

**Seasonal Variation in Chemical Composition
and Biological Activities of Essential Oils from
Artemisia annua L.**

Thesis

Submitted to the



**G. B. Pant University of Agriculture & Technology
Pantnagar- 263145, Uttarakhand, India**

By

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**IN PARTIAL FULFILLMENT OF THE REQUIREMENTS
FOR THE DEGREE OF**

**Master of Science
(Chemistry)**

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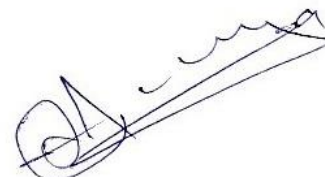

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CERTIFICATE-I

This is to certify that the thesis entitled “**Seasonal Variation in Chemical Composition and Biological Activities of Essential Oils from *Artemisia annua* L.**” submitted in partial fulfillment of the requirements for the degree of **Master of Science** with major in **Chemistry**, of the College of Post Graduate Studies, G.B. Pant University of Agriculture and Technology, Pantnagar, is a record of bonafide research carried out by **Ms. Akanksha Malhotra**, Id. No. **55503**, under my supervision, and no part of the thesis has been submitted for other degree or diploma.

The assistance and help received during the course of this investigation have been duly acknowledged.

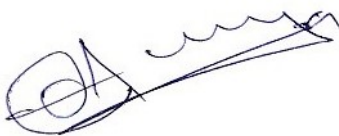
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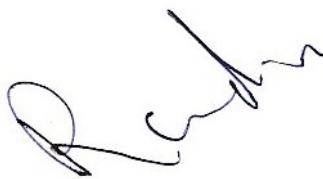
We, the undersigned, members of the Advisory Committee of **Ms. Akanksha Malhotra**, Id. No. **55503**, a candidate for the degree of **Master of Science** with major in **Chemistry**, agree that the thesis entitled “**Seasonal Variation in Chemical Composition and Biological Activities of Essential Oils from *Artemisia annua* L.**” may be submitted in partial fulfilment of the requirements for the degree.



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LIST OF ABBREVIATIONS

%	Percentage
μL	Microliter
AAREO	<i>Artemisia annua</i> rainy EssentialOil
AAWEO	<i>Artemisia annua</i> winter EssentialOil
A _o	Absorption value of control sample
A _t	Absorption value of test sample
BHT	Butylated hydroxytoluene
cm	Centimeter
DH	Diterpenoid hydrocarbons
DPPH	2,2-diphenyl-1-picrylhydrazyl
EOs	Essential oils
g	Gram
GC	Gas chromatography
GC-MS	Gas chromatography-Mass Spectrometry
IB ₅₀	The half maximal inhibition of protein denaturation
IC ₅₀	The half maximal inhibitory concentration
IPP	Isopentyl pyrophosphate
M	Meter
MH	Monoterpenoid hydrocarbons
Min	Minute
MS	Mass spectrometry
Na ₂ EDTA	Sodium salt of ethylenediaminetetraacetic acid
OD	Oxygenated Diterpenoids
OM	Oxygenated monoterpenoid

OS	Oxygenated Sesquiterpenoid
pH	Potential of hydrogen
ppm	Parts per million
SEm=	Standard error
SH	Sesquiterpenoid hydrocarbons
Sq	Square
STD	Standard deviation
v/v	Percentage by volume
w/w	Percentage by mass



Introduction



Throughout human history, plant products have always played an important role in medicine. The use of herbal medicine is becoming increasingly popular around the world. Medicinal plants as we know are known for playing vital role in betterment of human health from time immemorial. India is thought to have a long history of using plants as medicines. Approximately 2500 plant species have been reported to have medicinal value, with 750 of these species known to be the active ingredients in 14,000 published species of Ayurvedic, Siddha, and Unani medicines (**Chauhan, 1999**).

Amongst 1,75,000 flowering plants which are believed to be found in India about 3000 of them are known for their medicinal uses (**Sharma et al., 1997**). People have been using compounds obtained from plants for treating variety of diseases since prehistoric times. over 50,000 plant species are used for medicinal purposes like Analgesic (e.g., morphine), antitussive (e.g., codeine), antihypertensive (e.g., reserpine), cardiotoxic (e.g., digoxin), or antimalarial (e.g., artemisinin) compounds, are only a few of the several irreplaceable medicinal products derived from plants (**Nessler, 1994**). About 25% of the new drugs sanctioned in the past 30 years are based on a molecule of plant origin, and about 50% of the top selling chemicals obtained from study on the plant secondary metabolism (**Terryn et al., 2006**).

Majority of civilizations in developing countries like India have traditionally relied on products obtained from plants, especially from forests, for treatment of various human and livestock ailments. Variety of aromatic plants are popular for domestic and commercial uses. Collectively they are called medicinal and aromatic plants (MAPs). Out of 4,22,000 plant species reported worldwide, 12.5% are reported to have medicinal values; but only a few of them are known to be in cultivation (**Rao et al., 2004**). The isolation of the hypertensive alkaloid from sarpgandha (*Rouwolfia serpentina*), priced for the treatment of insomnia, hypertension, and insanity, was the first and most notable contribution from ayurvedic medicine (**Dev, 1999**).

With new concepts like phytotherapy and veterinary medicinal uses, aromatherapy, nutraceuticals, cosmeceuticals, and animal welfare uses, MAPs have now evolved into "industrial products."(Máthé *et al.*, 2015). MAPs are of utmost importance in curing various diseases. These plants are the resources of traditional medicines and several modern medicines have been produced indirectly from plants. Over the last 2,500 years, there have been very strong traditional systems of medicine such as Chinese, Ayurvedic, and the Unani, born and practiced, more in the eastern continent (Rashid *et al.*, 2014).

Aromatic plants, also called as herbs and spices, have been used since ages as folk medicine and also as preservatives in foods. The best known aromatic plants, originated from the Mediterranean area constitute many biologically active compounds, like polyphenolics, which have been found to possess antimicrobial, antioxidant, antiparasitic, antiprotozoal, antifungal, and anti-inflammatory properties. In aromatic and scented plants, the volatile organic compounds originate from three categories of chemicals including phenolic compounds, fatty acid derivatives, and isoprenoids (Caissard *et al.*, 2004).

About 120 plant-derived chemical compounds have been developed into modern pharmaceuticals. Natural essential oils and their fragrance are among the most exceptional products of plant metabolism, and they have had an impact on human thoughts and emotions since the dawn of civilization. India is well known for its plant resources from a long time, and is one of the world's 12 bio-diversity centers sheltering over 45,000 different plant species. The diversity of India is remarkable due to the presence of 16 different agro climatic zones, 10 vegetation zones, 25 biotic provinces and 426 biomes. Out of these, about 15,000-20,000 species have good medicinal value (Thomas *et al.*, 1999).

The medicinal properties of plants are primarily due to the production of secondary metabolites in the plants (Dar *et al.*, 2017).The study of medicinal plants continues primarily for the purpose of discovering secondary metabolites, as secondary metabolites are responsible medicinal activity of the plants. Plant products have been used in phytomedicines. These can be derived from any part of the plant,

such as the bark, leaves, flowers, seeds, and so on; in other words, any part of the plant can contain active components (**Savithramma *et al.*, 2011**).

The secondary metabolites of plants have an immense application in human health and have commercial importance in pharmaceuticals, food additives and other industrial materials (**Naik and Khayri, 2016**). The secondary compounds of plants are generally grouped on the basis of their biosynthetic pathways. An excellent example of a metabolite family is given by phenolics: because these molecules are involved in the synthesis of lignin, they are common to all higher plants. Due to their biological activities; plant secondary metabolites have been used for centuries in traditional medicine (**Bourgaud *et al.*, 2001**). They provide plants with a distinct aroma, colour, flavour, and medicinal properties, as well as protection from biotic and abiotic stresses (**Weisshaar and Jenkins, 1998**). The basis of the classification of secondary metabolites is chemical structure, composition, their solubility in different solvents, or the pathway from which they are synthesized. The main classification system is comprised of three major groups: terpenoids, alkaloids and phenolics. Phenolic compounds from plants are considered as one of largest group of secondary metabolites synthesized by fruits, vegetables, teas, cocoa and other plants that have specific health benefits. They have antioxidant, anti-inflammatory, anti-carcinogenic and other biological properties, and may protect from oxidative stress and some other diseases. Simple phenolics possess bactericidal, antiseptic and anthelmintic properties (**Kabera *et al.*, 2014**). The plant secondary metabolites includes terpenes, phenolics, nitrogen containing compounds, sulphur containing compounds etc as shown in **figure 1.1**.

The literature studies revealed that certain derivatives of nitrogenous terpenes have the potent anti- hypertensive property and may indicate a new era in the field of medicine, where as due to some insecticidal and anti-microbial activities in other terpenoids, they are able to be used as fungicides and pesticides in agriculture (**Kataev *et al.*, 2011; Bohme *et al.* , 2014**).

