

**UTILIZATION OF PRODUCTS OBTAINED FROM
PYROLYSIS OF PADDY STRAW**

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**DEPARTMENT OF FORESTRY AND
ENVIRONMENTAL SCIENCE
UNIVERSITY OF AGRICULTURAL SCIENCES
BANGALORE**

2021

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Thesis submitted to the

UNIVERSITY OF AGRICULTURAL SCIENCES, BANGALORE

In partial fulfilment of the requirements for the award of the degree of

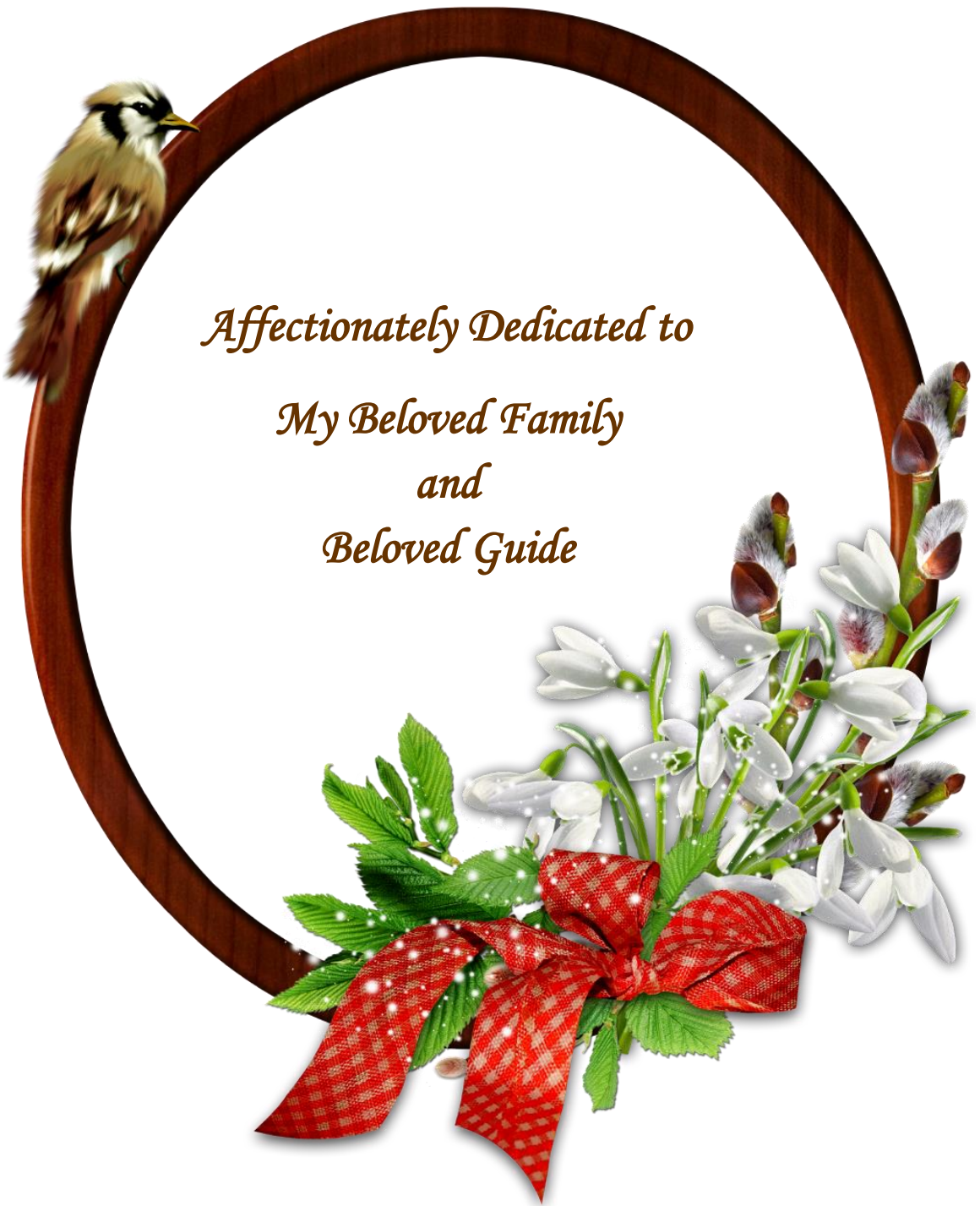
MASTER OF SCIENCE (Agriculture)

in

ENVIRONMENTAL SCIENCE

BENGALURU

OCTOBER, 2021



*Affectionately Dedicated to
My Beloved Family
and
Beloved Guide*

**DEPARTMENT OF FORESTRY AND
ENVIRONMENTAL SCIENCE
UNIVERSITY OF AGRICULTURAL SCIENCES
BANGALORE**

CERTIFICATE

This is to certify that the thesis entitled “UTILIZATION OF PRODUCTS OBTAINED FROM PYROLYSIS OF PADDY STRAW” submitted by Ms. ANUSHA, B. S., ID. No. PALB 9246 for the degree of MASTER OF SCIENCE (Agriculture) in ENVIRONMENTAL SCIENCE of the University of Agricultural Sciences, Bangalore is a *bona-fide* record of research work carried out by her during the period of her study in this University under my guidance and supervision and no part of the thesis has been submitted for the award of any other degree, diploma, associateship, fellowship or any other similar titles.

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(ANUSHA, B. S.)

UTILIZATION OF PRODUCTS OBTAINED FROM PYROLYSIS OF PADDY STRAW

ANUSHA, B. S.

ABSTRACT

A poly bag study was undertaken during December 2020 at RIOF, UAS, GKVK Bangalore 65. Pyrolysis of paddy straw was done at 500°C without using catalyst, 2.29 kg of biochar containing 0.41% Nitrogen, 0.04% Phosphorus, 3.83% Potassium, 12 ml Bio oil which is rich in phenol (27.23%) and 101.48 L of syngas was obtained. Further the biochar was used in packet experiment to evaluate the influence of biochar on performance of Field bean using Completely Randomized Design with twenty-four treatments replicated twice. The results revealed that application of 25 tons of biochar + FYM + Vermicompost + Ganajeevamrutha recorded higher growth parameters *viz.*, plant height (66 cm), number of leaves (80), number of branches (14) and yield attributes like number of pods/plant (16) and pod weight/plant (21g). Application of biochar recorded significant improvement in water holding capacity (42%) of the soil. The soil pH (9.5), electrical conductivity (0.19 dSm^{-1}) and organic carbon (6.42 g kg^{-1}) significantly increased, however decrease in bulk density (1.34 Mg m^{-3}) of soil was noticed in the same treatment. There was significant increase in Nitrogen (297 kg ha^{-1}), Phosphorus (63.25 kg ha^{-1}) and Potassium ($173.95 \text{ kg ha}^{-1}$) in post-harvest soil. The Iron and Manganese concentration decreased and increase in Zinc and Copper concentration in post-harvest soil was observed.

October, 2021

Department of Forestry and Environmental science
UAS, GKVK, Bengaluru

RINKU VERMA
Major Advisor

ಭತ್ತದ ಹುಲ್ಲಿನ ಪೈರಾಲಿಸಿಸ್ ನಿಂದ ನಿರ್ಬಂಧಿತವಾದ ಉತ್ಪನ್ನಗಳ ಬಳಕೆ

ಅನುಪಾ, ಬಿ.ಎಸ್ .

ಪ್ರಬಂಧ ಸಾರಾಂಶ

ಕೃಷಿ ವಿಶ್ವವಿದ್ಯಾಲಯ, ಜಿಕೆವಿಕೆ ಬೆಂಗಳೂರು 65 ರಲ್ಲಿ ಪ್ಯಾಕೆಟ್ ಅಧ್ಯಯನವನ್ನು ಡಿಸೆಂಬರ್ 2020 ರಲ್ಲಿ ಕೈಗೊಳ್ಳಲಾಯಿತು. ವೇಗವರ್ಧಕವನ್ನು ಬಳಸದೆ 500° ಸೆಲ್ಸಿಯಸ್‌ಯಲ್ಲಿ ಭತ್ತದ ಒಣಹುಲ್ಲಿನ ಪೈರಾಲಿಸಿಸ್‌ನಿಂದ, 0.41 ಶೇಕಡಾ ಸಾರಾಜನಕ, 0.04 ಶೇಕಡಾ ರಂಜಕ, 3.83 ಶೇಕಡಾ ಪೊಟ್ಯಾಸಿಯಮ್ ಹೊಂದಿರುವ 2.29 ಕೆಜಿ ಜೈವಿಕ ಇದ್ದಲು, 12 ಮಿಲಿ ಫಿನಾಲ್ (27.23 ಶೇಕಡಾ) ಸಮೃದ್ಧವಾದ ಜೈವಿಕ ತೈಲ ಮತ್ತು 101.48 ಲೀಟರ್ ಸಿಂಗ್ಯಾಸ್ ಪಡೆಯಲಾಗಿದೆ. ಮತ್ತಷ್ಟು ಜೈವಿಕ ಇದ್ದಲನ್ನು ಪ್ಯಾಕೆಟ್ ಪ್ರಯೋಗದಲ್ಲಿ ಬಳಸಲಾಗಿದ್ದು, ಅವರೆ ಬೆಳೆಯ ಕಾರ್ಯಕ್ಷಮತೆಯ ಮೇಲೆ ಜೈವಿಕ ಇದ್ದಲಿನ ಪ್ರಭಾವವನ್ನು ಮೌಲ್ಯಮಾಪನ ಮಾಡಲು ಸಂಪೂರ್ಣ ಯಾದೃಚ್ಛಿಕ ವಿನ್ಯಾಸವನ್ನು ಬಳಸಿ ಇಪ್ಪತ್ತಾಲ್ಪು ಉಪಚಾರಗಳನ್ನು ಎರಡು ಬಾರಿ ಪುನರಾವರ್ತಿತಿಸಲಾಗಿದೆ. ಫಲಿತಾಂಶಗಳು 25 ಟನ್ ಜೈವಿಕ ಇದ್ದಲು + ಕೊಟ್ಟಿಗೆ ಗೊಬ್ಬರ + ಎರೆಹುಳು ಗೊಬ್ಬರ + ಘನಜೀವಾಮೃತದ ಅನ್ವಯವು ಸಸ್ಯದ ಎತ್ತರ (66 ಸೆಂಮೀ), ಎಲೆಗಳ ಸಂಖ್ಯೆ (80), ಕೊಂಬೆಗಳ ಸಂಖ್ಯೆ ಪ್ರತಿ ಗಿಡಕ್ಕೆ (14) ಮತ್ತು ಇಳುವರಿ ಗುಣಲಕ್ಷಣಗಳು ಬೀಜಗಳ ಸಂಖ್ಯೆ ಪ್ರತಿ ಗಿಡಕ್ಕೆ (16) ಮತ್ತು ಬೀಜಗಳ ತೂಕ ಪ್ರತಿ ಗಿಡಕ್ಕೆ (21 ಗ್ರಾಂ) ಹೆಚ್ಚಿನ ಬೆಳವಣಿಗೆಯ ನಿಯತಾಂಕಗಳನ್ನು ದಾಖಲಿಸಿದೆ ಎಂದು ತಿಳಿದುಬಂದಿದೆ. ಜೈವಿಕ ಇದ್ದಲನ್ನು ಅನ್ವಯಿಸುವುದರಿಂದ ಮಣ್ಣಿನ ನೀರಿನ ಹಿಡಿದಿಡುವ ಸಾಮರ್ಥ್ಯದಲ್ಲಿ (42 ಶೇಕಡಾ) ಗಮನಾರ್ಹ ಸುಧಾರಣೆ ದಾಖಲಾಗಿದೆ. ಮಣ್ಣಿನ ರಸಸಾರ (9.5), ವಿದ್ಯುತ್ ವಾಹಕತೆ (0.19 ಡೆಸಿ ಸೀಮೆನ್ಸ್ ಪ್ರತಿ ಮೀಟರ್‌ಗೆ) ಮತ್ತು ಸಾವಯವ ಇಂಗಾಲ (6.42 ಗ್ರಾಂ ಪ್ರತಿ ಕೆಜಿಗೆ) ಗಣನೀಯವಾಗಿ ಹೆಚ್ಚಾಗಿದೆ. ಕೊಯ್ಲಿನ ನಂತರದ ಮಣ್ಣಿನಲ್ಲಿ ಸಾರಜನಕ (297 ಕೆಜಿ ಪ್ರತಿ ಹೆಕ್ಟಾರ್‌ಗೆ), ರಂಜಕ (63.25 ಕೆಜಿ ಪ್ರತಿ ಹೆಕ್ಟಾರ್‌ಗೆ) ಮತ್ತು ಪೊಟ್ಯಾಸಿಯಮ್ (173.95 ಕೆಜಿ ಪ್ರತಿ ಹೆಕ್ಟಾರ್‌ಗೆ) ಗಮನಾರ್ಹ ಹೆಚ್ಚಳ ಕಂಡುಬಂದಿದೆ. ಕಬ್ಬಿಣ ಮತ್ತು ಮ್ಯಾಂಗನೀಸ್ ಸಾಂದ್ರತೆಯು ಕಡಿಮೆಯಾಗಿದೆ ಮತ್ತು ಕೊಯ್ಲಿನ ನಂತರದ ಮಣ್ಣಿನಲ್ಲಿ ಸತುವು ಮತ್ತು ತಾಮ್ರದ ಸಾಂದ್ರತೆಯು ಹೆಚ್ಚಾಗಿದೆ.

ಅಕ್ಟೋಬರ್, 2021

ಅರಣ್ಯ ಮತ್ತು ಪರಿಸರ ವಿಜ್ಞಾನ ವಿಭಾಗ
ಕೃಷಿ ವಿಶ್ವವಿದ್ಯಾಲಯ, ಬೆಂಗಳೂರು

ರಿಂಕು ವರ್ಮಾ
ಪ್ರಧಾನ ಸಲಹೆಗಾರರು



Effect of paddy straw biochar on growth and yield parameters of field bean and soil properties

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Introduction

❖ Agricultural crop residues form a major source of biomass in India and annually about 69.7 million tonnes of crop residues are produced from the six major crops. Hence, the conversion of crop waste into biochar helps to offset CO₂ emission and increase the carbon sequestration in the soil under changing climate.

❖ Biochar is a carbon rich substance obtained from thermal decomposition of plant biomass through the process of pyrolysis. The special characteristics of the biochar are its effectiveness in retaining most of the nutrients and making them available to plants rather than other organic substances such as natural leaf litter, compost or manures. Biochar typically has a powerful aromatic structure and is therefore biochemically more recalcitrant than any other type of organic matter in soils and may last in the soil for thousands of years. Therefore it's feasible to use biochar as part of a change to organic farming methods to transform agriculture from a net carbon emitter to a method to restore carbon stocks in the soil.

❖ Objective:

- ✓ To evaluate the effect of paddy straw biochar on field bean and post harvest soil properties.

Material and Methods

❖ The paddy straw biochar was produced by the process of pyrolysis at 500°C and was ground to pass through 0.2 mm sieve for further application.

Experiment details:

Materials used: FYM, Vermicompost, Ganajeevamrutha, Paddy straw biochar

Location: Organic farming, GKVK, Bangalore.

Season: Rabi

Crop, Variety: Field bean, Hebbal Avarae

Design: CRD

No. of treatments: 24

No. of replications: 2

Initial soil properties:

pH – 7.2, EC – 0.22 dSm⁻¹, OC- 5.2 g kg⁻¹, N – 263.54 kg ha⁻¹, P₂O₅ – 46.37 kg ha⁻¹, K₂O – 196 kg ha⁻¹

Treatment details:

T₁: 15 tons of BC + FYM
 T₂: 15 tons of BC + VC
 T₃: 15 tons of BC + GJ
 T₄: 20 tons of BC + FYM
 T₅: 20 tons of BC + VC
 T₆: 20 tons of BC + GJ
 T₇: 25 tons of BC + FYM
 T₈: 25 tons of BC + VC
 T₉: 25 tons of BC + GJ
 T₁₀: 15 tons of BC + FYM+ VC
 T₁₁: 15 tons of BC + FYM+ GJ
 T₁₂: 15 tons of BC + VC + GJ
 T₁₃: 20 tons of BC + FYM + VC
 T₁₄: 20 tons of BC + FYM + GJ
 T₁₅: 20 tons of BC + VC + GJ
 T₁₆: 25 tons of BC + FYM + VC
 T₁₇: 25 tons of BC + FYM + GJ
 T₁₈: 25 tons of BC + VC + GJ
 T₁₉: 25 tons of BC + FYM + VC + GJ
 T₂₀: 25 tons of BC + FYM + VC + GJ
 T₂₁: 25 tons of BC + FYM + VC + GJ
 T₂₂: 15 tons of BC
 T₂₃: 20 tons of BC
 T₂₄: 25 tons of BC
 VC-Vermicompost, BC-Biochar, GJ-Ganajeevamrutha.

