively.

The effect of these treatments under ambient conduits of storage shows that total sugar, reducing sugar and titrable acidity values varied from 15.21 to 21.88%, 0.204 to 0.255%, 0.28 to 0.53%; 15.15 to 21.84%, 0.258 to 0.374%, 0.27 to 0.62% and 15.11 to 21.80%, 0.407 to 0.599% and 0.35 to 0.70 % at 7, 14 and 21 days of storage of sugarcane juice.

5.9 Effect of Potassium sorbate and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice.

In refrigerated condition at Figure 18 to 19 days of storage period of sugarcane juice the effect of potassium sorbate and pasteurization temperature on polyphenol oxidase, total soluble solid and phenol content, exhibits from 0.031 to 0.035; 18.30 to 20.70; 14.52 to 14.87% respectively. Same varied from 0.025 to 0.029; 18.00 to 20.50; 14.92 to 15.27% respectively on 14 days of storage period. Similarly upon 21 days of storage period it varied from 0.026 to 0.028; 17.80 to 20.20; 15.32 to 15.67% respectively.

Under ambient condition, effect of potassium sorbate and pasteurization temperature on polyphenol oxidase, total soluble solid and phenol on sugarcane juice at 7 days of storage period, shows that values varied from 0.029 to 0.035; 18.20 to 20.50; 14.12 to 14.47% respectively. On

14 days of storage period the same varied from 0.025 to 0.032; 18.00 to 20.20; 14.52 to 14.87 respectively. And upon 21 days of storage period the values of total sugar, reducing sugar and titrable acidity varied from 0.025 to 0.032; 18.00 to 19.50; 14.92 to 15.27 % respectively.

5.10 Effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice.

Figure 20 to 21 shows that under refrigerated condition 7 days of storage period of sugarcane juice the effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content, values varied from 75.40 to 81.20%; 13.73 to 16.27%; 0.2 to 0.4% respectively. Same varied from 76.54 to 84.60%; 13.59 to 15.70%; 0.2 to 0.6 % respectively on 14 days of storage period. And upon 21 days of storage period it varied from 76.24 to 82.25%; 13.73 to 17.23%; 0.2 to 0.6% respectively.

The effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content at ambient condition at 7 days of storage period of sugarcane juice the values varied from 76.32 to 78.00%; 13.64 to 16.67%; 0.2 to 0.4 % respectively. Upon 14 days of storage it varied from 78.00 to 84.60%; 13.59 to 16.53%; 0.2 to 0.4% respectively. Similarly, the same varied from 76.50 to 81.40%; 13.92 to 18.41%; 0.2 to 0.4% respectively 21 days of storage period. The present findings are in conformity with reported results of Chauhan et al. (1997) and Hesham et al. (2010)

5.11 Effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice.

Figure 22 to 23 indicates that under refrigerated condition at 7 days of storage period of sugarcane juice the effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity values varied from 15.63 to 22.62%; 0.204 to 0.247%; 0.26 to 0.51% respectively. Similarly, it varied from 15.58 to 22.57%; 0.258 to 0.370%; 0.35 to 0.60% respectively on 14 days of storage. And upon 21 days of storage

period its values varied from 15.47 to 22.53%; 0.443 to 0.679%; 0.43 to 0.67% respectively.

The effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity under ambient storage condition at 7 days of storage period of sugarcane juice, the values varied from 15.79 to 21.30%; 0.204 to 0.256%; 0.28 to 0.53% respectively. Similarly, its value varied from 15.74 to 21.25%; 0.275 to 0.351%; 0.27 to 0.62% respectively on 14 days of storage. Upon 21 days of storage period the values of total sugar, reducing sugar and titrable acidity varied from 15.70 to 21.20%; 0.407 to 0.608%; 0.35 to 0.70 %respectively. The present findings are in conformity with reported results of Chauhan et al. (1997) and Hesham et al. (2010)

5.12 Effect of ascorbic acid and pasteurization temperature on poly phenol oxidease (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice.

A cursory view of the findings figure 24 to 25, showed that under refrigerated condition on 7 days of storage period of sugarcane juice on the effect of ascorbic acid and pasteurization temperature on polyphenol oxidase, total soluble solid and phenol values varied from 0.029 to 0.034; 18.30 to 20.40; 14.62 to 14.76% respectively. Upon 14 days of storage same varied from 0.023 to 0.031; 18.00 to 19.80; 15.02 to 15.16% respectively. Similarly the values of total sugar, reducing sugar and titrable acidity varied from 0.022 to 0.029; 17.00 to 19.50; 15.42 to 15.56% respectively on 21 days of storage period.

Under ambient storage condition the effect of ascorbic acid and pasteurization temperature of stored sugarcane juice on the polyphenol oxidase, total soluble solid and phenol content, values varied from 0.049 to 0.054; 18.00 to 20.40; 14.22 to 14.36% respectively on 7 days of storage period. The same varied from 0.030 to 0.043; 17.70 to 20.10; 14.62 to 14.76 % respectively on 14 days of storage. And upon 21 days of storage period the values varied from 0.026 to 0.032; 17.30 to 19.70; 15.02 to 15.16 % respectively. The present findings are in conformity with reported results of Chauhan et al. (1997) and Hesham et al. (2010)

5.13 Effect of chemical preservatives and pasteurization temperature on total plate count of stored sugarcane juice.

Figure 26 shows that Citric acid and pasteurization temperature exhibits significant effect on total plate count at refrigerated and ambient condition. It varied from 1.72 to 1.78 cfu & 1.82 to 1.88 cfu; 1.9 to 1.98 cfu and ambient condition 1.78 to 2.12 cfu; and 1.86 to 2.06 cfu & 1.99 to 2.16 cfu respectively at 7, 14 & 21 days of storage period of sugarcane juice.

Figure 27 shows the effect of potassium metabisulphite and pasteurization temperature on total plate count at refrigerated and ambient condition on 7, 14 & 21 days of storage period of sugarcane juice, It varied from 1.78 to 2.12 cfu; 1.80 to 2.15cfu & 1.92 to 1.17 cfu; and 1.90 to 1.98 cfu & 14.94 to 15.25 cfu respectively.

The effect of potassium sorbate and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition for 7, 14 & 21 days varied from 1.60 to 1.68 cfu & 13.14 to 13.45 cfu; 1.70 to 1.78 cfu & 13.55 to 13.85 cfu; and 1.80 to 1.88 cfu & 13.94 to 14.25 cfu respectively (Figure 28).

The effect of ascorbic acid and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition for 7,14 and 21 days varied from 1.99 to 1.72 to 2.05;. 1.78 to 1.94 to 1.76 to 2.2 1.80 to2.4 respectively (Figure 29).

5.14 Effect of chemical preservatives and pasteurization temperature on sensor quality attributes of stored sugarcane juice.

5.14.1 Effect of citric acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice.

The effect of citric acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition is presented through Figure 30 (5^o and 30^oC). It varies from 8.46 to 8.86, 8.35 to8.78, 8.29 to 8.67 and 8.35 to 8.74, 8.26 to8.6 and 8.17 to 8.45 at 7, 14 and 21 days stored sugarcane juice under refrigerated and ambient condition respectively. The stored sugarcane juice under both

conditions of storage have received the sensory score above 8.0 which comes under category of liked very much category.

5.14.2 Effect of potassium metabisulfite and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice.

The effect of levels of potassium metabisulfite and pasteurization on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition varies from 8.85 to 8.90, 8.61 to 8.75 and 8.34 to 55 and 8.53 to 8.85, 8.34 to 8.73, 8.25 to 8.58 at 7, 14 and 21 days storage sugarcane juice under refrigerated and ambient conditions respectively. The table also reveals that stored sugarcane juice under both conditions of storage have again received the sensory score above 8.0 which comes under category of liked very much. Figure 31 (5⁰ and 30⁰C).

5.14.3 Effect of potassium sorbate and pasteurization temperature on Overall acceptability sensory score of stored sugarcane juice.

The results of effect of potassium sorbate and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition is presented through fig 32. It varies from 8.57 to 8.84, 8.34 to 8.71 and 8.20 to 8.52 at 7, 14 and 21 days store sugarcane juice under refrigerated conditions. While the mean value of sensory score at ambient condition of storage were found to be 8.54 to 8.76, 8.43 to 8.67 and 8.34 to 8.57 at 7, 14 and 21 days stored sugarcane juice. Figure 32 (5⁰ and 30⁰C).

5.14.4 Effect of ascorbic acid and pasteurization temperature on Overall acceptability sensory score of stored sugarcane juice.

A findings presented through figure33 (5⁰ and 30⁰C) indicates that effect of ascorbic acid and pasteurization temperature on overall acceptability sensory score of stored sugarcane juice at ambient and refrigerated condition. The mean value varies from 8.68 to 8.94, 8.67 to 8.85, 8.46 to 8.76 and 8.36 to 8.65, 8.25 to 8.57, 8.16 to 8.48 at 7.14 and 21 days storage of sugarcane juice respectively. The stored sugarcane juices under both conditions of storage have again received the sensory score above 8.0 which comes under category of liked very much. The present findings of acceptable

sensory score are in conformity with reported results of Chauhan et al. (1997) and Hesham et al. (2010)

In a nutshell it is reduced from the findings of present investigation and their justification that different treatment combination of preservatives with pasteurization temperature can be successfully utilized for development of preserved bottled sugarcane juice without sacrificing the quality of product. The preservation technology has got great potential for value addition in food processing industries. The standardized optimum condition of can be successfully translated at commercial scale for domestic and commercial application.

Fig.1 Chemical Composition of sugarcane juice.

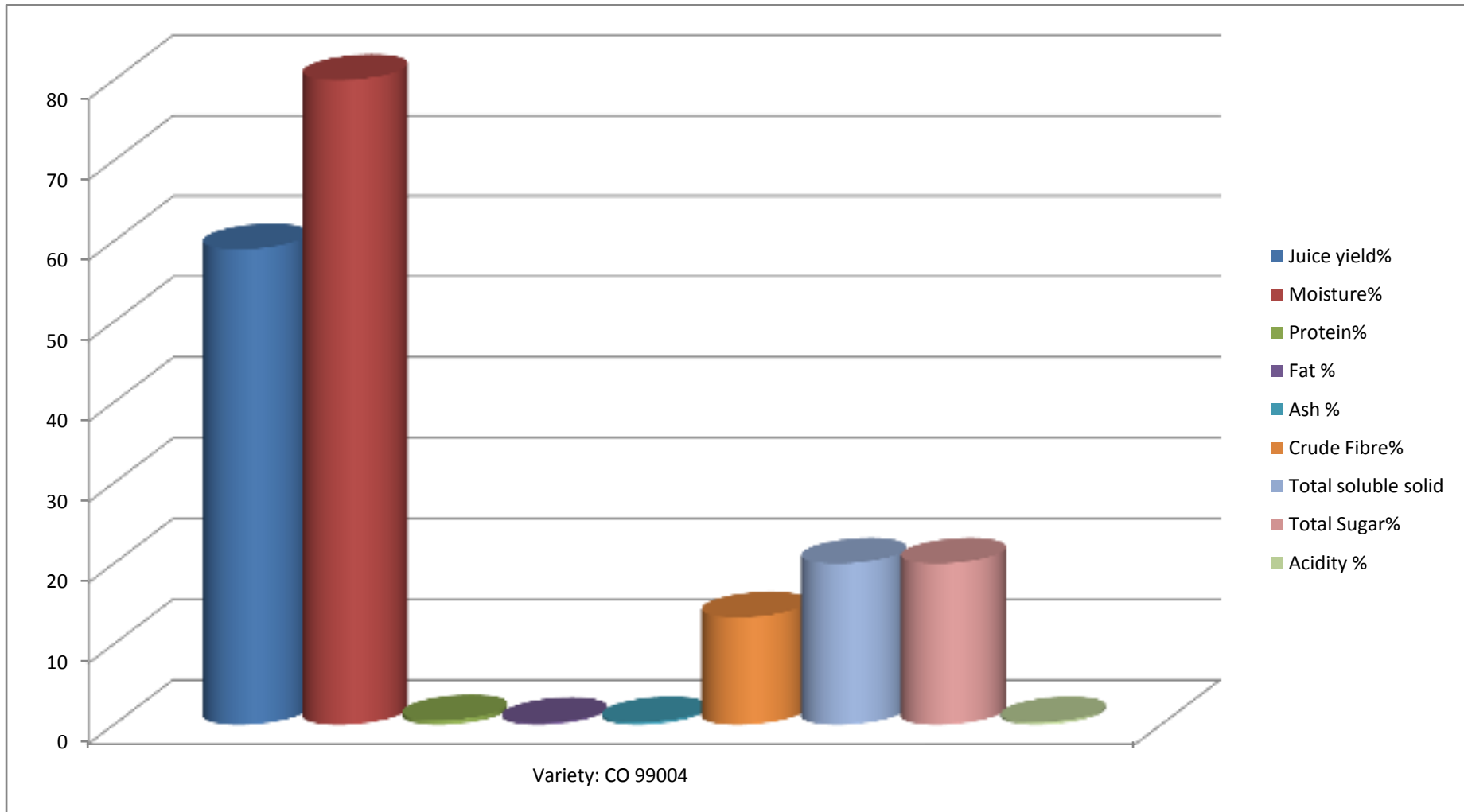


Fig. 2 Effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (5° C)

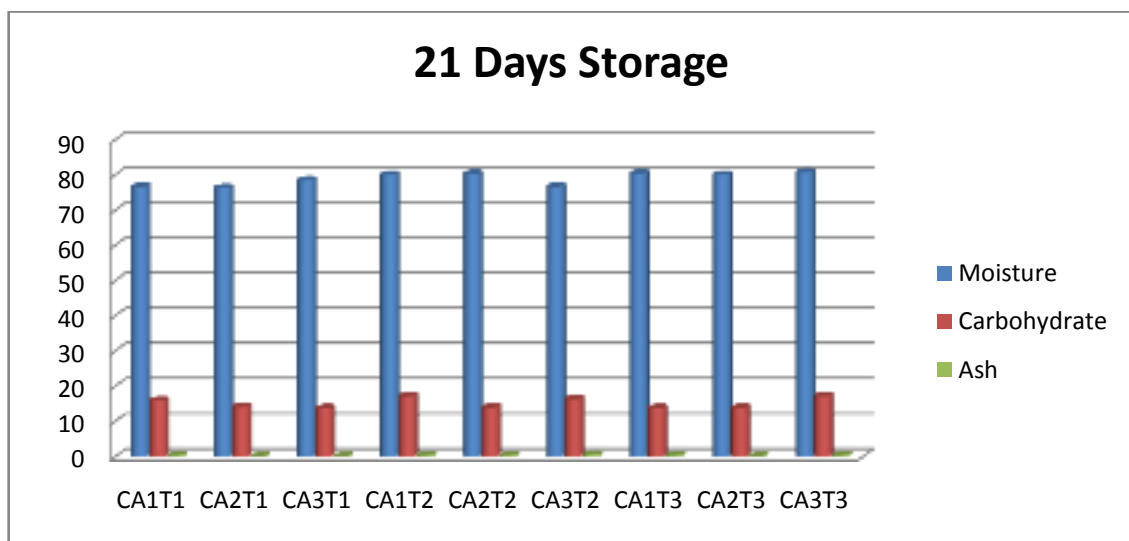
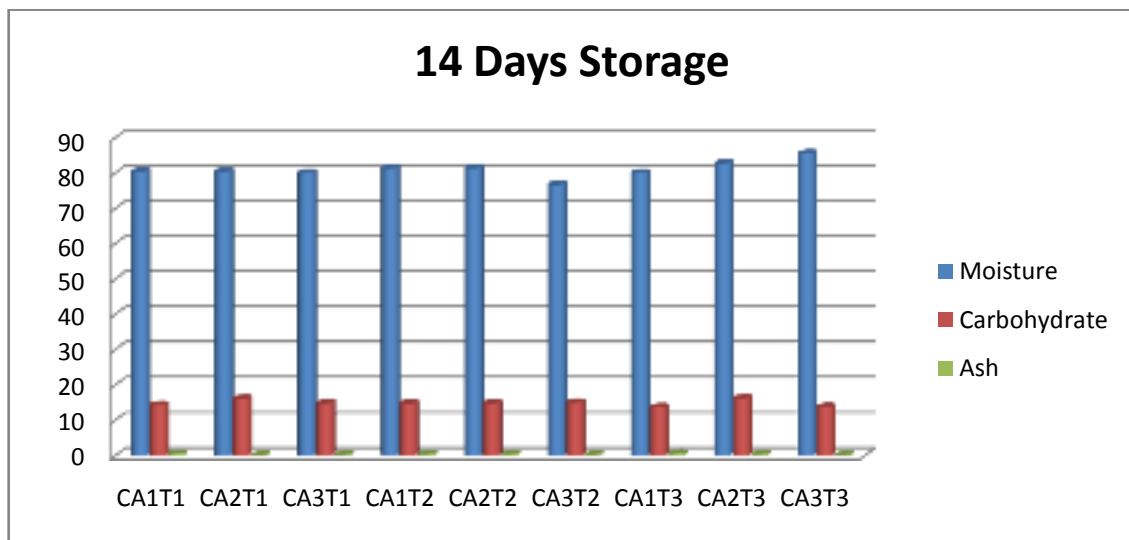
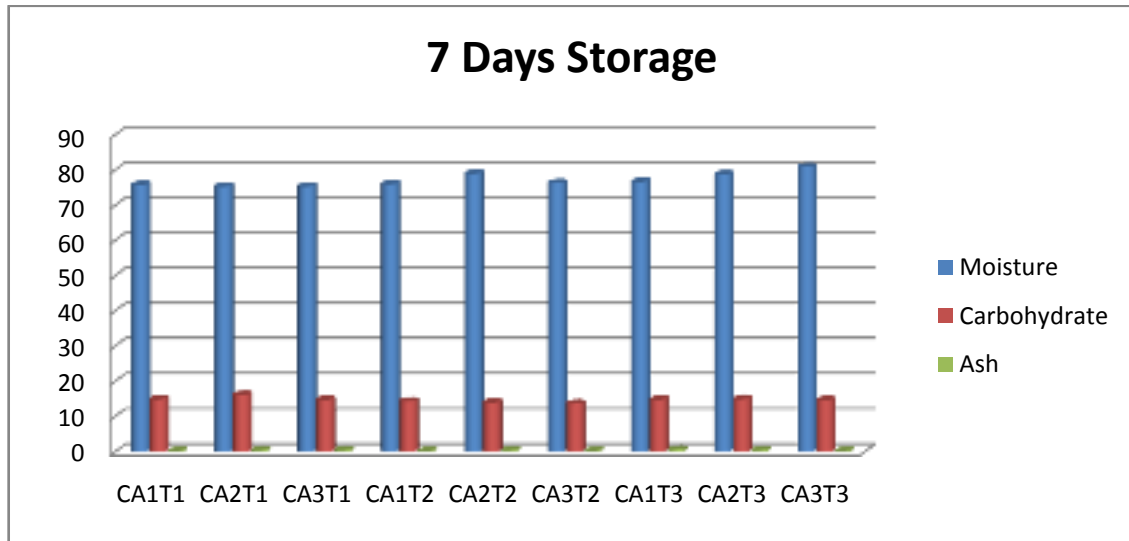


Fig. 3 Effect of citric acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at ambient condition (30° C)

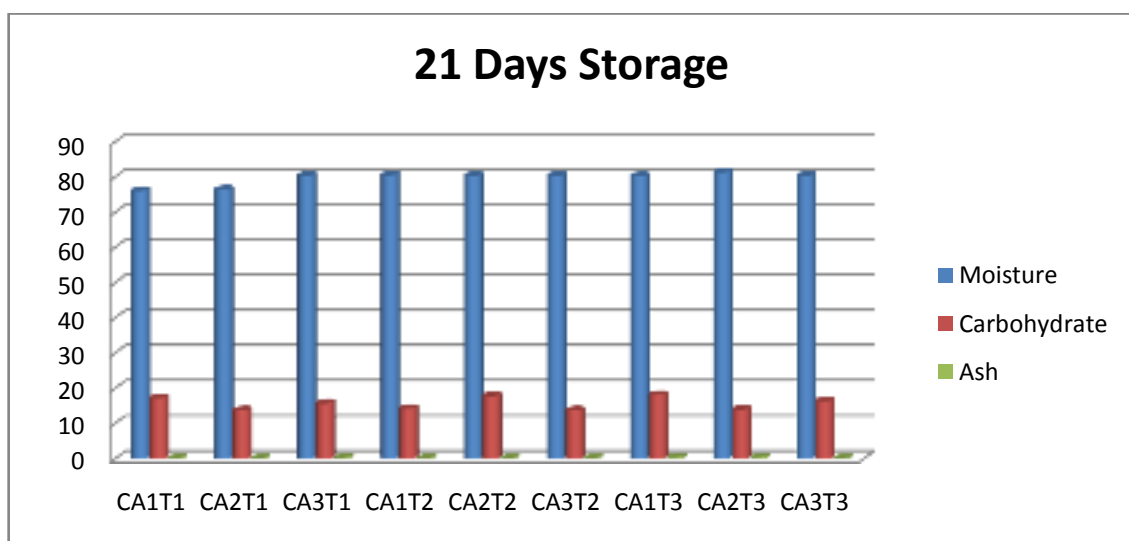
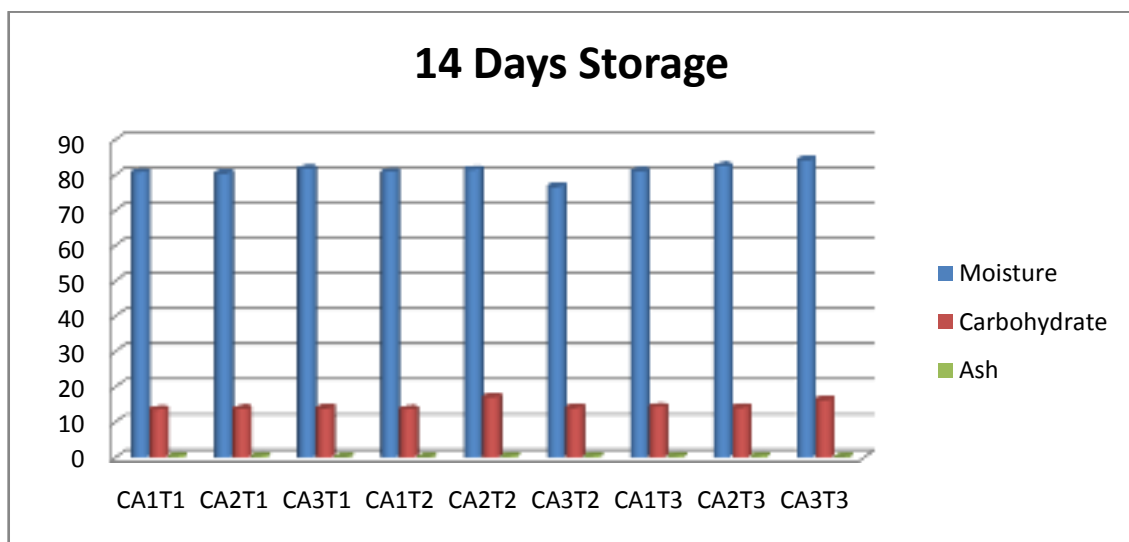
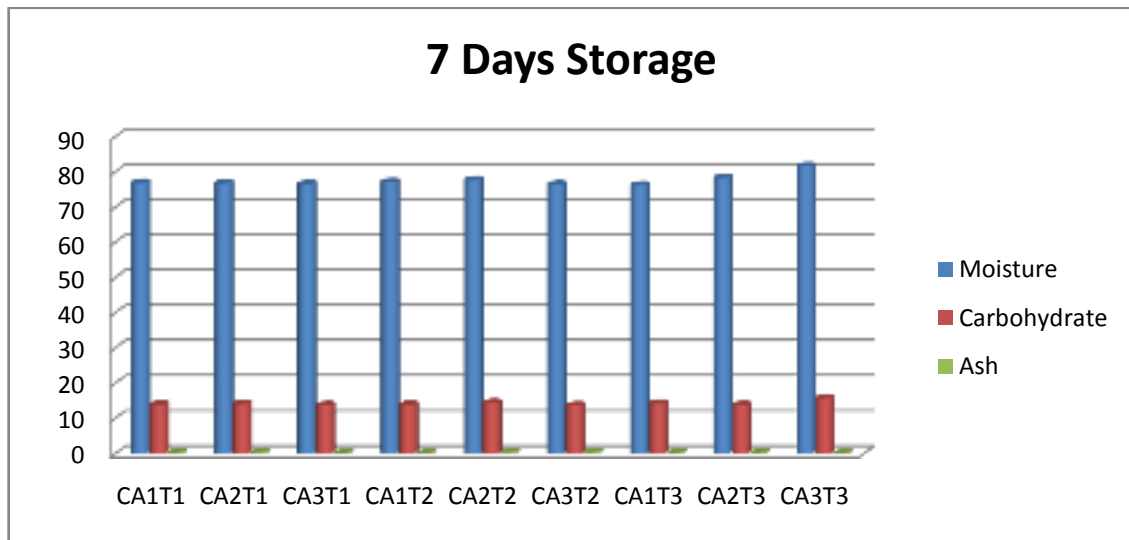