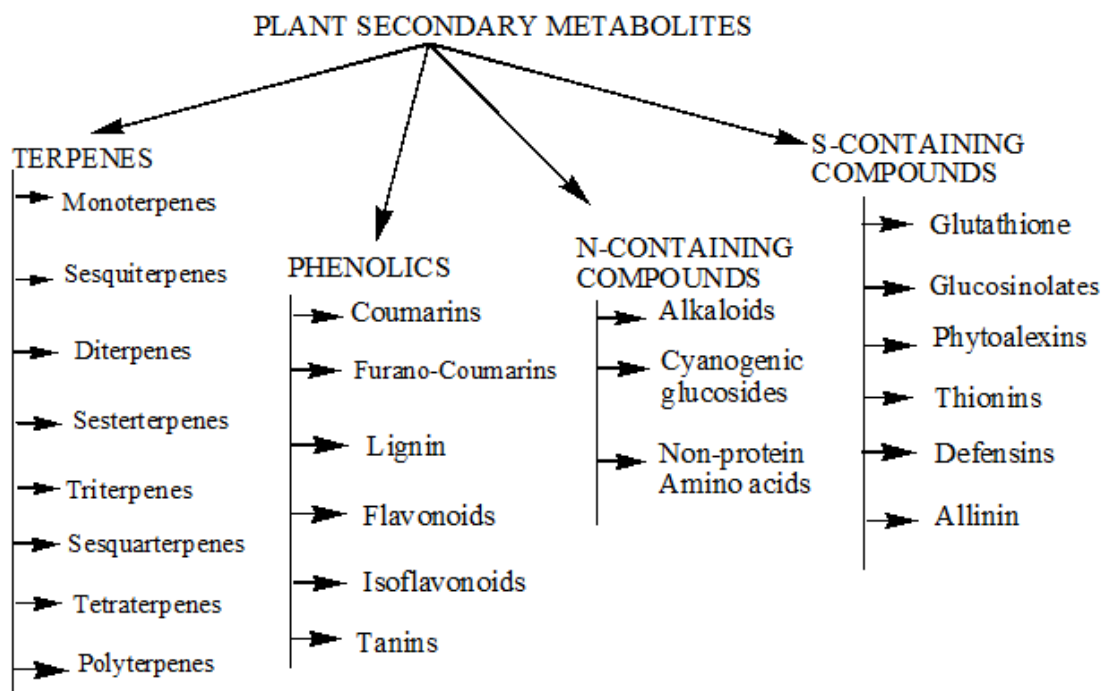


Fig. 1.1 Secondary metabolites of plants

Essential oils are a complex mixture of terpenoids formed from the secondary metabolism of plants and are responsible for their unique aroma. They are basically complex mixtures, comprised by terpenoid hydrocarbons, oxygenated terpenes and sesquiterpenes. These oils are extracted by the following methods like steam distillation, extractions with solvents and expression, of which steam distillation is the most common method as it permits the separation of slightly volatile, water immiscible substances due to low temperature distillation (**Chamorro *et al.*, 2012**). Essential oils generally help plants in protection against diseases and predators due to their strong fragrance. They are known to stimulate certain responses in the body for example lavender helps in relaxation, and *Mentha piperita* boosts exercise performance where Eucalyptus relieves helps in over coming congestion (**Meamarbashi, 2014 ; Tan and Johnston 2014**). They are volatile, natural and known to have a strong odour. They are liquid, volatile, limpid and rarely coloured, lipid soluble and soluble in organic solvents with a density less than that of water. They can usually be synthesized by all plant organs viz.. buds, flowers, leaves, stems, twigs, seeds, fruits, roots, wood or bark, and are stored in secretory cells, cavities,

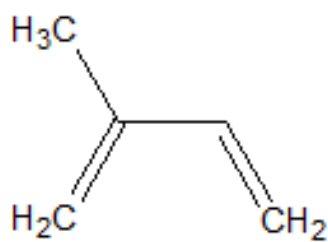
canals, epidermic cells or glandular trichomes. Essential oils are generally known for their antiseptic i.e. bactericidal, virucidal, fungicidal, and medicinal properties and their fragrance, they are often used in preservation of foods and as antimicrobial, analgesic, sedative, anti-inflammatory, and locally anesthetic remedies (**Bakkali et al., 2008**).

Chemical composition of essential oils

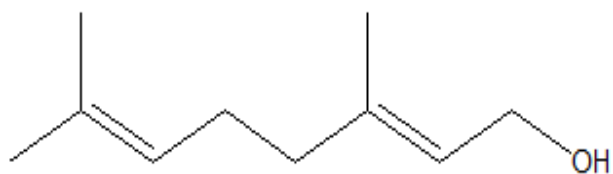
Terpenoid and phenylpropanoid derivatives are the two main components found in essential oils. Terpenoids account for about 80% of the terpenoids in most plant's essential oils. However, essential oils have significant flavour, odour, and are piquant due to the presence of phenylpropanoid derivatives. These two groups of compounds are derived from two different pathways from different primary metabolites (**Sangwan et al., 2001**). Classification of terpenoids on the basis of number of isoprene units which are important for biosynthetic pathway is given in **table 1.1**.

Table 1.1 The main classes of plant terpenoids

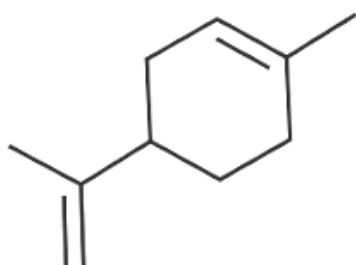
S No.	Classification	No. of carbon atoms	No. of isoprene units	Example
1.	Isoprene	5	1	Isoprene [1]
2.	Monoterpenes	10	2	Geraniol [2], Limonene [3]
3.	Sesquiterpenes	15	3	Farnesol [4]
4.	Diterpenes	20	4	Taxadiene [5], cafestol [6]
5.	Sesterterpenes	25	5	Ophiobolin [7]
6.	Triterpenes	30	6	β -amyrin [8], squalene [9]
7.	Sterols	27-30	5-6	Sitosterol [10]
8.	Tetraterpenes	40	8	β -carotene [11],lycopene [12]
9.	Polyterpenes	≥ 40	≥ 8	Gutapercha [13]



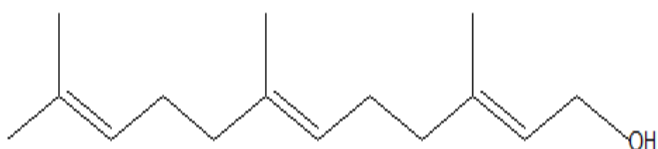
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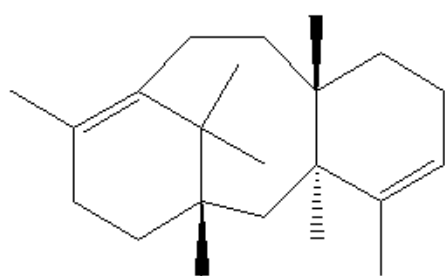
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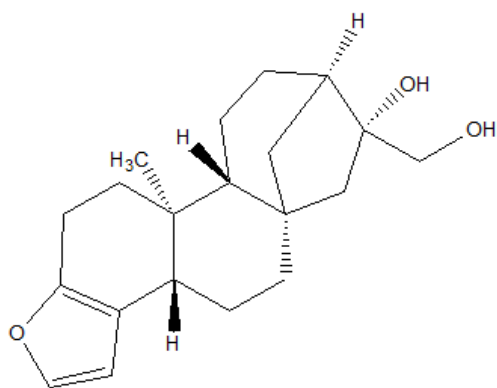
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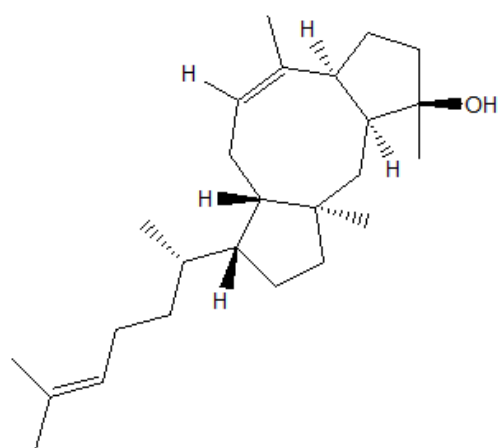
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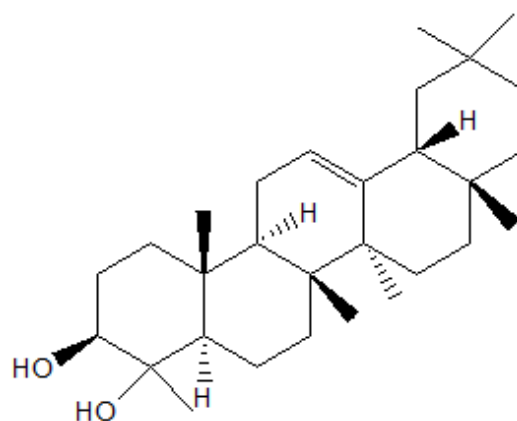
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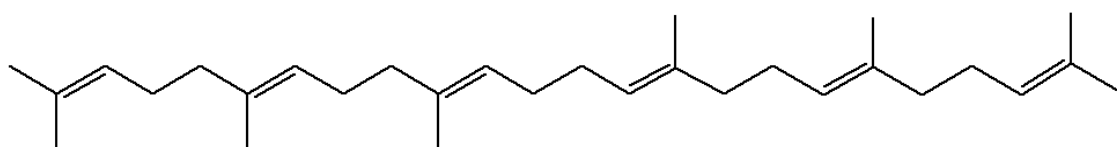
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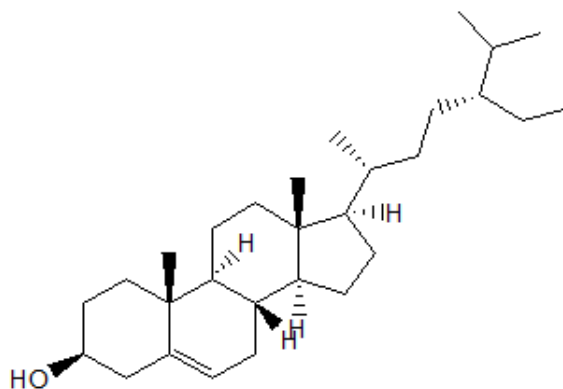
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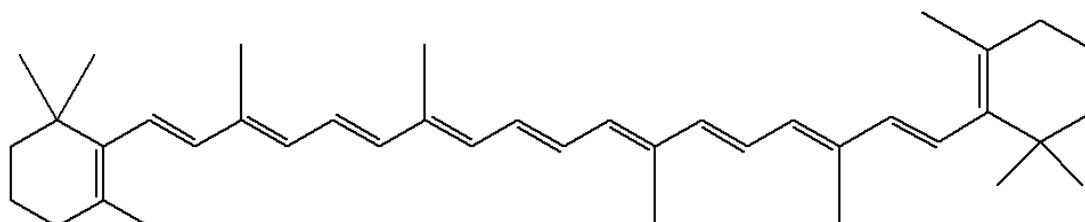
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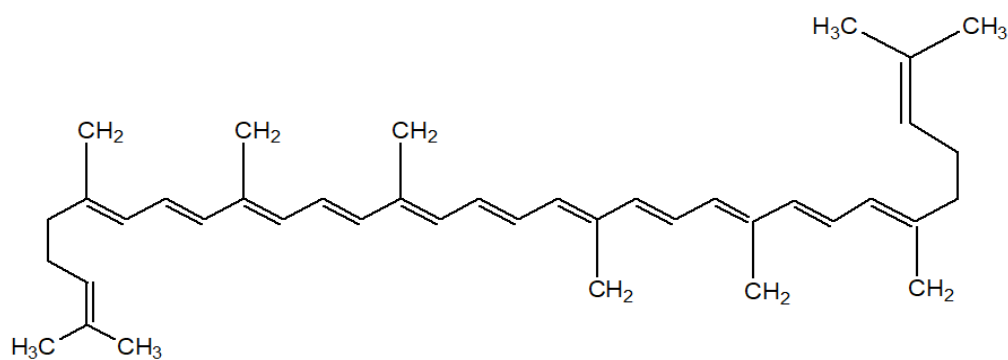
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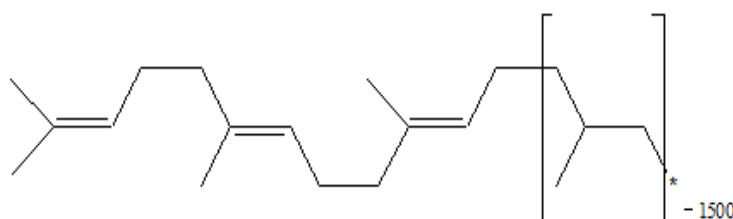
[10]