Experimental Results

Fig 1: Effect of paddy straw biochar on Field bean

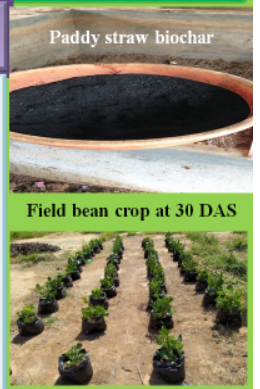
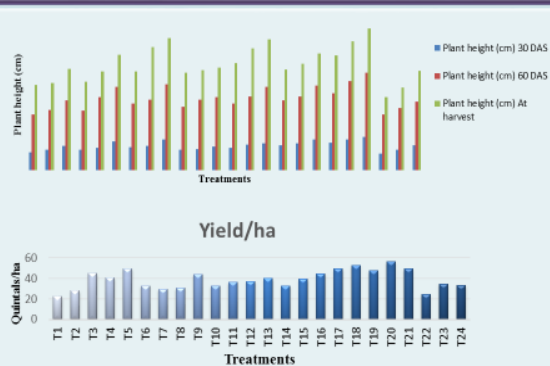


Table 1: Influence of biochar on soil properties after harvest of Field bean

| Treatment details | pH | EC (dSm ⁻¹) | OC (g kg ⁻¹) | N (kg ha ⁻¹) | P ₂ O ₅ (kg ha ⁻¹) | K ₂ O (kg ha ⁻¹) |
|-------------------|------|-------------------------|--------------------------|--------------------------|--|---|
| T ₁ | 7.20 | 0.10 | 5.48 | 265.50 | 53.00 | 159.00 |
| T ₂ | 7.97 | 0.11 | 5.46 | 269.00 | 55.50 | 159.00 |
| T ₃ | 8.02 | 0.11 | 5.40 | 270.50 | 56.75 | 160.70 |
| T ₄ | 7.52 | 0.12 | 5.60 | 271.00 | 56.11 | 159.00 |
| T ₅ | 7.33 | 0.12 | 5.66 | 273.00 | 58.60 | 163.50 |
| T ₆ | 7.85 | 0.13 | 5.72 | 273.00 | 60.13 | 166.00 |
| T ₇ | 7.24 | 0.14 | 5.86 | 274.00 | 59.11 | 165.80 |
| T ₈ | 7.44 | 0.16 | 5.98 | 277.50 | 60.91 | 167.90 |
| T ₉ | 8.20 | 0.17 | 6.11 | 278.50 | 61.60 | 167.70 |
| T ₁₀ | 7.47 | 0.12 | 5.63 | 266.00 | 55.00 | 169.10 |
| T ₁₁ | 7.77 | 0.11 | 5.46 | 265.00 | 56.25 | 167.50 |
| T ₁₂ | 7.92 | 0.11 | 5.45 | 268.50 | 54.25 | 169.00 |
| T ₁₃ | 7.53 | 0.14 | 5.87 | 272.00 | 58.50 | 168.10 |
| T ₁₄ | 7.30 | 0.14 | 5.83 | 272.50 | 59.35 | 169.50 |
| T ₁₅ | 7.26 | 0.16 | 6.15 | 271.50 | 59.95 | 170.50 |
| T ₁₆ | 7.52 | 0.16 | 6.12 | 274.50 | 59.00 | 168.30 |
| T ₁₇ | 7.41 | 0.15 | 5.88 | 277.00 | 60.25 | 170.10 |
| T ₁₈ | 7.36 | 0.16 | 6.18 | 279.50 | 61.60 | 169.50 |
| T ₁₉ | 7.60 | 0.16 | 6.13 | 286.00 | 60.10 | 169.80 |
| T ₂₀ | 8.98 | 0.18 | 6.20 | 291.50 | 63.10 | 172.90 |
| T ₂₁ | 9.50 | 0.19 | 6.42 | 297.00 | 63.25 | 173.90 |
| T ₂₂ | 6.76 | 0.10 | 5.47 | 255.50 | 48.85 | 158.00 |
| T ₂₃ | 7.65 | 0.12 | 5.72 | 258.00 | 52.45 | 159.20 |
| T ₂₄ | 8.03 | 0.16 | 6.11 | 261.50 | 53.15 | 158.50 |
| S.E.m± | 0.01 | 0.066 | 0.03 | 2.36 | 0.94 | 1.24 |
| C.D @ 5% | 0.02 | 0.193 | 0.09 | 6.89 | 2.75 | 3.64 |

Summary

- ✓ Application of biochar along with other organic inputs helps in improving nutrient availability in soil hence improves crop growth.
- ✓ Compared to all the treatments T₂₁ which received 25 tons of biochar + FYM + Vermicompost + Ganajeevamrutha showed best result followed by T₂₀ which received 20 tons of biochar + FYM + Vermicompost + Ganajeevamrutha.

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Discussion

➤ Increase in plant available N, P, K, total Organic carbon and soil pH is a function of biochar addition and in turn helps in increasing the crop yield. Similar results were observed by Pandit *et al.*, (2018).

➤ Increase in the dosage of biochar showed positive impact on growth and yield parameters of field bean and in turn it helps in improving the physical and chemical properties of soil. Similar observation were recorded by Karer *et al.*, (2013).

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LIST OF ABBREVIATIONS

| Abbreviation | Full form | Abbreviation | Full form |
|--------------------------------|--------------------------------------|---|------------------------------------|
| % | Per cent | KCl | Potassium chloride |
| @ | At the rate of | DAS | Days after sowing |
| °C | Degrees Celsius | KMnO ₄ | Potassium permanganate |
| ha | hectare (s) | Mg | Magnesium |
| BD | Bulk Density | kg | Kilogram |
| CaCl ₂ | Calcium chloride | Mg m ⁻³ | Mega gram per cubic meter |
| CaCO ₃ | Calcium carbonate | ml | Millilitre |
| CD | Critical difference | mm | Millimetre |
| cm | Centimetre | Mn | Manganese |
| C mol (p+) kg ⁻¹ | Centimole per kilogram | NaOH | Sodium hydroxide |
| Cu | Copper | ZnSO ₄ | Zinc sulphate |
| dS m ⁻¹ | Decisiemen per meter | NPK | Nitrogen, phosphorus and potassium |
| kg ha ⁻¹ | Kilogram per hectare | NS | Non-significant |
| DTPA | Diethylenetriamine Penta acetic acid | OC | Organic carbon |
| HCl | Hydrochloric acid | RCBD | Randomized complete blockdesign |
| EBT | Erichrome black – T | P ₂ O ₅ | Phosphorus pentoxide |
| <i>et al.,</i> | and other people | PD | Particle density |
| EC | Electrical conductivity | Ppm | parts per million |
| <i>etc.,</i> | and so on | SOC | Soil organic carbon |
| Fig. | Figure | FYM | Farm yard manure |
| g | Gram | K ₂ Cr ₂ O ₇ | Potassium dichromate |
| H ₂ SO ₄ | Sulphuric acid | K ₂ O | Potassium oxide |

I INTRODUCTION

Every year approximately 92 metric tons of crop waste is burned in India, causing excessive particulate matter emission and air pollution. Burning of Crop residue is a major environmental problem responsible health issues and leads to Global warming. Harvesting of crops generates large volume of residue both on and off farm. Crop residues is not a waste but rather a useful natural resource. Frequent residue burning of paddy straw leads to complete loss of microbial population and causes environmental pollution. Smoke from this burning leads to a cloud of particulates and produce toxic cloud resulting in severe air pollution. Including emission of greenhouse gases that contribute to global warming increase levels of particulate matter (PM) and smog cause health hazard, loss of biodiversity of agricultural lands and deterioration of soil fertility. During October month this problem with smog and air pollution are witnessed in Delhi due to straw burning in northern parts of India especially in Punjab and Haryana.

Crop residues are India's major source of biomass, with approximately 69.7 million tonnes of residue produced annually by 6 major crops in India (NPMCR 2014); So, the conversion of crop waste into biochar under changing climate helps to offset CO₂ emission and increase the carbon sequestration in the soil. One way to reduce this is by pyrolysis / thermal depolymerisation which is the decomposition or cracking of materials in absence of oxygen. There are two types of pyrolysis, one is slow pyrolysis and the other fast pyrolysis. Pyrolysis yields Biochar, Biooil and Syngas. The present work will look into the potential utility of the above via slow pyrolysis.

Biochar is a carbon rich substance obtained from thermal decomposition of plant biomass through the process of pyrolysis. The special characteristics of the biochar are its effectiveness in retaining most of the nutrients and maintaining them available to plants rather than other organic substances such as natural leaf litter, compost or manures. Biochar typically has a powerful aromatic structure and is therefore biochemically more recalcitrant than any other type of organic matter in soils and may last in the soil for thousands of years. Therefore, it's feasible to use biochar as a part of change to organic farming methods to transform agriculture from a net carbon emitter to a method to restore

carbon stocks in the soil. Further, it also helps in improving the productivity of soil resulting in higher yields. This technology is also a measure for controlling Greenhouse Gas emission and ways to effectively enhance natural rates of carbon sequestration in the soil.

Biochar, which is produced by the thermo-chemical decomposition of organic material (crop residues and wood waste) under partial or full exclusion of oxygen (pyrolysis) at a relatively higher temperature range of 450°C to 600°C and it has been the subject of researchers for many years. Converting inexpensive, available crop straws into biochar and adding them to soils can have substantial agricultural and environmental benefits (Wang *et al.*, 2013). There is sufficient evidence that due to its particular structure and composition, biochar land use may theoretically increase carbon sequestration, boosts soil health and crop productivity by contributing to sustainable management of organic residues (Lehmann and Joseph, 2009).

Application of biochar improved the soil health, increase the carbon capture and storage, reduce the greenhouse gas emission and enhance the crop yield with sustained soil health, which enables to meet the food grain needs of the ever growing population (Kannan *et al.*, 2013). Recently, biochar applications have gained momentum with features such as: carbon sequestration, enhancing nutrient uptake from the soil, reducing soil compaction, improves soil physical condition, and helps to reduce nitrous oxide emission (Lehmann and Rondon, 2005; Lehmann, 2007).

Biochar is extremely porous, typically alkaline, and has a wide specific surface area (Glaser *et al.*, 2002; Downie *et al.*, 2009). Because of these intrinsic chemical and physical properties, biochar can theoretically affect a variety of soil characteristics, including soil pH, porosity, bulk density and ability to retain water (Glaser *et al.*, 2002; Chan *et al.*, 2007). In addition, biochar absorbs ions from soil solution by a mixture of electrostatic, complexing and capillary pressures on their surfaces and in pores (Major *et al.*, 2010). This biochar properties can significantly decrease soil nutrient leaching (Lehmann *et al.*, 2003) as well as the accessibility of ions to soil microorganisms. Thus preliminary research by scientists and environmentalists in various parts of the world

shows the beneficial impact of biochar in agriculture to increase the production of different crops.

The use of biochar in agricultural systems is a viable option that can improve the natural rate of carbon sequestration in the soil, reduce agricultural waste and improve soil quality. Initial results of the application of biochar suggest that it improves soil health and crop yields. Biochar has a lasting impact on soil physical properties, availability of nutrients, soil microbial activity, carbon sequestration capacity, crop yield/production and mitigation of Greenhouse gas. It is one of the abundant sources for the production of paddy straw biochar from farm waste that can be used locally, and its application to the soil has a great impact on the physical, chemical and biological properties of the soil. Rice straw biochar produced at 500 °C can be used locally at the same price as Farmyard Manure (FYM), and their role in plant growth, yield, and maintaining soil health are immeasurable compared to FYM. Thus keeping this in view, to evaluate the performance of Field bean which is grown throughout India for its green pods as vegetable and dry seeds as pulse as influenced by paddy straw biochar and soil properties, the present research entitled **“Utilization of products obtained from pyrolysis of paddy straw.”** was conducted with following objectives.

Objectives of research work

- 1) To study the characterization of paddy biochar.
- 2) To evaluate the performance of paddy biochar application on Field bean.
- 3) To study the characterization of paddy bio oil.

II REVIEW OF LITERATURE

The present literatures are gone through as a basic structure for the further studies on Utilization of products obtained from pyrolysis of paddy straw. The review related to the present investigation is given in the following headings:

2.1 To study the characterization of paddy biochar.

2.2 To evaluate the performance of paddy biochar application on Field bean.

2.3 To study the characterization of paddy bio oil.

2.1 To study the characterization of paddy biochar.

Schmidt and Noack (2000), researched biochar as a carbonaceous substance that includes a number of certain specific classes of polycyclic aromatic hydrocarbons.

Antal *et al.* (2003) reported that low temperatures of pyrolysis that produce torrefied-like biochars containing semi-degraded cellulose and hemicellulose compounds because 300 to 400 °C is the critical temperature range for their structural breakdown. Compared to 500 °C, biochars that were formed at 250 °C potentially include more stable classes of carboxylic acid.

It was found that specific biochar materials can decompose to differing degrees, based on the physical and chemical properties of a given biochar as reported by Hamer *et al.* (2004). Researchers also observed that two biochars formed at different temperatures released 0.3 and 0.8 per cent of initial C as CO₂.

Brown *et al.* (2006) recorded that at lowest (300 °C) and maximum (> 1000 °C) pyrolysis temperature, lower surface area (~10 m² g⁻¹) of biochars was formed. Although higher surface area (~400 m² g⁻¹) was reported at 650 to 850 °C as intermediate pyrolysis temperature (Lua and Yang, 2004).

Gaskin *et al.* (2008) observed that the overall feedstock nutrients extractable and Mehlich I is concentrated after pyrolysis in biochar. Some biochar may act as a fertilizer as well as a carbon sequester depending on the feedstock.

Biochar and charcoal have a low volume density of about 300 kg m^{-3} compared to a normal density of 1300 kg m^{-3} in soil. Particles may be of rather small quality, and structures may be hydrophobic in comparison Deluca and Aplet (2008). A biochar's porosity and surface area have a greater effect of surface binding on the nutrient retention ability.

Novak *et al.* (2009) reported that higher pyrolytic temperature ($>400^\circ\text{C}$) was observed to produce biochars with alkaline pH.

Downie *et al.* (2009) observed that the biochar porosity, which defines its surface region, has an extremely variable pore size range that includes nano-(< 0.9 nm), micro-(< 2 nm) to macro-pores (> 50 nm) and biochar has lower bulk density than mineral soil with organic soil values.

Lei *et al.* (2009) these biochar properties are influenced by pyrolytic parameters and feedstock type.

Study on physical and chemical properties of different forms of biochar from peanut hulls, peacan shells, poultry litter, and switch grass were carried out. Novak *et al.* (2009) found higher pyrolysis temperatures resulted in lower biochar mass recovery, greater surface areas, elevated pH and higher ash contents.

Rodriguez *et al.* (2009) noted that, in a maize trial in Colombia, the biochar produced from sugarcane bagasse increased soil pH from 4.0–4.5 to 6.0–6.5. Biochar is found predominantly in soil organic matter (SOM) fractions that reside in small clusters of soil particles, or soil aggregates, rather than as free organic matter.

Steinbeiss *et al.* (2009) stated that biochars with low temperatures were found to have a less compact C structure and are predicted to have greater reactivity in soil than biochars with higher temperatures and to contribute more to soil fertility.

Singh *et al.* (2010) proposed that wood biochars had higher total C, lower ash output, lower total N, P, K, S, Ca, Mg, Al, Na and Cu components, and lower possible

cation exchange capacity (CEC) and exchangeable cations than manure-based biochars, and that leaf biochars were usually intermediate. Higher pyrolysis temperature raised the basicity of the biochar, pH of the substrate and lowered acidity.

The fined biochars are melted easily and rapidly than pelleted. These changes will affect the ability of biochar mixing into soils, also some significant physical parameters such as specific surface area, surface charge, particle size distribution, porosity, bulk density and specific gravity, water-holding capacity, and hydraulic conductivity as reported by Zimmerman (2010).

As investigated by Wu *et al.* (2012) pyrolysis temperature had a greater influence than residence time on the chemical composition and structure of rice straw-derived biochar produced at low heating rate. The rice straw-derived biochar especially produced at 400 °C had high alkalinity and cation exchange capacity, and high levels of available phosphorus and extractable cations.

Ippolito *et al.* (2012) Biochar is characterized with its lower specific surface area, higher water-holding capacity, lower pH, more carboxyl and phenol groups, and higher cation exchange capacity (CEC), when pyrolysis happens at low temperature.

Wang *et al.* (2013) observed that rising biochar pyrolysis temperature from 500 to 700 °C enhanced ash volume, surface area, pH, total P and Ca content and decreased biochar output, CEC, total acid and nitrogen content. Excessive residence period from 4 to 8 or 16 hours, increased the BET surface area and biochar ash amount but decreased biochar yield. Study of Fourier-transform infrared spectroscopy (FTIR) found that, at a higher temperature, more recalcitrant and aromatic compounds were developed in the biochar.

Analysis carried out by Jindo *et al.* (2014) revealed the chemical and physical properties, which are strongly related to the form of the initial substance used and the conditions of pyrolysis, and the results showed that the biochar obtained at 600 °C has a high recalcitrant character while that obtained at 400 °C maintains flexible and easily labile compounds. Owing to the inclusion of silica elements into its chemical structure,

the biochar derived from rice materials (rice husk and rice straw) exhibited high yield and special chemical properties. The biochar obtained from wood products was lower in carbon content and high in absorption.

Wan *et al.* (2014) showed that pH, carbonate content, base cation, and biochar alkalinity increased with temperature rise in pyrolysis. Biochar's elevated pH was due to hydrolysis of the Ca, Mg, and K salts.

In a study estimated the bulk density values of biochars produced from different feedstock and it ranged from 0.25 to 0.6 g cm⁻³ Brewer *et al.* (2014). Low-temperature pyrolysis produced high biochar yields; in contrast, high-temperature pyrolysis led to biochar with a high C content, large surface area, and high adsorption characteristics as reported by Jindo *et al.* (2014).