Fig. 4 Effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5° C)

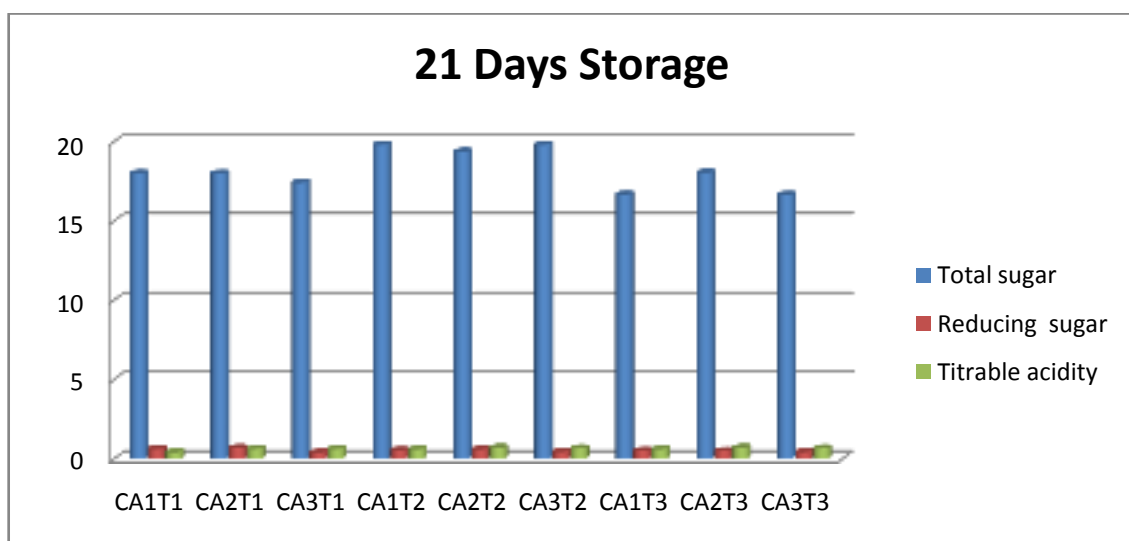
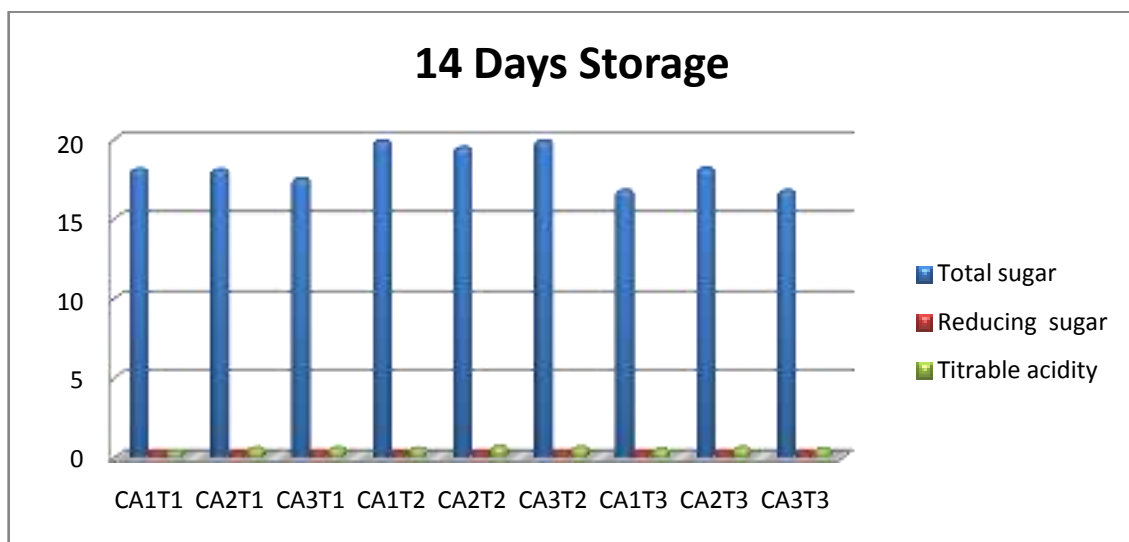
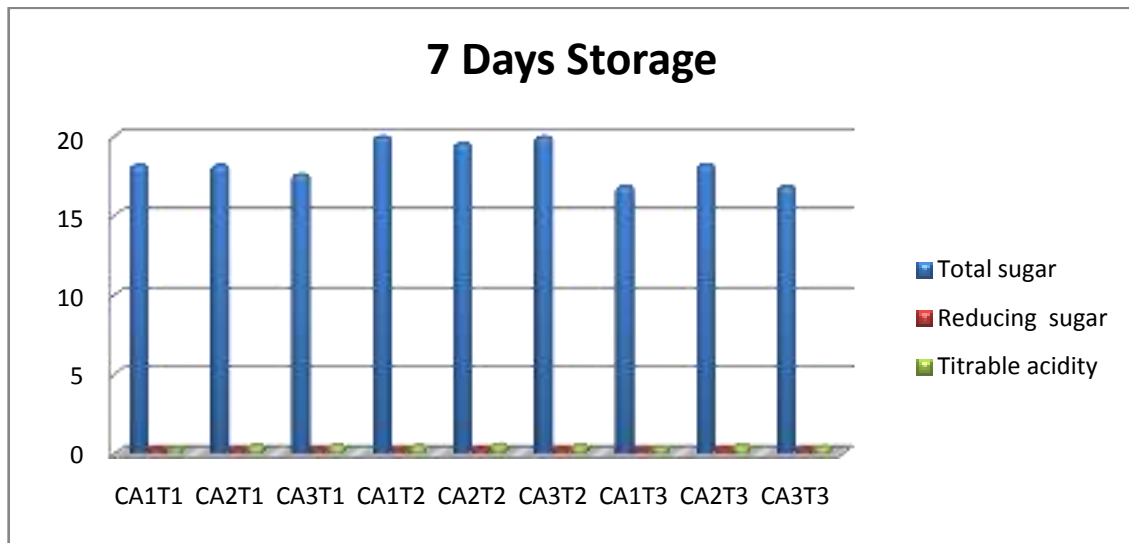


Fig. 5 Effect of citric acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30° C)

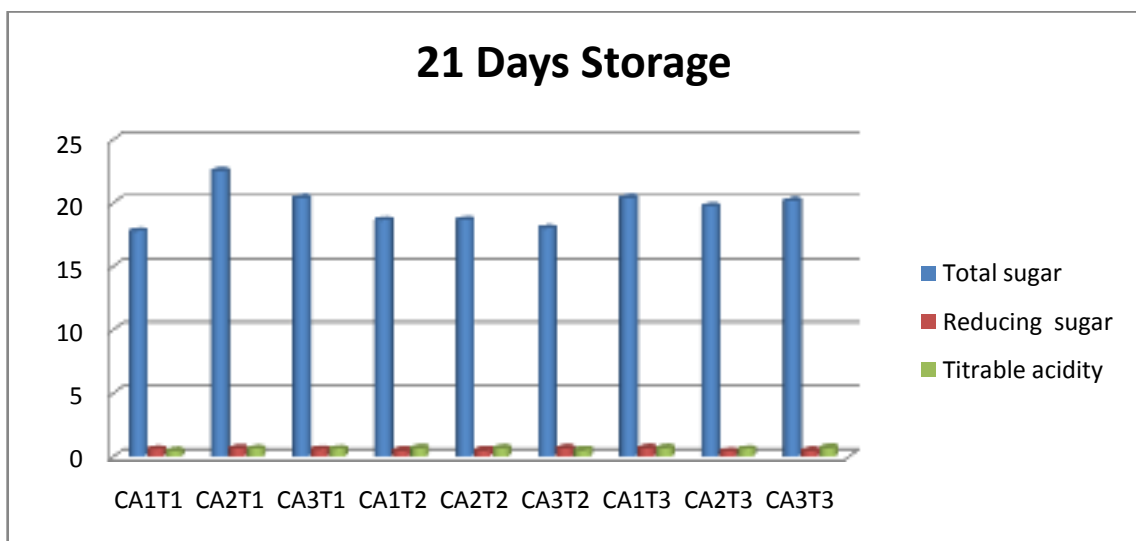
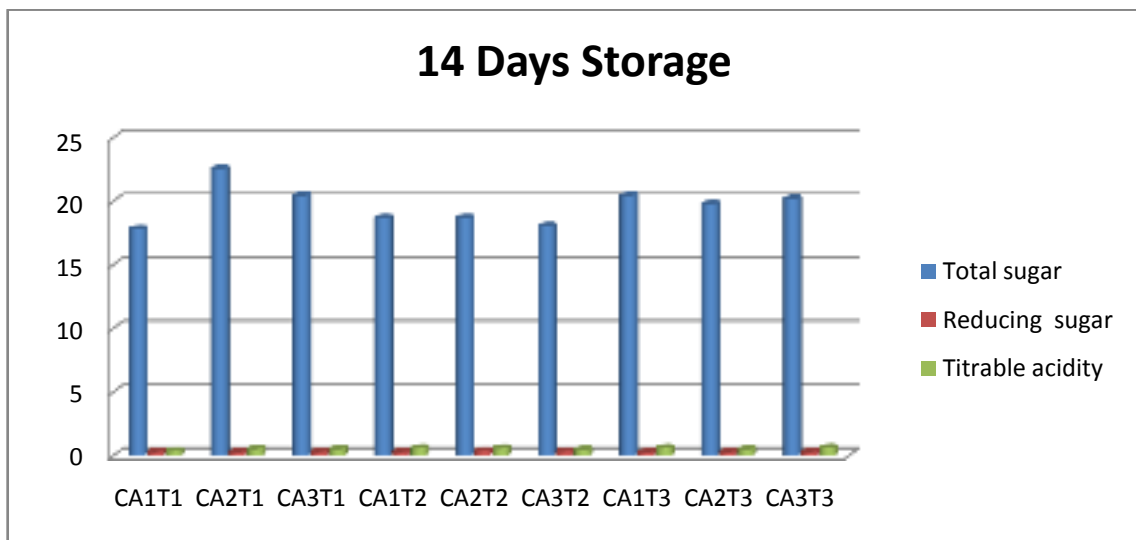
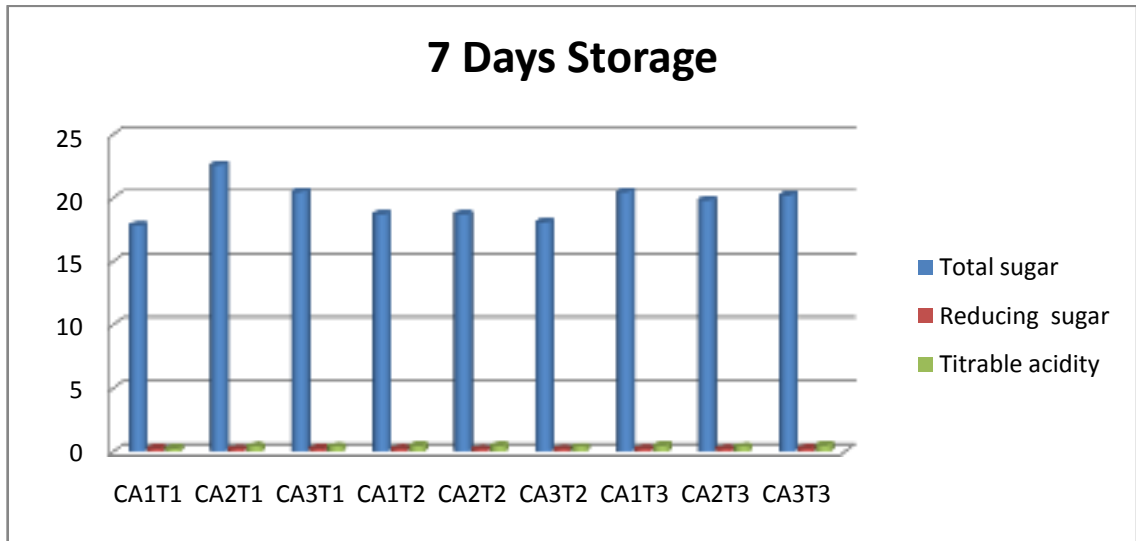


Fig. 6 Effect of citric acid and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5° C)

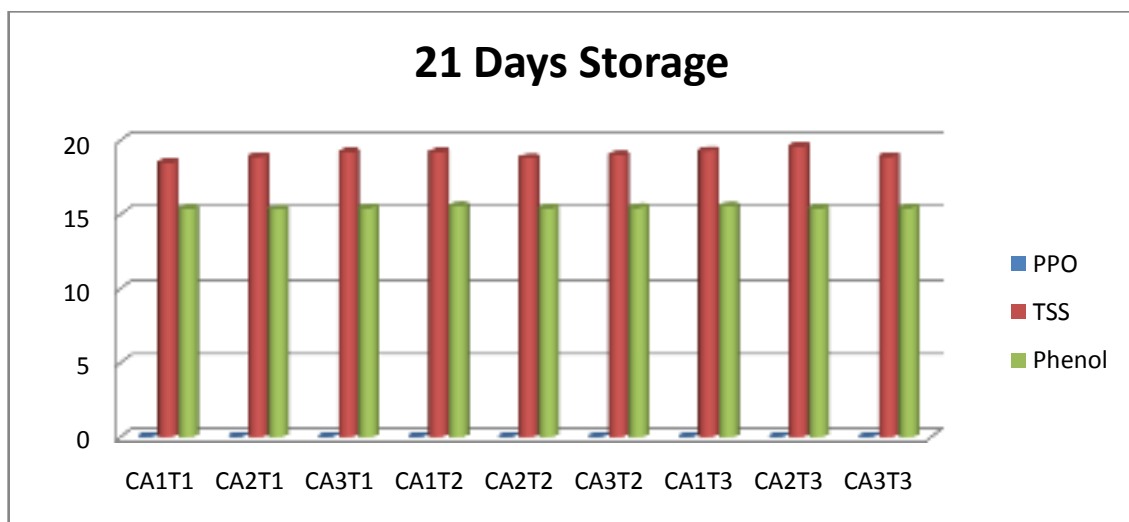
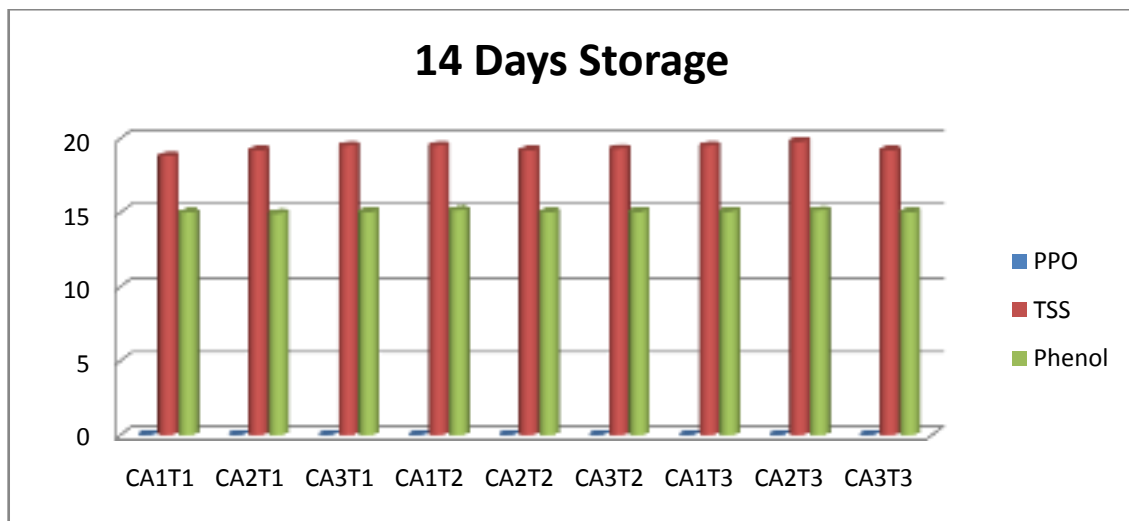
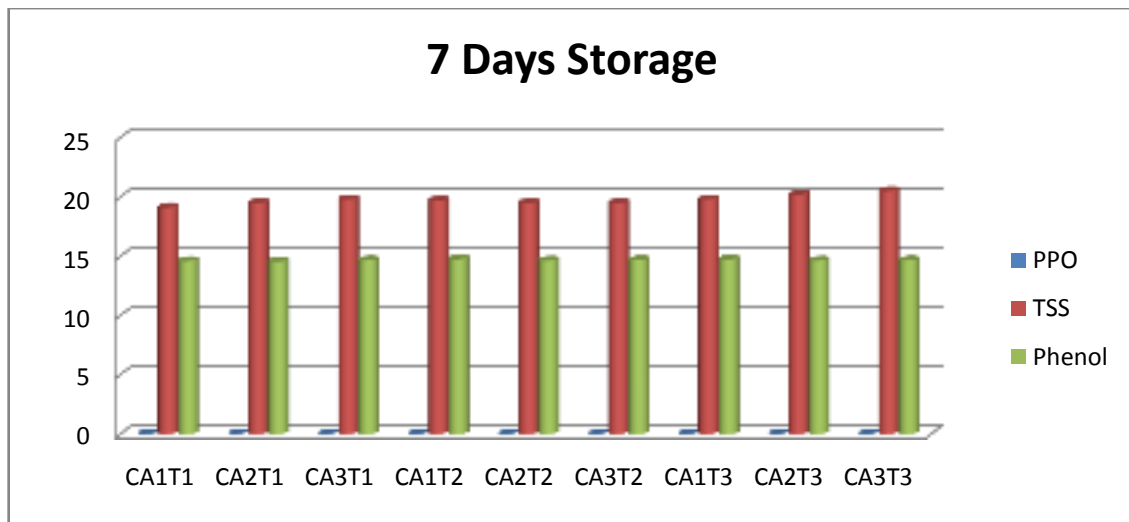


Fig. 7 Effect of citric acid and pasteurization temperature on poly phenol oxidase (PPO) total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30°C)

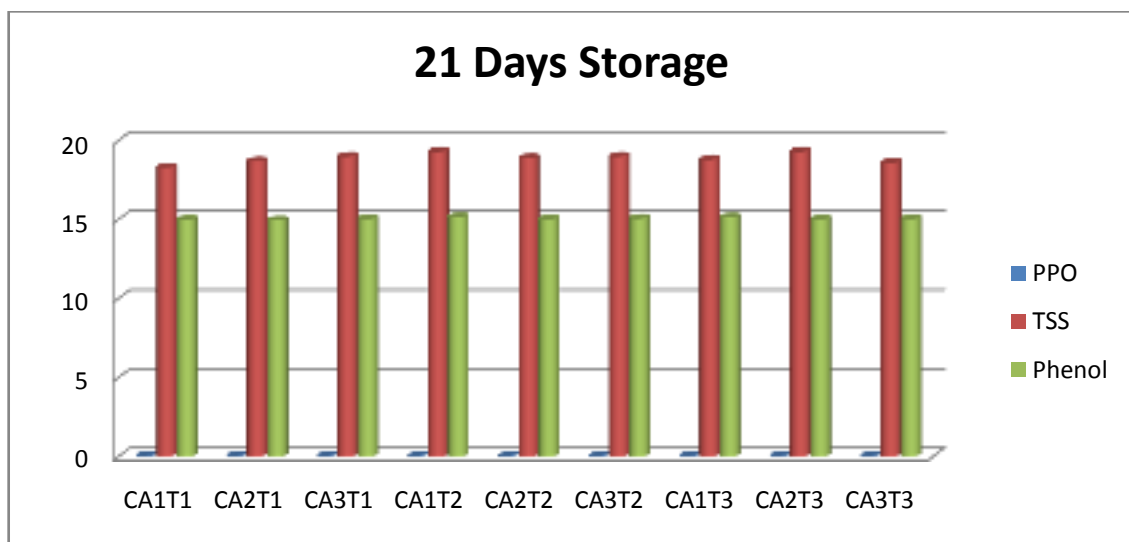
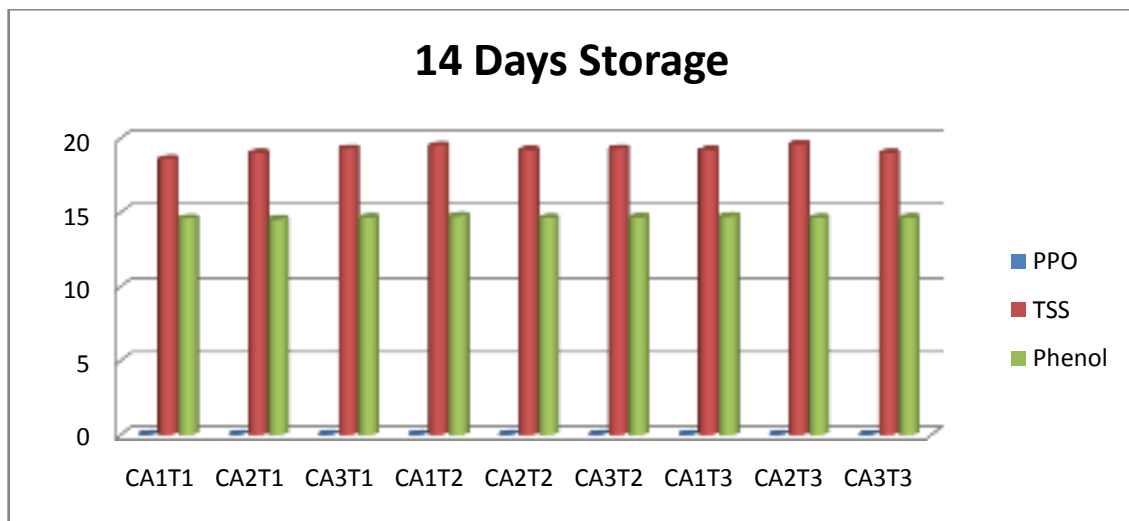
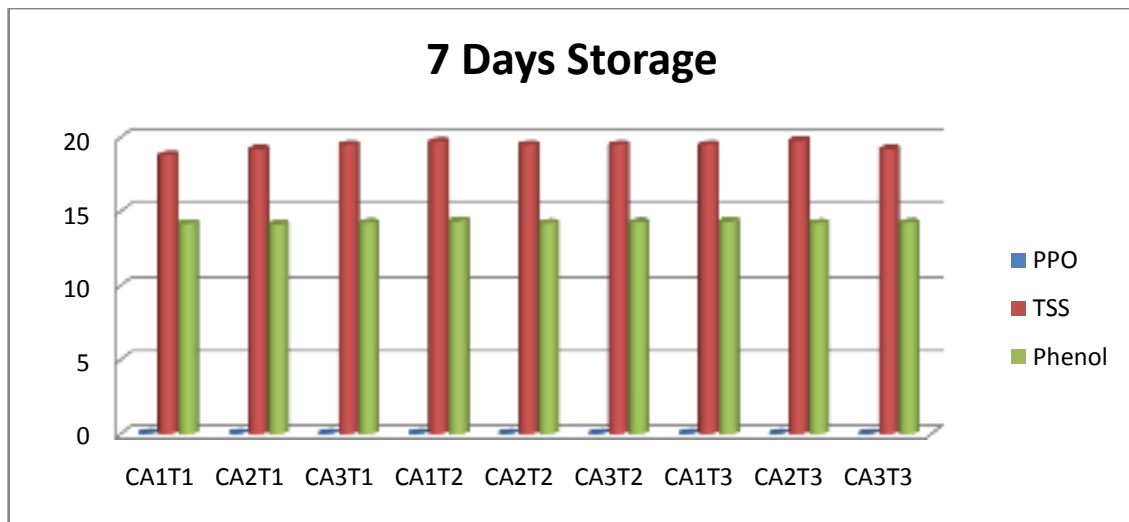


Fig. 8 Effect of potassium metabisulfite and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (5°C)