[11]



[12]



[13]

In spite of the huge structural differences between terpenoids, they all have been derived from the same C₅ skeleton of isoprene [1]. The backbone of the terpenoids are formed from two precursors: isopentenyl pyrophosphate (IPP) and dimethyl allyl pyrophosphate (DMAPP) via a series of repeats, rearrangement, and cyclization reactions. Two distinct biosynthetic pathways for the synthesis of these universal precursors have been identified, the classical mevalonate (MVA) pathway and the most recently characterized 2C-methyl-D-erythritol-4-phosphate (MEP) pathway, also known as the 1-deoxy-D-xylulose-5-phosphate (DXP) pathway. The MVA pathway is present in eukaryotes (all mammals, the cytosol and mitochondria of plants, fungi), archaea, and some eubacteria while the non-mevalonate pathway is found to occur in eubacteria, algae, cyanobacteria, and the chloroplasts of plants. The MVA pathway consists of mainly seven enzymatic reactions to convert the precursor acetyl-CoA to IPP and DMAPP, whereas while the MEP pathway converts the starting materials, pyruvate and glyceraldehyde-3-phosphate, to IPP and DMAPP via eight enzymatic reactions **fig.1.1 (Abdallah and Quax, 2017)**. There are two pathways that lead to the synthesis of isoprene. These two pathways occur in the different parts of the specialized cell. The mevalonic acid pathway occurs in the cytosol, whereas the MEP pathway occurs in the plastid (**Fokou *et al.*, 2020**).

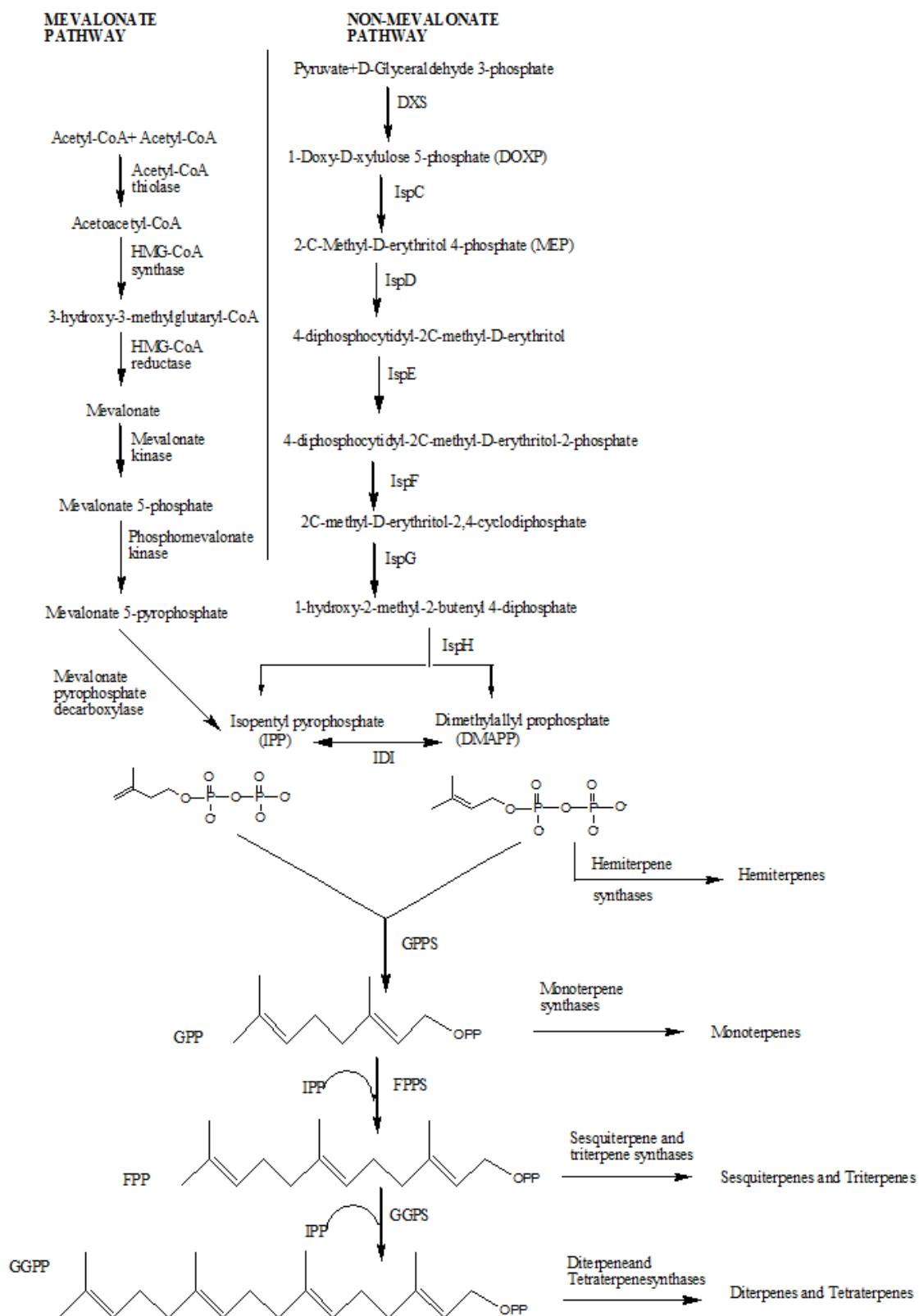


Fig. 1.2 Biosynthetic pathways for terpenoid production

The Essential oils also possess the following significant biological activities:

Oil repellent efficiency- Plant essential oils and their individual metabolites have a high potential for repellent activity against a variety of insect and arthropod species. It has been reported that among the various metabolites isolated from essential oils, the ones with the best repellent activity are oxygenated, being the hydroxyl group part of the molecules with better activities (**Nerio *et al.*, 2010**).

Culinary use- Due to the presence of several chemical compounds like cinnamaldehyde, eugenol, chavicol, safrole etc, the essential oils provide aroma to the food items (**Bakkali *et al.*, 2008**).

In Perfumery- Some constituents of essential oils like D-limonene, geranyl acetate or D-carvone are used in perfumes, creams, soaps, as fragrances for cleaning products in household and as industrial solvents (**Bakkali *et al.*, 2008**).

In Aroma therapy- Aroma therapy is one of the commendatory therapies which utilize essential oils as the major therapeutic factor in treatment of various diseases. This therapy has a calming effect on person's mind and the body. (**Ali *et al.*, 2015**).

As a food Preservative- Several EOs have been used as a food preservative for meat and meat products, vegetables and fruits as well as for dairy products (**Falleh *et al.*, 2020**).