Batool *et al.* (2015) showed that in stressed plants, stomatal conductance and transpiration decreased and nearly reached zero. On the other side, comparison with untreated plants, water use efficiency (WUE) has increased in stressed plants that produce biochar and gypsum. Biochar alone is a better strategy for fostering plant growth and WUE in particular.

Rumi *et al.* (2015) reported that biochar is a carbon rich material contains 50 per cent of its original carbon, which is highly recalcitrant nature, which are produced during pyrolysis process. Along with the biochar some amount of bio-oil and gases are also produced.

Figueredo *et al.* (2017) discovered that the biochar, a material of organic waste pyrolysis, can be used as a soil conditioner and also as a solid waste management option. The results showed the existence of C, H, N, O, ashes, macro and micronutrients, mineral process characterization, and functional groups. Relationships between O/C and H/C decreased with rising pyrolysis temperature, which determines greater biochar C stability.

2.2 To evaluate the performance of paddy Biochar application on crops.

Lehmann and Rondon (2005) showed that depending on the amount of biochar added, significant improvements in plant productivity were achieved ranging from 20 to 220 per cent. Increase the yields of maize and peanut due to changes in P availability through increased root colonization of arbuscular mycorrhizal fungi was recorded by Yamato *et al.* (2006).

Cumulative yield increase of rice and sorghum on a Brazilian Oxisol after three repeated biochar applications at the rate 7 t ha⁻¹ for two years were reported by Steiner *et al.* (2007). The progressive increase in beneficial effects of biochar overtime was due to increased NPK availability in soil.

Asai *et al.* (2009) conducted a study to know the effect of rate of biochar application on rice with varied level of N and P at ten different sites with two cultivars. Biochar application resulted in higher grain yields at the sites where P availability was low. Whereas in the absence of N fertilizer and at the highest rate of biochar application, there was reduction in leaf chlorophyll content indicating biochar application is highly dependent on soil fertility and fertilizer management.

Van Zwiterten *et al.* (2010) reported that plant biomass increased significantly with increasing biochar addition when applied in combination with lower N application rates. Application of biochar at the highest rate resulted in significantly more NO₃-N in the soil. Application of biochar also increased microbial activity and available P in the soil.

Feng *et al.* (2014) reviewed a 2009 long-term field trial to assess the impact of biochar on crop yield of beans and soil properties in an alkaline environment. Five treatments were incorporation of straw, incorporation of straw with inorganic fertilizer and incorporation of straw with inorganic fertilizer, and biochar 30, 60, and 90 t ha⁻¹, respectively. The study showed significantly increase in average yield over the first four growing seasons. The findings indicated that biochar could be used without yield degradation or major impacts on resource supply in alkaline soils.

A field experiment conducted by Coumaravel *et al.* (2015) with hybrid maize, the result revealed that application of biochar at 10 t ha⁻¹ along with the recommended dose of fertilizer + FYM 12 t ha⁻¹ and Azospirillum at 2 kg ha⁻¹ significantly increased yield and nutrient uptake.

Alie Kamara *et al.* (2015) observed that the use of biochar increased the supply of phosphorus, exchangeable cations and CEC when opposed to soil management without biochar. Plant height, number of tillers, and dry biomass weight of rice varieties grown in rice straw biochar adjusted soils were substantially higher than those of untreated soils. Biomass yield and tiller figures represented the most significant improvement in plant growth characteristics as a part of biochar introduction to soil.

Haider *et al.* (2016) reported that aged biochar was more effective than fresh biochar for nutrient capture and delivery, a property which may lead to increased crop yield over time.

The six biochars differed in pH, ash quality, equivalent Calcium carbonate (CaCO₃), total base cations, CEC, and other properties (pore scale, functional surface groups). Desmodium's gross dry weights is raised 2 to 4-fold over the control or lime treatment after biochar applications. Ultimately, Arnoldus and Nguyen (2016), concluded that comparable CaCO₃ and complete simple cations were among the biochar's most significant properties responsible for improving the productivity of acid soils and plant productivity.

Kalyani *et al.* (2016) reported increased plant height, number of leaves and yield of the bean, fenugreek and mint with biochar in the combination of organic manure than biochar alone or compost alone.

A field trial at Alasulige village Sakleshpur taluk, district of Hassan, Karnataka conducted by Mamatha *et al.* (2018) to use the simple slag as a soil modification and nutrient source in acid soil relative to lime. Results of the experiment showed that significantly higher yields of paddy grain (50.32 q ha⁻¹) and straw (65.93 q ha⁻¹) were

observed in treatment T₉ (100% RDF + simple slag at 2 t ha⁻¹) followed by T₁₀ (100% RDF + lime at 2 t ha⁻¹).

2.3 To study the characterization of Paddy Bio oil

Mokhlisse, and Chanaa. (1999) experimental results show that the bio-oil yield of steam pyrolysis is significantly higher than that of normal and nitrogen pyrolysis. Bio-oil yield increases with increasing steam velocity. This result is consistent with the theory that for higher velocities secondary reactions such as char formation are prevented.

Ozbay *et al.* (2001) reported the kinetics of pyrolysis reactions shows that the primary pyrolysis reactions takes place in the temperature range of 200–400°C which results in formation of solid product, char. Temperatures above 400°C, solid residue slowly undergoes further chemical and physical transformations which leads to the formation of liquid and gas products. Pyrolysis temperatures up to 600°C maximizes the production of bio-oils and temperatures above 700°C maximizes gaseous products while minimizing char formation.

Pütin *et al.* (2004) experimented that Bio-oil yield was 33.41% when steam velocity was 1.3 cm/s and this value increased to 35.86% with the velocity of 2.7 cm/s. The opposite was observed for the solid product; char yield decreased from 17.05% to 14.82% when steam velocity was decreased.

Tsai *et al.* (2007) reported that temperatures above 550°C are favorable in the successive aromatization reactions, such as the cracking of the light organics, oligomerization, cyclization, hydrogen or hydride transfer, and aromatization, and Bio-oil produced by fast pyrolysis of cellulosic biomass is an emulsion of water (~20% w/w) and a wide range of oxygenated organic compounds including organic acids, aldehydes, alcohols, phenols, carbohydrates, and lignin-derived oligomers.

Goyal *et al.* (2008) stated that product yields from slow pyrolysis are approximately 35% biochar, 30% bio-oil, and 35% syngas by mass.

Fahmi *et al.* (2008) observed that Cellulosic feedstocks, bio-oil yield decreases with increasing ash content and the average molecular weight of the bio-oil fraction increases with lignin content of the feedstock.

Laird *et al.* (2009) observed that Pyrolysis of biomass, acids are partitioned into the bio-oil fraction while bases – primarily K_2O , CaO , and MgO – are partitioned into the biochar fraction and lighter aromatic molecules, including most single ring and smaller polycyclic aromatic hydrocarbons (PAHs) are distilled into the bio-oil fraction during pyrolysis.

Zhang *et al.* (2011) reported that four distinct conditions required for the pyrolysis processes: Medium temperatures, High heating rates, Short vapor residence times and fast condensation. CC bonds of the biomass functional groups broken to create bio-oil, and biochar, and residual non-condensable compounds with C-O bonds.

Report by Huang *et al.* (2012) found that biomass rich in cellulose or hemicellulose contains higher olefins yield in the bio-oil produced than the feedstocks with higher lignin content.

Choi *et al.* (2012) stated that in association with catalyst loading, pyrolysis variant conditions also influence the yield and composition of the pyrolysate. The conditions (operating parameters) are tempered to direct the pyrolysis reaction to the desired quality and quantity of bio-oil. The typical operating parameters for fixed, fluidized bed, batch and auger pyrolyzers can include; Biomass loading, Biomass particle size, Nitrogen flow rate, Pyrolysis temperature and Time.

Shi *et al.* (2012) evaluated the influence of the acid washed straw forms slightly more bio-oil compared to gaseous products than untreated straw, over a temperature range of 300–700°C. Similarly, in a temperature range of 500–700°C, the untreated straw yields more char than water for the acid treated straw. The visible differences in the products distribution between the treated and untreated straw indicates the influence the alkali and alkali earth metals had on the rice straw pyrolysis.

Shun *et al.* (2013) observed the catalysts upgrades the quality of the pyrolysis oil through cracking, deoxygenation, oligomerization, cyclization, aromatization, alkylation, isomerization and polymerization. The catalytic upgrade mainly increases the heating value (energy density) of the bio-oil. Catalysts tailor thermally driven fast pyrolysis reactions to yield high quality and more stable bio-oil.

Jacobson *et al.* (2013) observed that the bio-oil contains complex heterogeneous mixtures of aromatics and aliphatic oxygenated compounds as Phenolics, Carbohydrates, Alcohol, Aldehydes, Ketones and Carboxylic acids.

Jacobson *et al.* (2013) explained the pretreated biomass has higher lignin content than the virgin biomass, more phenolic compounds are produced upon catalytic pyrolysis over the less acidic silicate catalysts. Remarkably, pore blockade by bio-oil macromolecules and active site deactivation are avoided by the use of aluminosilicate during bio-oil upgrading process.

The catalysts significantly alter the composition of the non-catalytic pyrolysis oil by eliminating acids and reducing phenol as reported by Zhang *et al.* (2014). Also, reduction of linear aldehydes and anhydro sugars (mainly laevoglucose) and the formation of cycloalkanes, aromatics and other light hydrocarbon compounds substantially change the bio-oil chemical composition. The capabilities of the catalysts confirm to upgrade the fuel properties of bio-oil and enhance the formation of some valuable chemicals.

A study by Naqvi *et al.* (2014) on both catalytic and non-catalytic pyrolysis of paddy husk biomass in a drop type fixed bed pyrolyzer. They studied the effect of temperature (300 – 600 °C) and catalyst loading on the pyrolysis of the biomass in a drop type fixed bed pyrolyzer. Thermal pyrolysis oil yield increased from 350 to 450 °C, and maximum yield was obtained at 450 °C. However, the paddy husk is catalytically pyrolysis at 450 °C and varying catalyst loading ratio from 0.5 to 2. A catalyst loading of 0.5 favors highest de-oxygenation rate and the catalytic pyrolysis oil yield was highest.

Shen *et al.* (2015) explained that individual biomass macromolecules (cellulose, hemicellulose and lignin) pyrolyzed under varying range of temperatures contribute the essential functionalities to the bio-oil. Cellulose decomposes to laevoglucose, and then fragmented to linear compounds as furfural, acetone, alcohols and ketones.

Veses *et al.* (2015) reported that the catalyst loading critically influences bio-oil yield; higher loading ratio discourages bio-oil yield but increases the biochar and gas production. Small catalyst dosage is often recommended because it led to a high yield of bio-oil, rich in phenol derived compounds and less in low molecular weight oxygenates.

III MATERIAL AND METHODS

The research entitled “Utilization of products obtained from pyrolysis of paddy straw” was undertaken during 2020-2021. In this study, a potted experiment was conducted in the GKVK at Bangalore. The details of the material used and methodologies adopted during the investigation are described in this chapter.

3.1 General description of the product

3.1.1 Biochar

Biochar prepared from paddy straw was used. The paddy straw biochar was produced at 500°C was used in the present experiment and burnt material was ground to pass through 0.2 mm sieve.

Characterization of biochar

The biochar used in this present study was characterized for its various physicochemical properties such as bulk density, water holding capacity, pH, EC, and total elemental composition by adopting standard analytical procedures. The analytical procedures employed for analysis of different parameters are given in Table 3.1.

3.1.2 Ganajeevamrutha

10 kg of desi cow dung was mixed with 500 g of pulse flour, 500 ml desi cow urine, 100 g jaggery and a hand full of soil and shade dried for 3 days and then powdered and used for the experiment.

3.2 Weather during crop growth period

During crop growth period, a total rainfall of 358.2 mm was received from September to December. Maximum temperature ranged from 27.2°C to 34.5°C and minimum temperature ranged from 19.0°C to 20.5°C.

Table 3.1: Methods employed for the analysis of biochar

| Sl. No. | Parameter | Method | Reference |
|----------------|-------------------------|--|----------------------------|
| 1. | pH | Potentiometry | Jackson (1973) |
| 2. | EC (dSm ⁻¹) | Conductometry | Jackson (1973) |
| 3. | MWHC (%) | Keen's cup method | Pipper (1966) |
| 4. | Bulk density (g/cc) | Keen's cup method | Pipper (1966) |
| 5. | Total carbon (%) | Dry combustion method (CHNS, LECO) | Page <i>et al.</i> 1982 |
| 6. | Nitrogen (%) | Kjeldahl digestion & distillation method | Jackson (1973) |
| 7. | Phosphorus (%) | Diacid digestion & vanadomolybdate method | Jackson (1973) |
| 8. | Potassium (%) | Diacid digestion & vanadomolybdate method | Jackson (1973) |
| 9. | Calcium (%) | Complexometric titration method | Jackson (1973) |
| 10. | Magnesium (%) | Complexometric titration method | Jackson (1973) |
| 11. | Sulphur (ppm) | 0.15% CaCl extraction & Turbidity | Black (1965) |
| 12. | Iron (ppm) | Diacid digestion & Atomic Absorption Spectrophotometry | Lindsay and Norwell (1978) |
| 13. | Copper (ppm) | | |
| 14. | Manganese (ppm) | | |
| 15. | Zinc (ppm) | | |

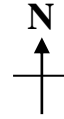
3.3 Design and layout of the experiment

The experiment was laid out in a Completely Randomized Design (CRD) having twenty four treatment combinations and replicated twice on a net packet. The plan of layout of the experiment is shown in Fig 1.

3.3.1 Experimental details

Table 3.2: Experimental details

| | |
|---------------------|--|
| Crop | Field bean |
| Variety | Hebbal avare |
| Season | Rabi |
| Polythene bag size | 18 x 18 inches |
| Date of sowing | 18/12/2020 |
| Date of harvest | 1/4/2021 |
| Source of nutrients | Paddy straw Biochar, FYM, Vermi compost, Ganajeevamrutha |



| R₁ | R₂ | R₁ | R₂ |
|-----------------------|-----------------------|-----------------------|-----------------------|
| T₁₂ | T₈ | T₁₆ | T₆ |
| T₂₀ | T₁ | T₂₃ | T₃ |
| T₅ | T₁₀ | T₆ | T₂₄ |
| T₁₁ | T₂ | T₁₀ | T₁₆ |
| T₁ | T₅ | T₂₃ | T₉ |
| T₁₃ | T₁₉ | T₃ | T₂ |
| T₄ | T₁₇ | T₁₅ | T₁₄ |
| T₉ | T₁₂ | T₁₇ | T₄ |
| T₁₄ | T₇ | T₂₁ | T₁₈ |
| T₂₂ | T₂₀ | T₂₄ | T₂₁ |
| T₇ | T₁₅ | T₈ | T₁₁ |
| T₁₈ | T₁₃ | T₁₉ | T₂₂ |

Fig. 3.1: Arrangement of packets



PADDY STAW BIOCHAR



BIO OIL

Plate 1: Products obtained from pyrolysis



PACKET FILLING



SOWING OF FIELD BEAN

Plate 2: A view of Packet filling and sowing of Field bean

3.3.2 Treatment details

Table 3.3: Treatment details

| Treatments | Details |
|-------------------|---|
| T ₁ | 15 tons of Biochar + FYM |
| T ₂ | 15 tons of Biochar + VC |
| T ₃ | 15 tons of Biochar + Ganajeevamrutha |
| T ₄ | 20 tons of Biochar + FYM |
| T ₅ | 20 tons of Biochar + VC |
| T ₆ | 20 tons of Biochar + Ganajeevamrutha |
| T ₇ | 25 tons of Biochar + FYM |
| T ₈ | 25 tons of Biochar + VC |
| T ₉ | 25 tons of Biochar + Ganajeevamrutha |
| T ₁₀ | 15 tons of Biochar + FYM+ VC |
| T ₁₁ | 15 tons of Biochar + FYM+ Ganajeevamrutha |
| T ₁₂ | 15 tons of Biochar + VC + Ganajeevamrutha |
| T ₁₃ | 20 tons of Biochar + FYM +VC |
| T ₁₄ | 20 tons of Biochar + FYM + Ganajeevamrutha |
| T ₁₅ | 20 tons of Biochar + VC + Ganajeevamrutha |
| T ₁₆ | 25 tons of Biochar + FYM + VC |
| T ₁₇ | 25 tons of Biochar + FYM + Ganajeevamrutha |
| T ₁₈ | 25 tons of Biochar + VC+ Ganajeevamrutha |
| T ₁₉ | 15 tons of Biochar + FYM + VC + Ganajeevamrutha |
| T ₂₀ | 20 tons of Biochar + FYM + VC + Ganajeevamrutha |
| T ₂₁ | 25 tons of Biochar + FYM + VC + Ganajeevamrutha |
| T ₂₂ | 15tons of Biochar |
| T ₂₃ | 20tons of Biochar |
| T ₂₄ | 25tons of Biochar |