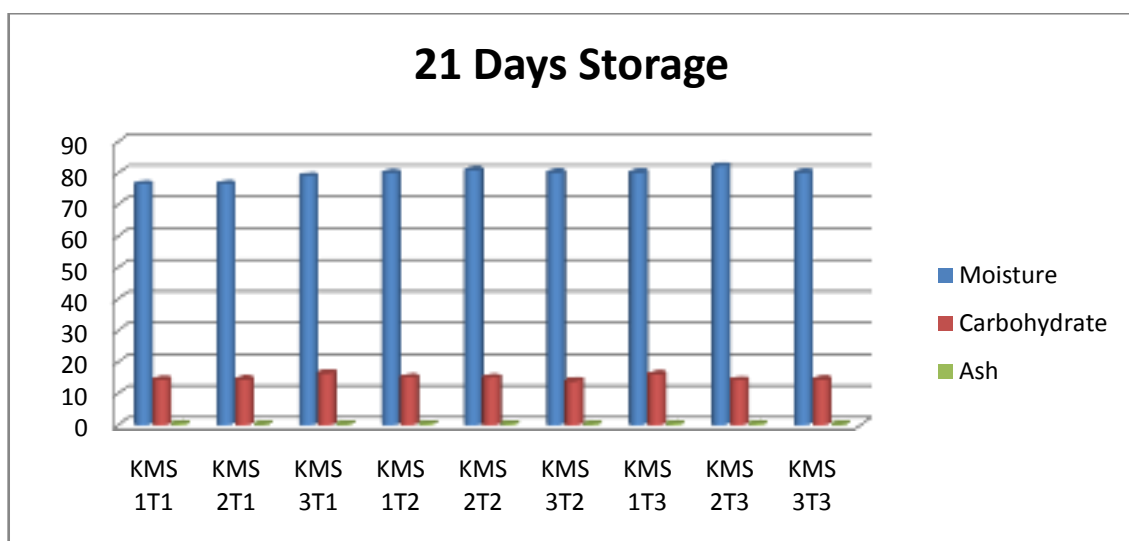
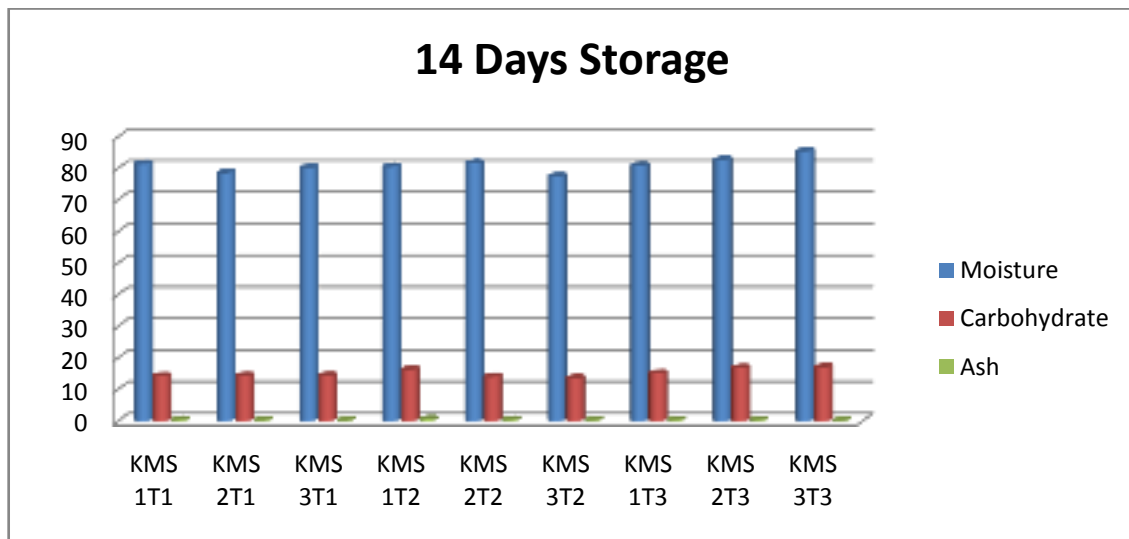
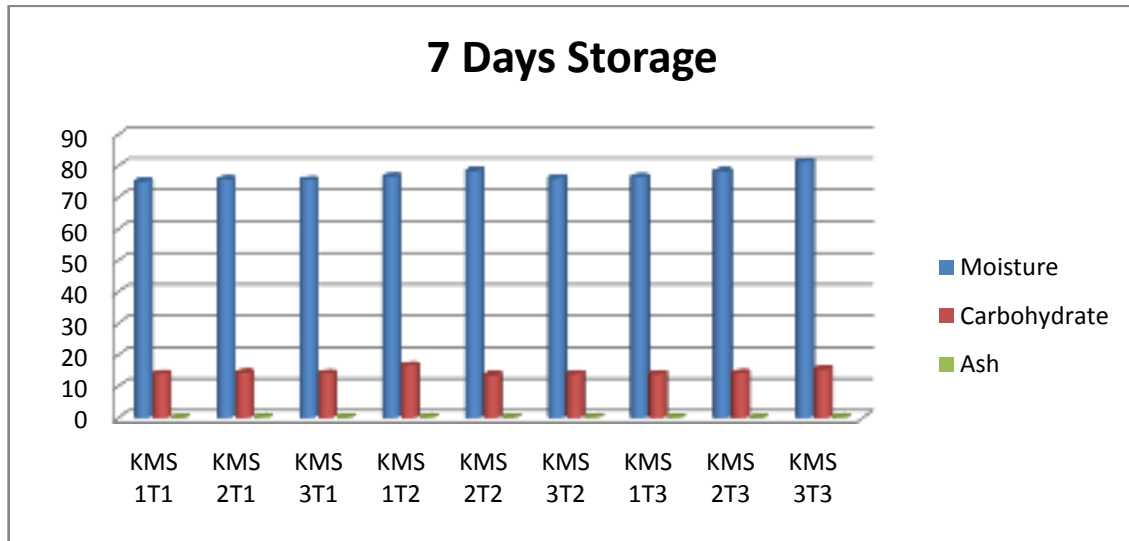


Fig. 9 Effect of potassium metabisulfite and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (30 °C)

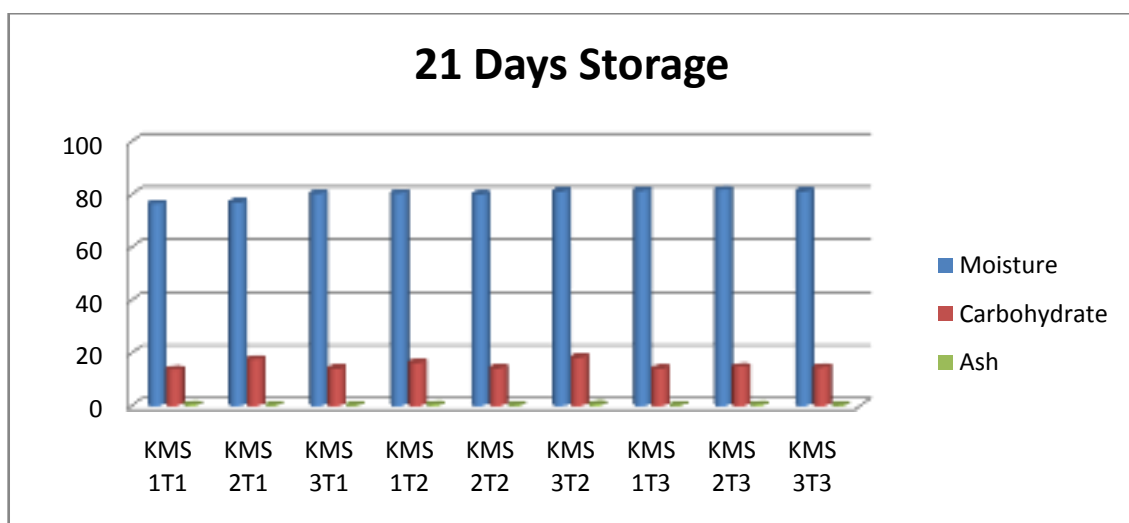
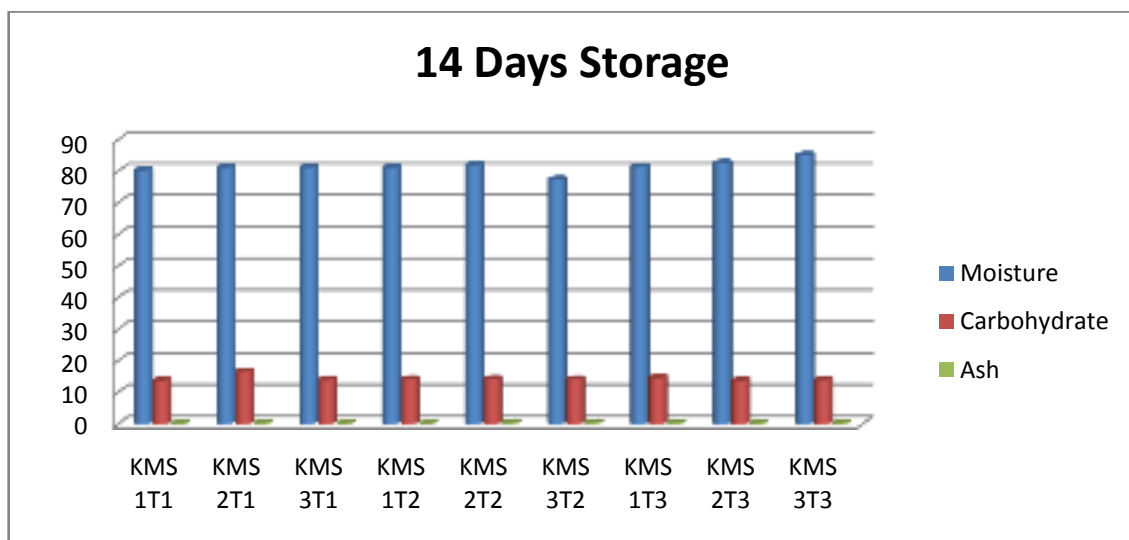
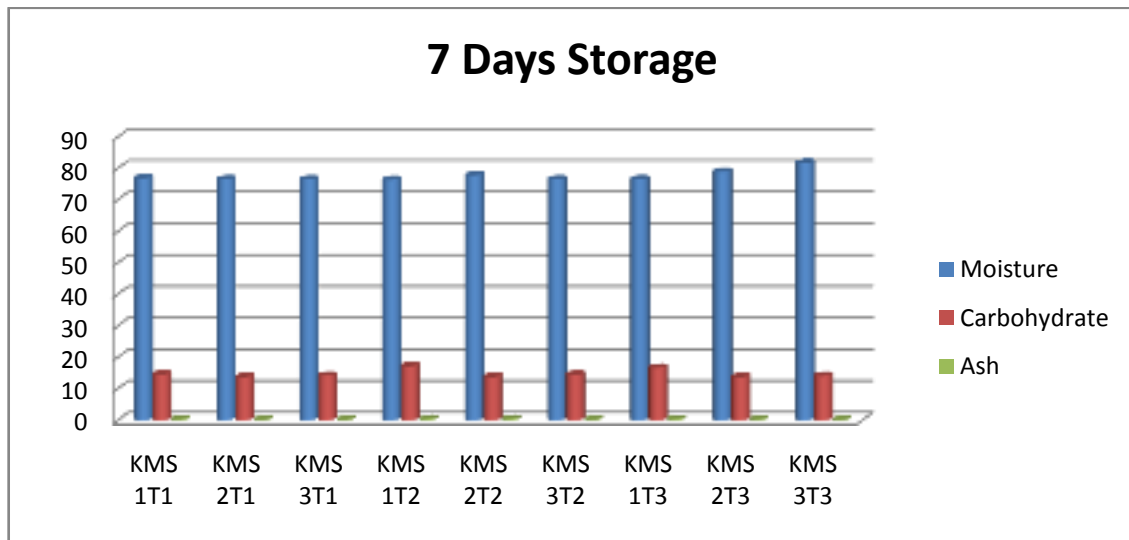


Fig. 10 Effect of potassium metabisulfite and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5° C)

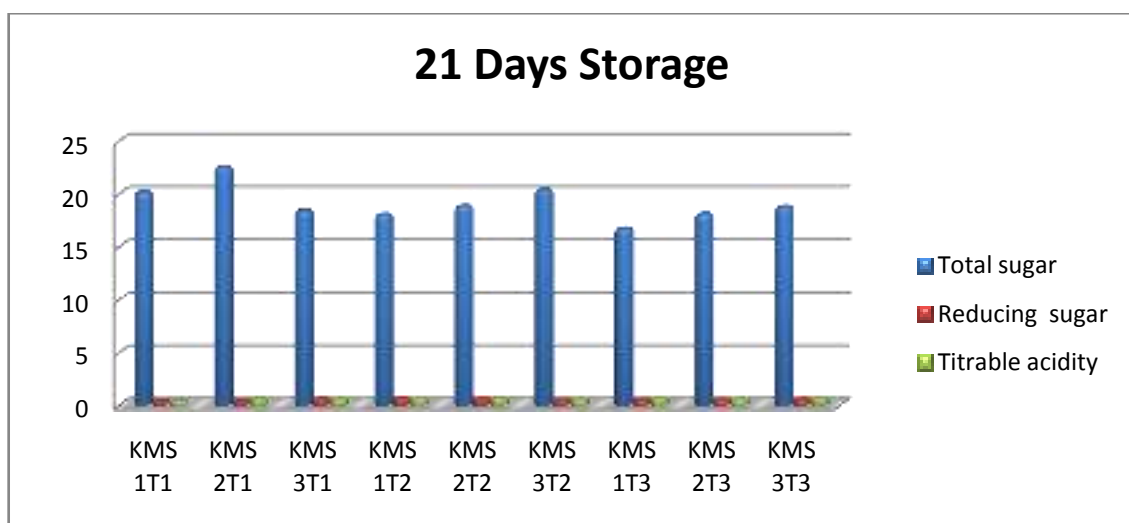
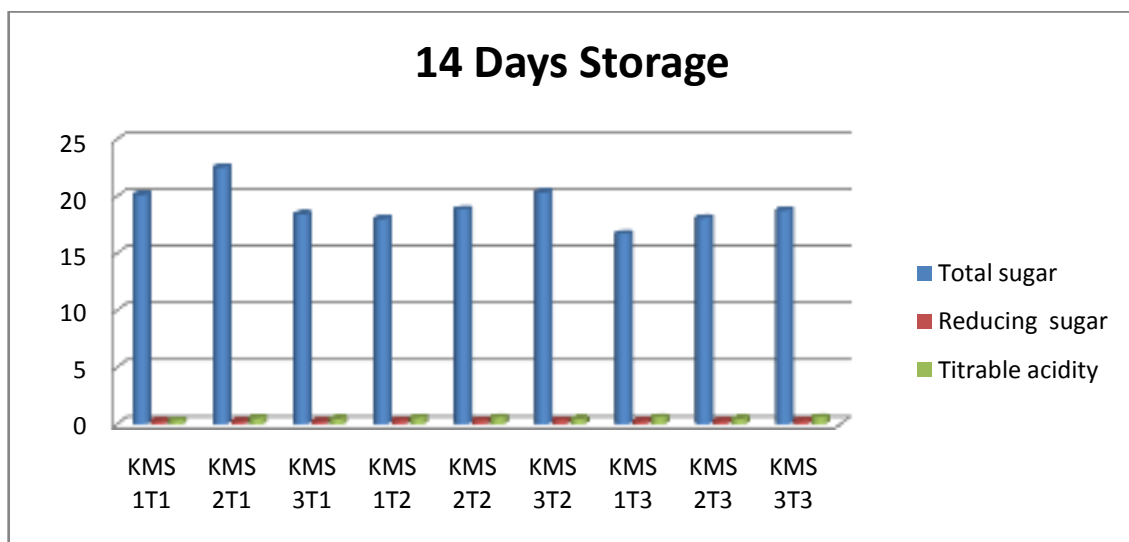
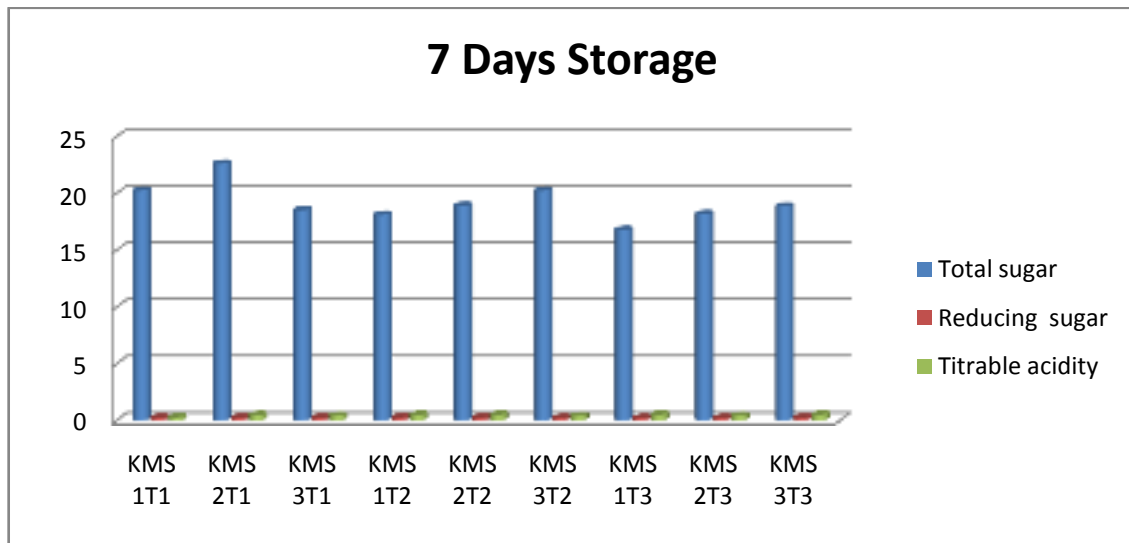


Fig. 11 Effect of potassium metabisulfate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30° C)

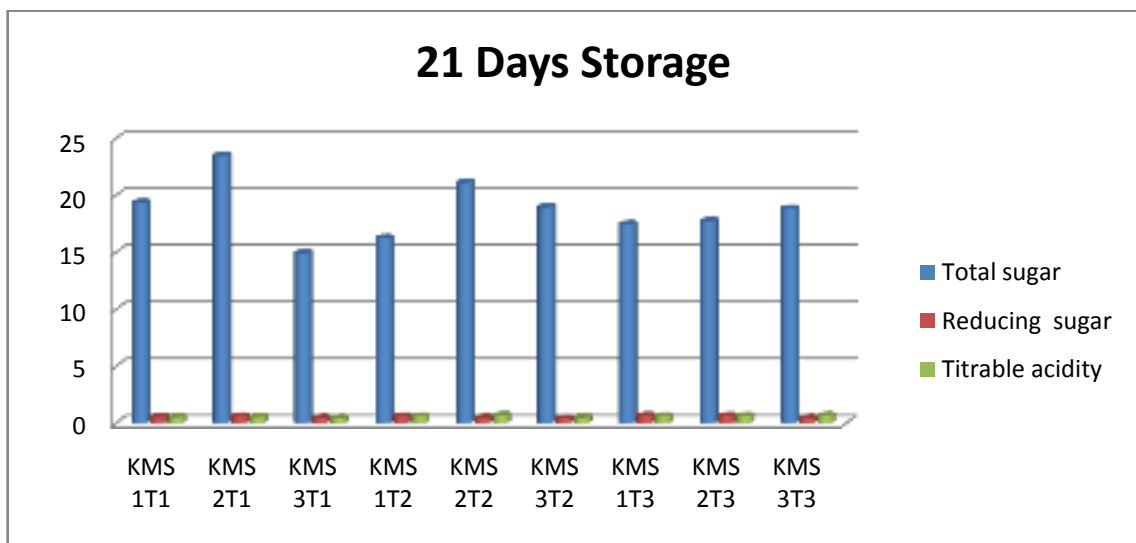
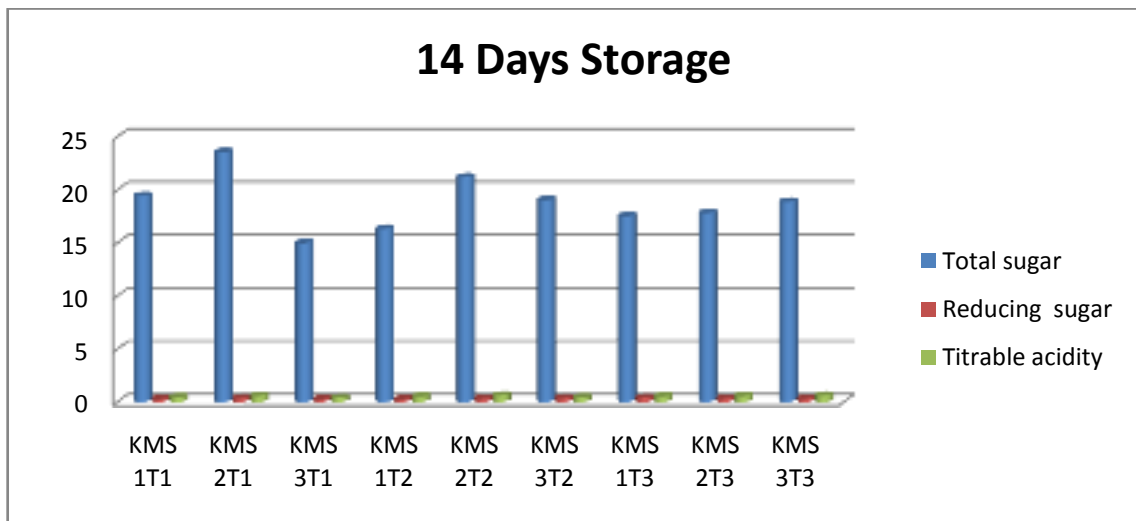
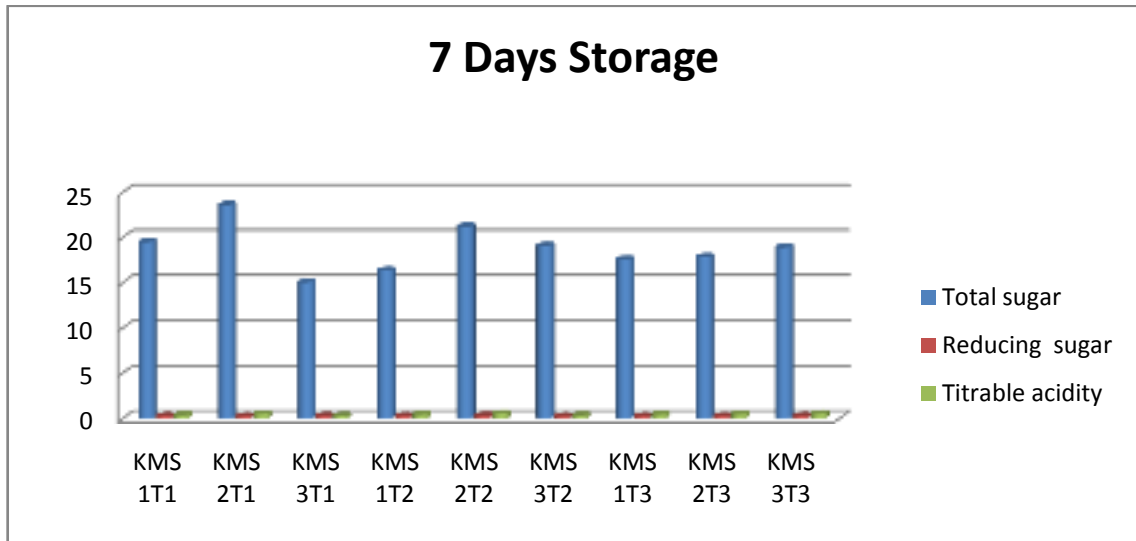


Fig. 12 Effect of potassium metabisulfite and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5⁰C)