Allelopathic affects- Allelopathy is mechanism of interference in which plants produce certain chemical compounds that effect other plants (**Gholami *et al.*, 2011**). It has been reported from various studies that chemicals found in essential oils of plants have allelopathic effects on germination of seeds (**Rahimi *et al.*, 2013**).

As pheromones- Several essential oil components are utilised by insects as major or minor components of their pheromones and thus attract conspecifics to them. Certain species of termites *Reticulitermes* were found to possess (-)- α -Pinene, (-)- β -pinene, (-)-camphene, myrcene, (-)-limonene and both (Z)- and (E)-ocimene, (+)- γ -cadinene,

, (-)-germacrene A and B, γ -himachalene and β -bisabolene in their pheromones (Muller and Buchbauer, 2011).

Apart from the above mentioned activities, the essential oils have been known to possess many more biological activities as depicted in **fig.1.3**

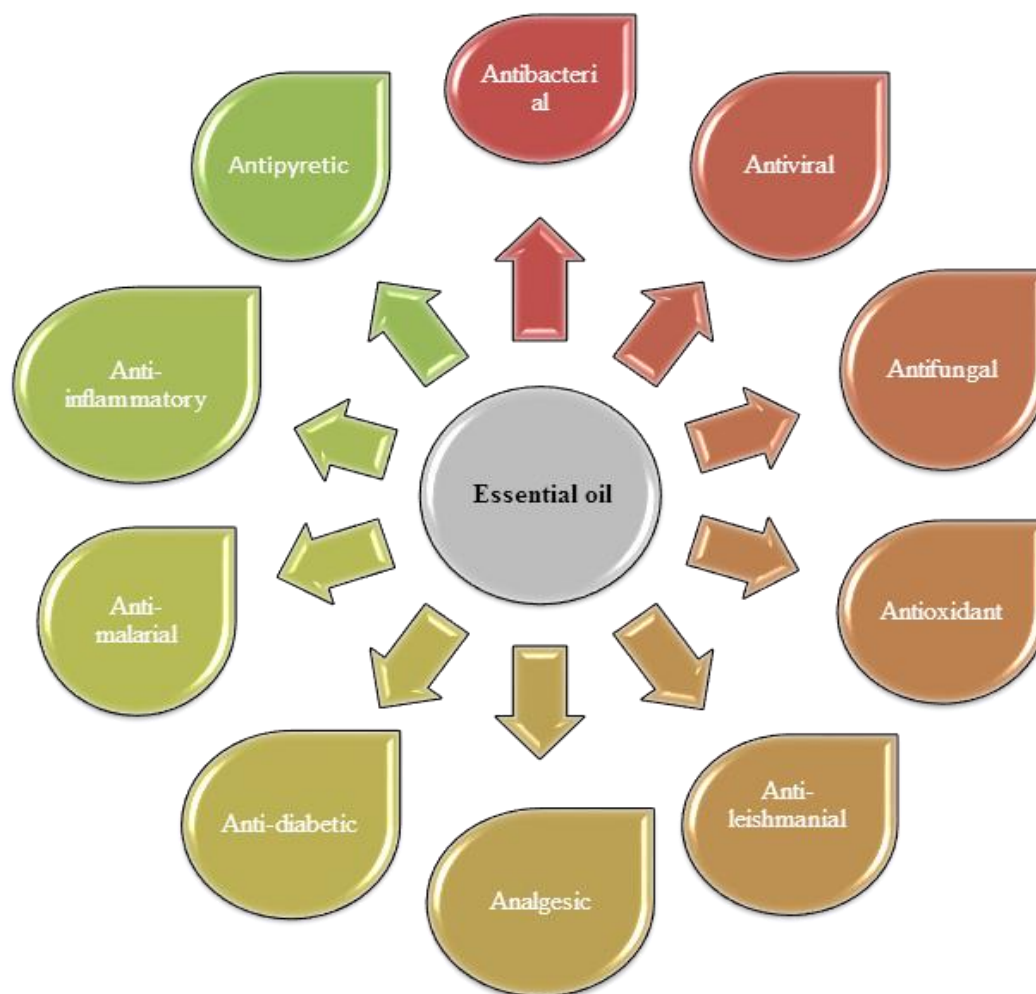


Fig 1.3 Different biological activities of essential oils

Based on the thorough search of literature and ethanobotanical applications, its traditional applications, it becomes necessary to generate more data based on phytochemical analysis and various biological potential of essential oil from aerial parts of *A.annua*, I have taken this herb for the study of its seasonal variation in chemical composition of essential oils and determination of different biological activities with following objectives:

OBJECTIVES

1. Collection of fresh plant material from its natural habitat.
2. Taxonomic identification of collected plant material and maintenance of herbarium.
3. Extraction of essential oil by using hydrodistillation/steam distillation through a Clevenger type apparatus.
4. Phytochemical analysis of essential oil using gas chromatography and gas chromatography- mass spectrometry.

Different biological activities (anti-inflammatory, anti-oxidant, anti-diabetic, antifeedant and nematicidal activities will be studied using standard methods reported and generally are being practiced by various researchers).



*Review
of
Literature*



2.1 Family: Asteraceae

The Asteraceae Bercht. & J. Presl (= compositae Giseke) family is also known as ‘daisy’, ‘aster’ or ‘composite’ family (Xu and Chang, 2017). The Asteraceae family is broadly divided into two large subfamilies: Cichorioideae (syn. Lactucoideae, multisieae, Cardueae) and Asteroideae (Inuleae, Astereae, Anthemideae). It is one of the most numerous plant families. This family is found all over the world, but it is especially prevalent in the Mediterranean, Eastern Europe, and Asia Minor. There are approximately 25,000 species in the family under about 1,568 genera. (Mabberley, 2017). The majority of the species are herbaceous, but some are shrubs or even trees, and they are mainly found in the tropical regions of North and South America, Africa, and Madagascar, as well as on the islands of the Atlantic and Pacific Oceans. Generally the species of Asteraceae have a capitulum or head inflorescence, an inferior, unilocular ovary with one ovule, and with few exceptions fused anthers surrounding the style (Harris, 1995). Some of these species, such as Chamomile (*Matricaria recutita* L.), Yarrow (*Achillea millefolium* L.) and Wormwood (*Artemisia nilagirica* (C.B Clarke Pamp.)) are reported to have medicinal properties. Essential oils extracted from the Asteraceae species have been reported to be used against several ailments and sufferings.

The majority of percentage of this family is herbaceous, while about 2 percent are trees or shrubs (Lawrence, 1973). Asteraceae members typically have leaves that are alternate, rarely opposite, exstipulate, and rarely stipulate; inflorescence capitulum or head surrounded by involucre of bracts; ray and disc florets, flower tubular or ligulate, flowers bisexual or unisexual or outer male or female, pentamerous, actinomorphic or zygomorphic, calyx modified topappus, corolla gamopetalous, petallobes 5, stamens 5, epipetalous, usually ditheous, filament free and anthers united i.e. syngenesious, introrse, ovary unilocular, inferior with solitary ovule in basal placentation, style slender stigma bifid; fruit cypsela. The stem can be erect, or prostrate, herbaceous or woody (*Artemisia*), hairy, sometimes with latex. Stem tubers are also

present (*Helianthus*); tubers are edible (*H. tuberosus*); cylindrical, solid or fistular, stem may be leaf-like (*Baccharis*). The anatomical features that can be usually seen in Asteraceae are: (a) presence of different types of glandular or covering trichomes; (b) papillae (c) anomocytic, anisocytic and rarely heliocytic stomata; (d) presence of hydathodes; (e) presence of hypoderm; (f) homogeneous or heterogeneous mesophyll; and (g) vascular bundles with parenchymatic sheath composed by large cells (**Metcalfe and Chalk, 1950**). This family contains species that are extremely important in the fields of nutrition, cosmetics and pharmacy due to the production of essential oils (**Milan et al, 2006**). The systematic position of family Asteraceae is given in **table 2.1** and the diversity of the Family Asteraceae is depicted in **table 2.2** respectively.

Table 2.1 Systematic position of family Asteraceae

Kingdom	Plantae
Sub-Kingdom	Tracheobionta
Division	Magnoliophyta
Sub-division	Angiospermae
Class	Dicotyledons
Sub-class	Asteridae
Order	Asterales
Family	Asteraceae

Table 2.2 Diversity of the family Asteraceae

SN.	Distribution	No. of genera	No. of species	References
1.	World	1568-1,911	25,000-32,913	The Plant List (2013); Christenhusz and Byng (2016); Mabberely (2017)
2.	India	193	999	Karthikeyan et al., (2009)
3.	Uttarakhand	125	376	Uniyal et al., (2007)

2.2 Genus *Artemisia* L.

Review of Literature

Artemisia L. is a large, diverse and economically important genus of the family Asteraceae. It has more than 540 species in the world (Mabberley, 2017). *Artemisia* is a wind pollinated cosmopolitan genus that is primarily found in temperate areas of the northern hemisphere's mid to high latitudes, colonising arid and semiarid landscapes, with just a few representatives in the southern hemisphere. Its centre of diversification is Central Asia, with the Mediterranean region and North West America serving as secondary areas of speciation (Hayat *et al.*, 2009). The geographical distribution of genus *Artemisia* illustrated in the **fig. 2.1**.