Table 3.3.1: Treatment details

| Treatments | Details | Quantity of inputs added per packet (g) | | | |
|-----------------|---|---|------|-------|-----------------|
| | | Biochar | FYM | VC | Ganajeevamrutha |
| T ₁ | 15 tons of Biochar + FYM | 42.75 | 34.2 | - | - |
| T ₂ | 15 tons of Biochar + VC | 42.75 | - | 10.26 | - |
| T ₃ | 15 tons of Biochar + Ganajeevamrutha | 42.75 | - | - | 2.85 |
| T ₄ | 20 tons of Biochar + FYM | 57 | 34.2 | - | - |
| T ₅ | 20 tons of Biochar + VC | 57 | - | 10.26 | - |
| T ₆ | 20 tons of Biochar + Ganajeevamrutha | 57 | - | - | 2.85 |
| T ₇ | 25 tons of Biochar + FYM | 71.25 | 34.2 | - | - |
| T ₈ | 25 tons of Biochar + VC | 71.25 | - | 10.26 | - |
| T ₉ | 25 tons of Biochar + Ganajeevamrutha | 71.25 | - | - | 2.85 |
| T ₁₀ | 15 tons of Biochar + FYM+ VC | 42.75 | 34.2 | 10.26 | - |
| T ₁₁ | 15 tons of Biochar + FYM+ Ganajeevamrutha | 42.75 | 34.2 | - | 2.85 |
| T ₁₂ | 15 tons of Biochar + VC + Ganajeevamrutha | 42.75 | - | 10.26 | 2.85 |
| T ₁₃ | 20 tons of Biochar + FYM +VC | 57 | 34.2 | 10.26 | - |
| T ₁₄ | 20 tons of Biochar + FYM + Ganajeevamrutha | 57 | 34.2 | - | 2.85 |
| T ₁₅ | 20 tons of Biochar + VC + Ganajeevamrutha | 57 | - | 10.26 | 2.85 |
| T ₁₆ | 25 tons of Biochar + FYM + VC | 71.25 | 34.2 | 10.26 | - |
| T ₁₇ | 25 tons of Biochar + FYM + Ganajeevamrutha | 71.25 | 34.2 | - | 2.85 |
| T ₁₈ | 25 tons of Biochar + VC+ Ganajeevamrutha | 71.25 | - | 10.26 | 2.85 |
| T ₁₉ | 15 tons of Biochar + FYM + VC + Ganajeevamrutha | 42.75 | 34.2 | 10.26 | 2.85 |
| T ₂₀ | 20 tons of Biochar + FYM + VC + Ganajeevamrutha | 57 | 34.2 | 10.26 | 2.85 |
| T ₂₁ | 25 tons of Biochar + FYM + VC + Ganajeevamrutha | 71.25 | 34.2 | 10.26 | 2.85 |
| T ₂₂ | 15tons of Biochar | 42.75 | - | - | - |
| T ₂₃ | 20tons of Biochar | 57 | - | - | - |
| T ₂₄ | 25tons of Biochar | 71.25 | - | - | - |

* Each packet consisted of 10kg each consisting of 6kgs of soil added initially and to the remaining 4kgs of soil, desired quantity of the inputs as per the treatments have been added finally achieving a total of 4kgs (soil + inputs (depending on the treatment the following were added : Biochar /FYM / VC/ Ganajeevamrutha).



30 DAYS AFTER SOWING OF FIELD BEAN



60 DAYS AFTER SOWING OF FIELD BEAN

Plate 3: General view of experimental site at different growth stages of Field bean

Table 3.4: Initial physico-chemical properties of the soil

| PARAMETERS | VALUES |
|---|------------|
| Sand | 62.73 |
| Silt | 20.41 |
| Clay | 16.86 |
| Textural class | Sandy loam |
| Bulk density (g/cc ⁻¹) | 1.41 |
| Maximum water holding capacity (%) | 34.1 |
| Soil pH | 7.2 |
| Electrical conductivity (dS m ⁻¹) | 0.22 |
| Organic carbon (%) | 0.72 |
| Available Nitrogen (kg ha ⁻¹) | 263.54 |
| Available Phosphorus (kg ha ⁻¹) | 46.37 |
| Available Potassium (kg ha ⁻¹) | 196 |
| Available Sulphur (ppm) | 31.98 |
| Exchangeable Calcium (c mol (p+) kg ⁻¹) | 2.56 |
| Exchangeable Magnesium (c mol (p+) kg ⁻¹) | 1.2 |
| DTPA Zn (ppm) | 0.43 |
| DTPA Fe (ppm) | 18.9 |
| Cu (ppm) | 0.21 |
| Mn (ppm) | 9.98 |
| B (ppm) | 0.45 |

Table 3.5: Methods employed for analysis of soil samples

| Sl. No | Parameter | Method | Reference |
|---------------------------------|---|---|----------------------------|
| <u>Physical analysis</u> | | | |
| 1. | Texture | International pipette method | Pipper (1966) |
| 2. | MWHC (%) | Keen's cup method | Pipper (1966) |
| 3. | Bluk density (g/cc) | Keen's cup method | Pipper (1966) |
| <u>Chemical analysis</u> | | | |
| 4. | pH | Potentiometry | Jackson (1973) |
| 5. | EC (dSm ⁻¹) | Conductometry | Jackson (1973) |
| 6. | Organic carbon (%) | Wet oxidation | Walkley & Black (1934) |
| 7. | Available Nitrogen (kg ha ⁻¹) | Kjeldahl digestion & distillation method | Jackson (1973) |
| 8. | Available Phosphorus (kg ha ⁻¹) | Diacid digestion & vanadomolybdate method | Jackson (1973) |
| 9. | Available Potassium (kg ha ⁻¹) | Diacid digestion & vanadomolybdate method | Jackson (1973) |
| 10. | Exchangeable Calcium (c mol (p+) kg ⁻¹) | Complexometric titration method | Jackson (1973) |
| 11. | Exchangeable Magnesium (c mol (p+) kg ⁻¹) | Complexometric titration method | Jackson (1973) |
| 12. | Available Sulphur (ppm) | 0.15% CaCl extraction & Turbidity | Black (1965) |
| 13. | DTPA Fe (ppm) | Atomic Absorption Spectrophotometry | Lindsay and Norwell (1978) |
| 14. | Cu (ppm) | | |
| 15. | Mn (ppm) | | |
| 16. | Zn (ppm) | | |

3.4 Effect of different levels of paddy straw biochar on growth and yield parameters of Field bean

3.4.1 Plant height

The plant height was measured at 30, 60 days after sowing and at harvest and the mean plant height was worked out and expressed in centimetre (cm).

3.4.2 Number of branches per plant of Field bean

The number of branches per plant was counted at 30, 60 days after sowing and at harvest stages and the average number of branches per plant was documented.

3.4.3 Number of leaves per plant of Field bean

The number of leaves per plant was counted at 30, 60 days after sowing and at harvest stages and the average number of leaves per plant was documented.

3.4.4 Number of pods per plant of Field bean

Number of pods per plant is counted.

3.4.5 Pod weight per plant of Field bean

Pods per plant were weighed and the data was documented.

3.5 Collection of soil samples

3.5.1 Collection of soil samples

Soil samples at a plough layer depth (0-15 cm depth) were obtained from each of the experimental twenty-four treatment pots after the crop's harvest. The samples obtained were dried in shade, rendered with a pestle and mortar to ground, passed through 2 mm sieve, and placed in polythene bags.

3.6 Preparation of plant samples

The plant samples obtained from the crop were dried powdered and examined for macronutrients (N, P, K, Ca, Mg, and S) and micronutrients (Fe, Zn, Mn and Cu).

3.6.1 Methods employed for plant sample analysis of Paddy straw

3.6.1.1 Nitrogen (%)

0.5 grams of plant samples were digested with conc. sulphuric acid (H_2SO_4) and digestion mixture (K_2SO_4 : $CuSO_4 \cdot 5H_2O$: selenium in the ratio of 100:20:1) until green residue was collected. The digested content was distilled by the Micro Kjeldhal. The ammonia produced was trapped in boric acid (H_3BO_4) and was then measured by titrating against standardized sulphuric acid (Pipper, 1966).

3.6.1.2 Digestion of plant samples for nutrients estimation

One gram of the dried and ground samples was pre-digested with 10 ml HNO_3 (62%) for 24 hours, then digested in a vacuumed chamber at $85^\circ C$ on sand bath with the following steps: the pre-digested samples were treated with 10 ml di-acid mixture reagent ($HNO_3 + HClO_4$ at a ratio of 10:4) and held on sand bath until white precipitate was left in the bottom. After filtration, the digested samples were diluted with distilled water and volume formed to a defined concentrations. Using normal techniques this extract was used to estimate P, K, Ca, Mg and micronutrients (Fe, Cu, Mn, and Zn).

3.6.1.3 Phosphorus (%)

The measurement of phosphorus content in plant by vanado molybdo phosphoric yellow colour method (Pipper, 1966) was rendered using a suitable aliquot of the extracted sample.

3.6.1.4 Potassium (%)

The plant samples were tested for potassium content by feeding the diacid digested samples to the flame photometer (Pipper, 1966) after diluting it to appropriate concentration.

3.6.1.5 Calcium and magnesium (%)

As described by Pipper (1966), the di-acid digested plant samples were tested for Calcium and Magnesium using versenate titration process. 3.6.1.6 Sulphur (%) content of

field bean In di-acid digested plant sample, sulphur was measured by using the process called turbidometry. The turbidity intensity produced in the sample was calculated at 420 nm wavelength using a spectrophotometer, as outlined by Pipper (1966).

3.6.1.7 Micronutrients (mg kg⁻¹) content of paddy straw

Micronutrients (Zn, Cu, Mn and Fe) in plants were measured by feeding the digested extract samples after being diluted to an appropriate concentration of the Perkin Elmer atomic absorption spectrophotometer using correct hollow cathode lamps and presented as mg kg⁻¹ in plant and seed sample (Jackson 1973).

3.7 Methodology for soil analysis

The soil samples collected initially (Table 3) and after the harvest of field bean were analyzed for physical and chemical characteristics employing standard methods of analysis.

3.7.1 Physical properties

3.7.1.1 Particle size analysis

Particle size distribution in soil samples was determined by international pipette method as described by Pipper (1966) using sodium hydroxide as a dispersing agent. From the dispersed suspension an aliquot of clay + silt and clay were pipetted out from specified depth at their specific time intervals depending on the suspension temperature. The total sand obtained by repeated decantation of silt and clay was passed through 0.05 mm sieve. The fraction that was finer than 0.05 mm was added to silt determined initially by pipetting to have particle size classes as per USDA system.

3.7.2 Chemical properties

3.7.2.1 pH and Electrical conductivity

The soil pH was determined in 1: 2.5 soil: water suspensions using digital pH meter with glass electrode (Jackson, 1973). The electrical conductivity of soil was

determined using clear extract of soil: water suspension using conductivity bridge (Jackson, 1973).

3.7.2.2 Organic carbon (%)

The dry soil samples were powdered using pestle and mortar and passed through 0.2 mm sieve. A known weight of finely powdered sample was treated with excess known volume of standard $K_2Cr_2O_7$ and concentrated H_2SO_4 . The unused $K_2Cr_2O_7$ was quantified by back titration with standard ferrous ammonium sulphate using ferroin as an indicator (Jackson, 1973).

3.7.2.3 Available nitrogen

Available nitrogen was determined by macro distillation of the sample following alkaline permanganate method as suggested by Subbiah and Asija (1956).

3.7.2.4 Available phosphorus

Available phosphorus was extracted with Bray's No.1 extractant (0.03 N NH_4F + 0.025 N HCl). The phosphorus in the extract was determined by Chlorostannous reduced molybdo-phosphoric blue colour method in HCl acid medium. The intensity of blue colour was read at 660 nm using a spectrophotometer (Bray and Kurtz, 1945).

3.7.2.5 Available potassium

Available potassium was determined flame photometrically after extracting the soil with neutral normal ammonium acetate (Jackson, 1973).

3.7.2.6 Exchangeable calcium and magnesium

Calcium and magnesium in the digested plant materials were determined by Versenate titration method as outlined by Pipper (1966).

3.7.2.7 Available sulphur

Sulphur in the digested plant materials was determined turbidometrically. The intensity of turbidity was measured using spectrophotometer at 420 nm of wavelength as outlined by Pipper (1966).

3.7.2.8 DTPA extractable micronutrients

The method developed by Lindsay and Norwell (1978) using DTPA extractant (Diethylene triamine penta acetic acid) was followed for the estimation of Zn, Cu, Mn and Fe. Ten grams of soil was shaken with 20 ml of DTPA extractant for 2 hours for the extraction of micronutrient cations. Atomic absorption spectrophotometer with appropriate hollow cathode lamp was used for measuring the concentration.

3.7.2.9 Exchangeable aluminium

For the same solution, one drop of 0.1 N HCl was added to bring back to colourless condition. Then NaF was added to the solution which formed a stable Sodium fluoroaluminate liberating equivalent quantity of NaOH. Titration was made with 0.1 N HCl which gave an estimate of exchangeable aluminium in the extract.

3.8 Statistical analysis of data

The comparative study of experimentally collected results was carried out by implementing Fisher's system of measurement of variance as described by Gomez and Gomez (1984). The significance level used in the 'F' evaluation was offered at 5 per cent. Critical difference (CD) values are presented at a significance level of 5 per cent in the table, wherever the 'F' measure was found to be relevant at 5 per cent.

IV RESULTS AND DISCUSSION

The research work entitled “**Utilization of products obtained from pyrolysis of paddy straw**” was carried out during 2020-2021. The experiment was carried out in organic farming field, UAS, GKVK, Bangalore – 560065. The results and discussion of the study are presented in this chapter.

4.1 Characterization of biochar

The Paddy straw was procured from Punjab, India and biochar was prepared at a higher temperature (500 °C) in oxygen limited condition / absence of oxygen.

Characterization of biochar was done by various standardized analytical procedures for its proximate and ultimate analysis and other properties such as bulk density, water holding capacity, pH, EC. and biooil was analyzed by GC-MS.

Representative samples were taken from paddy straw biochar. The paddy straw biochar was ground to move through a sieve of 2 mm and tested for various chemical parameters and findings are shown in Table 4.1.

Table 4.1: Pyrolysis details

| Parameter | Value |
|------------------------------|----------|
| Temperature | 500°C |
| Time | 6 hours |
| Straw filled in the unit | 3.4 kg |
| Biochar | 2.29 kg |
| Bio oil | 12 ml |
| Syngas | 101.48 L |
| Conversion ratio for biochar | 67.35 % |

The Paddy straw biochar was characterized for various physical (Bulk Density, Maximum Water Holding Capacity) and chemical (pH and Electrical Conductivity) properties, and total concentration of nutrients. The properties of the Paddy straw biochar used in the present study greatly varied. The variation in Paddy straw biochar's physico-chemical properties may be due to influence of the type of feedstock and the temperature of pyrolysis. Gaskin *et al.* (2008) and Singh *et al.* (2010) published similar findings of variation of physico-chemical properties according to type of feedstock and the temperature of pyrolysis.

The paddy straw biochar was found as 13% Ash content, 9.5% moisture content, 71.4% volatile matter and 6.1% fixed carbon. In agreement with Wang *et al.* (2014) and Phuong *et al.* (2015), who reported similar results in proximate analysis.

The pH of paddy straw biochar was found to be 10.29 is organically alkaline with EC value 3.2 dSm⁻¹ Kannan *et al.* (2013) reported the pH of the biochar produced from different agricultural feed stock materials ranged from 7.9 to 11.8 which are of alkaline range. Bagreev *et al.* (2001) recorded that the biochar of *Prosopis juliflora* had an 8.57 pH and a paddy straw biochar had a higher pH of 9.68. For biochar produced at 200 and 400°C, Wabel *et al.* (2013) observed EC values of 0.76 and 1.34 dSm⁻¹, respectively which shows increase in pyrolysis temperature increases the EC of biochar.

The lower bulk density of 0.56 g cm⁻³ was recorded in the biochar produced from paddy straw. Similar results of BD of biochar produced from different feedstocks ranged from 0.25 to 0.6 g cm⁻³ were also recorded by Pastor-Villegas *et al.* (2006) and Brewer *et al.* (2014) in a study from natural fire site.

In this study the concentration of total nitrogen, phosphorus and potassium were found to be 0.41, 0.04 and 3.83 per cent, respectively. The biochar contains total calcium, magnesium, sulphur content of 2.39, 0.24, and 0.07 per cent, respectively.