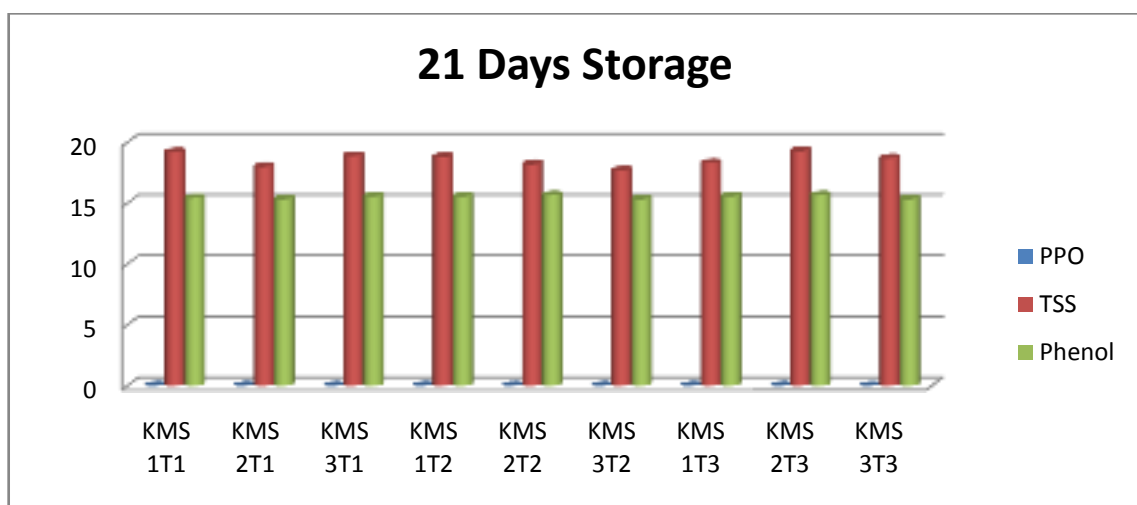
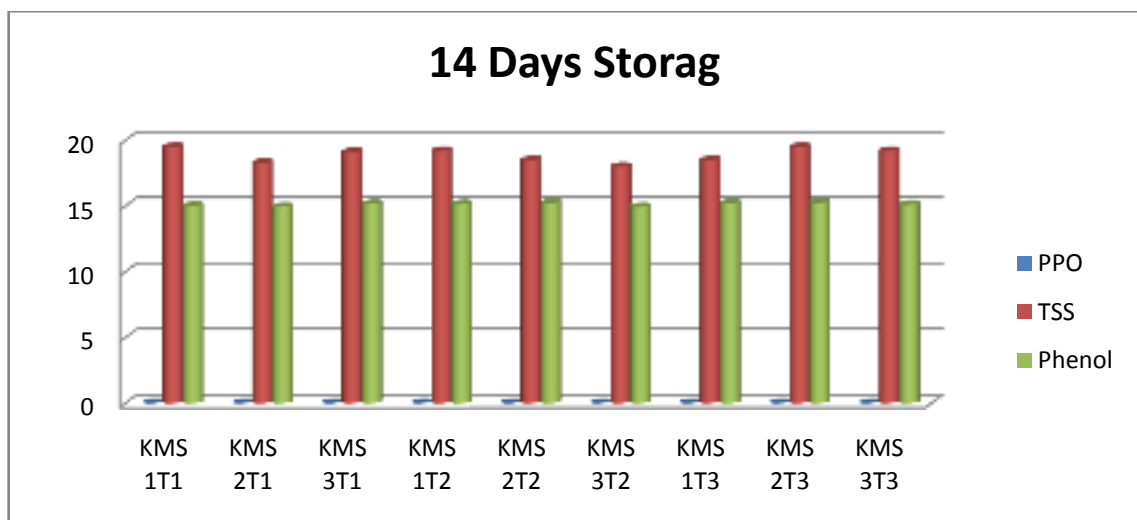
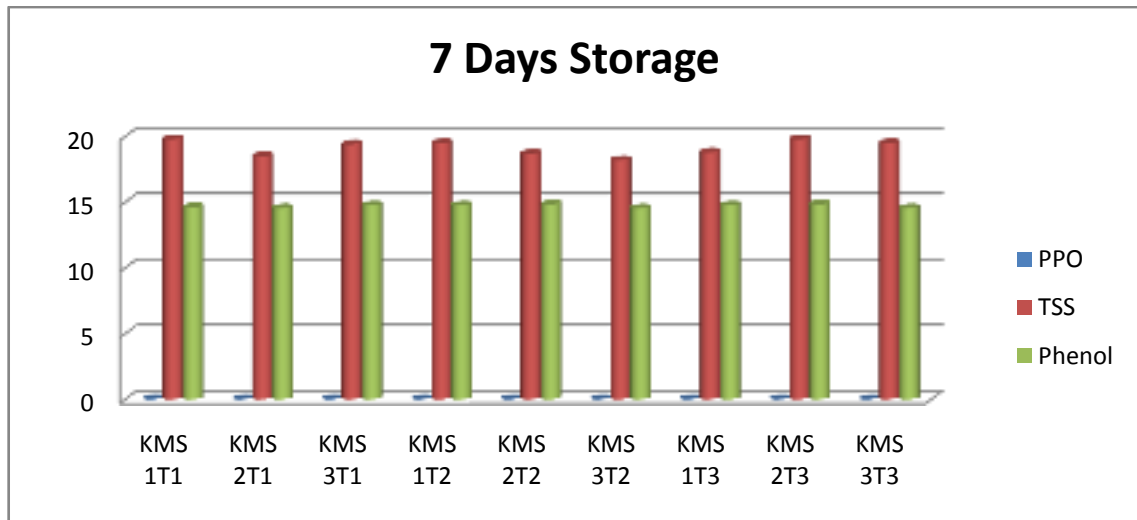


Fig. 13 Effect of potassium metabisulfite and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30° C)

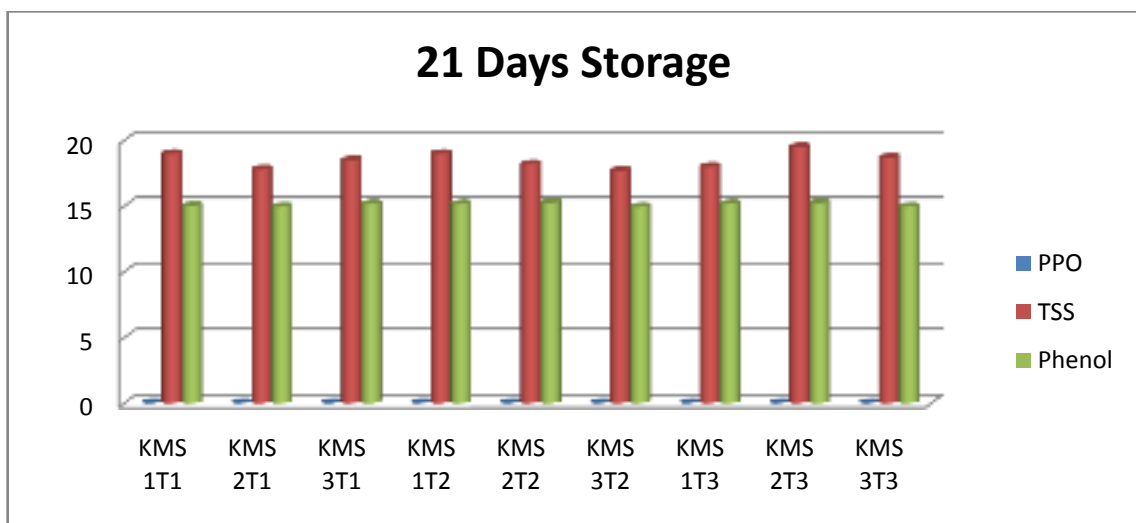
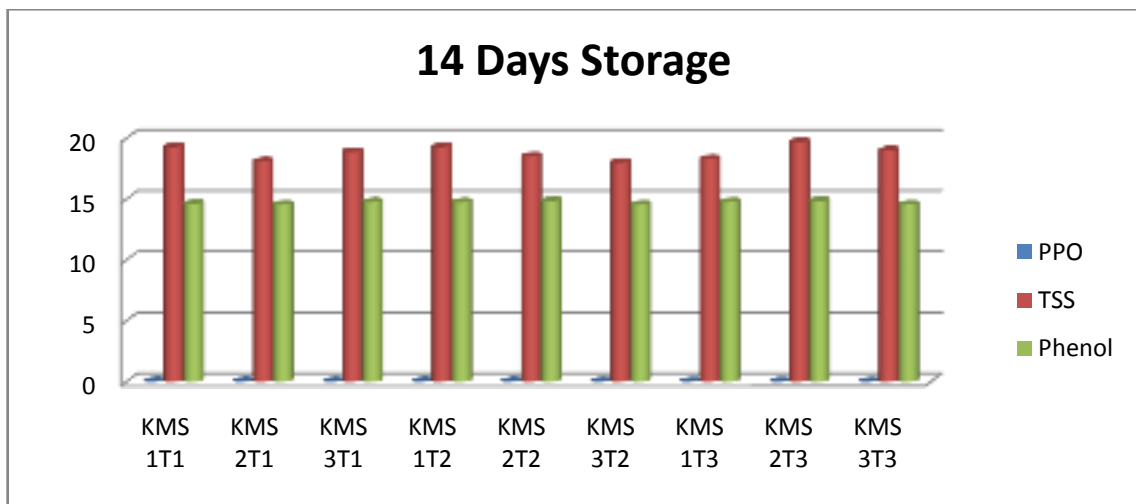
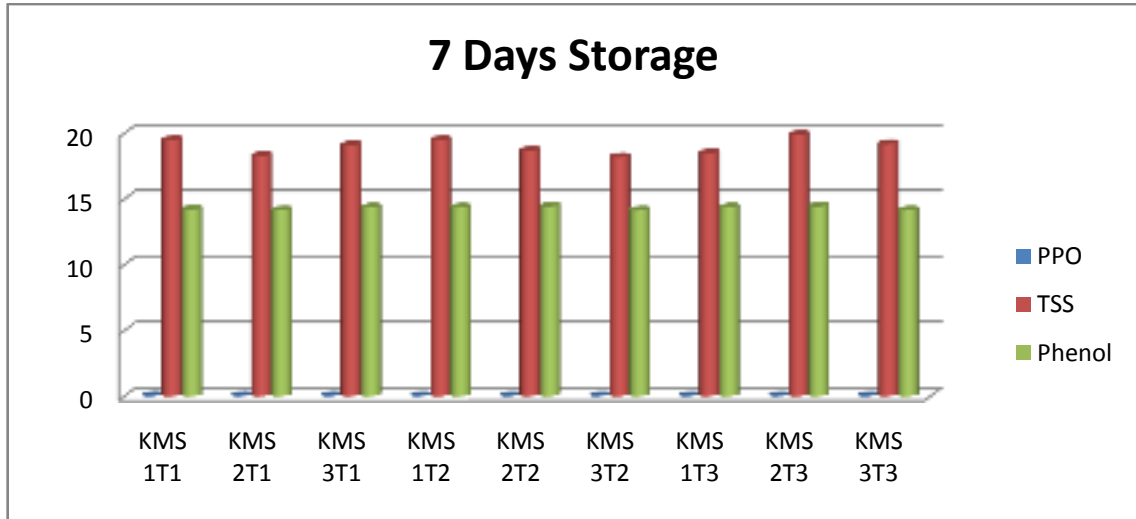


Fig. 14 Effect of potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash of stored sugarcane juice at refrigerated condition (5°C)

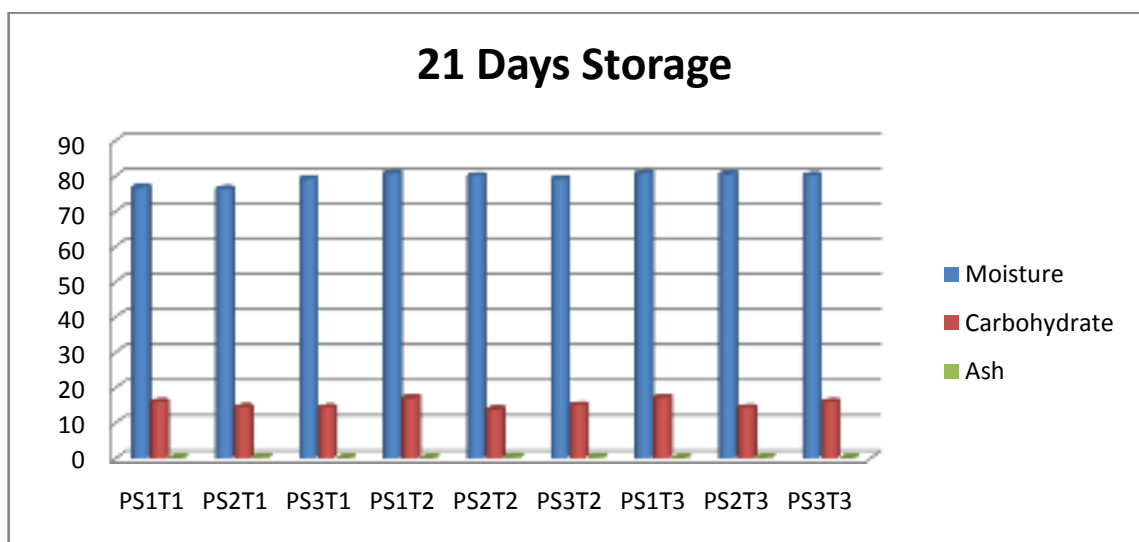
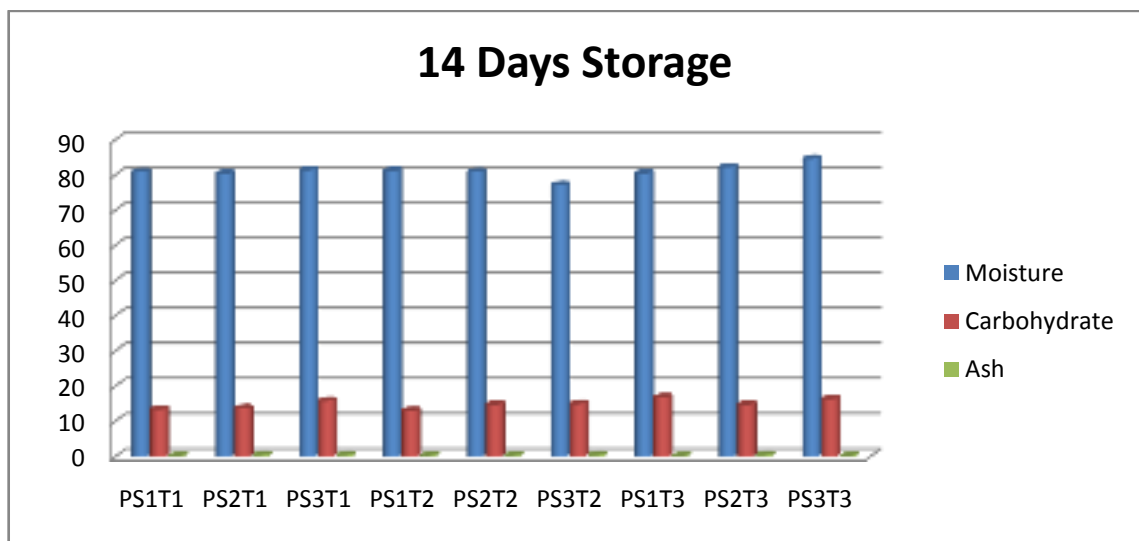
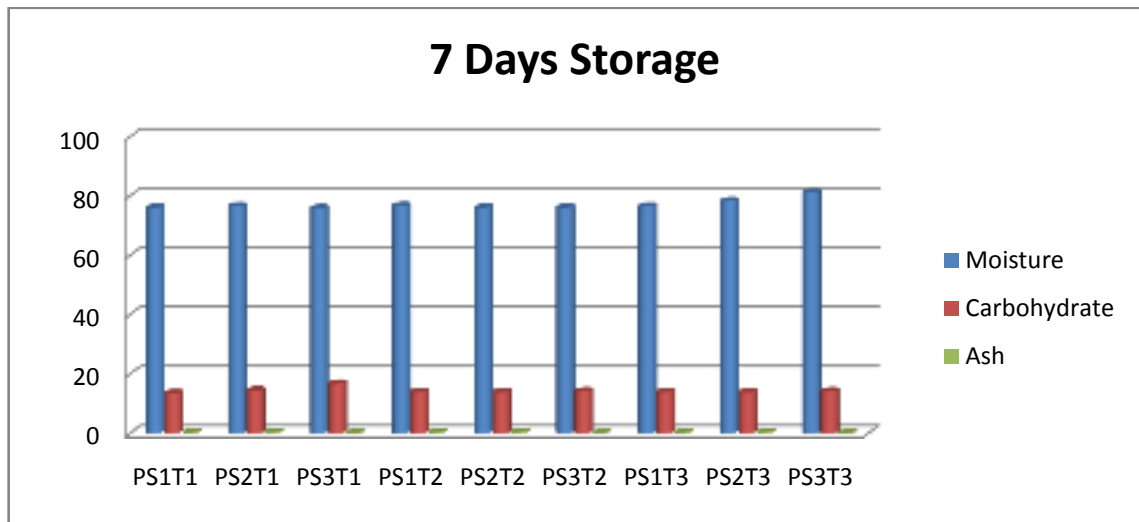


Fig. 15 Effect of potassium sorbate and pasteurization temperature on moisture, carbohydrate and ash of stored sugarcane juice at ambient condition (30°C)

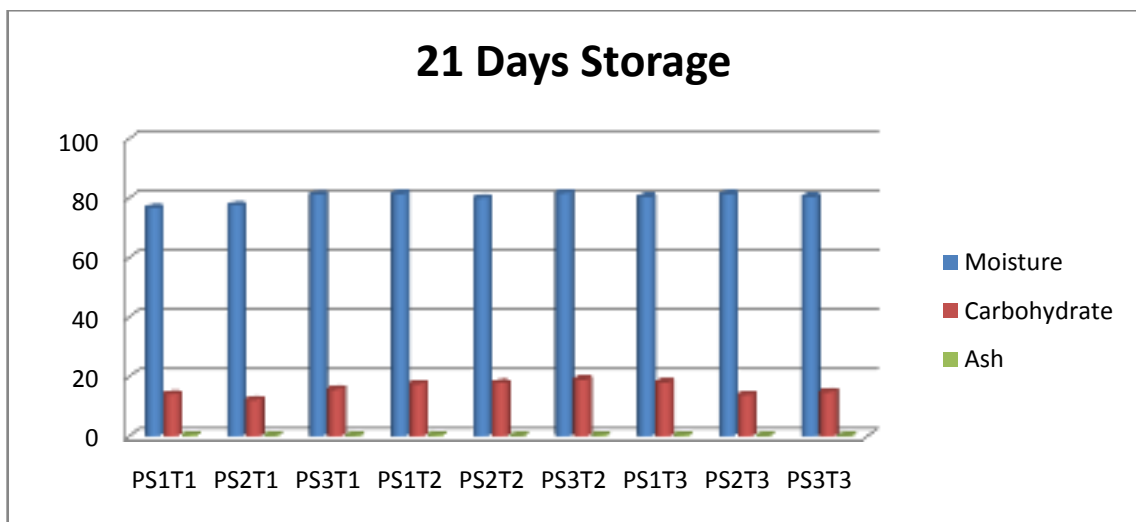
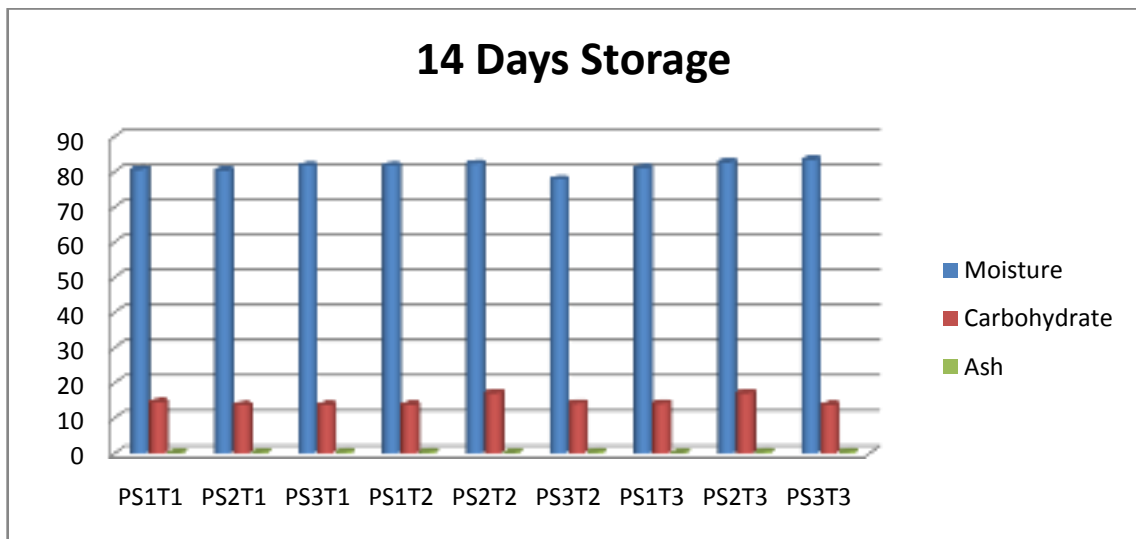
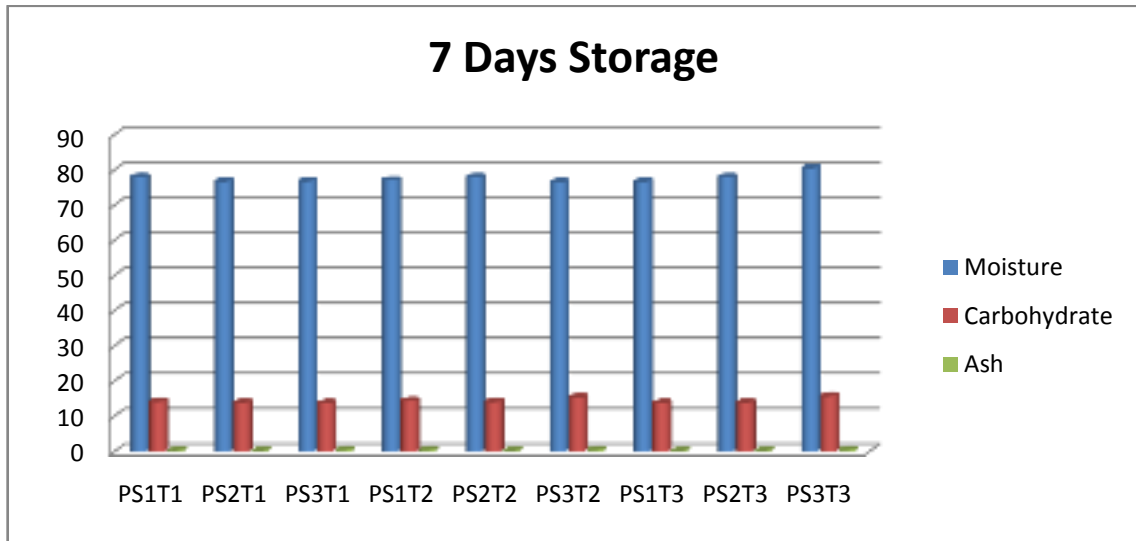


Fig. 16 Effect of potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5°C)