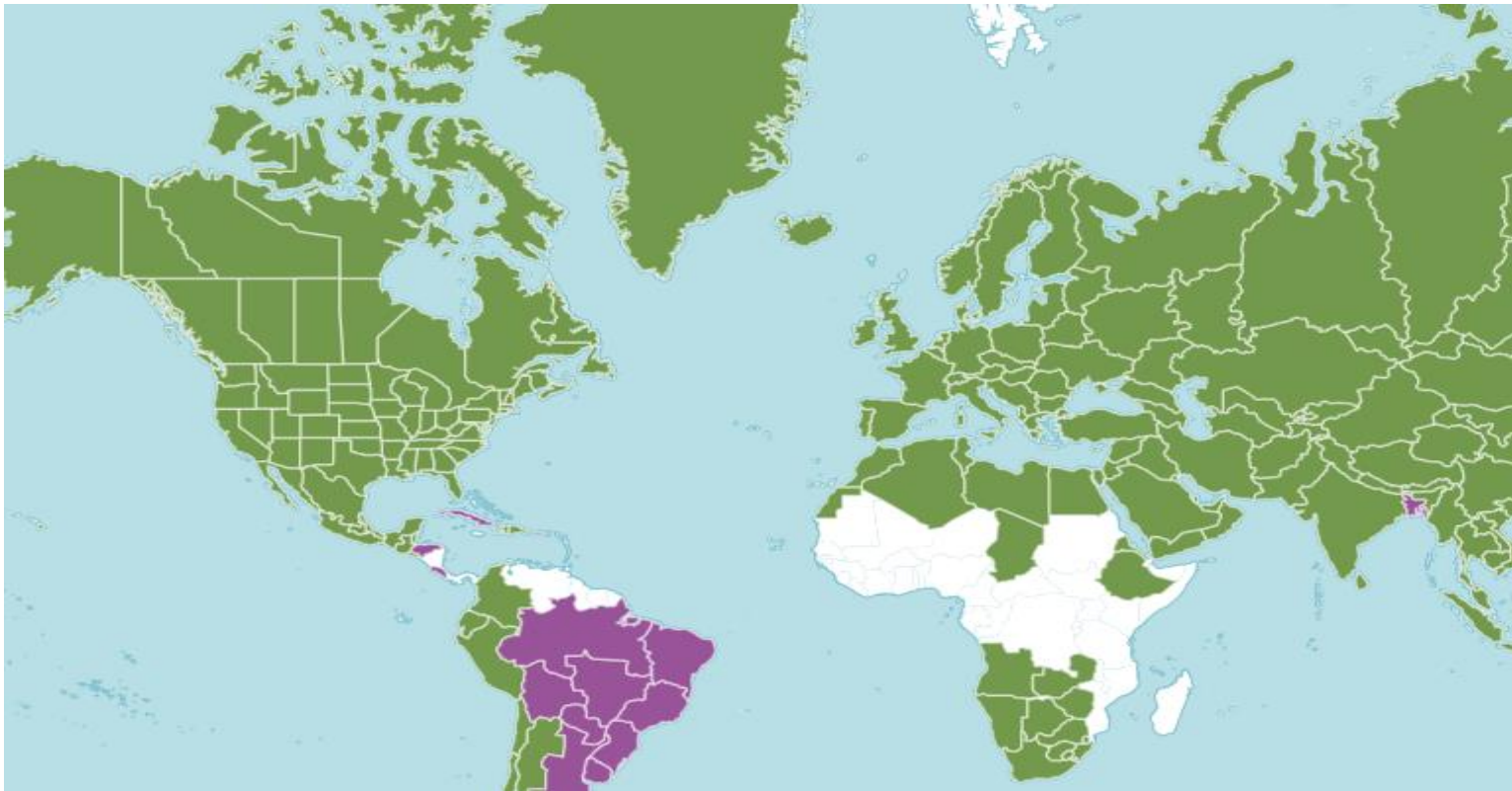
The species of *Artemisia* are generally perennial, biennial and annual herbs or small shrubs. The genus *Artemisia* is characterized by a wide range of morphological and phytochemical variability, which is associated with different geographical origins of the samples. The genus displays a huge ecological plasticity, with species occurring from sea level to high mountains and from arid zones to wetlands. Phytochemical analysis of the genus *Artemisia* revealed that the *Artemisia* species consists mainly terpenoids, flavonoids, coumarins, caffeoylquinic acids, sterols and acetylenes. Amongst numerous species of *Artemisia*, *A. absinthium*, *A. afra* Jacq., *A. annua* L., *A. maritima* L. and *A. scoparia* Waldst. & Kit. are principally rich in terpenoids. A new class of highly effective antimalarials has been discovered using artemisinin, which is an endoperoxide sesquiterpene lactone isolated from the Chinese medicinal plant *A. annua*. Combination therapies based on artemisinin are now widely regarded as the best treatment against *Plasmodium falciparum* malaria (Bora and Sharma, 2011).

Species of the genus *Artemisia* are mainly known by the common names such as mugwort, wormwood and sagebrush. Due to the presence of terpenoids and sesquiterpene lactones, most of the species possess strong aromas and bitter tastes, which discourage herbivory, and may have had a selective advantage. Major components consisted of several terpenes, terpenoids and phenolic compounds, where as 1, 8-cineole, beta-pinene, thujone, artemisia ketone, camphor, caryophyllene, camphene and germacrene-D were reported dominant in several species of *Artemisia*. The different essential oils from the genus *Artemisia* and their compounds have been reported as effective antimicrobial, insecticidal and antioxidant agents. Antioxidant activity found in oils is basically due to presence of phenolic compounds (Pandey and Singh, 2017).

The genus *Artemisia*, which belongs to the Anthemideae tribe, contains important medicinal plants that are currently the focus of phytochemical research due to their biological and chemical diversity, as well as the production of essential oils. The strong and aromatic odour of some *Artemisia* species is caused mainly by high levels of volatile terpenes, which are constituents of their essential oils, notably in leaves and flowers. The chemical composition of essential oils derived from the *Artemisia* genus has been extensively researched in a wide range of species from around the world. Many studies have shown that *Artemisia* species show significant intraspecific variations in the terpene constituents of their essential oils. It has been reported that the quality and yield of essential oils obtained from *Artemisia* species is generally due to the influence of the harvesting season, fertilizer and pH of soils, the choice and stage of drying conditions, the geographic location, chemotype or subspecies, choice of plant part or genotype, or extraction method (Abad *et al.*, 2012). The genus *Artemisia* L. is one of the most important genera of the Asteraceae, as it is used all over the world in traditional medicine, as a source for spices (e.g., *Artemisia dracunculus* L.), ingredient in liquors (e.g., *Artemisia absinthium* L.), and as a source for artemisinin (*Artemisia annua* L.) (Panda *et al.*, 2019). Joshi (2013) reported that the chemical composition of various species of *Artemisia* from different origins exhibited the presence of 1, 8- cineole, α -thujone, β -thujone, chamazulene, davanone, artemisia ketone, germacrene-D, β -caryophyllene and caryophyllene oxide. The systematic classification of genus *Artemisia* is depicted in table 2.3 where as table 2.4 contains the *Artemisia* Species that are reported from India and Uttarakhand.

Table 2.3 Systematic classification of genus *Artemisia* L.

Kingdom	Plantae
Clade	Tracheophytes
Clade	Angiosperms
Clade	Eudicots
Clade	Asterids
Order	Asterales
Family	Asteraceae
Subfamily	Asteroideae
Supertribe	Asterodae
Tribe	Anthimideae
Genus	<i>Artemisia</i> L.



■ Native ■ Introduced

Fig. 2.1 Map showing distribution of genus *Artemisia* across the world (POWO, 2019)

Table 2.4 *Artemisia* species reported from India and Uttarakhand (Karthikeyan *et al.*, 2009; Uniyal *et al.*, 2007)

SN.	Species	SN.	Species
1.	* <i>Artemisia absinthium</i> L.	2.	* <i>Artemisia annua</i> L.
3.	* <i>Artemisia austrohimalyana</i> Ling & Puri	4.	* <i>Artemisia biennis</i> Willd.
5	* <i>Artemisia capillaris</i> Thunb.	6.	* <i>Artemisia caruifolia</i> Buch.-Ham.ex Roxb.
7.	* <i>Artemisia dracunculus</i> L.	8.	* <i>Artemisia dubia</i> Wall.
9.	* <i>Artemisia edgeworthii</i> N.P. Balakr.	10.	* <i>Artemisia elegantissima</i> Pamp.
11.	* <i>Artemisia eriocephala</i> Pamp.	12.	* <i>Artemisia filiformilobulata</i> Y.R Ling & Puri
13.	* <i>Artemisia gmelinii</i> Weber	14.	* <i>Artemisia hedinii</i> Ostenf. & Paulson
15.	* <i>Artemisia incia</i> pamp.	16.	* <i>Artemisia indica</i> Willd.
17.	* <i>Artemisia japonica</i> Thunb	18.	* <i>Artemisia macrocephala</i> Jacquem.
19.	* <i>Artemisia maritima</i> L.	20.	* <i>Artemisia myriantha</i> Wall.
21.	* <i>Artemisia nilagirica</i> (C.B. Clarke) Pamp.	22.	* <i>Artemisia roxburghiana</i> Wall.
23.	* <i>Artemisia scoparia</i> Waldst. and Kit	24.	<i>Artemisia abrotanum</i> L.
25.	<i>Artemisia amygdalina</i> Decne.	26.	<i>Artemisia austroyunnanensis</i> Ling & Y.R. Ling
27.	<i>Artemisia brunnea</i> Pamp.	28.	<i>Artemisia campbellii</i> Hook.f.
29.	<i>Artemisia demissa</i> Krasch.	30.	<i>Artemisia demissa</i> Krasch.
31.	<i>Artemisia desertorum</i> Spreng.	32.	<i>Artemisia dolichocephala</i> Pamp.
33.	<i>Artemisia dubia</i> Wall.ex Besser	34.	<i>Artemisia falconeri</i> C.B. Clarke
35.	<i>Artemisia glaucasensu</i> Hook.f.	36.	<i>Artemisia grandis</i> Pamp.
37.	<i>Artemisia hypoleuca</i> Edgew.	38.	<i>Artemisia jacquemontiana</i> Besser
39.	<i>Artemisia kohatica</i> (Klatt) Pamp	40.	<i>Artemisia laciniata</i> Willd.
41.	<i>Artemisia lactiflora</i> Wall.	42.	<i>Artemisialancea</i> Vaniot.
43.	<i>Artemisia lavandulaefolia</i> DC.	44.	<i>Artemisia leptostachya</i> DC.