Table 4.2: Characterization of biochar

| Sl. No. | PARAMETERS | VALUES |
|---------------------------|------------------------|----------------------------|
| PROXIMATE ANALYSIS | | |
| 1 | ASH (%) | 13 |
| 2 | FIXED CARBON (%) | 6.1 |
| 3 | MOISTURE CONTENT (%) | 9.5 |
| 4 | VOLATILE MATTER (%) | 71.4 |
| ULTIMATE ANALYSIS | | |
| 1 | CARBON (%) | 35.15 |
| 2 | NITROGEN (%) | 0.41 |
| 3 | PHOSPHORUS (%) | 0.04 |
| 4 | POTASSIUM (%) | 3.83 |
| 5 | OXYGEN (%) | 46.21 |
| 6 | CALCIUM (%) | 2.39 |
| 7 | MAGNESIUM (%) | 0.24 |
| 8 | SULFUR (%) | 0.07 |
| 9 | HYDROGEN (%) | 4.16 |
| 10 | CARBON:NITROGEN (C:N) | 85.7 |
| 11 | O:C | 1.31 |
| 12 | H:C | 0.12 |
| 13 | C:N:P | 35.15: 0.41: 0.04 |
| OTHER PROPERTIES | | |
| 1. | pH | 10.29 |
| 2. | EC | 3.287 (dSm ⁻¹) |
| 3. | Bulk density | 0.56 (Mg m ⁻³) |
| 4. | Water holding capacity | 64.35% |

Table 4.3: Characterization of input materials

| Particulars | Gana jeevamrutha | Vermi compost | FYM | Biochar |
|---|-------------------------|----------------------|------------|----------------|
| pH | 6.8 | 7.06 | 6.6 | 10.29 |
| Electrical conductivity (dS m ⁻¹) | 7.03 | 3.26 | 3.17 | 3.287 |
| Organic carbon (%) | 14.2 | 13.6 | 11.8 | 35.15 |
| Available Nitrogen (%) | 3.81 | 2.63 | 1.03 | 0.41 |
| Available Phosphorus (%) | 10.7 | 9.17 | 6.59 | 0.04 |
| Available Potassium (%) | 3.48 | 2.95 | 1.55 | 3.83 |
| Exchangeable Calcium (ppm) | 55 | 49 | 38.6 | 2.39% |
| Exchangeable Magnesium (ppm) | 21.8 | 14.5 | 11.6 | 0.24% |
| Available Sulphur (ppm) | 0.18 | 0.13 | 0.11 | 0.07% |

The paddy straw biochar contained higher carbon content exchangeable bases, Mn and Cu. Gaskin *et al.* (2008) and Singh *et al.* (2010) found similar values of different nutrients in biochar similar to those reported in literature. The N content of 0.41 per cent was recorded in the paddy straw biochar, which might be attributed to the heat used during pyrolysis. Similar results were reported by Cheng *et al.* (2006), Novak *et al.* (2009) and Rondon *et al.* (2007).

4.2 To evaluate the performance of paddy Biochar Application on Field Bean.

4.2.1 Quality analysis:

The pH of inputs used for research were neutral pH of 6.8, 7.06 and 6.6 in Ganajeevamrutha, VC and FYM respectively. Ganajeevamrutha contain more soluble salts resulted in higher EC value.

Available nitrogen, phosphorus, potassium are rich in Ganajeevamrutha compared to other sources of input with 3.81, 10.7, 3.48 per cent respectively.

4.2.2 Growth parameters:

4.2.2.1 Plant height

Table 4.4 presents data retrieved for field bean plant height (cm) affected by various levels of biochar at different growth stages. Plant height differed significantly in all growth stages with various treatments. Plant height gradually increased with increasing crop age up to 90 days after sowing.

Significantly higher plant height (15.5 cm) was recorded at 30 days after sowing (DAS) in T₂₁ which received 25 tons of Biochar + FYM + VC + Ganajeevamrutha. On par data was found with T₂₀ (14.5 cm) , T₁₈ (14.5 cm) and T₉ (14.5 cm) which received 20 tons of Biochar + FYM + VC + Ganajeevamrutha, 25 tons of Biochar + VC+ Ganajeevamrutha and 25 tons of Biochar + Ganajeevamrutha respectively. Followed by T₆ (13.5 cm) which received 20 tons of Biochar + Ganajeevamrutha. Lowest height was recorded in T₂₂ (7.75 cm) which received 15tons of Biochar alone.

Table 4.4: Effect of biochar on plant height (cm) of field bean at 30 DAS, 60 DAS and at harvest stages.

| Treatment details | Plant height (cm) | | |
|--|-------------------|--------|------------|
| | 30 DAS | 60 DAS | At harvest |
| T ₁ : 15 tons of BC + FYM | 8.50 | 26.00 | 39.75 |
| T ₂ : 15 tons of BC + VC | 9.75 | 28.25 | 40.75 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 11.50 | 32.50 | 47.25 |
| T ₄ : 20 tons of BC + FYM | 9.75 | 28.00 | 41.25 |
| T ₅ : 20 tons of BC + VC | 10.50 | 34.25 | 46.00 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 13.50 | 39.00 | 54.00 |
| T ₇ : 25 tons of BC + FYM | 10.75 | 31.00 | 46.00 |
| T ₈ : 25 tons of BC + VC | 11.50 | 33.00 | 57.50 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 14.50 | 40.25 | 61.50 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 9.75 | 29.50 | 45.50 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 10.00 | 33.00 | 46.75 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 11.00 | 34.00 | 48.00 |
| T ₁₃ : 20 tons of BC + FYM +VC | 10.50 | 31.00 | 50.00 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 12.00 | 34.50 | 57.00 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 12.50 | 39.00 | 61.00 |
| T ₁₆ : 25 tons of BC + FYM + VC | 11.75 | 32.50 | 47.00 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 12.50 | 34.50 | 49.75 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 14.50 | 39.50 | 54.50 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 13.00 | 36.00 | 53.50 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 14.50 | 41.50 | 60.00 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 15.50 | 45.50 | 66.00 |
| T ₂₂ : 15 tons of BC | 7.75 | 26.00 | 34.00 |
| T ₂₃ : 20 tons of BC | 9.50 | 29.00 | 38.50 |
| T ₂₄ : 25 tons of BC | 11.75 | 32.00 | 46.50 |
| S. Em ± | 0.78 | 1.02 | 1.77 |
| C.D@ 5% | 2.28 | 3.00 | 5.20 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure, DAS-Days after sowing.

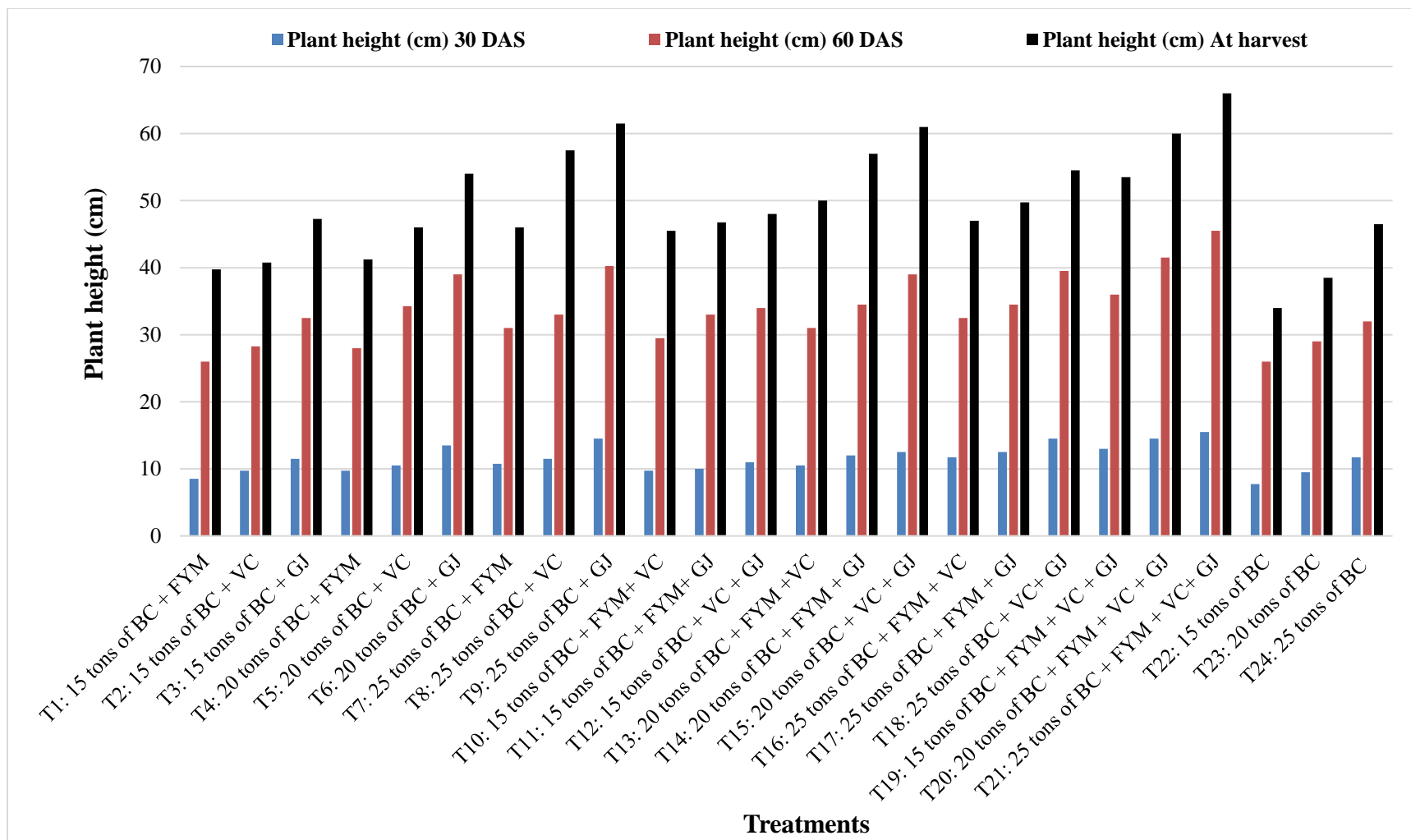


Fig. 4.1: Effect of biochar on plant height (cm) of field bean at 30 DAS, 60 DAS and at harvest stages

Note: BC-Biochar, VC-Vermicompost, FYM- Farm yard manure and GJ-Ganajeevamrutha

The plant height significantly varied with different treatments having different combinations of biochar levels at 60 DAS. Among which T₂₁ (45.5 cm) showed higher value which received 25 tons of Biochar + FYM + VC + Ganajeevamrutha, followed by T₂₀ (41.5 cm) received 20 tons of Biochar + FYM + VC + Ganajeevamrutha and T₉ (40.25 cm) received 25 tons of Biochar + Ganajeevamrutha. Lowest value was recorded in T₂₂ (26 cm) which received 15tons of Biochar alone.

Significantly higher plant height was observed in T₂₁ (66 cm) at harvest stage which received 25 tons of Biochar + FYM + VC + Ganajeevamrutha and was on par with T₉ (61.5 cm) which received 25 tons of Biochar + Ganajeevamrutha. The lower value was recorded in T₂₂ (34 cm) which received 15tons of Biochar.

4.2.2.2 Number of leaves

Influence of different dosage of biochar on number of leaves of field bean at different stages of plant growth is presented in Table 4.5.

Application of 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁) recorded highest number of leaves (32.5) followed by T₁₅ (28) and T₂₄ (28) which received 20 tons of BC + VC + Ganajeevamrutha and 25tons of BC respectively followed by T₉ (26.5) with 25 tons of BC + Ganajeevamrutha at 30 DAS. The lower value was recorded in T₁₇ (14) which received 25 tons of BC + FYM + Ganajeevamrutha.

At 60 DAS treatment with 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁) recorded significantly higher number of leaves (72) followed by T₁₅ (68) and T₂₀ (66.5) which received 20 tons of BC + VC + Ganajeevamrutha and 20 tons of BC + FYM + VC + Ganajeevamrutha, respectively. The lower value was recorded in T₂₂ (43.5) which received 15tons of Biochar.

During harvest, data on the number of leaves shows a similar pattern to that of 60 days. Maximum number of leaves was noticed in T₂₁ (80) with 25 tons of BC + FYM + VC + Ganajeevamrutha.

Table 4.5: Effect of biochar on number of leaves at 30 DAS, 60 DAS and at harvest stages

| Treatment details | NUMBER OF LEAVES | | |
|--|------------------|--------|------------|
| | 30 DAS | 60 DAS | At harvest |
| T ₁ : 15 tons of BC + FYM | 18.50 | 46.50 | 51.00 |
| T ₂ : 15 tons of BC + VC | 23.00 | 54.00 | 60.00 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 24.00 | 55.50 | 64.50 |
| T ₄ : 20 tons of BC + FYM | 21.50 | 49.50 | 55.50 |
| T ₅ : 20 tons of BC + VC | 23.00 | 55.50 | 61.50 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 26.50 | 58.50 | 64.50 |
| T ₇ : 25 tons of BC + FYM | 22.00 | 51.00 | 55.50 |
| T ₈ : 25 tons of BC + VC | 24.00 | 54.00 | 60.00 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 26.50 | 63.00 | 67.50 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 21.50 | 55.50 | 63.00 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 23.00 | 60.00 | 67.50 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 24.50 | 61.50 | 63.00 |
| T ₁₃ : 20 tons of BC + FYM +VC | 18.50 | 58.50 | 64.50 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 24.50 | 61.50 | 69.00 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 28.00 | 68.00 | 70.50 |
| T ₁₆ : 25 tons of BC + FYM + VC | 19.00 | 63.00 | 61.50 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 14.00 | 54.00 | 64.50 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 21.50 | 58.50 | 66.00 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 19.50 | 64.50 | 72.00 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 26.00 | 66.50 | 74.00 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 32.50 | 72.50 | 80.00 |
| T ₂₂ : 15 tons of BC | 21.50 | 43.50 | 51.00 |
| T ₂₃ : 20 tons of BC | 22.50 | 46.50 | 55.50 |
| T ₂₄ : 25 tons of BC | 28.00 | 49.50 | 54.00 |
| S. Em ± | 1.68 | 2.29 | 2.25 |
| C.D@ 5% | 4.92 | 6.68 | 6.58 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure, DAS-Days after sowing.

On par with T₂₀ (74) which received 20 tons of Biochar + FYM + VC + Ganajeevamrutha followed by T₁₉ (72) received 15 tons of BC+FYM+VC+ Ganajeevamrutha. The lower value was recorded in T₂₂ (51) which received 15tons of Biochar alone.

4.2.2.3 Number of branches

Application of biochar in combination with other organic manures had significant difference between the treatments. However, application of 20 tons of BC + FYM + VC + Ganajeevamrutha (T₂₀) and 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁) recorded the highest number of branches (6) compared to T₂₂ (3) which had 15tons of BC showing lowest number of branches.

Treatment receiving of 20 tons of BC + FYM + VC + Ganajeevamrutha (T₂₀) and 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁) recorded significantly higher numbers of branches per plant (12) followed by T₁₇ (11) received 25 tons of BC + FYM + Ganajeevamrutha and the lowest value was recorded in T₄, T₇ and T₂₂ (7.5) that received 20 tons of BC + FYM, 25 tons of BC + FYM and 15tons of BC respectively.

At the time of harvest, treatment receiving recommended 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁), the number of branches per plant (14) was significantly higher, followed by T₂₀ (13.5) received 20 tons of BC + FYM + VC + Ganajeevamrutha and the lowest value was recorded in T₄ (9) that received 20 tons of BC + FYM.

Due to application of 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁), significant increase in growth parameters such as plant height, number of branches of fieldbean plant was reported. With biochar use in combination with other organic manures, there were several variables relating to the crop growth attributes. Those factors will operate either individually or in combination needs to be addressed. Indeed, a reduction in soluble Al and Fe led to increase in pH, Regulated and gradual nutrient release and because of biochar specific physical-chemical properties with low bulk density. This will improve the porosity and aeration status of the soil, which would

Table 4.6: Effect of biochar on number of branches at 30 DAS, 60 DAS and at harvest stages

| Treatment details | No. of branches | | |
|--|-----------------|--------|------------|
| | 30 DAS | 60 DAS | At harvest |
| T ₁ : 15 tons of BC + FYM | 3.50 | 9.00 | 10.50 |
| T ₂ : 15 tons of BC + VC | 4.50 | 8.50 | 10.00 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 5.00 | 10.50 | 12.50 |
| T ₄ : 20 tons of BC + FYM | 3.50 | 7.50 | 9.00 |
| T ₅ : 20 tons of BC + VC | 4.50 | 8.50 | 10.00 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 5.00 | 8.00 | 10.50 |
| T ₇ : 25 tons of BC + FYM | 4.50 | 7.50 | 12.50 |
| T ₈ : 25 tons of BC + VC | 5.00 | 9.50 | 11.50 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 4.50 | 8.50 | 9.50 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 3.50 | 9.50 | 10.50 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 5.50 | 10.50 | 12.00 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 4.50 | 10.00 | 12.00 |
| T ₁₃ : 20 tons of BC + FYM +VC | 3.50 | 10.00 | 11.00 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 4.50 | 10.00 | 12.00 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 4.50 | 9.50 | 10.50 |
| T ₁₆ : 25 tons of BC + FYM + VC | 5.50 | 8.50 | 12.00 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 5.50 | 11.00 | 11.50 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 4.50 | 9.50 | 11.50 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 5.50 | 10.00 | 11.50 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 6.00 | 12.00 | 13.50 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 6.00 | 12.00 | 14.00 |
| T ₂₂ : 15 tons of BC | 3.50 | 7.50 | 10.50 |
| T ₂₃ : 20 tons of BC | 4.00 | 8.50 | 10.00 |
| T ₂₄ : 25 tons of BC | 4.50 | 9.50 | 12.00 |
| S. Em ± | 0.47 | 0.83 | 0.86 |
| C.D@ 5% | 1.39 | 2.43 | 2.52 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure, DAS-Days after sowing.

indirectly increase the available plant water and boost microbial function, should have led to the improvement of growth parameters. Most of these changes in soil's physical and chemical properties in biochar transformed soils similar research findings were also found (Warnock *et al.* 2007, Amonette and Joseph 2009).

4.2.3 Yield parameters

The data pertaining to yield and yield parameters as influenced by the levels of paddy straw biochar on soil are presented in Table 4.7 and Fig 4.2.