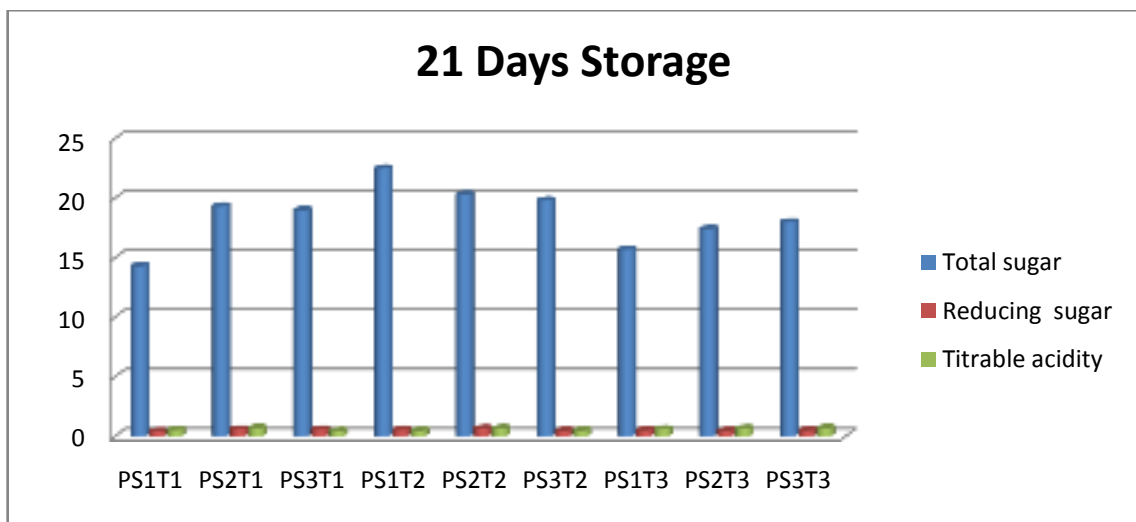
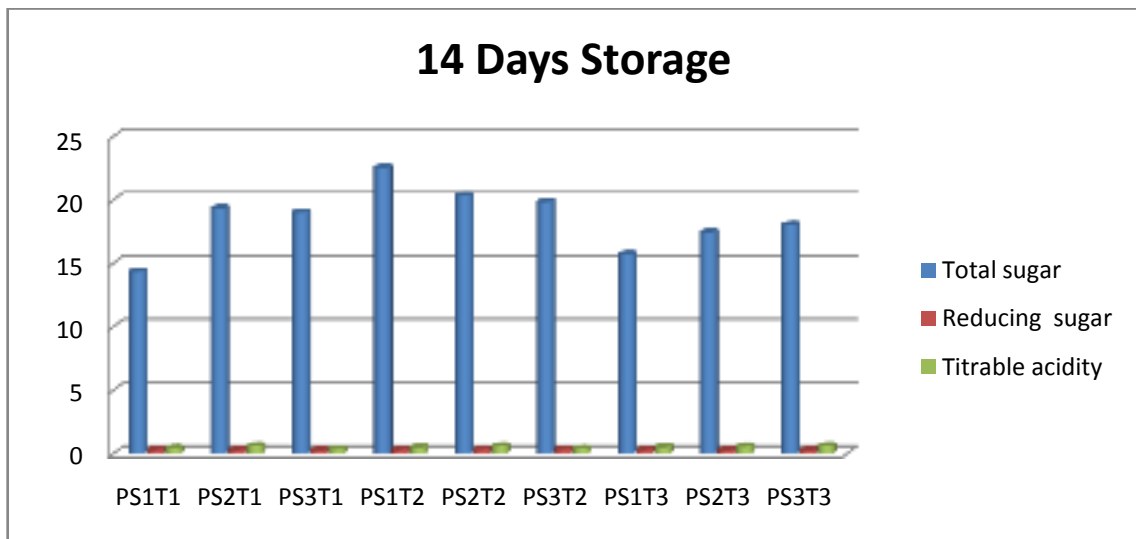
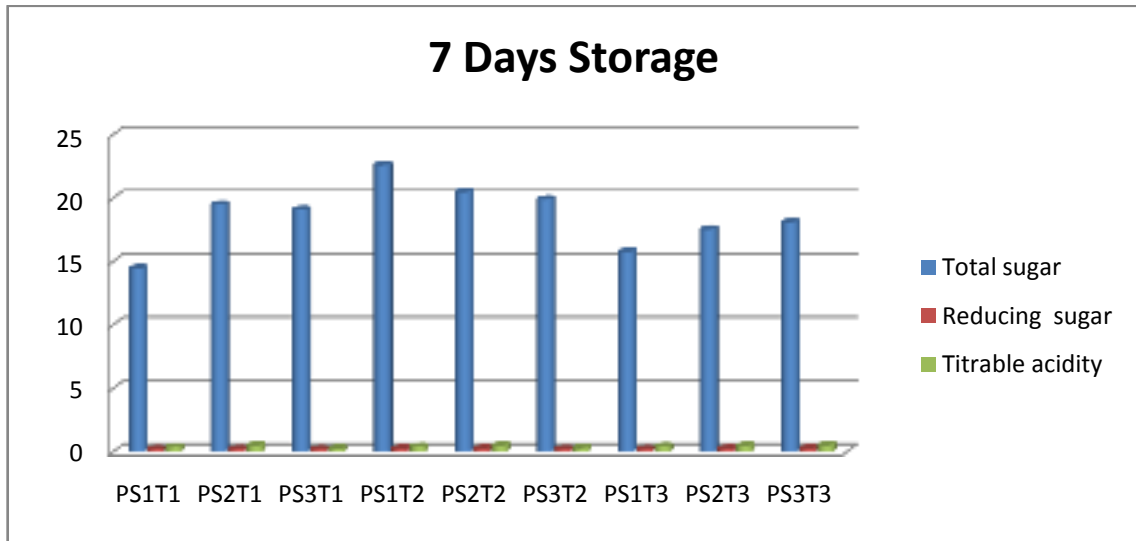


Fig. 17 Effect of potassium sorbate and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30°C)

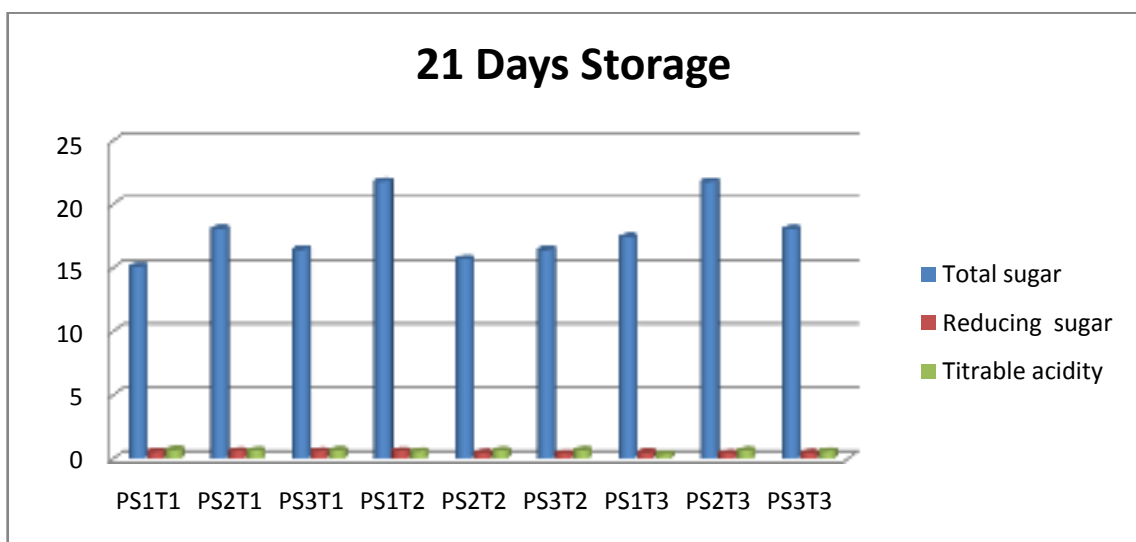
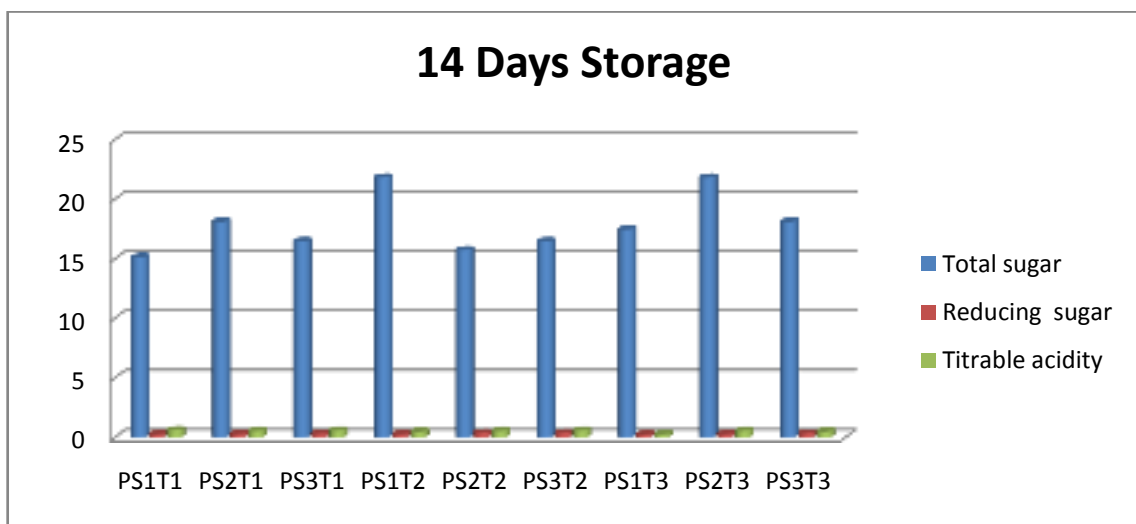
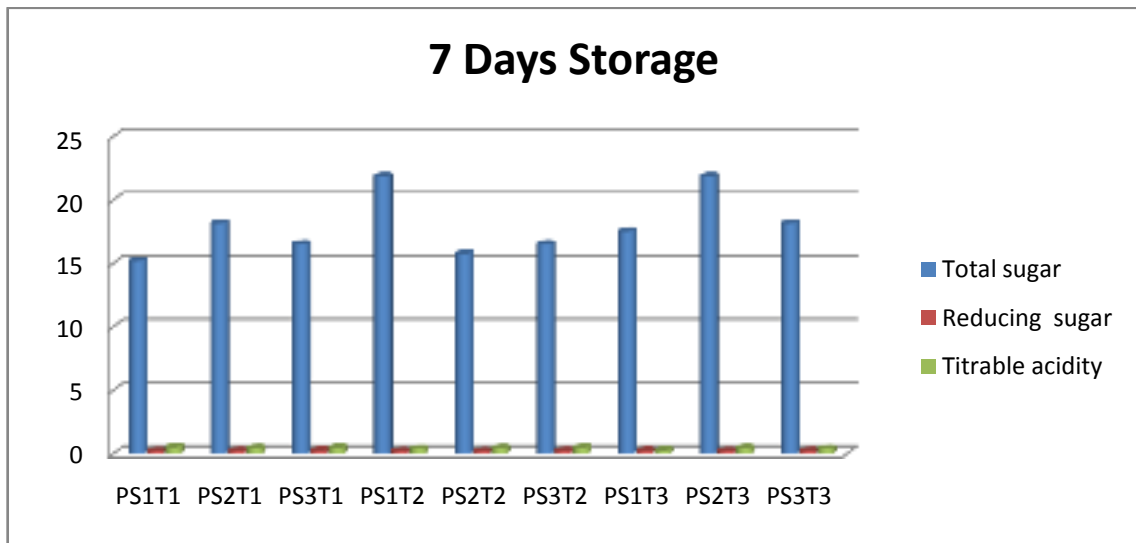


Fig. 18 Effect of potassium sorbate and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5°C)

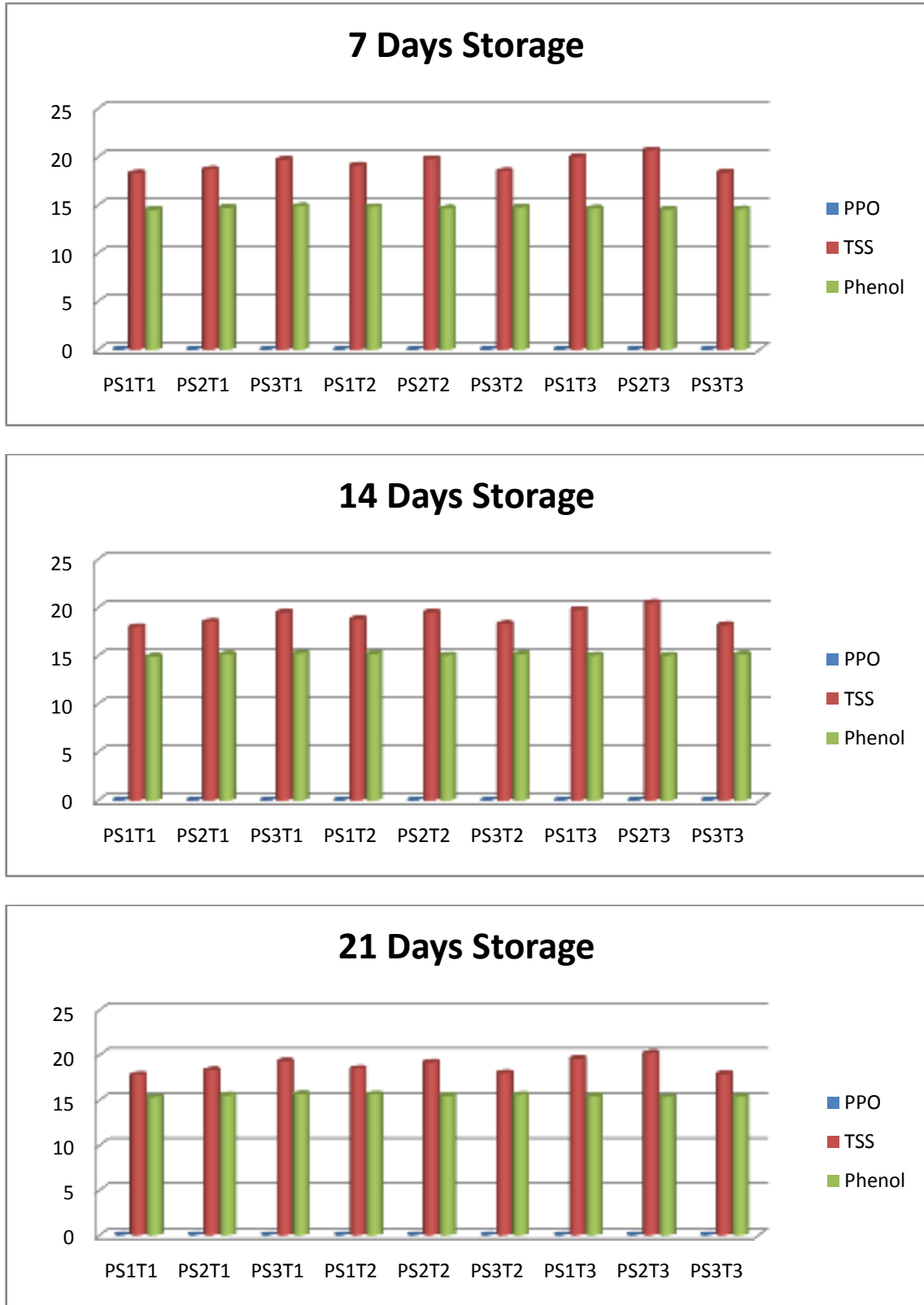


Fig. 19 Effect of potassium sorbate and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30° C)

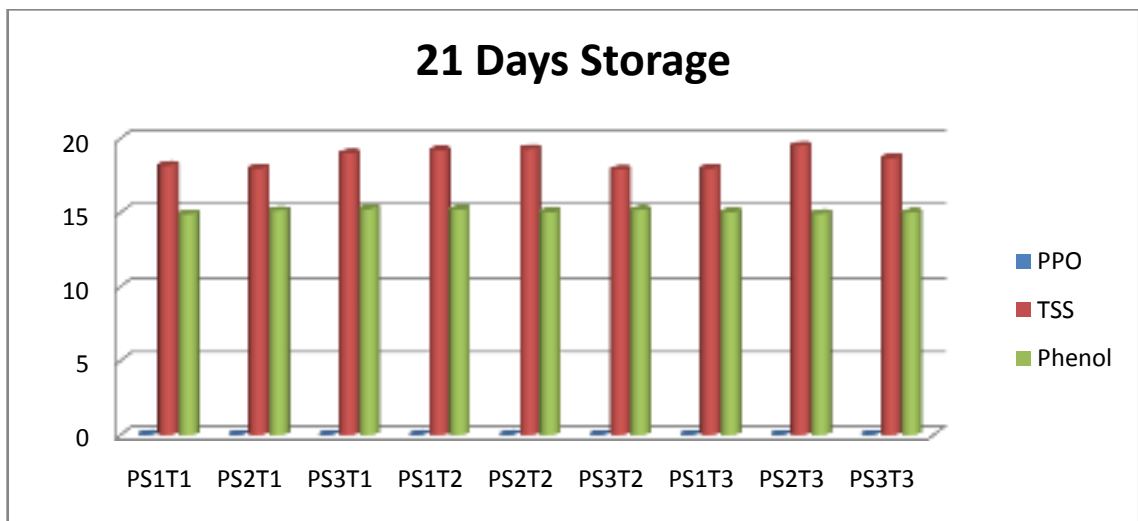
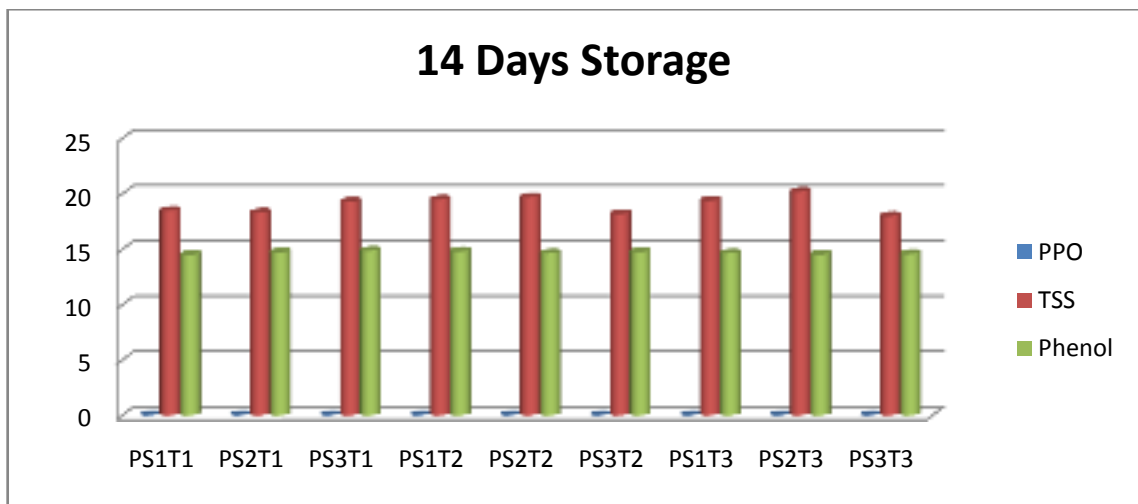
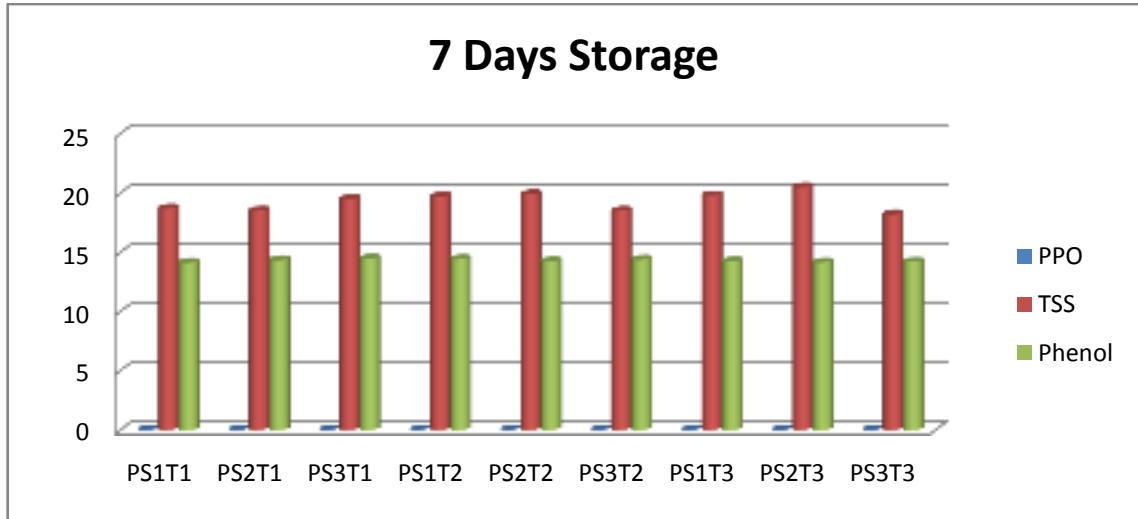


Fig. 20 Effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at refrigerated condition (5°C)

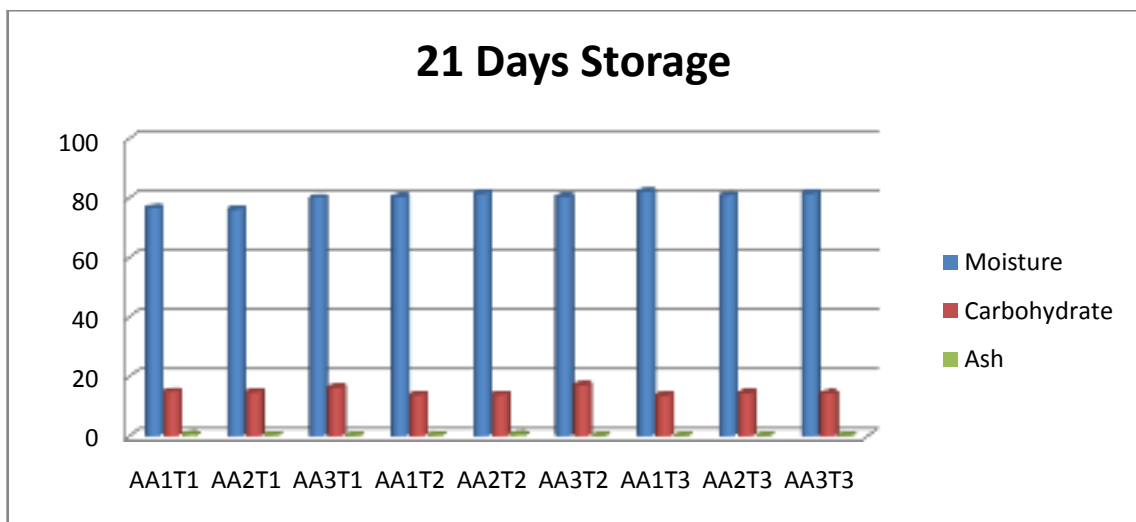
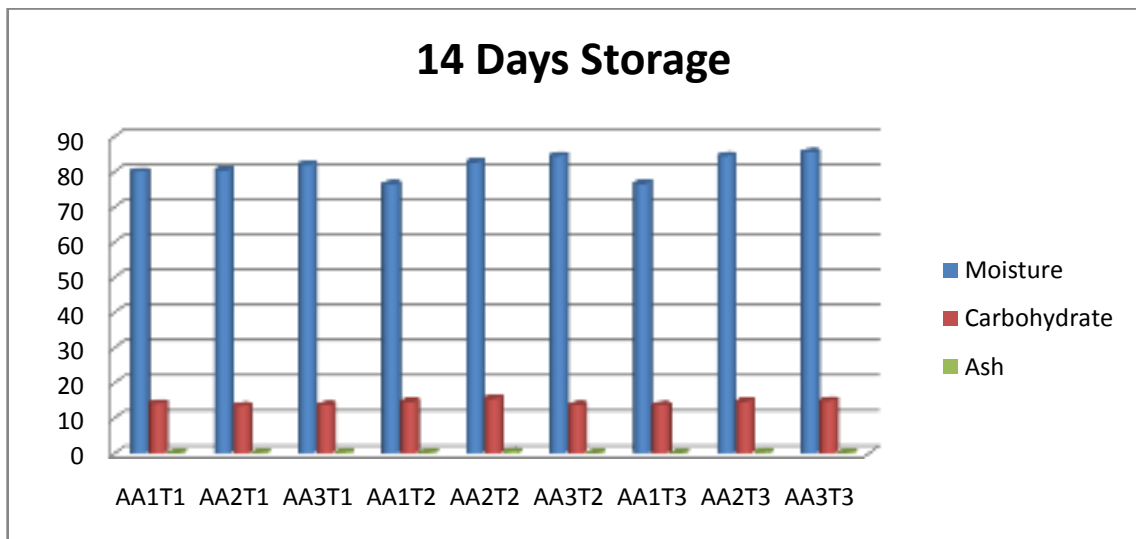
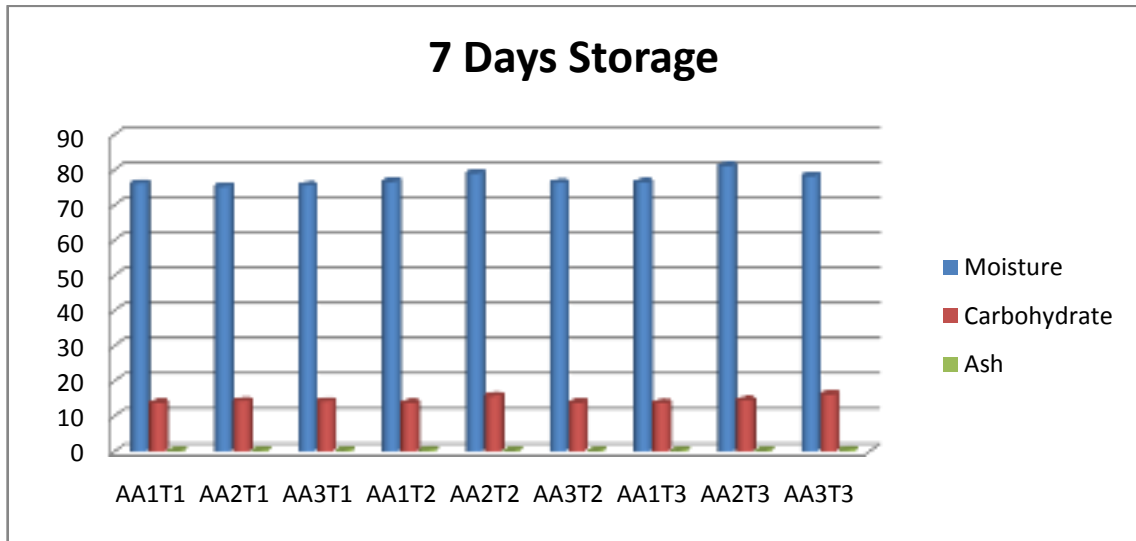