45.	<i>Artemisia macrocephala</i> Jacquem.	46.	<i>Artemisia minor</i> Jacquem.
47.	<i>Artemisia mollissima</i> DC.	48.	<i>Artemisia moorcroftiana</i> Wall. ex DC
49.	<i>Artemisia nepalensis</i> Nees	50.	<i>Artemisia pamirica</i> C.Winkl.
51.	<i>Artemisia pallens</i> Wall.	52.	* <i>Artemisia persica</i> Boiss.
53.	<i>Artemisia parviflora</i> Roxb.	54.	<i>Artemisia princeps</i> Pamp.
55.	<i>Artemisia pleiocephala</i> Pamp.	56.	<i>Artemisia revolute</i> Edgew.
57.	<i>Artemisia purpurascens</i> Jacquem.	58.	* <i>Artemisia roxburghiana</i> Bess.
59.	* <i>Artemisia robusta</i> (Pamp.) Ling & Y. R.	60.	* <i>Artemisia roxburghiana</i> var. <i>hypoleuca</i> Pamp.
61.	* <i>Artemisia roxburghinia</i> var. <i>acutiloba</i> Pamp.	62.	<i>Artemisia royleana</i> DC.
63.	* <i>Artemisia roxburghinia</i> var. <i>purpurascens</i> (Jacq.) Hook.f.	64.	* <i>Artemisia salsoloides</i> Willd.
65.	<i>Artemisia sacrorum</i> Ledeb.	66.	<i>Artemisia schmidtiana</i>
67.	<i>Artemisia santolinifolia</i> Turcz.	68.	<i>Artemisia simplicifolia</i> Pamp.
69.	<i>Artemisia sericea</i> (Besser) Weber	70.	<i>Artemisia stewartii</i> C.B. Clarke
71.	* <i>Artemisia sieversiana</i> Ehrh.	72.	* <i>Artemisia strongylocephala</i> Pamp
73.	<i>Artemisia stelleriana</i> Besser.	74.	<i>Artemisia tainingensis</i> Hand. – Mazz
75.	<i>Artemisia stricta</i> Edgew.	76.	<i>Artemisia tenuifolia</i> Y.R. Ling & Puri
77.	<i>Artemisia subdigitata</i> Mattf.	78.	<i>Artemisia thomsonii</i> C.B. Clarke
79.	<i>Artemisia tainingensis</i> var. <i>nitida</i> (Pamp) Y.R. Ling	80.	<i>Artemisia tridentata</i> (USA)
81.	<i>Artemisia thellungiana</i> Pamp.	82.	<i>Artemisia velutina</i> Pamp.
83.	<i>Artemisia tournefortiana</i> Rhcb.	84.	* <i>Artemisia vestita</i> Wall.
85.	<i>Artemisia tukuchaensis</i> Kitam.	86.	<i>Artemisia vulgaris</i> T.
87.	* <i>Artemisia verlotiorum</i> Lamotte	88.	<i>Artemisia wallichiana</i> Bess.
89.	<i>Artemisia vulgaris</i> Hook.f.	90.	<i>Artemisia wadei</i> Edgew.
91.	<i>Artemisia wellbyi</i> Hemsl. & H.Pearson	92.	* <i>Artemisia elegantissima</i> var. <i>Kumaunensis</i> Pamp

*species reported from Uttarakhand

2.3 *Artemisia annua* L.

Artemisia annua L., also known as sweet wormwood, sweet sagewort and annual wormwood (Chinese: qinghao), is a common type of wormwood that is native to temperate Asia, but naturalized throughout the world. *A. annua* is a large shrub often reaching more than 2.0 m in height, usually single-stemmed with alternate branches. The leaves are aromatic with deeply dissected lamina and range from 2.5 to 5 cm in length. Leaves and flowers contain both 10-celled biseriate trichomes and 5 cell filamentous trichomes. *A. annua* is well-known medicinal plant for being as a source of antimalarial compound artemisinin (Qinghaosu), which is a cadinane-type sesquiterpene lactone with an endoperoxide bridge that currently is the most potent and effective compound against chloroquine and quinine-resistant *Plasmodium falciparum* and other malaria-causing parasites (Das, 2012).

A.annua grows naturally as a part of steppe vegetation in northern parts of Chatar and Suiyan province in China at 1,000–1,500 m above sea level.. The stem is generally cylindrical and branched. Leaves can be seen as alternate, dark green, or brownish green. Odour is characteristic and aromatic while the taste is bitter. The Chinese name of the plant is Qinghao (or Qing Hao or Ching-hao which means green herb (Dhingra *et al.*, 1999). The plant is cropped on a large scale in China, Vietnam, Turkey, Iran, Afghanistan, and Australia. In India, it is cultivated on an experimental basis in the Himalayan regions, as well as temperate and subtropical conditions (Bhakuni *et al.*, 2001).

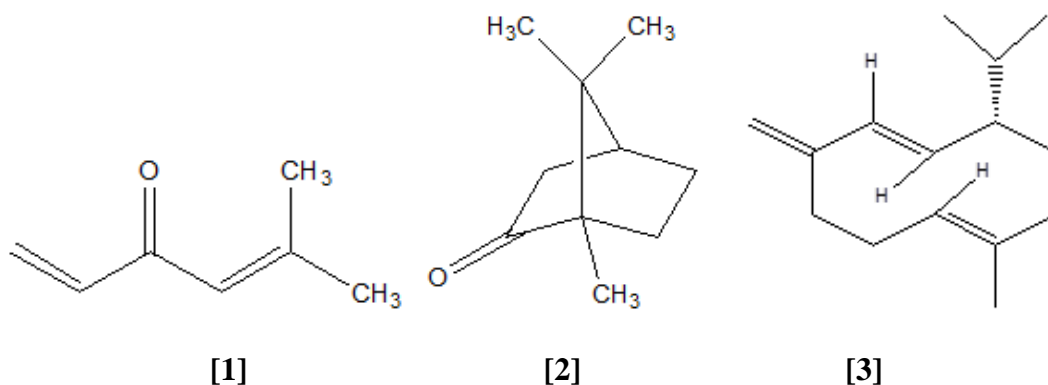
Artemisia annua is a potent antimalarial plant and used in folks, Homeopathic and Ayurvedic system of medicine. According to the literature search, in China, aqueous preparation of the dried herb was applied against fever, malaria, skin diseases, jaundice and haemorrhoids (Brisibe *et al.*, 2008; Nofal *et al.*, 2009; Rezelman and Goris, 2008). Leaves of *Artemisia annua* are widely used for the treatment of malaria and as anti-inflammatory in most part of the world (Das, 2012).

According to reports, the essential (volatile) oil of *A. annua* can yield up to 85 kg/ha. It is produced by secretory cells, particularly those in the plant's uppermost foliar portion, which has nearly twice the number of cells as the lower leaves. According to reports, capitate glands, which contain terpenoidic volatile constituents, cover 35% of the mature leaf surface. Essential oil from *A. annua* is distributed, with 36% of the total from the upper third of the foliage, 47% from the middle third, and 17% from the lower third, with only trace amounts in the main stem side shoots and roots. The yield of the oil generally ranges between 0.3 and 0.4% but it can reach 4.0% (V/W) from selected genotypes. Several studies have concluded that the *A. annua* crop can be harvested much before flowering to obtain high artemisinin yields and must be allowed to mature to obtain high essential oil yields. (Bilia *et al.*, 2014). Whole flowering plant of *A.annua* have been reported to be antihelminthic, antipyretic, antiseptic, antispasmodic, carminative, stimulant, tonic, and stomachic (Sadiq *et al.*, 2014). It is considered as an economically important crop. Due presence of Artemisinin in medicinal plant *Artemisia annua*, It has proved to be an effectual anti-malarial agent (Miraj and Alesaeidi, 2016).