4.2.3.1 Number of pods

The highest number of pods per plant was observed in T₂₀ (16) which received 20 tons of BC + FYM + VC + Ganajeevamrutha similar result was found in T₂₁ received 25 tons of BC + FYM + VC + Ganajeevamrutha which was on par with T₁₈ (15) which received 25 tons of BC + VC+ Ganajeevamrutha which is on par with T₁₇ (14) receiving 25 tons of BC + FYM + Ganajeevamrutha and the lowest number of pods per plant was recorded in T₂₂ (6), which did not receive an external source of nutrients but 15 tons of biochar only.

The highest amount of pod weight was observed in T₂₀ (24) which received 20 tons of BC + FYM + VC + Ganajeevamrutha followed by T₁₈ (22.5) which received tons of 25 BC + VC+ Ganajeevamrutha and the lowest pods yield was recorded in T₁ (9.6), which received 15 tons of BC + FYM only.

Significant increase in yield parameters such as number of pods and pod weight was reported highest using 20 tons of BC + FYM + VC + Ganajeevamrutha (T₂₀). It could be attributed to improved nutrient demand and supply through the introduction of elevated biochar levels. Biochar has been found to improve the productivity and fertilizer usage as stated by Dong *et al.* (2015) and Taghizadeh- Toosi *et al.* (2012). Paddy straw biochar contains a large amount of total carbon, potassium, phosphorus, calcium and magnesium, which improves the physical, chemical and biological properties of the soil, increasing the performance of yield parameters significantly. Chan *et al.* (2007) took into account identical results.

Table 4.7: Effect of biochar on number pods and pod weight.

| Treatment details | No. of pods | Pod weight (g/plant) | Yield (q/ha) |
|--|-------------|-----------------------|--------------|
| T ₁ : 15 tons of BC + FYM | 7.00 | 9.60 | 22.53 |
| T ₂ : 15 tons of BC + VC | 7.50 | 12.00 | 28.16 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 12.00 | 19.25 | 45.17 |
| T ₄ : 20 tons of BC + FYM | 11.00 | 17.50 | 41.06 |
| T ₅ : 20 tons of BC + VC | 13.00 | 21.25 | 49.86 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 10.00 | 14.00 | 32.85 |
| T ₇ : 25 tons of BC + FYM | 9.00 | 12.60 | 29.57 |
| T ₈ : 25 tons of BC + VC | 9.50 | 13.00 | 30.50 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 13.50 | 18.80 | 44.11 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 10.00 | 14.00 | 32.85 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 11.00 | 15.40 | 36.14 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 11.50 | 15.90 | 37.31 |
| T ₁₃ : 20 tons of BC + FYM +VC | 12.00 | 17.00 | 39.89 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 10.00 | 14.00 | 32.85 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 11.50 | 16.75 | 39.30 |
| T ₁₆ : 25 tons of BC + FYM + VC | 12.00 | 19.00 | 44.58 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 14.00 | 21.00 | 49.28 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 15.00 | 22.50 | 52.80 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 13.50 | 20.25 | 47.52 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 16.00 | 24.00 | 56.32 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 15.00 | 21.00 | 49.28 |
| T ₂₂ : 15 tons of BC | 6.00 | 10.50 | 24.64 |
| T ₂₃ : 20 tons of BC | 10.00 | 14.75 | 34.61 |
| T ₂₄ : 25 tons of BC | 9.50 | 14.25 | 33.44 |
| S. Em ± | 1.25 | 1.82 | 4.28 |
| C.D@ 5% | 3.66 | 5.32 | 12.50 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure.

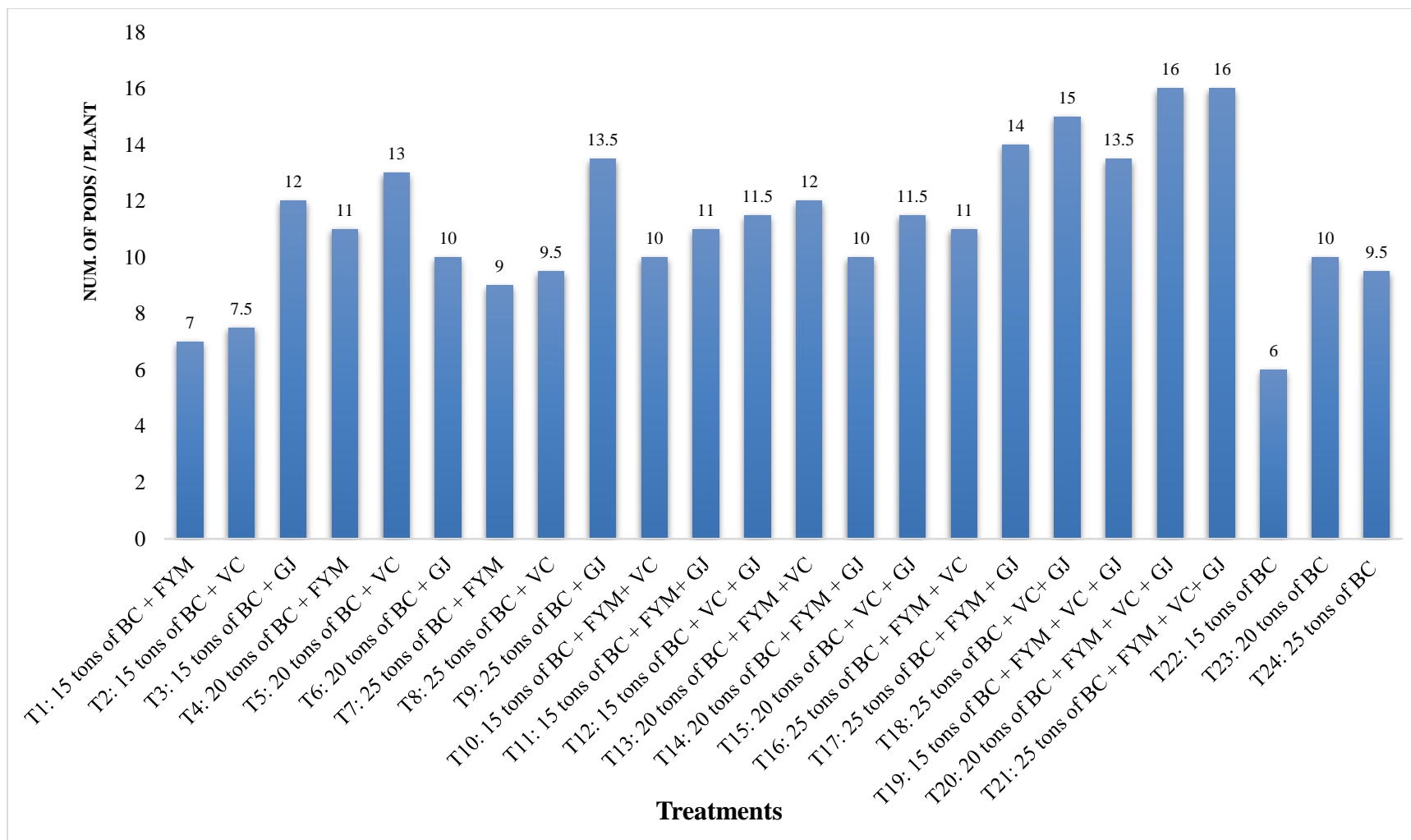


Fig. 4.2: Effect of biochar on number of pods / plant

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure., GJ-Ganajeevamrutha

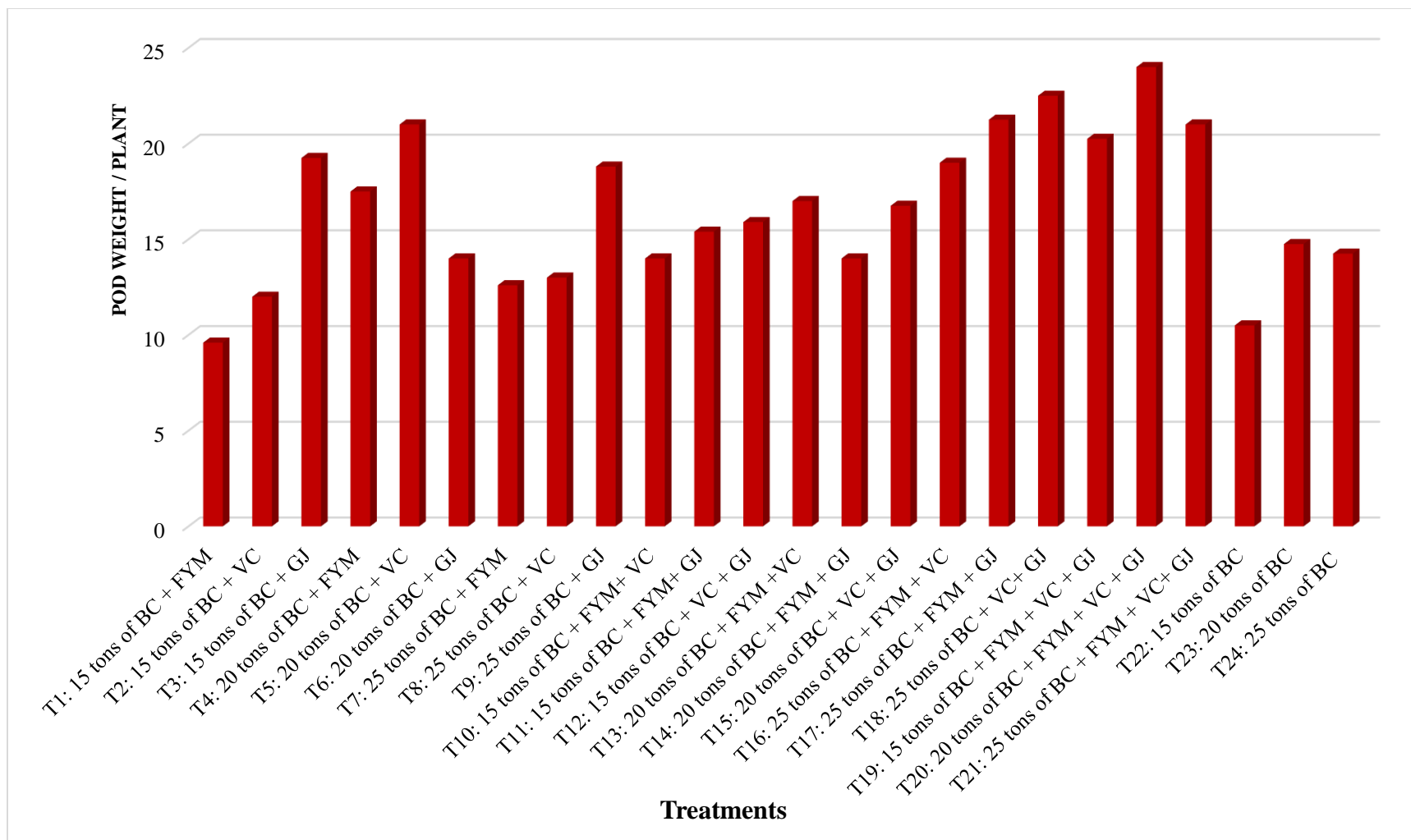


Fig. 4.3: Effect of biochar on pod weight / plant

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure., GJ-Ganajeevamrutha

4.2.4 Physical parameters of soil

Table 4.8 represents the effect of different dosage of biochar on bulk density and maximum water holding capacity of soil. Significantly higher soil bulk density (1.4 Mg m^{-3}) was recorded in T₂₂, which received 15tons of BC alone. Significantly lower BD (1.29 Mg m^{-3}) was observed in T₇ (25 tons of BC + FYM), T₉ (25 tons of BC + Ganajeevamrutha), T₁₈ (25 tons of BC + VC+ Ganajeevamrutha), followed by T₁₇ showing 1.30 Mg m^{-3} (25 tons of BC + FYM + Ganajeevamrutha).

Significantly higher maximum water holding capacity was recorded in T₁₈ (42.03%) 25 tons of BC + VC+ Ganajeevamrutha, followed by T₂₂ (42.00%) which received 25 tons of BC + FYM + VC + Ganajeevamrutha, and lower soil water holding capacity was observed in T₂₂ (34.5%) which received 15tons of BC alone.

Biochar application (Table 4.8) has had a significant influence on the physical properties of soil like bulk density and maximum water holding capacity. Among the various treatments, the treatment which received 25 tons of BC + VC+ Ganajeevamrutha recorded lower bulk density and higher maximum water holding capacity values over the rest of treatments This may be due to the high carbon content in biochar, which acts as a cementitious material when stable soil aggregates are formed. The porous structure of biochar has shown to impact soil, with reference to water retention capacity and adsorption efficiency (Yu *et al.* 2006, Gundale and Deluca 2006.) biochar has a large density that is much lower than that of mineral soils and therefore biochar application can reduce the overall density of the soil. The biochar application reduced the bulk density by 12 to 25 per cent and the water holding capacity was increased compared to at all biochar application rates (Emmanuel *et al.*, 2010).

4.2.5 Chemical parameters of soil

The data in Table 4.9 revealed that after harvest of the Field bean, there was a substantial difference in soil pH condition. pH of the soil increased with increasing rate of biochar and combination with lime has slight increase than biochar alone.

Table 4.8: Effect of biochar on soil physical properties.

| Treatment details | BD (Mg m ⁻³) | MWHC (%) |
|--|--------------------------|----------|
| T ₁ : 15 tons of BC + FYM | 1.40 | 35.65 |
| T ₂ : 15 tons of BC + VC | 1.39 | 36.36 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 1.39 | 35.60 |
| T ₄ : 20 tons of BC + FYM | 1.37 | 35.13 |
| T ₅ : 20 tons of BC + VC | 1.38 | 35.05 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 1.38 | 35.31 |
| T ₇ : 25 tons of BC + FYM | 1.29 | 41.77 |
| T ₈ : 25 tons of BC + VC | 1.32 | 41.07 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 1.30 | 42.03 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 1.36 | 36.25 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 1.36 | 36.82 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 1.39 | 35.77 |
| T ₁₃ : 20 tons of BC + FYM +VC | 1.36 | 36.34 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 1.37 | 36.60 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 1.38 | 36.46 |
| T ₁₆ : 25 tons of BC + FYM + VC | 1.30 | 41.31 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 1.30 | 40.50 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 1.30 | 42.00 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 1.39 | 35.80 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 1.36 | 36.70 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 1.34 | 39.64 |
| T ₂₂ : 15 tons of BC | 1.40 | 34.50 |
| T ₂₃ : 20 tons of BC | 1.38 | 35.03 |
| T ₂₄ : 25 tons of BC | 1.37 | 36.50 |
| S. Em ± | 0.0075 | 0.207 |
| C.D@ 5% | 0.0218 | 0.605 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure.

Table 4.9: Effect of biochar application on soil pH, EC and Organic carbon content of post-harvest soil of field bean crop.

| Treatment details | pH | EC (dSm ⁻¹) | OC (g kg ⁻¹) |
|--|------|-------------------------|--------------------------|
| T ₁ : 15 tons of BC + FYM | 7.20 | 0.10 | 5.48 |
| T ₂ : 15 tons of BC + VC | 7.97 | 0.12 | 5.46 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 8.02 | 0.11 | 5.40 |
| T ₄ : 20 tons of BC + FYM | 7.52 | 0.12 | 5.60 |
| T ₅ : 20 tons of BC + VC | 7.33 | 0.12 | 5.66 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 7.85 | 0.13 | 5.72 |
| T ₇ : 25 tons of BC + FYM | 7.24 | 0.15 | 5.86 |
| T ₈ : 25 tons of BC + VC | 7.44 | 0.16 | 5.98 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 8.20 | 0.18 | 6.11 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 7.47 | 0.12 | 5.63 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 7.77 | 0.12 | 5.46 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 7.92 | 0.12 | 5.45 |
| T ₁₃ : 20 tons of BC + FYM +VC | 7.53 | 0.14 | 5.87 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 7.30 | 0.14 | 5.83 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 7.26 | 0.16 | 6.15 |
| T ₁₆ : 25 tons of BC + FYM + VC | 7.52 | 0.16 | 6.12 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 7.41 | 0.15 | 5.88 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 7.36 | 0.17 | 6.18 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 7.60 | 0.16 | 6.13 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 8.98 | 0.18 | 6.20 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 9.50 | 0.19 | 6.42 |
| T ₂₂ : 15 tons of BC | 6.76 | 0.10 | 5.47 |
| T ₂₃ : 20 tons of BC | 7.65 | 0.12 | 5.72 |
| T ₂₄ : 25 tons of BC | 8.03 | 0.16 | 6.11 |
| S. Em ± | 0.01 | 0.0066 | 0.03 |
| C.D@ 5% | 0.02 | 0.0193 | 0.09 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure.

Significantly higher pH (9.5) was reported in treatment receiving 25 tons of BC + FYM + VC + Ganajeevamrutha and was on par with T₂₀ (8.98) receiving 20 tons of BC + FYM + VC + Ganajeevamrutha. In addition, the lower pH value in soil was reported in T₂₂ (6.76) followed by T₁ (7.20) which was 15 tons of BC + FYM.