Fig. 21 Effect of ascorbic acid and pasteurization temperature on moisture, carbohydrate and ash content of stored sugarcane juice at ambient condition (30°C)

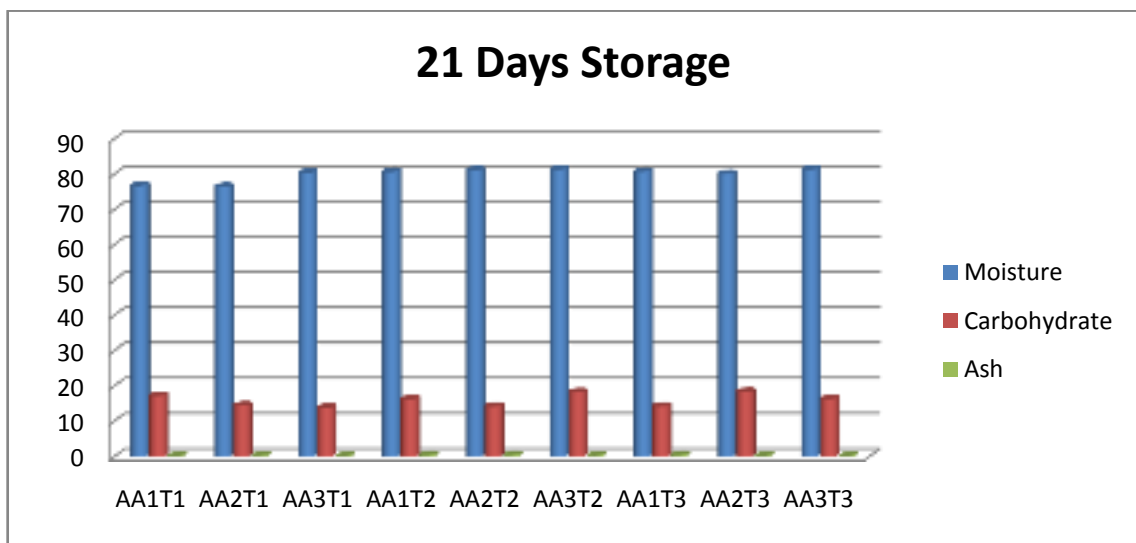
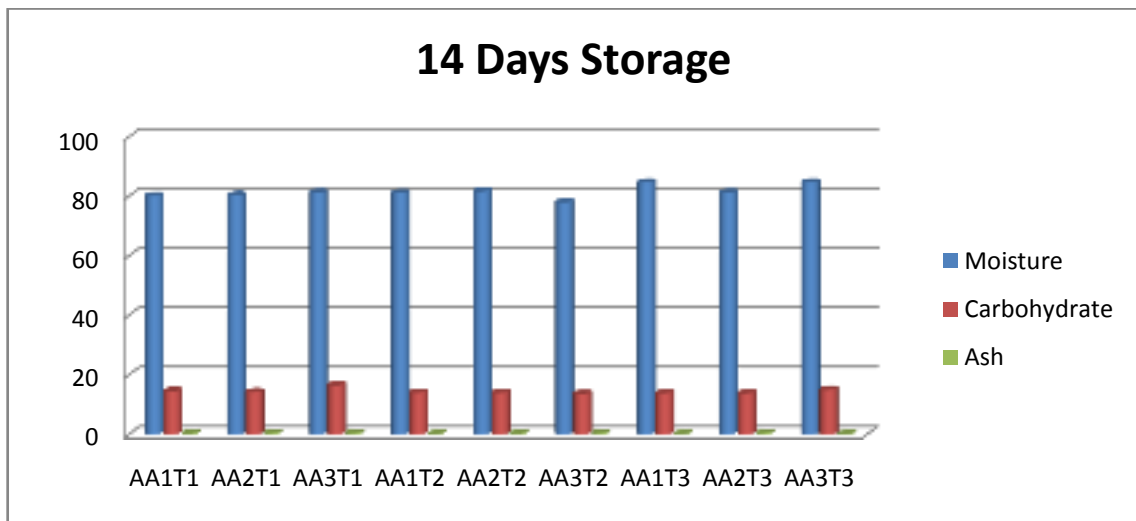
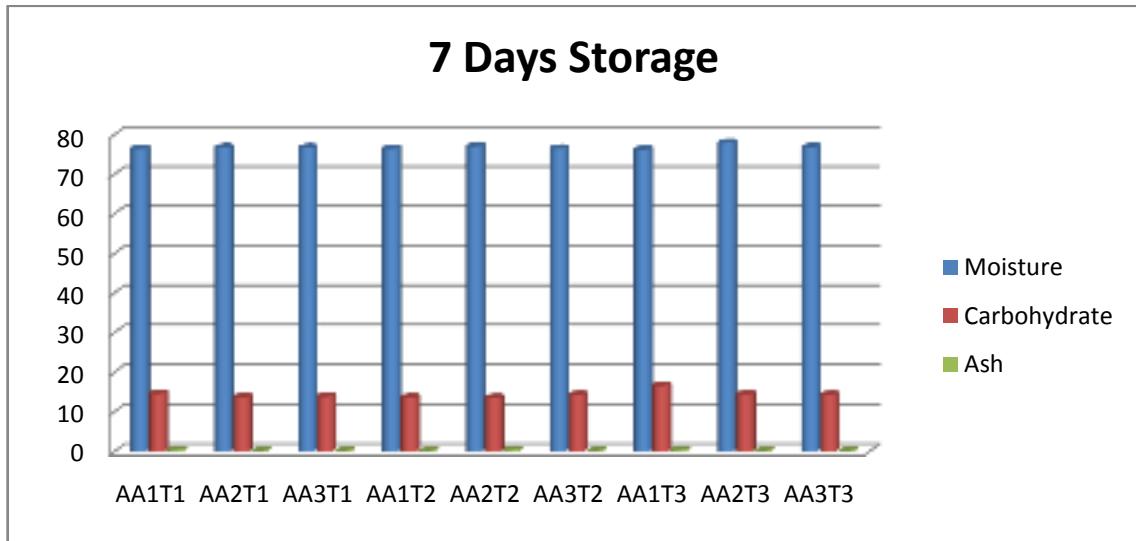


Fig. 22 Effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at refrigerated condition (5°C)

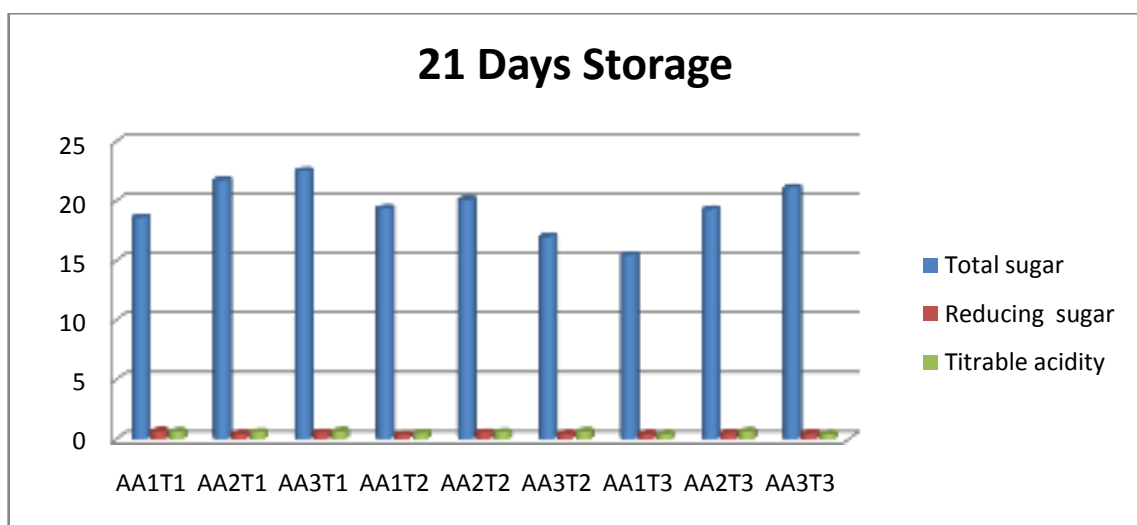
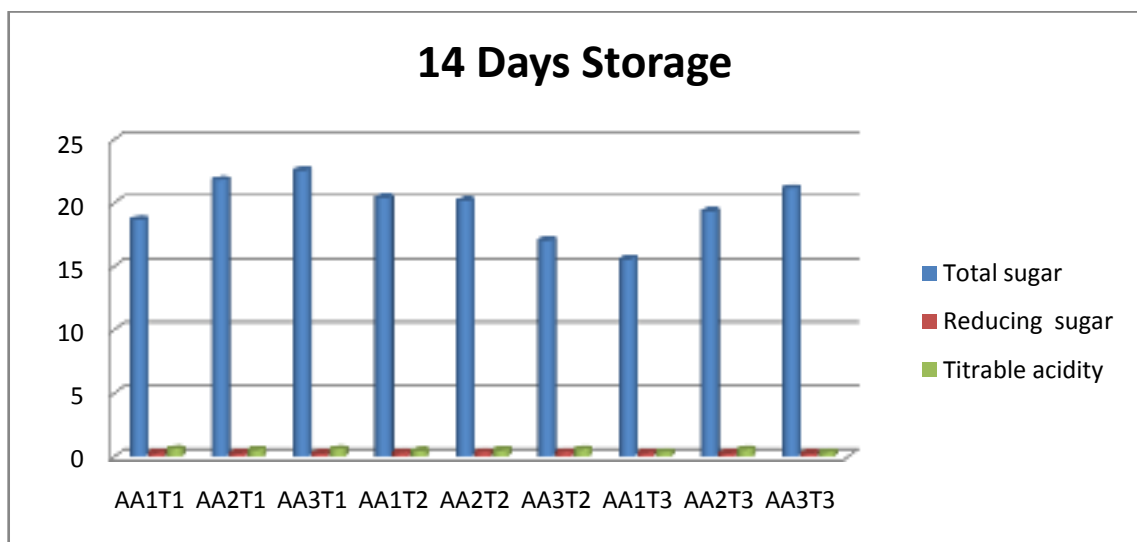
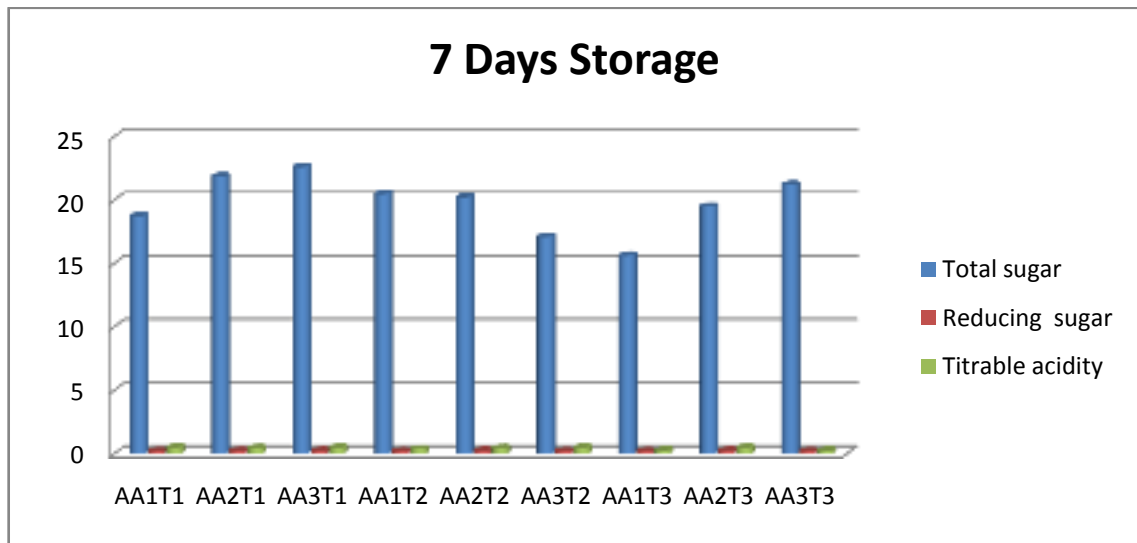


Fig. 23 Effect of ascorbic acid and pasteurization temperature on total sugar, reducing sugar and titrable acidity of stored sugarcane juice at ambient condition (30°C)

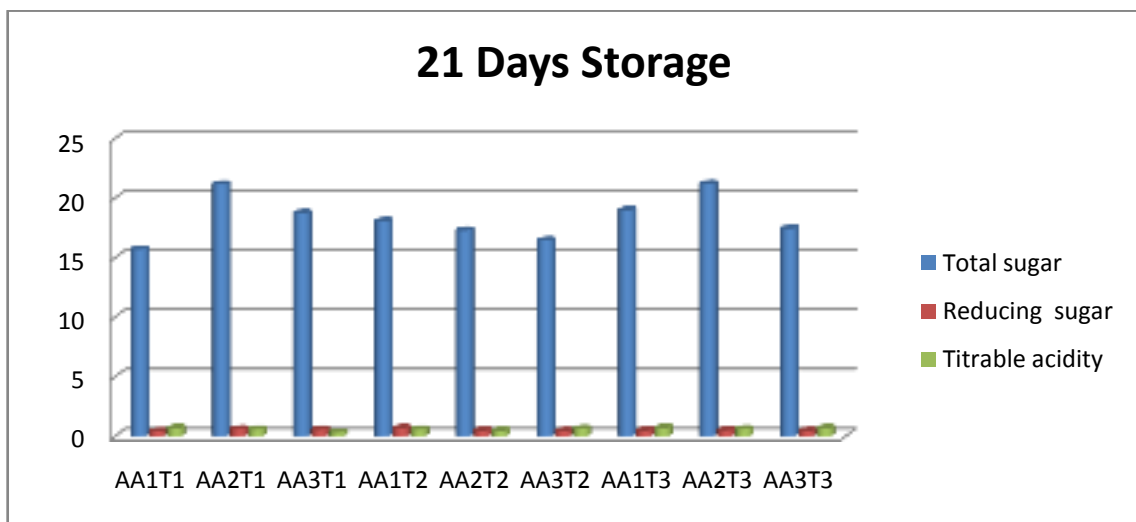
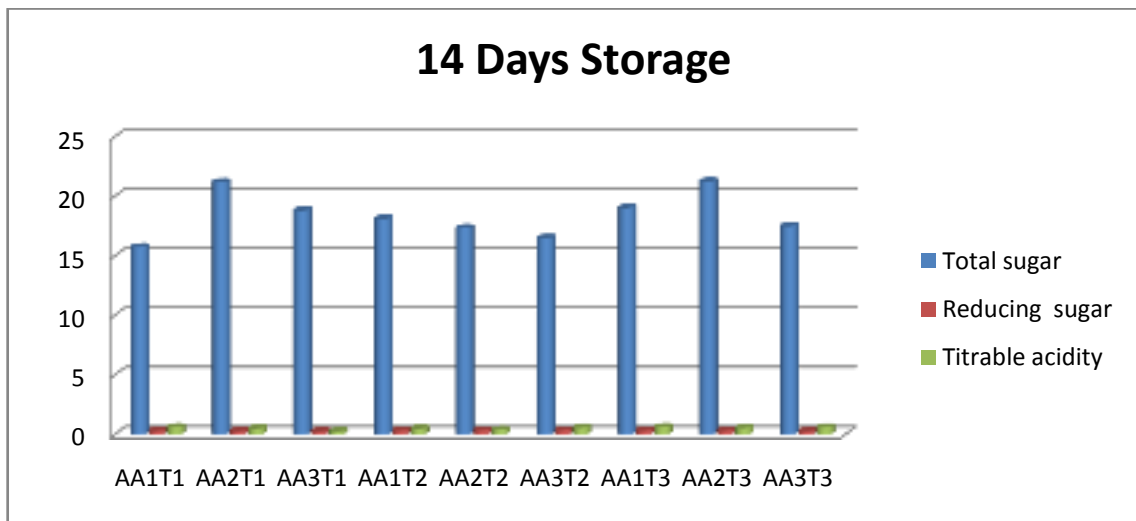
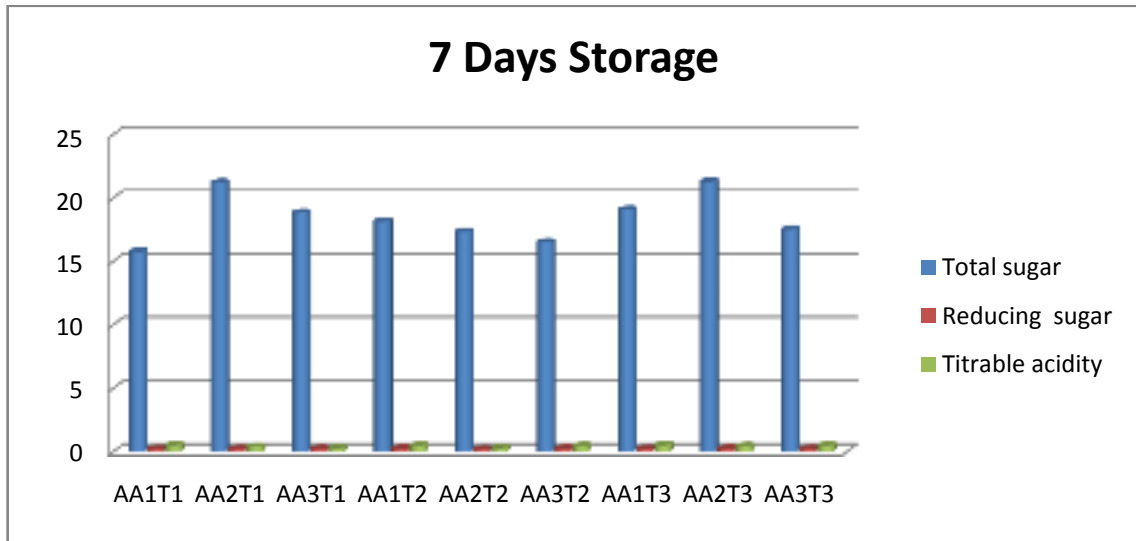


Fig. 24 Effect of ascorbic acid and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at refrigerated condition (5°C)

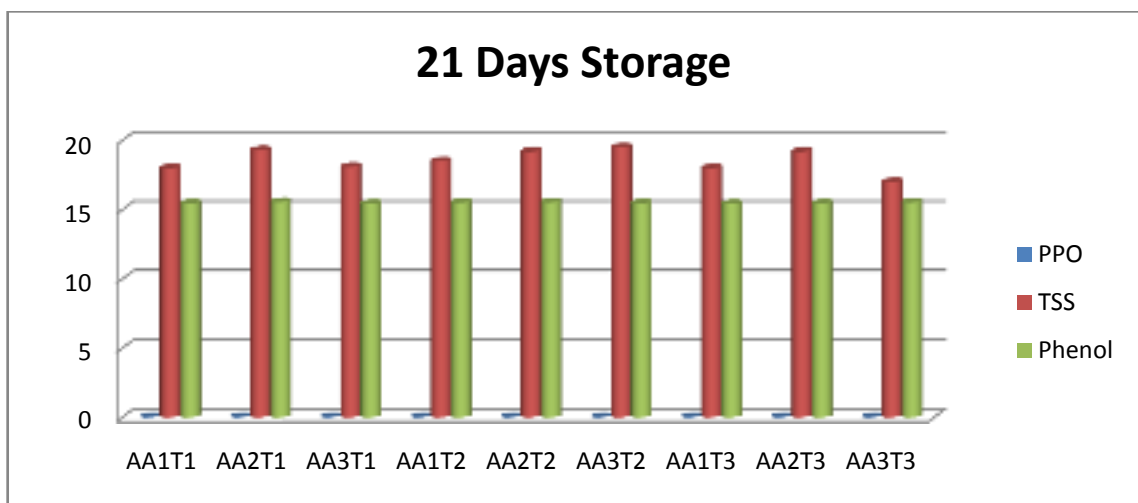
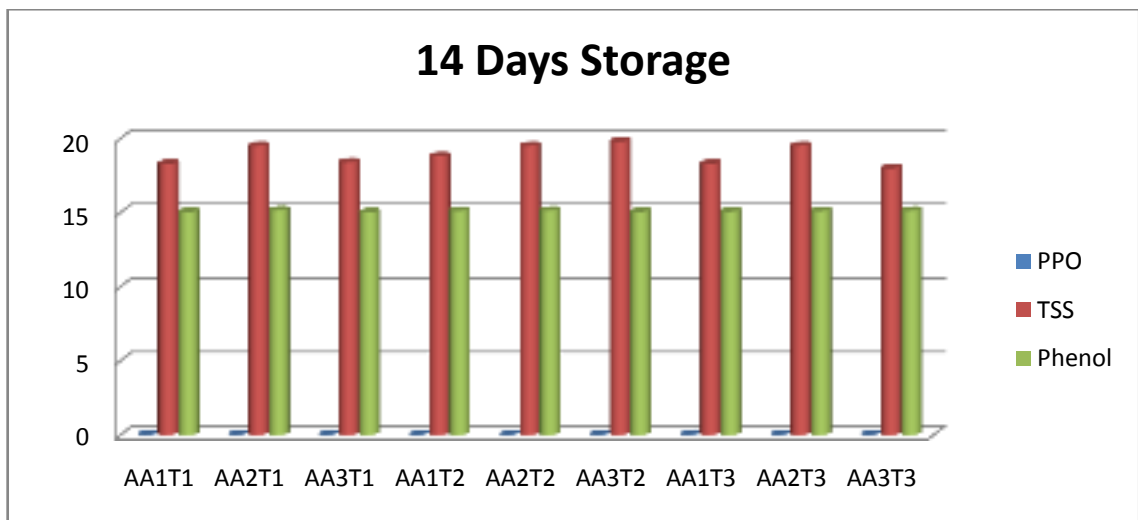
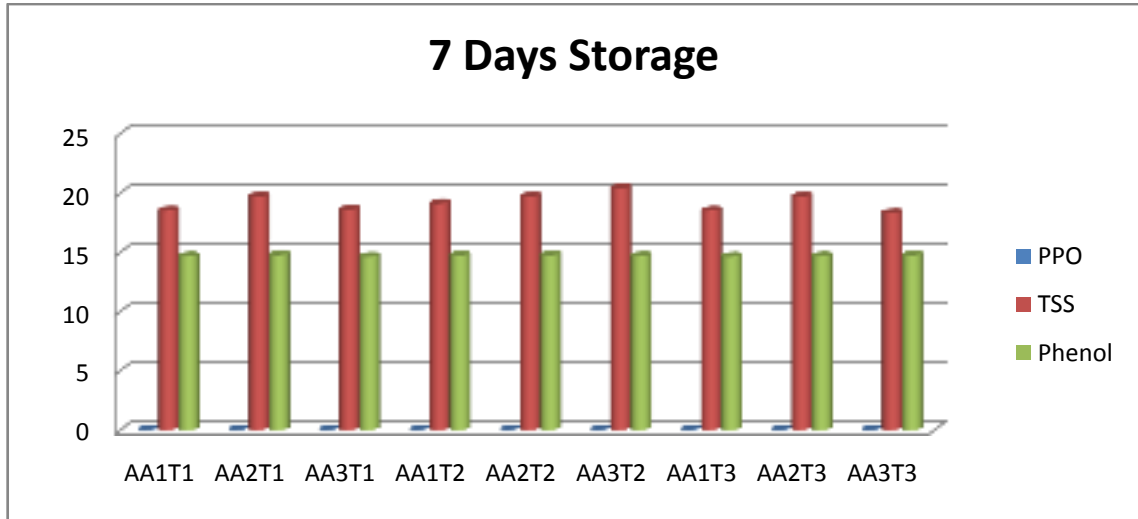


Fig. 25 Effect of ascorbic acid and pasteurization temperature on poly phenol oxidase (PPO), total soluble solids (TSS) and phenol content of stored sugarcane juice at ambient condition (30°C)