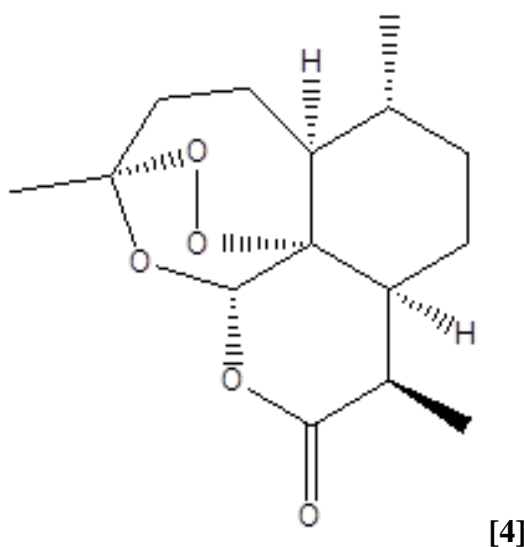
2.3.1 Phytochemical analysis

A detailed updated review of literature on chemical composition of *A. annua* is being given chronologically as under:

The volatile constituents like artemisia ketone [1], camphor [2] and germacrene-D [3] have reported in *A.annua* from China and Vietnam (Woerdenbag *et al.*, 1993).



The phytochemistry of *A. annua* has been reported to be dominated by terpenoids flavonoids, coumarins and other shikimate metabolites. The sesquiterpene, artemisinin [4], is however unique to *A. annua* (Klayman *et al.*, 1984; Balachandran *et al.*, 1987).

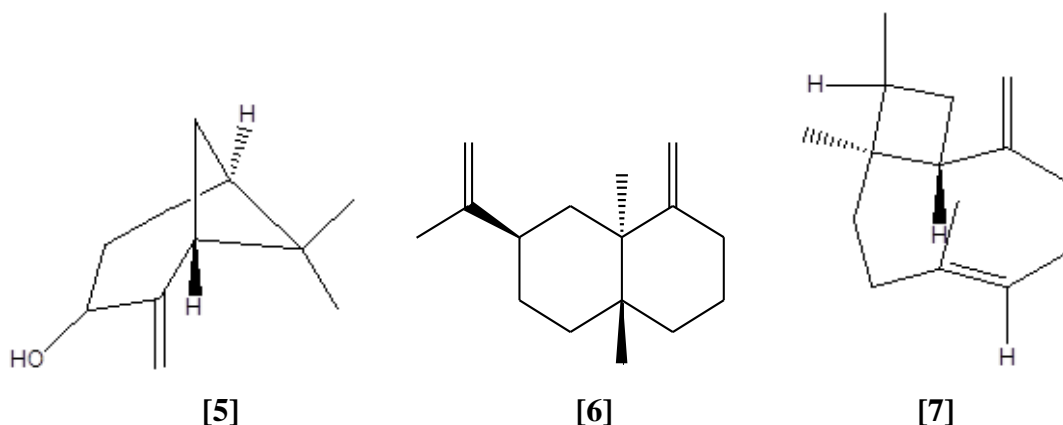


Brown (2010) reported that the main intermediate in the biosynthesis of artemisinin [4] was the bicyclic sesquiterpene-amorpha-4,11-diene, which is formed from FPP (farnesyl di phosphate) by the action of the sesquiterpene cyclase, amorpha-4,11-diene synthase (ADS). Two molecules of isopentenyl phosphate (IPP) and one dimethyl allyl phosphate (DMAPP) are condensed by farnesyl di phosphate synthase (FPPS/FPS) into farnesyl di phosphate (FPP), the C₁₅ Sesquiterpenoid precursor (Weathers *et al.*, 2006; Brown, 2010; Wen and Yu, 2011). FPP is converted to amorpha-4,11-diene by amorpha-4,11-diene synthase (ADS) via carbocation formation and cyclization (Ikram and Simonsen, 2017).

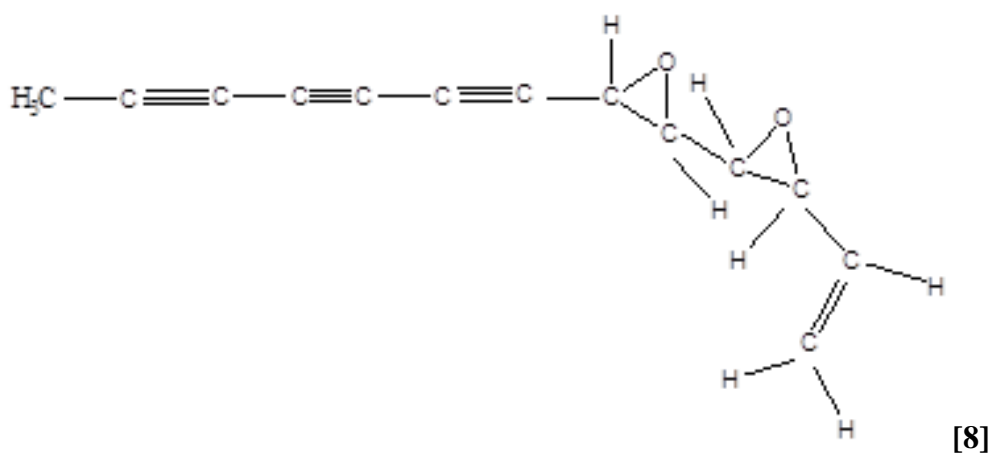
The biosynthetic pathway of artemisinin, the major constituents present in *Artemisia annua* is being depicted in **fig 2.2**.

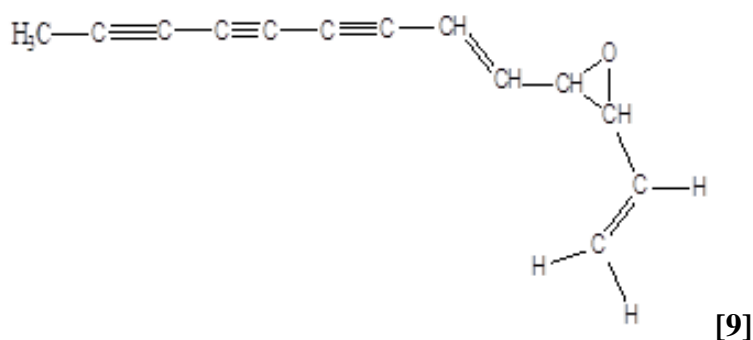
A cDNA for (3R)-linalool synthase, which converts geranyl pyrophosphate (GPP) to (3R)-linalool (250) by ionization of the pyrophosphate group, has been reported from *A. annua* (Jia *et al.*, 1999).

The phytochemical investigation of essential oil composition of *A.annua* revealed the presence camphor [2] (44%), germacrene D [3] (16%), *trans*-pinocarveol [5] (11%), β -selinene [6] (9%), β -caryophyllene [7] (9%) and artemisia ketone [1] (3%) (Juteau *et al.*, 2002).



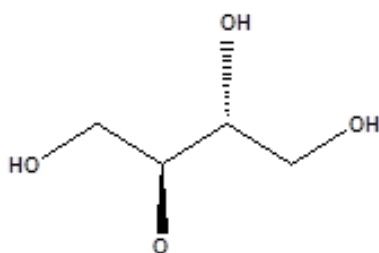
Two compounds like annuadiepoide [8] and ponticepoide [9] from leaves and seeds of *A.annua* have been reported. Few polyacetylenes have also been reported from *A. annua* (Manns and Hartmann, 1992; Brown *et al.*, 2003).



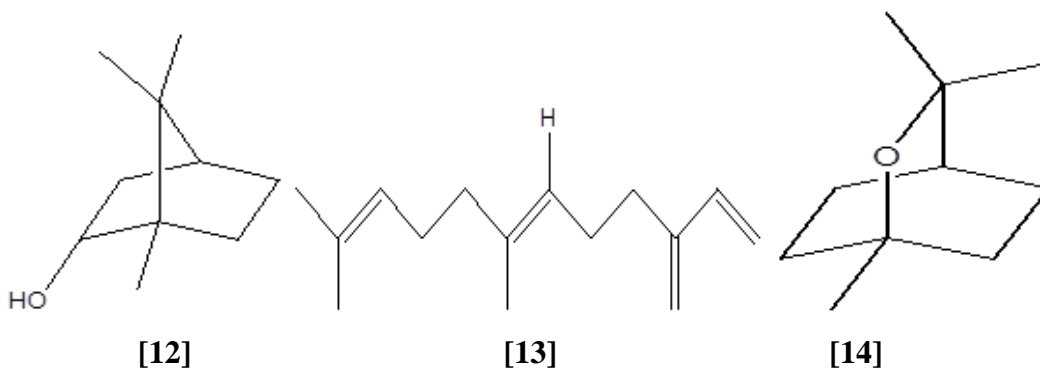


Literature search revealed that most phytochemical investigations of *A. annua* have been reported on the aerial parts (leaves and/or stems - sometimes also including the flowers) although some report revealed the analysis on the seeds (**Brown *et al.*, 2003**).

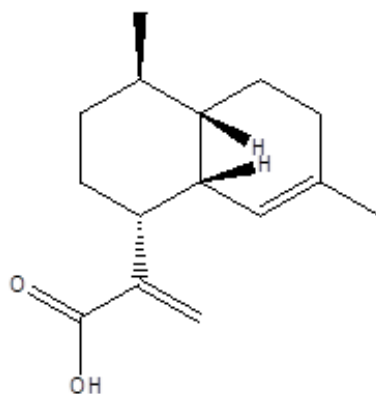
Erythritol (50.30 %) [10], camphor (7.25 %) [2], pinocarveol (4.13 %) [5], and diethoxyethane (2.18 %) [11] have reported from the non-volatile components of *A.annua* (**Haghighian *et al.*, 2008**).



It has been reported that *A. annua* from China contained borneol (15.9%) [12], and (*Z*)- β -farnesene (12.9%) [13] while same species cultivated in Iran has been reported to be very rich in camphor (36.7–48.0%) [2] and 1,8-cineole (9.4–13.9%) [14] (**Mohammadreza, 2008**).