The increase in pH of the soil that is very much in excess is due to the application of the biochar, the high surface area and the porous nature of the biochar can increase the cation exchange capacity (CEC) of the soil. It is also due to the presence of the interchangeable alkaline and high bases of the biochar. It is known that bio-accumulation increases the pH of acidic and efficient acidic soil improvement. The basic cations in biochar are interchangeable with the interchangeable to Al³⁺ and H⁺ in soil exchange complexes, thus reducing replaceable acidity in acid and neutral soils. Such an observation that matches Chintala *et al.* (2014), Anteneh *et al.* (2014) and Hass *et al.* (2012) revealed the alkalinity of Biochar is an important factor that affects its capacity for limitation, and the largest alkalinity of the biochar led to a greater decrease in acidity. The incorporation of biochar reduces the Replaceable soil acidity, increasing the base of replaceable and improves base saturation, thus improving soil fertility.

The data on electrical conductivity (EC) in Table 4.9 showed that there was significant difference among the different treatments with respect to electrical conductivity of soil after harvest. The EC (0.19 dSm⁻¹) was significantly higher in treatment imposed with 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁). Further there was a gradual increase in EC with respect to increasing levels of biochar.

Numerically higher value of EC was recorded in T₂₁ received 25 tons of BC + FYM + VC + Ganajeevamrutha and lowest EC was recorded in 15 tons of BC(T₂₂) and 15 tons of BC + FYM (T₁). This may be due to the increase in salt concentration and exchangeable cations in paddy straw biochar which can increase the EC in treated plot compared to untreated plot. Significant increase in EC with varied levels of biochar application was often reported by Chintala *et al.* (2014), Gundale and De luca (2006) and Chan *et al.* (2008).

The organic carbon data in Table 4.9 indicated that there was a significant difference in soil organic carbon content among the various treatments after field bean harvest. Soil organic carbon content after field bean harvest varied from 5.47 (g kg⁻¹) to 6.42 (g kg⁻¹). Numerically higher organic carbon value (6.42 g kg⁻¹) was recorded in T₂₁ with 25 tons of BC + FYM + VC + Ganajeevamrutha followed by T₂₀ (6.2 g kg⁻¹) receiving 20 tons of BC + FYM + VC + Ganajeevamrutha. The lowest soil organic carbon value (5.4 g kg⁻¹) was reported in T₃ which received 15 tons of BC + Ganajeevamrutha.

Increasing the amount of biochar applied increased the organic carbon content in the soil. Higher levels of organic carbon in biochar-treated soils indicate that recalcitrant organic carbon in biochar is not affected by mineralization and is further lost. Soil organic carbon storage called carbon sequestration will be improved. Lehmann and Randon (2005) and Solomon *et al.* (2009) also reported high organic carbon content in soils using biochar. The special properties of stable carbon in biochar and the high surface area, the high charge per unit area, the appearance of specific functional surface groups and the ash content have beneficial effects on the chemical properties of the soil. The application of biochar increased the SOC, pH, EC, CEC and interchangeable alkalis in the soil to which the biochar was applied. (Hass *et al.*, 2012 and Anteneh *et al.*, 2014).

4.2.6 Nutrient status of the soil after the harvest of field bean.

4.2.6.1 Available nitrogen

The nutrient status of available N, P₂O₅, K₂O, exchangeable Ca and Mg, available S and DTPA extractable micronutrients Zn, Cu, Mn and Fe of the soil after harvest of field bean was analyzed and the results are presented in Table 4.10, 4.11, 4.12.

After the harvest of field bean, the available nitrogen content in soil differed significantly due to different rates of biochar application (Table 4.10). Significantly higher available nitrogen (297 kg ha⁻¹) was recorded in T₂₁ where 25 tons of BC + FYM + VC + Ganajeevamrutha was added. The lowest available nitrogen content was reported T₂₂ (255.5 kg ha⁻¹) where 15 tons of BC. This might be due to incorporation of biochar in

combination with organic manures has made nitrogen available to the soil. Lehmann and Rondon (2005) reported that biochar changes soil dynamics of N. The abundance and intensity of organic nitrogen mineralization contained in biochar added to the soil provides an indicator of biochar's potential as a slow-release nitrogen fertilizer stated by Chan and Xu (2013) and Steiner *et al.* (2007).

4.2.6.2 Available phosphorus

The phosphorus content available in soil after field bean harvest varied significantly among the treatments imposed (Table 4.10). Significantly higher available soil phosphorus content was observed in T₂₁ (63.25 kg ha⁻¹) which received 25 tons of BC + FYM + VC + Ganajeevamrutha compared to T₂₀ (63.1 kg ha⁻¹) which received 20 tons of BC + FYM + VC + Ganajeevamrutha. That may be due to the high levels of available P found in the biochar. Van Zwieten *et al.* (2010) also showed an increase in available P to the soil after biochar application. The reason that the abundance of P₂O₅ is improved by the application of the soil biochar may be due to the presence of soluble exchangeable phosphates in the soil. Parvage *et al.* (2013) and Hass *et al.* (2012) also reported that the addition of biochar increased the available amount of P₂O₅.

4.2.6.3 Available potassium

Due to biochar application the available potassium content in soil increased significantly (Table 4.10). Application of 25 tons of BC + FYM + VC + Ganajeevamrutha (T₂₁) substantially reported significantly higher available potassium (173.95 kg ha⁻¹) that was on par with treatment T₂₀ which received 20 tons of BC + FYM + VC + Ganajeevamrutha (172.95 kg ha⁻¹) compared to T₁₅ (170.5 kg ha⁻¹) that 20 tons of BC + VC + Ganajeevamrutha. The lowest value was recorded in absolute control T₂₂ (158 kg ha⁻¹) where it received 15 tons of BC.

The increased proportion of biochar combined with other organic manures increased the potassium content of in the soil, which can be attributed to the high K concentration present in the biochar (Chan *et al.* 2007).

Table 4.10: Effect of biochar application on available major nutrient status of soil after the harvest of field bean

| Treatment details | N (kg ha ⁻¹) | P ₂ O ₅ (kg ha ⁻¹) | K ₂ O (kg ha ⁻¹) |
|--|-----------------------------|---|--|
| T ₁ : 15 tons of BC + FYM | 265.50 | 53.00 | 159.05 |
| T ₂ : 15 tons of BC + VC | 269.00 | 55.50 | 159.00 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 270.50 | 56.75 | 160.75 |
| T ₄ : 20 tons of BC + FYM | 271.00 | 56.12 | 159.00 |
| T ₅ : 20 tons of BC + VC | 273.00 | 58.60 | 163.50 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 273.00 | 60.14 | 166.00 |
| T ₇ : 25 tons of BC + FYM | 274.00 | 59.12 | 165.85 |
| T ₈ : 25 tons of BC + VC | 277.50 | 60.92 | 167.95 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 278.50 | 61.60 | 167.75 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 266.00 | 55.00 | 169.10 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 265.00 | 56.25 | 167.50 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 268.50 | 54.25 | 169.00 |
| T ₁₃ : 20 tons of BC + FYM +VC | 272.00 | 58.50 | 168.10 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 272.50 | 59.35 | 169.50 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 271.50 | 59.95 | 170.50 |
| T ₁₆ : 25 tons of BC + FYM + VC | 274.50 | 59.00 | 168.35 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 277.00 | 60.25 | 170.10 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 279.50 | 61.60 | 169.50 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 286.00 | 60.10 | 169.80 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 291.50 | 63.10 | 172.95 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 297.00 | 63.25 | 173.95 |
| T ₂₂ : 15 tons of BC | 255.50 | 48.85 | 158.55 |
| T ₂₃ : 20 tons of BC | 258.00 | 52.45 | 159.20 |
| T ₂₄ : 25 tons of BC | 261.50 | 53.15 | 158.00 |
| S. Em ± | 2.36 | 0.94 | 1.24 |
| C.D@ 5% | 6.89 | 2.75 | 3.64 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure.

Increased potassium is primarily responsible for the immediate beneficial effects of nutrient-rich biochar input (Lehmann *et al.* 2003). Since biochar produces high ash, it has higher potassium content than other major nutrients, so adding ash-rich biochar to the soil significantly improved potassium levels.

4.2.6.4 Exchangeable calcium

The exchangeable calcium significantly increased from 2.56 (c mol (p+) kg⁻¹) due to the application of paddy straw biochar. Significantly higher exchangeable calcium content was recorded in T₂₁ (4.22 c mol (p+) kg⁻¹) 25 tons of BC + FYM + VC + Ganajeevamrutha. However, it was found on par with T₂₀ (4.07 c mol (p+) kg⁻¹) which received 20 tons of BC + FYM + VC + Ganajeevamrutha and followed by T₁₈ (3.85 c mol (p+) kg⁻¹) received 25 tons of BC + VC + Ganajeevamrutha. The lower value was recorded in T₂₂ (3.07 c mol (p+) kg⁻¹) received 15 tons of BC alone.

The exchangeable bases such as soil Ca content differed significantly due to the high cation exchange capacity in the application of various levels of biochar. Elevated biochar levels increased the calcium content of the soil post-harvest. This is likely due to the elevated Ca concentration and the exchangeable biochar base. It is likely due to the high porosity and surface/volume, and likely to increase the supply of Ca (Chan *et al.* 2007 and Yamato *et al.* 2006).

4.2.6.5 Exchangeable magnesium

Data on exchangeable soil magnesium content showed substantial difference between treatments due to the application of different soil biochar levels (Table 4.11). However, a numerically higher value of exchangeable magnesium in soil (2.98 c mol (p+) kg⁻¹) was found in T₂₁, which received 25 tons of BC + FYM + VC + Ganajeevamrutha, comparable to T₂₀ (2.96 c mol (p+) kg⁻¹), which received 20 tons of BC + FYM + VC + Ganajeevamrutha. The numerically lower amount of exchangeable magnesium (1.91 c mol (p+) kg⁻¹) was found in application of 15 tons of BC alone.

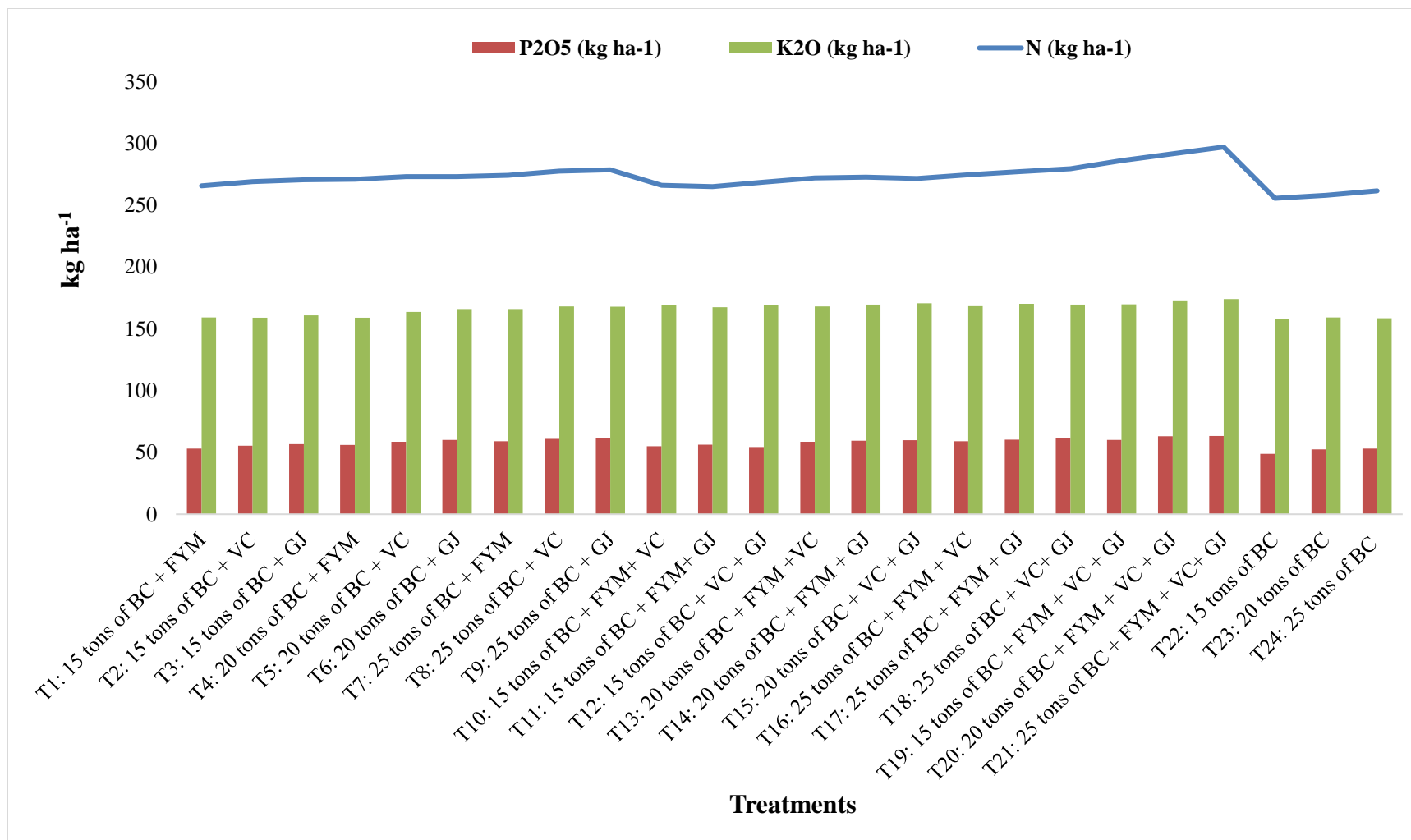


Fig. 4.4: Effect of biochar application on available major nutrient status of soil after the harvest of field bean

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure., GJ-Ganajeevamrutha

Table 4.11: Effect of biochar application on secondary nutrient status of soil after the harvest of field bean.

| Treatment details | Ca (c mol (p+) kg ⁻¹) | Mg (c mol (p+) kg ⁻¹) | S (mg kg ⁻¹) |
|--|---|---|--------------------------------|
| T ₁ : 15 tons of BC + FYM | 3.23 | 2.03 | 32.80 |
| T ₂ : 15 tons of BC + VC | 3.28 | 2.16 | 33.00 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 3.43 | 2.17 | 33.12 |
| T ₄ : 20 tons of BC + FYM | 3.57 | 2.20 | 33.15 |
| T ₅ : 20 tons of BC + VC | 3.56 | 2.19 | 33.65 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 3.60 | 2.35 | 34.15 |
| T ₇ : 25 tons of BC + FYM | 3.85 | 2.39 | 33.76 |
| T ₈ : 25 tons of BC + VC | 3.73 | 2.45 | 34.41 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 3.81 | 2.40 | 34.49 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 3.49 | 2.10 | 34.55 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 3.51 | 2.13 | 34.54 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 3.53 | 2.16 | 34.73 |
| T ₁₃ : 20 tons of BC + FYM +VC | 3.64 | 2.43 | 34.73 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 3.62 | 2.51 | 35.10 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 3.70 | 2.74 | 35.55 |
| T ₁₆ : 25 tons of BC + FYM + VC | 3.78 | 2.85 | 36.30 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 3.80 | 2.87 | 36.78 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 3.85 | 2.85 | 36.90 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 3.82 | 2.95 | 37.75 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 4.07 | 2.96 | 37.76 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 4.22 | 2.98 | 37.82 |
| T ₂₂ : 15 tons of BC | 3.07 | 1.91 | 32.55 |
| T ₂₃ : 20 tons of BC | 3.16 | 1.93 | 32.60 |
| T ₂₄ : 25 tons of BC | 3.20 | 1.94 | 32.75 |
| S. Em ± | 0.043 | 0.07 | 0.11 |
| C.D@ 5% | 0.12 | 0.21 | 0.34 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure.

Exchangeable bases such as the Mg content of the soil varied considerably with the application of different levels of biochar. The increase in exchangeable bases (Mg) in soil could be due to the release of essential cations from biochar at different time intervals. The ash from biochar easily releases free bases such as Ca, Mg and K into the soil solution, thus increasing not only the pH of soil, but also the exchangeable bases. Lehmann *et al.* (2003) and Chan *et al.* (2008) also documented such observations.

4.2.6.6 Exchangeable sulphur

As per the application of various amounts of biochar, the available soil sulphur content was found to vary substantially between the treatments (Table 4.11). Significantly higher value of available sulphur was reported in T₂₁ (37.82 mg kg⁻¹) which received 25 tons of BC + FYM + VC + Ganajeevamrutha compared to T₂₀ (37.76 mg kg⁻¹) which received 20 tons of BC + FYM + VC + Ganajeevamrutha. In the treatment 15tons of BC(T₂₂) the lowest value was recorded (32.55 mg kg⁻¹).

Soil sulphur concentration varies considerably with the application of various biochar amounts. This may be due to the contribution of the available sulphur to soil after mineralization of organic sulphur in biochar. The findings indicate that biochar also increases Sulphur bioavailability, which is primarily based on mineralizing organic sources of sulphur in soils. (Deluca *et al.* 2009).

4.2.6.7 DTPA extractable iron

The iron content of post-harvest soil showed significant difference due to different levels of biochar application represented in Table 4.12. The numerically lower value of Fe (18.34 mg kg⁻¹) was observed in T₂₁ which received 25 tons of BC + FYM + VC + Ganajeevamrutha, followed by T₁₈ (18.35 mg kg⁻¹) received 25 tons of BC + VC+ Ganajeevamrutha. The highest value of Fe (19.60 mg kg⁻¹) was found in found 15tons of BC alone (T₂₂).