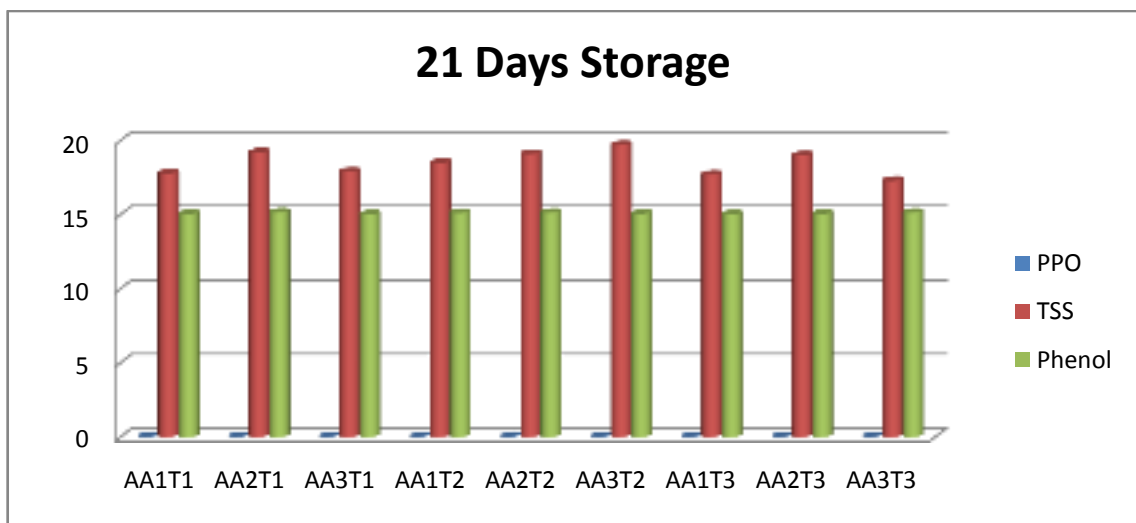
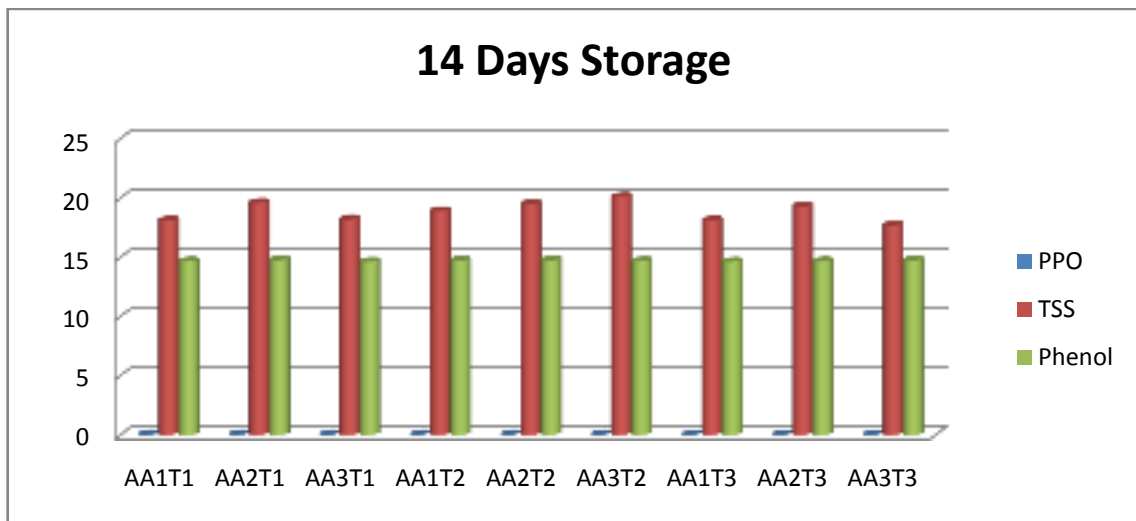
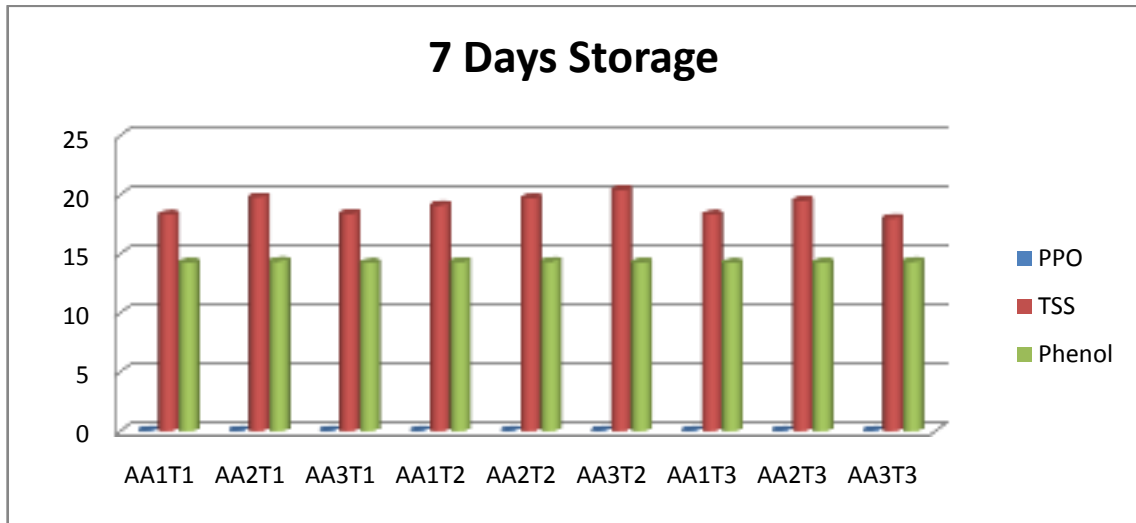


Fig. 26 Effect of citric acid and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

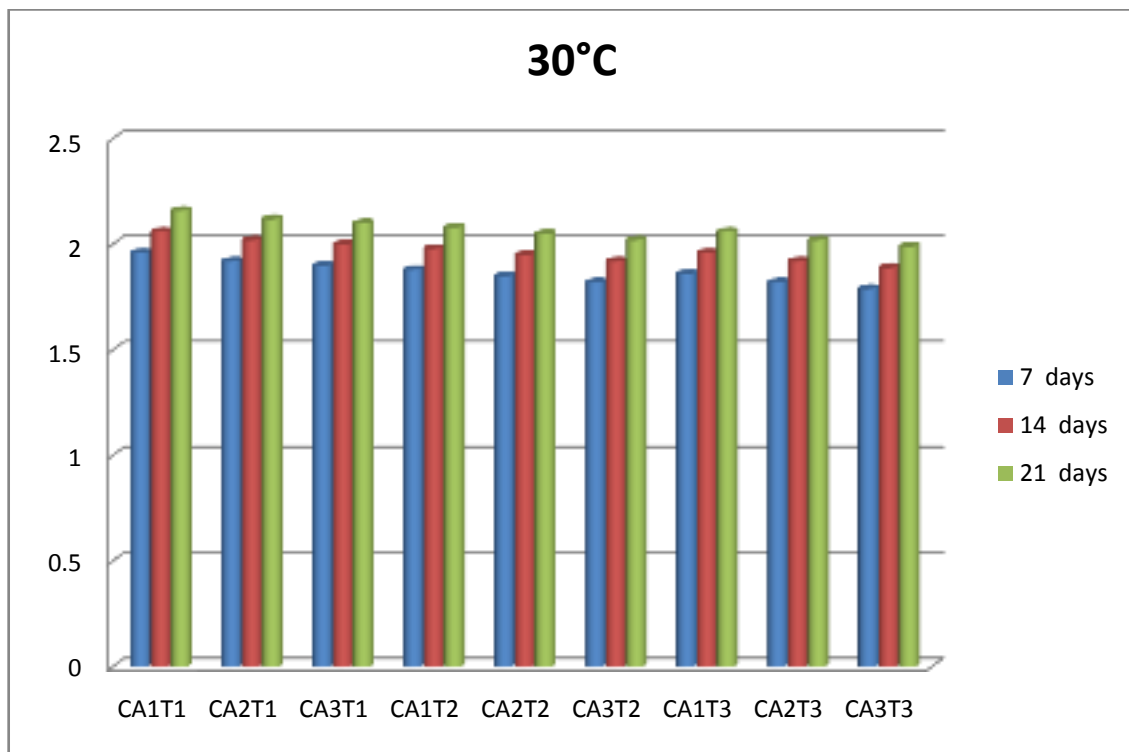
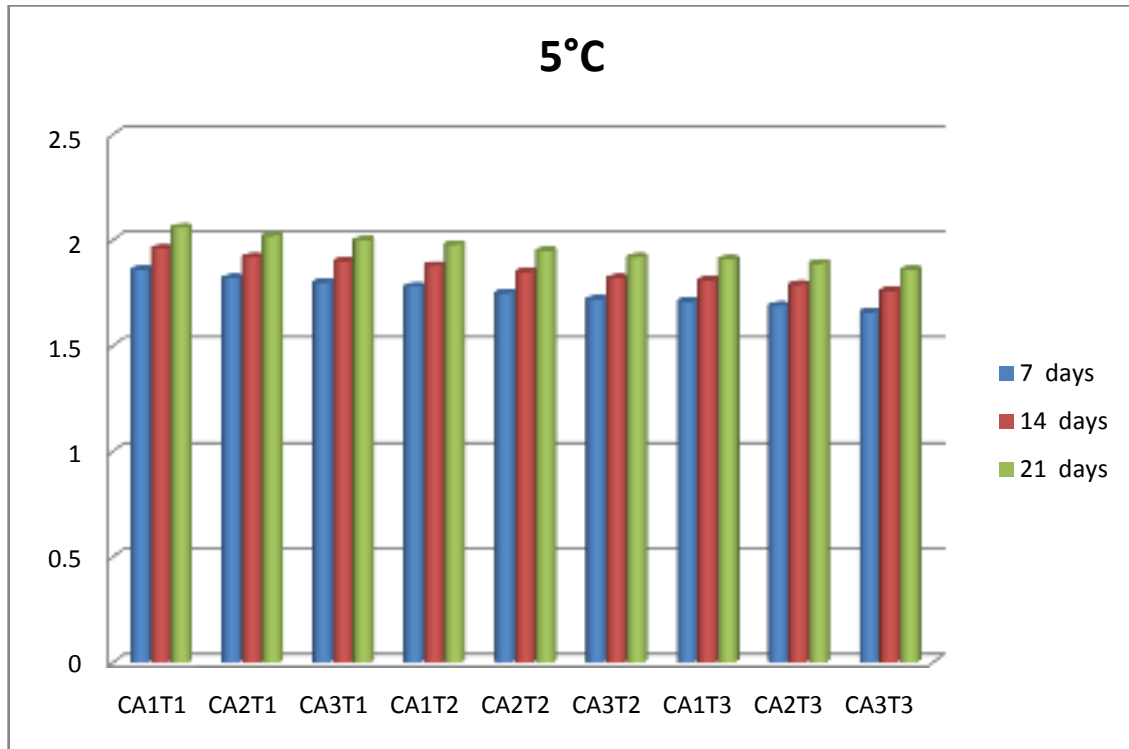


Fig. 27 Effect of potassium metabisuphite and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

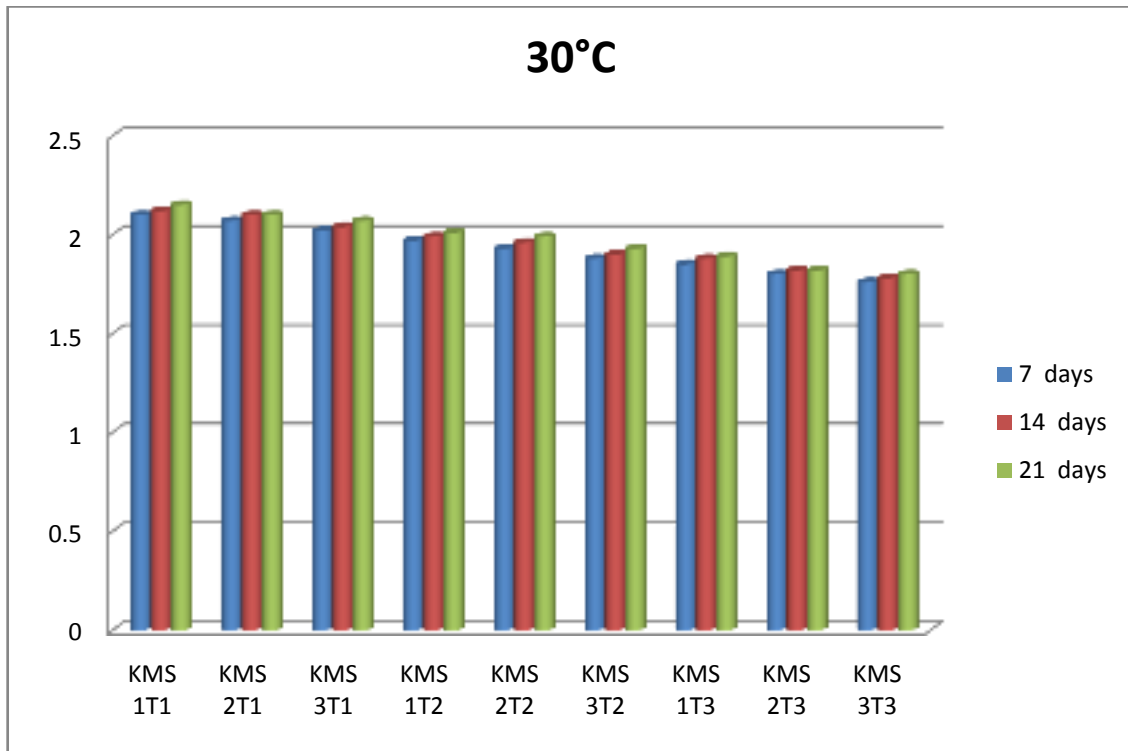
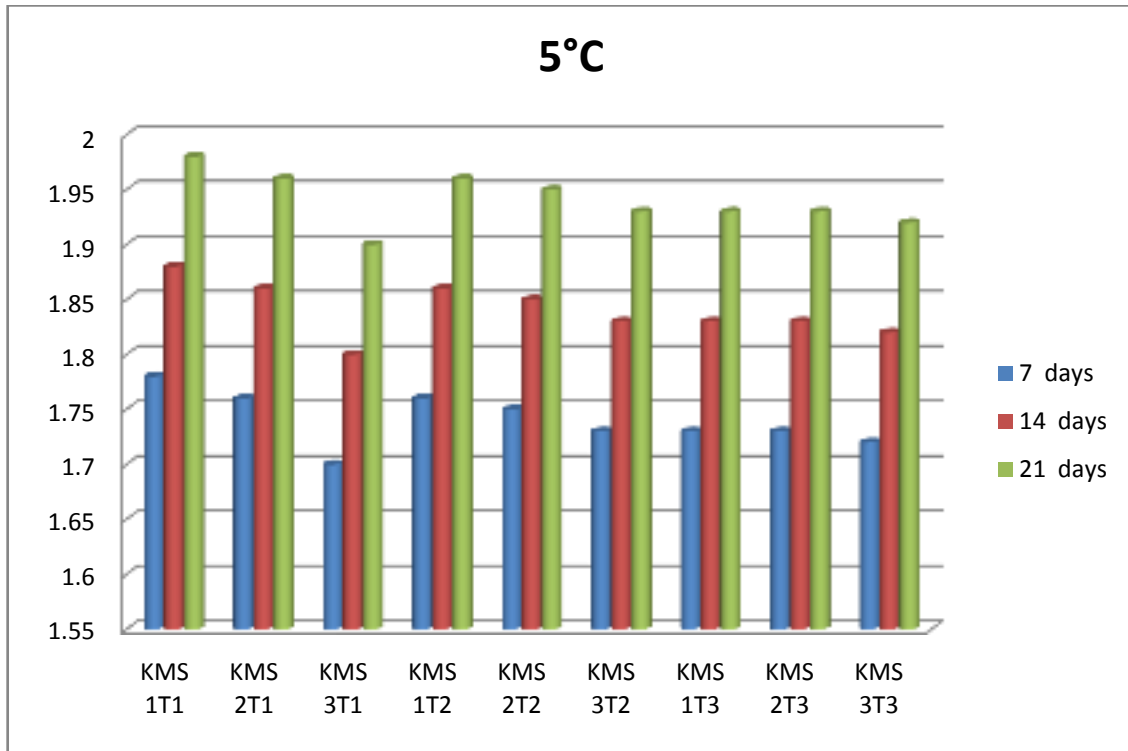


Fig. 28 Effect of potassium sorbate and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

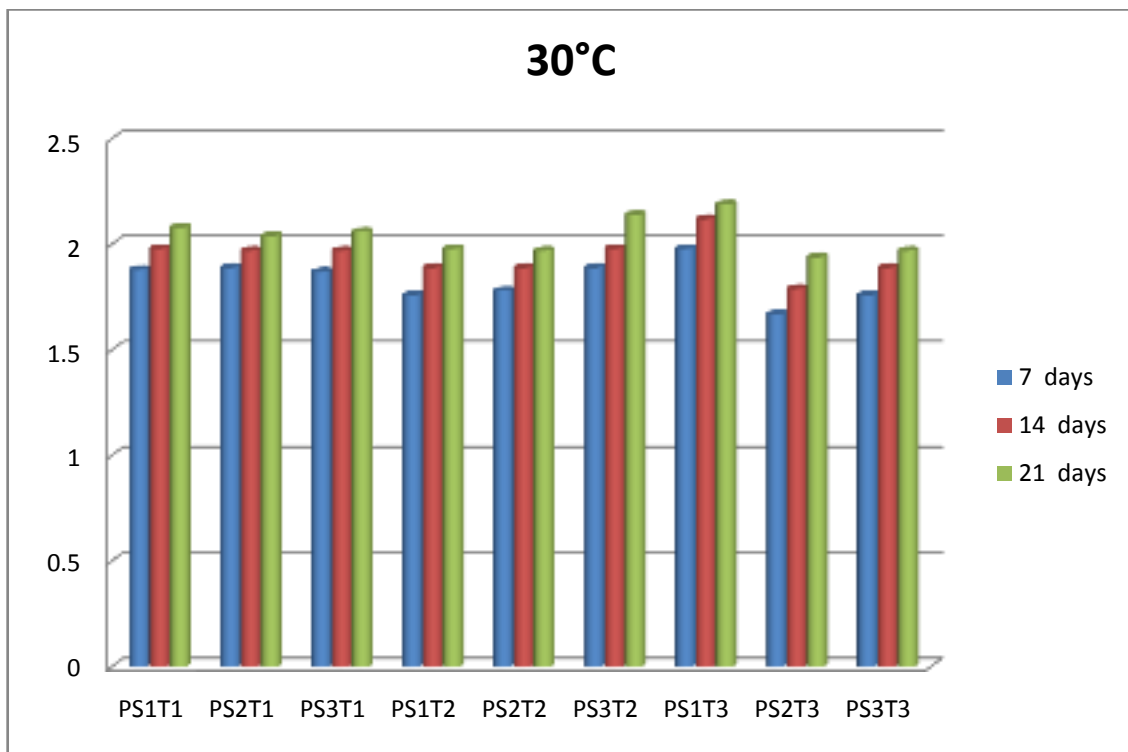
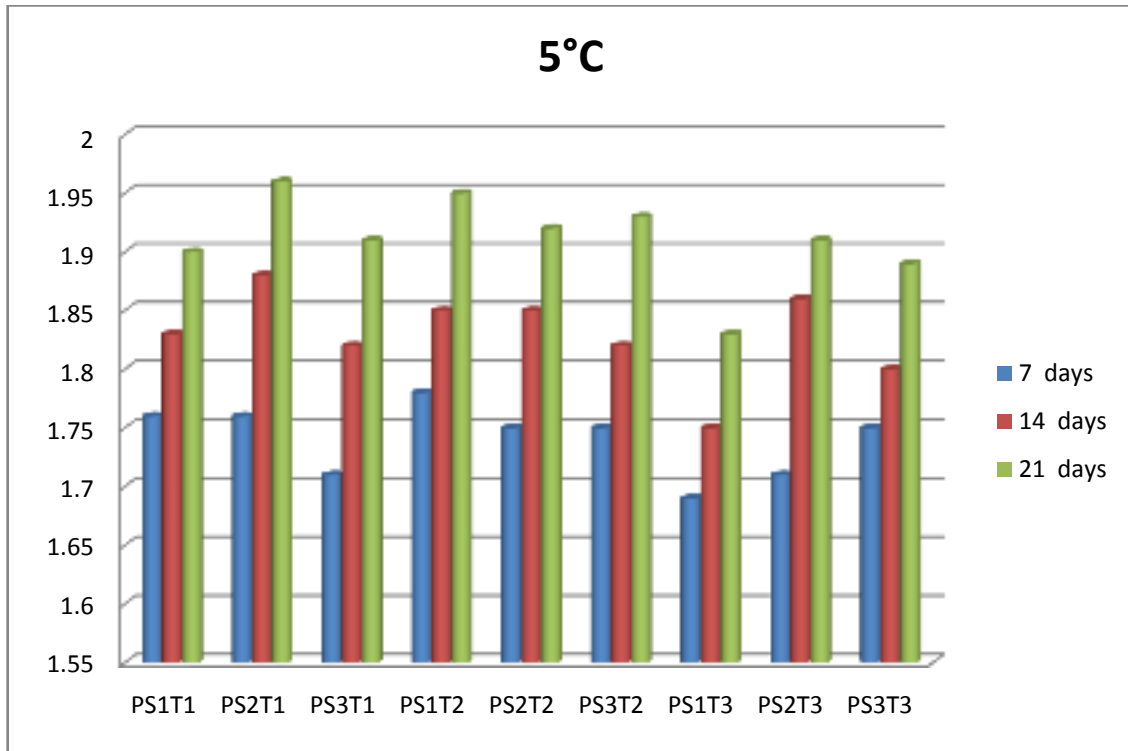


Fig. 29 Effect of ascorbic acid and pasteurization temperature on total plate count of stored sugarcane juice at refrigerated and ambient condition

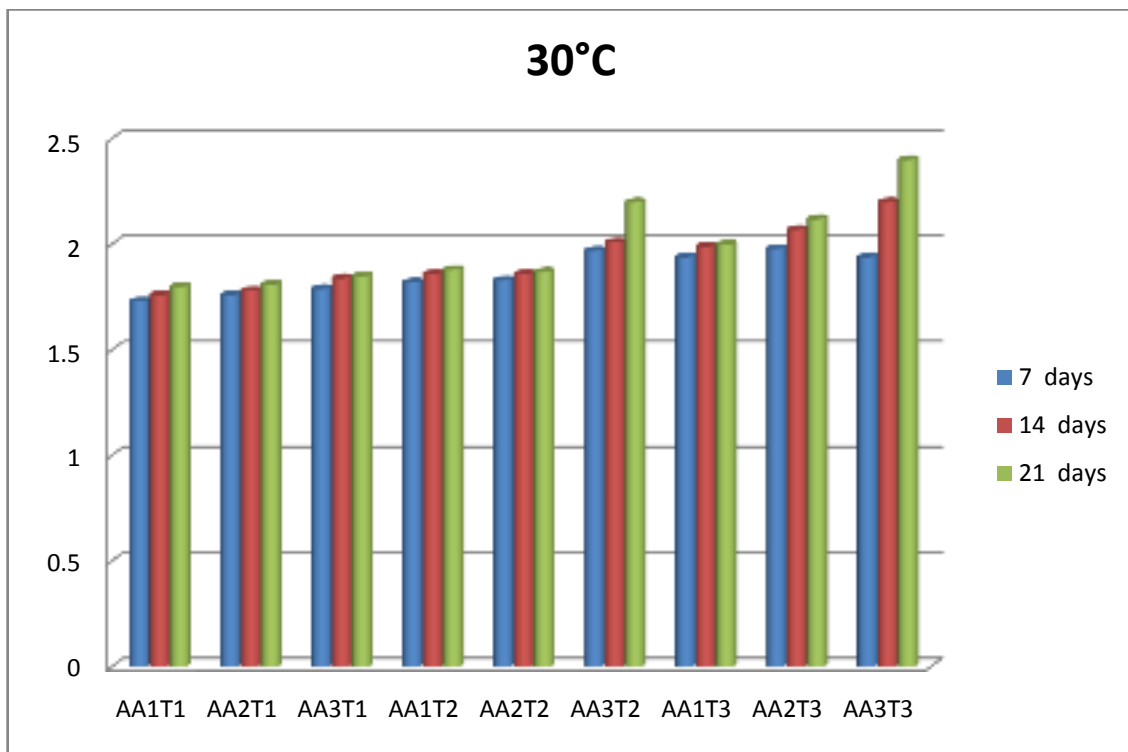
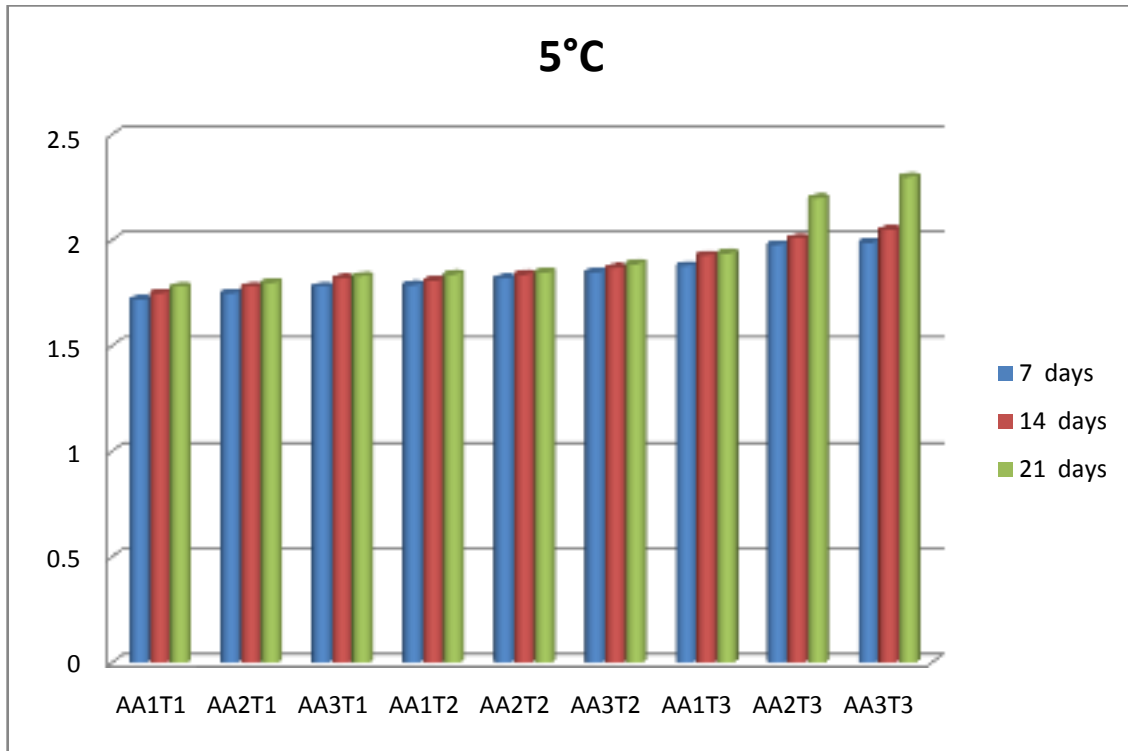