Table 4.12: Effect of biochar application DTPA micronutrient status of soil after the harvest

| Treatment details | Fe (mg kg ⁻¹) | Mn (mg kg ⁻¹) | Zn (mg kg ⁻¹) | Cu (mg kg ⁻¹) |
|--|---------------------------------|---------------------------------|---------------------------------|---------------------------------|
| T ₁ : 15 tons of BC + FYM | 19.35 | 9.87 | 0.38 | 0.19 |
| T ₂ : 15 tons of BC + VC | 19.35 | 9.83 | 0.42 | 0.26 |
| T ₃ : 15 tons of BC + Ganajeevamrutha | 19.00 | 9.85 | 0.43 | 0.22 |
| T ₄ : 20 tons of BC + FYM | 18.83 | 9.77 | 0.42 | 0.22 |
| T ₅ : 20 tons of BC + VC | 18.82 | 9.77 | 0.42 | 0.26 |
| T ₆ : 20 tons of BC + Ganajeevamrutha | 18.65 | 9.55 | 0.40 | 0.20 |
| T ₇ : 25 tons of BC + FYM | 18.45 | 9.35 | 0.39 | 0.21 |
| T ₈ : 25 tons of BC + VC | 18.27 | 9.43 | 0.39 | 0.20 |
| T ₉ : 25 tons of BC + Ganajeevamrutha | 18.50 | 9.40 | 0.39 | 0.21 |
| T ₁₀ : 15 tons of BC + FYM+ VC | 18.65 | 8.85 | 0.42 | 0.26 |
| T ₁₁ : 15 tons of BC + FYM+ Ganajeevamrutha | 18.55 | 8.47 | 0.41 | 0.22 |
| T ₁₂ : 15 tons of BC + VC + Ganajeevamrutha | 18.60 | 8.50 | 0.43 | 0.22 |
| T ₁₃ : 20 tons of BC + FYM +VC | 18.35 | 9.30 | 0.44 | 0.27 |
| T ₁₄ : 20 tons of BC + FYM + Ganajeevamrutha | 18.50 | 8.80 | 0.37 | 0.22 |
| T ₁₅ : 20 tons of BC + VC + Ganajeevamrutha | 18.45 | 8.60 | 0.36 | 0.19 |
| T ₁₆ : 25 tons of BC + FYM + VC | 18.60 | 8.55 | 0.43 | 0.25 |
| T ₁₇ : 25 tons of BC + FYM + Ganajeevamrutha | 18.45 | 9.84 | 0.34 | 0.17 |
| T ₁₈ : 25 tons of BC + VC+ Ganajeevamrutha | 18.35 | 9.25 | 0.35 | 0.18 |
| T ₁₉ : 15 tons of BC + FYM + VC + Ganajeevamrutha | 18.55 | 8.95 | 0.36 | 0.19 |
| T ₂₀ : 20 tons of BC + FYM + VC + Ganajeevamrutha | 18.50 | 8.89 | 0.39 | 0.20 |
| T ₂₁ : 25 tons of BC + FYM + VC+ Ganajeevamrutha | 18.34 | 8.45 | 0.39 | 0.22 |
| T ₂₂ : 15 tons of BC | 19.60 | 9.95 | 0.41 | 0.26 |
| T ₂₃ : 20 tons of BC | 19.40 | 9.88 | 0.42 | 0.26 |
| T ₂₄ : 25 tons of BC | 19.25 | 9.77 | 0.40 | 0.25 |
| S. Em ± | 0.079 | 0.11 | 0.02 | 0.012 |
| C.D@ 5% | 0.23 | 0.33 | NS | 0.036 |

Note: BC-Biochar, VC-Vermicompost and FYM- Farm yard manure.

4.2.6.7 DTPA extractable manganese

After the harvest of field bean, soil manganese content revealed the significant gap between the treatments at various biochar levels. Manganese's numerically higher significance (9.95 mg kg^{-1}) was observed in T₂₂ treatment where 15 tons of BC was added. The lower manganese content (8.45 mg kg^{-1}) was found in treatment (T₂₁), which received 25 tons of BC + FYM + VC + Ganajeevamrutha followed by T₁₁ (8.47 mg kg^{-1}), which received 15 tons of BC + FYM + Ganajeevamrutha.

4.2.6.8 DTPA extractable zinc

The zinc content of post-harvest soil did not differ significantly due to application of varied levels of biochar (Table 4.12). The higher value of Zn (0.44 mg kg^{-1}) was found in treatment T₁₃, where 20 tons of BC + FYM + VC was applied followed by T₁₆ (0.43 mg kg^{-1}) where 25 tons of BC + FYM + VC was applied. The lower zinc content (0.34 mg kg^{-1}) was noticed in treatment (T₁₇) which received 25 tons of BC + FYM + Ganajeevamrutha followed by T₁₈ (0.35 mg kg^{-1}) received 25 tons of BC + VC + Ganajeevamrutha.

4.2.6.9 DTPA extractable copper

The DTPA extractable copper content was found to be significantly varying between the treatments in soil due to the application of varied levels of biochar (Table 4.12). Significantly higher value of available copper was recorded in T₁₃ (0.27 mg kg^{-1}) received 20 tons of BC + FYM + VC and was on par with T₂₂ (0.26 mg kg^{-1}) received 15 tons of BC. The lower value of copper was recorded in T₁₇ (0.17 mg kg^{-1}) which received 25 tons of BC + FYM + Ganajeevamrutha alone.

The significant increase in copper content of soil by application of biochar could increase the soluble organic carbon; thereby resulting in the mobilization of Cu. Copper is strongly chelated by organic carbon and is less subjected to adsorption process. Beesley and Marmiroli (2011) also reported dependence of Cu content on soluble C and pH.

There was no clear trend in micronutrient content associated with the application of various types and levels of biochar modified in neutral pH soils. Changes in micronutrient content in soils to which various biochars have been applied can be attributed to physical and chemical properties. Biochars have high surface area, high metal affinity, high nutrient retention capacity, presence of acidic and basic functional groups, which can immobilize micronutrients in soil due to their ability to alkalinize soil. The mechanism of metal immobilization by applying these biochars were also reported by Park *et al.* (2011), Vithanage *et al.* (2014), Cao *et al.* (2009), Novak *et al.* (2009) and PazFerreiro *et al.* (2014).

4.3 To study the characterization of Paddy Bio oil.

Bio oil was separated from the liquid mixture using 4 ml toluene and the bio oil was analyzed using GCMS and the result is represented in the Table 4.13.

Wang *et al.* (2004) analyzed the composition of bio-oils from the pyrolysis of *Fraxinus mandshurica* by gas chromatography-mass spectrometry (GC-MS) and illustrated the similarities of the main contents, which are flakes, such as furfural, dimethyl phenol, 2 methoxy 4 methyl phenol, eugenol, cedrol, furanone, etc. constitute a large proportion of each bio-oil. Most of the components identified were phenol with the ketone and aldehyde groups attached, and almost all of the functional groups showed the widespread presence of oxygen. On the other hand, the analysis showed that the abundant aldehydes and ketones make the substance particularly hydrophilic and highly hydrated, which makes it difficult to remove water. Luo *et al.* (2004) showed that the relevant information on rice straw bio oil was not available, the bio-oil from *Prunus indicus* comprised of laevoglucose, furfural, phenols (with methyl, methoxy and/or propenyl groups), aldehydes (including benzaldehyde with methyl and/or hydroxyl) and vanillin according to the GC-MS analysis.

Table 4.13: Relative content of main compounds in organic composition of bio-oil produced from Paddy straw by GC-MS.

| Peak No. | Wt % | Name |
|----------|-------|---------------------------------------|
| 1 | 2.24 | 2-Cyclopenten-1-one, 2-methyl- |
| 2 | 2.85 | Butanoic acid, 4-hydroxy- |
| 3 | 4.50 | 2-Cyclopenten-1-one, 3-methyl- |
| 4 | 27.23 | Phenol |
| 5 | 1.77 | 2-Furanone, 2,5-dihydro-3,5-dimethyl- |
| 6 | 4.76 | 2-Cyclopenten-1-one, 2,3-dimethyl- |
| 7 | 6.50 | 2-methyl- Phenol |
| 8 | 16.95 | p-Cresol |
| 9 | 8.41 | Phenol, 2-methoxy- |
| 10 | 1.93 | 4-Hexen-3-one, 4,5-dimethyl- |
| 11 | 1.64 | 2,3-dimethyl- Phenol |
| 12 | 8.99 | 4-ethyl- Phenol |
| 13 | 2.12 | Phenol,3,5-dimethyl- |
| 14 | 1.79 | 2-Methoxy-5-methylphenol |
| 15 | 1.52 | L-.alpha.-Terpineol |
| 16 | 1.97 | 4-ethyl-2-methoxy- Phenol |
| 17 | 3.25 | 2,6-dimethoxy- Phenol |
| 18 | 1.59 | Apiol |

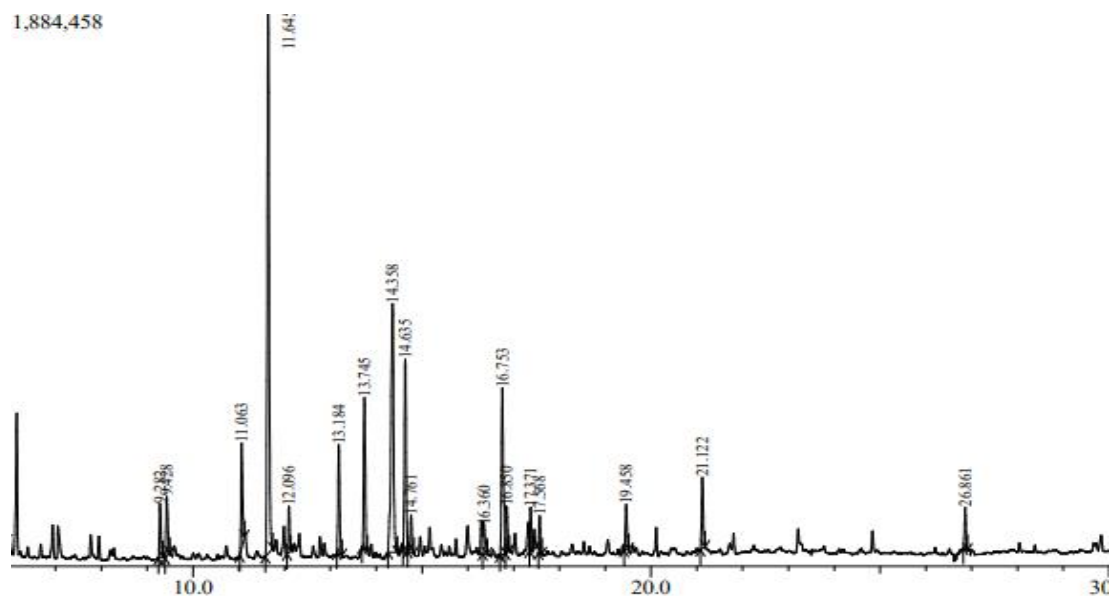


Fig. 4.5: GC-MS graph for bio oil.

The bio oil indicates that the bio-oil is a complicated organic compound that mainly consists of water, acids and heterocyclic substances and found to be rich in phenol with 27.23% by weight followed by p-cresol 16.95% on par with 4-ethyl- Phenol having 8.99%, and the lowest amount of L-.alpha.-Terpineol (1.52%) was recorded.

Bio-oils have received extensive recognition from international energy organizations around the world for their characteristics as fuels used in combustors, engines or gas turbines and resources in chemical industries. Many substances can be extracted from biooil, such as phenol used in the plastics industry, volatile organic acid to form a film forming agent, laevoglucose, hydroxyl acetaldehyde and some additives are applications in the pharmaceutical industry, synthetic fibres or fertilizers and flavouring agents in food products.

Yan *et al.*, (2001) The 99.7% of bio-oil, a complex mixture containing carbon, hydrogen and oxygen, is composed of acids, alcohols, aldehydes, esters, ketones, sugars, phenols, guaiacols, syringols, furans, lignin derived phenols and extractible terpene with multi-functional groups.

V SUMMARY

The research entitled “**Utilization of products obtained from pyrolysis of paddy straw.**” was undertaken during 2020-2021. In this study, a potted experiment was conducted in the GKVK at Bangalore. The observations obtained in the investigation are presented below.

The chemical study revealed that paddy straw biochar reported a strongly alkaline fraction having pH of 10.29 with an electrical conductivity of 3.287 dSm^{-1} and a fixed C content of 6.1 per cent. Nitrogen, phosphorus, and potassium concentrations were 0.41, 0.04 and 3.83 per cent, respectively. Paddy straw biochar also recorded good Ca, Mg and S values of 2.39, 0.24 and 0.07 percent respectively.

Different soil physical properties namely bulk density (BD) and maximum water holding capacity (MWHC) varied significantly due to different biochar dosage application. Between the various treatments, treatments receiving 25 tons of BC + VC+ Ganajeevamrutha documented lower bulk density (1.29 Mg m^{-3}) and higher water holding capacity (42.03%) over the rest of the other treatments.

The chemical properties of soil have been significantly influenced due to the use of paddy straw biochar at varied levels in combination with organic manures. The soil pH (9.50) and EC (electrical conductivity) (0.19 dSm^{-1}) were recorded significantly higher in T₂₁ treatment, which received 25 tons of BC + FYM + VC+ Ganajeevamrutha in the present research.

The available organic carbon per cent in the present study has significantly recorded high in treatment T₂₁ (6.42 g kg^{-1}) which received 25 tons of BC + FYM + VC+ Ganajeevamrutha. The lower percentage was recorded in T₃ (5.4 g kg^{-1}) which received 15 tons of BC + Ganajeevamrutha as an amendment.

Available soil nitrogen, phosphorus and potassium content differed considerably due to application of paddy straw biochar. The available nitrogen (297 kg ha^{-1}), phosphorus (63.25 kg ha^{-1}) and potassium ($173.95 \text{ kg ha}^{-1}$) recorded the highest in T₂₁

treatment, which received 25 tons of BC + FYM + VC+ Ganajeevamrutha. Significantly lower amount of nitrogen, phosphorus, and potassium were observed in T₂₂ (255.5, 48.85, 158 kg ha⁻¹) which received 15 tons of BC alone.

Exchangeable calcium, magnesium and available sulphur contents of soil varied significantly due to paddy straw biochar. Highest exchangeable Calcium (4.22 c mol (p+) kg⁻¹), Magnesium (2.98 c mol (p+) kg⁻¹) and available Sulphur (37.82 mg kg⁻¹) concentration observed in treatment receiving nutrients 25 tons of BC + FYM + VC+ Ganajeevamrutha (T₂₁) and significantly low value was observed in which received 15 tons of BC only.

Significant variation in micronutrient content was observed with application of varied levels of paddy straw biochar amended in neutral pH soil. Application of paddy straw biochar in neutral soil reduced micronutrient. Significantly, higher value of Iron (19.6 mg kg⁻¹), manganese (9.94 mg kg⁻¹) was recorded in T₂₂ which received 15 tons of BC only, zinc (0.44 mg kg⁻¹), and Cu (0.27 mg kg⁻¹) was recorded in T₁₃ receiving 20 tons of BC + FYM + VC.

In the present experiment, growth parameters like plant height, number of leaves and number of branches of field bean were significantly recorded highest at 30, 60 and at harvest of the crop growth stage in the treatment T₂₁ receiving 25 tons of BC + FYM + VC+ Ganajeevamrutha.

The number of pods per plant (16) was recorded significantly highest in treatment (T₂₁) which consists of 25 tons of BC + FYM + VC+ Ganajeevamrutha and weight of pods per plant (24g) was recorded significantly highest in T₂₀ which received 20 tons of BC + FYM + VC + Ganajeevamrutha and significantly lowest value was recorded in T₂₂ (6), which did not receive an external source of nutrients but 15 tons of biochar and T₁ (9.6g) received 15 tons of BC + FYM respectively.

This study shows the effects of paddy straw biochar combined with organic inputs on neutral soil and crop yield providing positive responses. For other soil physical and chemical parameters, nutrient uptake and crop yield in all treatment combinations

showed, higher the dosage of biochar, lead to increase in the soil nutrients and the yield as observed. T₂₁ with 25 tons BC + FYM + VC + Ganajeevamrutha had the best results of all treatments. However, the field bean, in the T₂₀ treatment which received 20 tons of BC + FYM + VC + Ganajeevamrutha had a higher overall yield. The use of paddy straw biochar in soil management and agricultural production has been shown to achieve higher field bean yields, improving the physical and chemical properties of the soil by increasing fertility and productivity of the soil.

Practical utility:

*T₂₁ with 25 tons BC + FYM + VC + Ganajeevamrutha and T₂₀ which received 20 tons of BC + FYM + VC + Ganajeevamrutha are required to be further studied under field condition.

*Along with that T₅ (20 tons of Biochar + VC), T₉ (25 tons of Biochar + Ganajeevamrutha), T₁₇ (25 tons of Biochar + FYM + Ganajeevamrutha), T₁₈ (25 tons of Biochar + VC+ Ganajeevamrutha), T₁₉ (15 tons of Biochar + FYM + VC + Ganajeevamrutha) should be individually studied further.

Future line of work

- The effect of catalyst on pyrolysis at the same temperature need to be done with same feed stock.
- Much more study on Ganajeevamrutha influence on crop production and soil parameters in field condition has to be studied.
- Field experiments of same treatments can help in formulating the practical utility of the research and this has to be studied.
- There is need to evaluate the long-term effect of biochar on plant growth and yield.

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