Fig. 30 Effect of citric acid and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

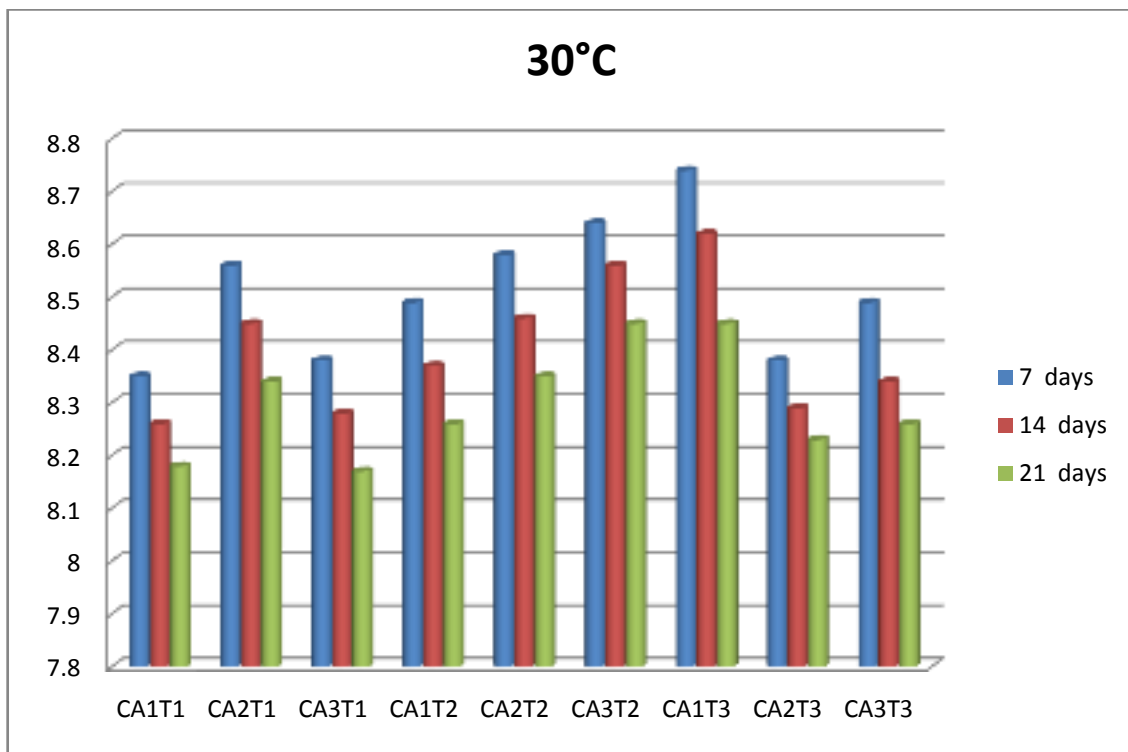
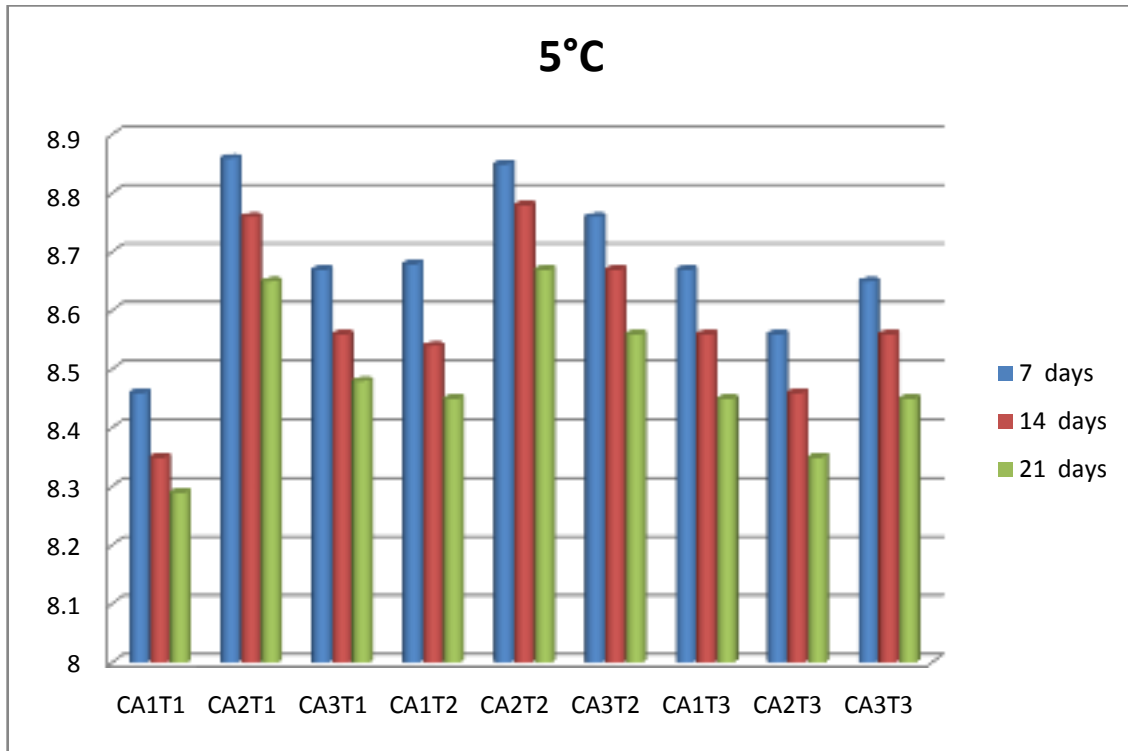


Fig. 31 Effect of potassium metabisulfite and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

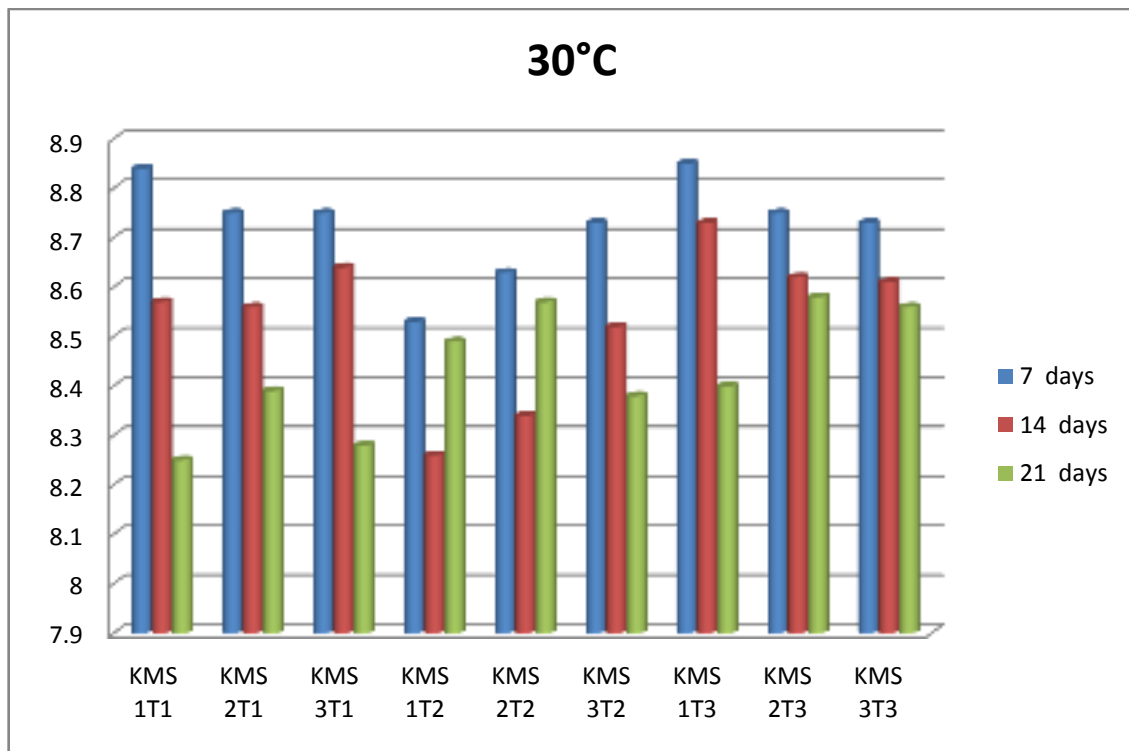
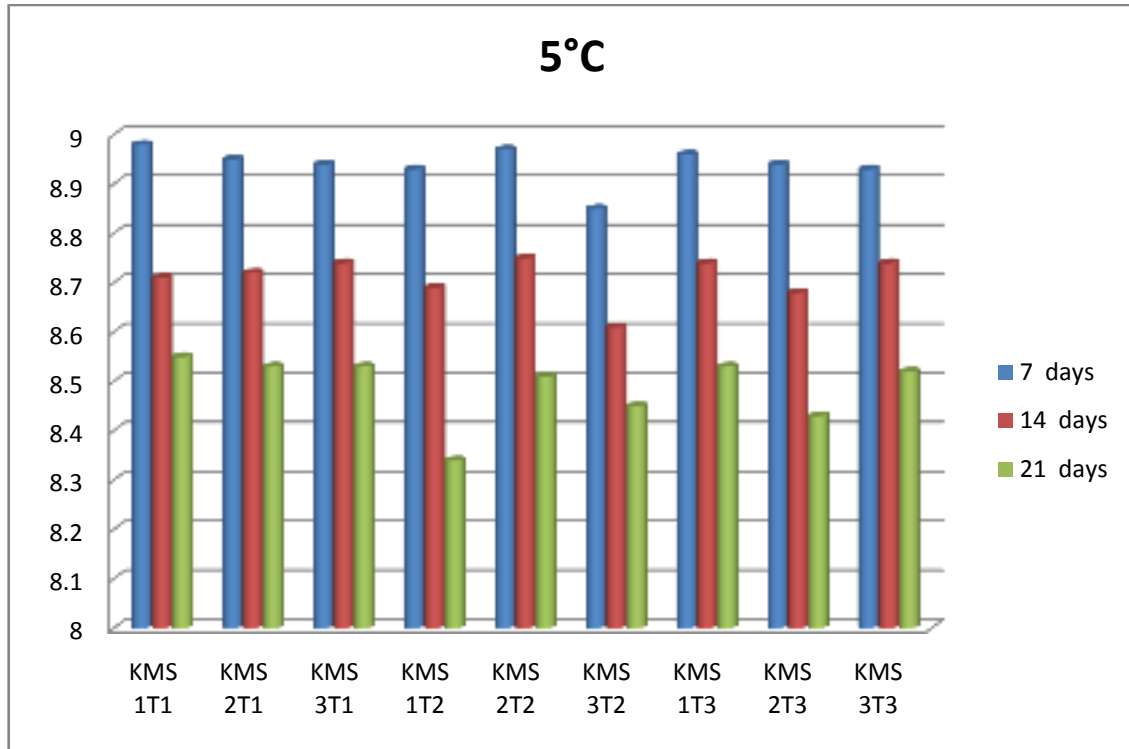


Fig. 32 Effect of potassium sorbate and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition

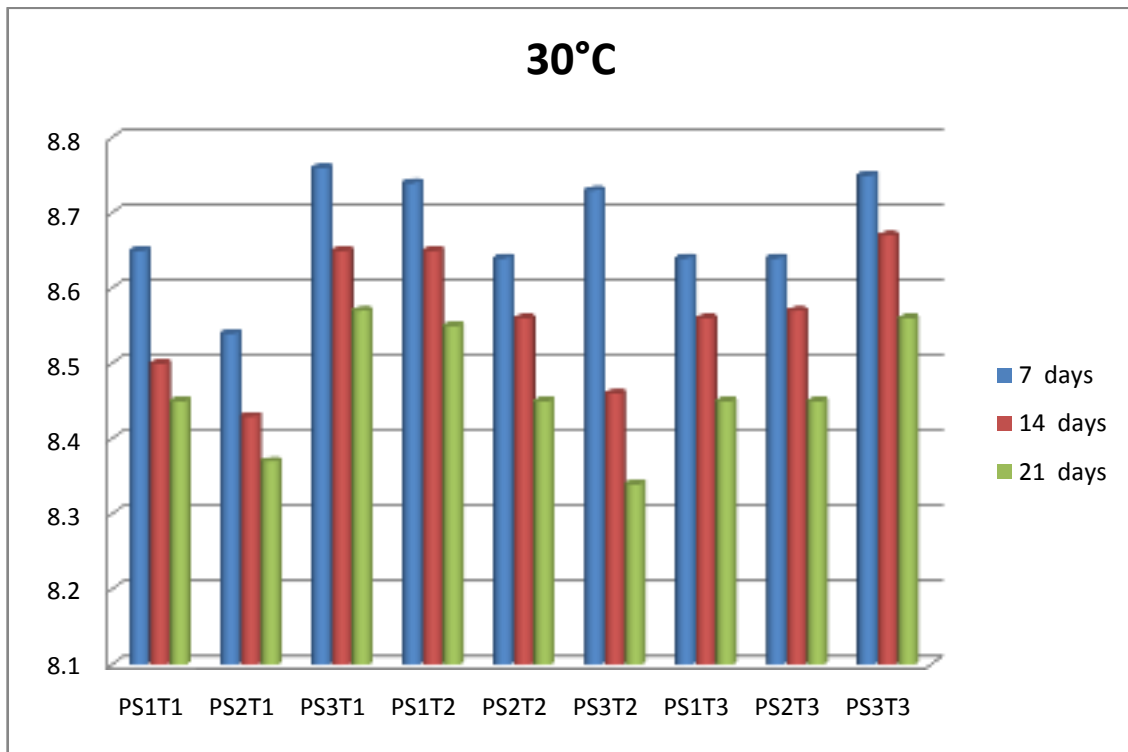
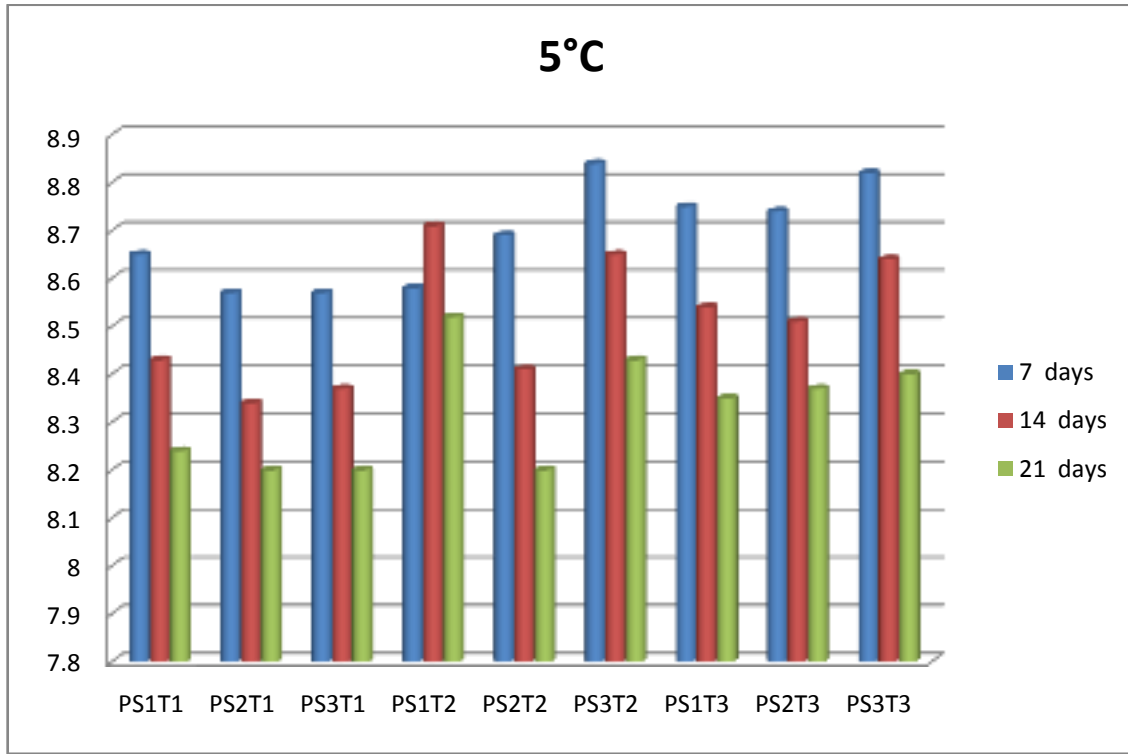
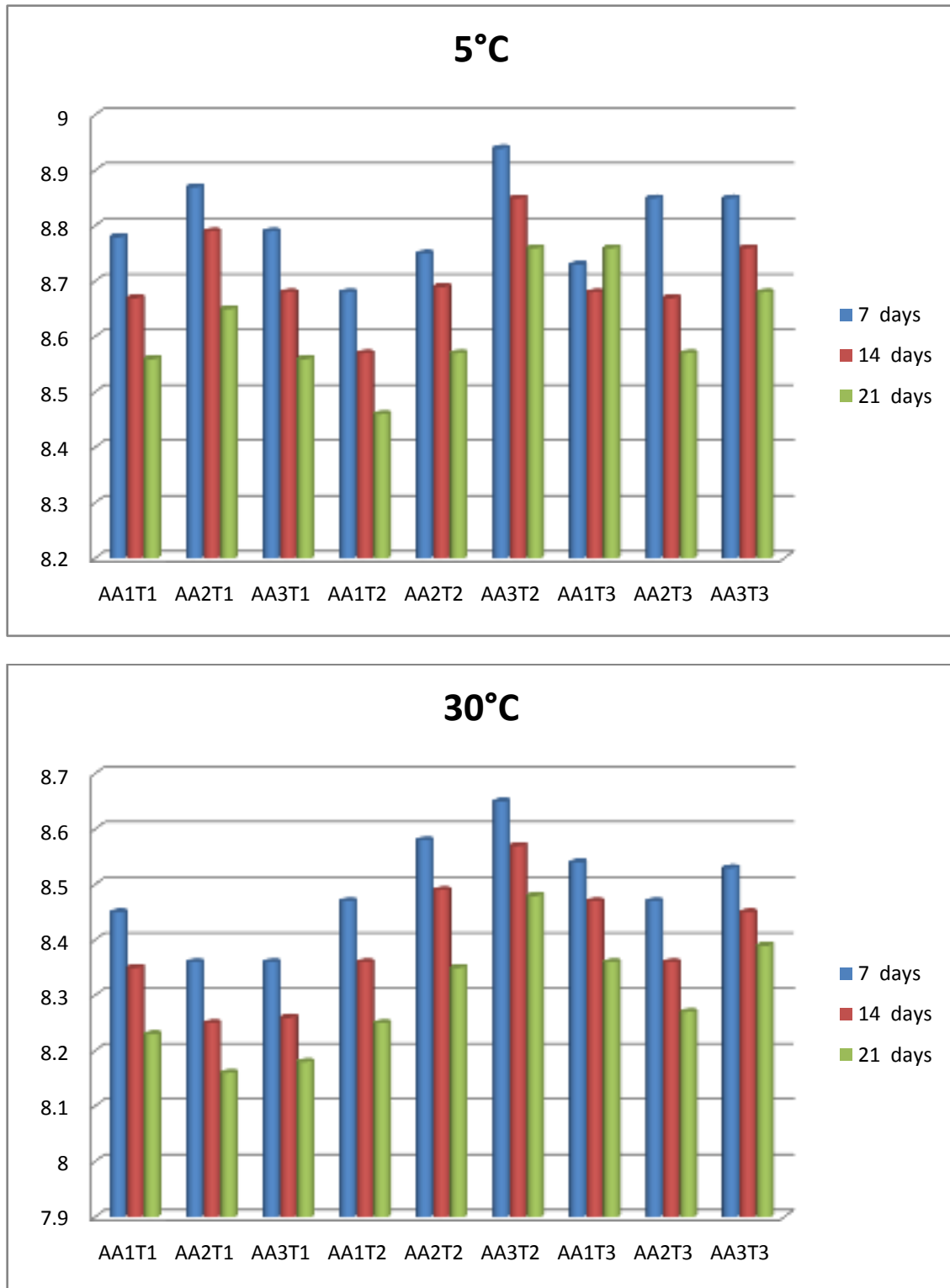


Fig. 33 Effect of ascorbic acid and pasteurization temperature on overall acceptability (Colour, Taste and Flavour) sensory score of stored sugarcane juice at refrigerated and ambient condition



Effect of Citric acid, KMS, Potassium sorbate & Ascorbic acid bottled sugarcane juice stored at 5⁰ C temperature



Effect of Citric acid, KMS, Potassium sorbate & Ascorbic acid bottled sugarcane juice stored at 30⁰ C temperature



FRESH JUICE



Stored at 5⁰ C temperature



Citric acid



Potassium metabisulphite



Potassium Sorbate



Ascorbic acid

Stored at 30⁰ C temperature



Citric acid



Potassium metabisulphite



Potassium Sorbate



Ascorbic acid



Crown cap



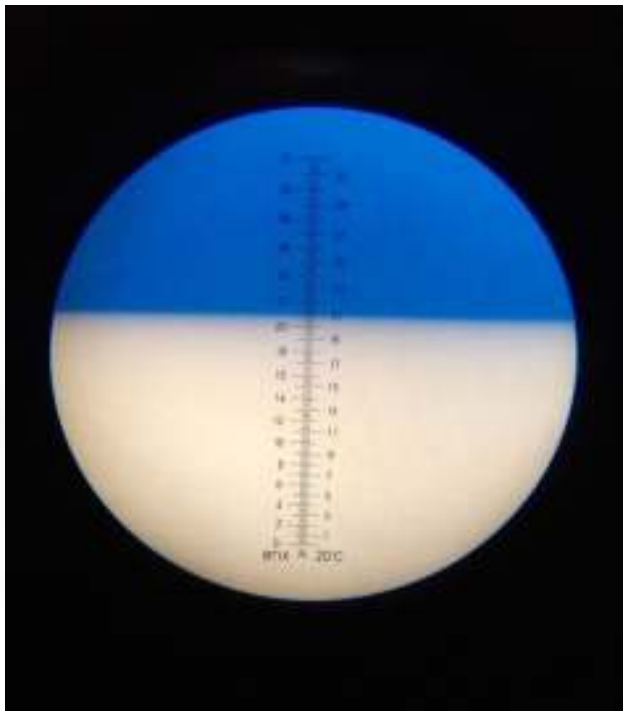
Glass bottles



Crown capping machine



Refractometer



Refractometer reading



Tharmameter

Microbial count





Autoclave