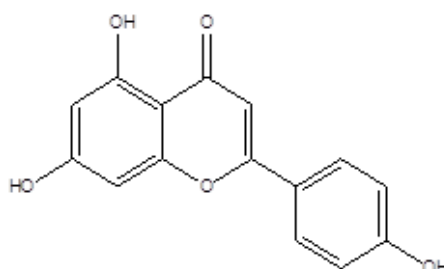


Sesquiterpenoids, including artemisinin [4], artemisinin I, artemisinin II, artemisinin III, artemisinin IV, artemisinin V, artemisnic acid [15], artemisilactone, artemisinol and epoxyarteannuinic acid have been reported as the main chemical constituents of *Artemisia annua* (Squires and Ferreira, 2009).



[15]

Ivanescu *et al.*, (2010) reported that flavonoid apigenin [16] has been reported in *A. annua*.



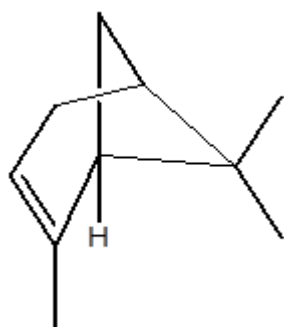
[16]

Reports suggested that the terpenoids is by far the largest group of metabolites from *A. annua*, which are biosynthetically derived from a branched isoprenoid unit (C5.); and higher terpenoids and flavonoids, which are biosynthetically derived from this same C6C3 precursor, are considered as the second largest group of metabolites from *A. annua* (Brown, 2010).

The phytochemical studies on the *A.annua* from different countries like France, Finland, Hungary, Romania, Kazakhstan, Iran, India, China and Vietnam has

been reported. It has also been reported that *A.annua* contains large group of secondary metabolites biosynthesized from (C5) units (**Brown, 2010**).

Bora and Sharma (2011) reported the constituents like α -pinene [17], camphene [18], camphor [2], germacrene D [3], artemisia ketone [1], borneol [12]and cineole[14]in the essential oil of *Artemisia annua* along with the higher yield of essential oil.

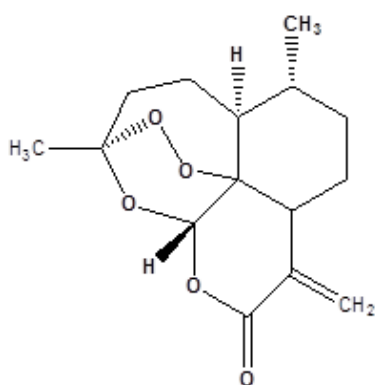


[17]

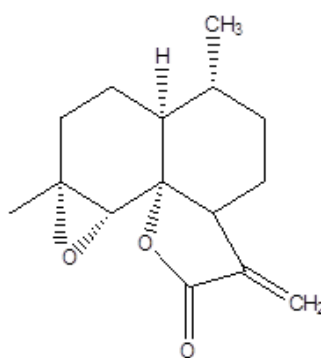


[18]

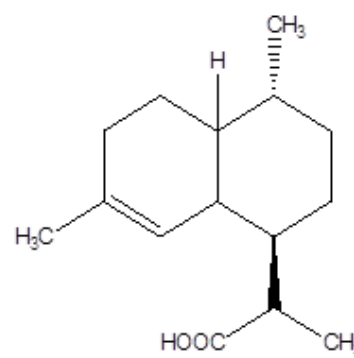
Carbonara et al., (2012) reported that the genus *Artemisia* represents a high reservoir of terpenes, phenols and acetylenes, but coumarins and flavonoids have also been identified. The most peculiar compounds are however the artemisinin related sesquiterpenes such as artemisitene [19], deoxyartemisinin, arteannuin B [20], artemisinic acid [15] and dihydroartemisinic acid [21]



[19]

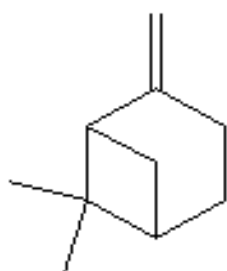


[20]

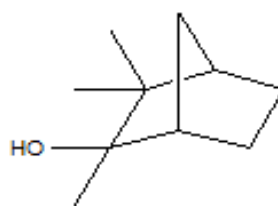


[21]

The documentation revealed that the chemical composition of *Artemisia annua* comprises of volatile and non-volatile constituents. The main compounds, which account for about 70% of the essential oils, are reported to be camphene [18], isoartemisia ketone [1], camphor [2], β -caryophyllene [7] and β -pinene [22]. In addition, other minor ingredients, such as artemisia ketone [1], 1, 8- cineole [14], and camphene hydrate [23] have also been reported in volatile parts of *Artemisia annua* (Das, 2012).

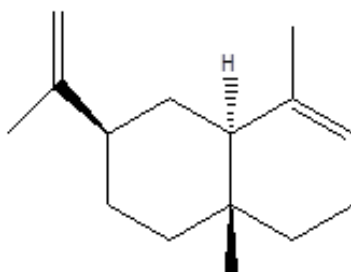


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[23]

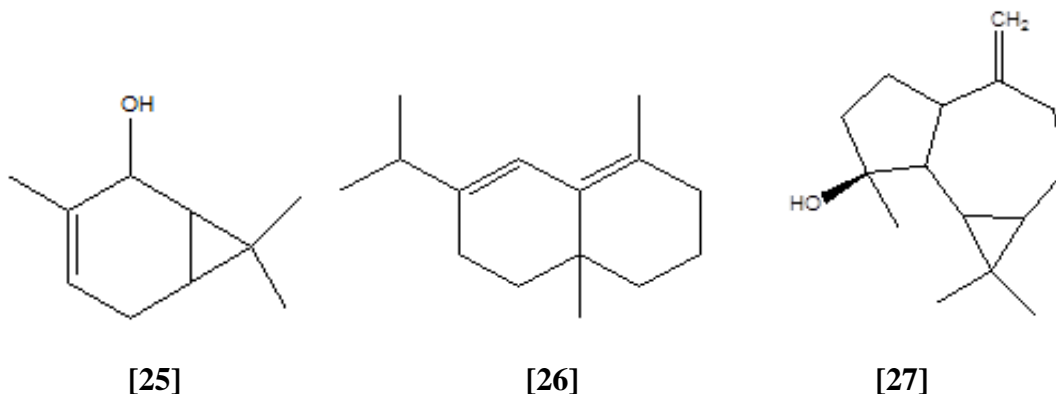
Camphor (16.30 %) [2], β -selinene (10.41 %) [6], artemisia ketone (8.79 %) [1], germacrene-D (7.14 %) [3], α -pinene (4.79 %) [17], 1,8-cineole (4.38 %) [14] and α -selinene (4.09 %) [24] have been reported as the major components in the essential oil hydrodistilled from aerial part of *A.annua*. Whereas α -pinene (12.03 %) [17], artemisia ketone (11.24 %), β -selinene (8.73 %) [6], 1,8-cineol (6.76 %) [14], camphene (4.90 %) [18] and α -selinene (4.85 %) [24] have been reported in the essential oil extracted by HS-SPME (head space-solid phase microextraction) technique (Nekoei *et al.*, 2012).



[24]

Saponins, phytosterols, carbohydrates, proteins, amino acid, and flavonoids have been reported in the alcoholic extract of *Artemisia annua* (Ashok and Upadhyaya, 2013).

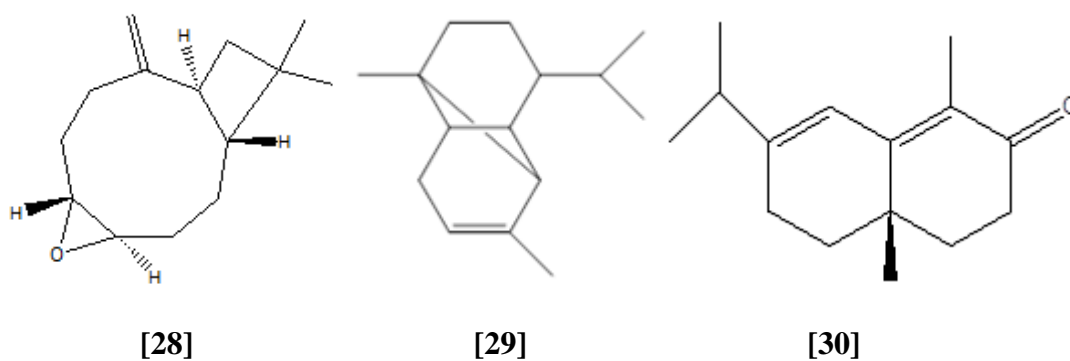
Trans-3(10)-caren-4-ol (22.3%) [25], artemisia ketone (18.6%) [1], 1,8-cineole (14.9%)[14], δ -selinene (13.0%) [26] and α -pinene (8.2%) [17], the major compounds have been reported in the seeds of essential oils of *A. annua*. where as camphor (48.00%)[2], 1,8-cineole (9.39%)[14], camphene (6.98%)[18] and spathulenol (4.89%) [27] have been reported as the major components in the essential oil derived from the dried flowering aerial parts of *A. annua* (Habibi *et al.*, 2013).



Artemisia ketone (35.7%) [1], α -pinene (16.5%) [17], 1,8-cineole (5.5%) [14] have been reported from aerial parts of *A.annua* from Serbia (Radulovic *et al.*, 2013).

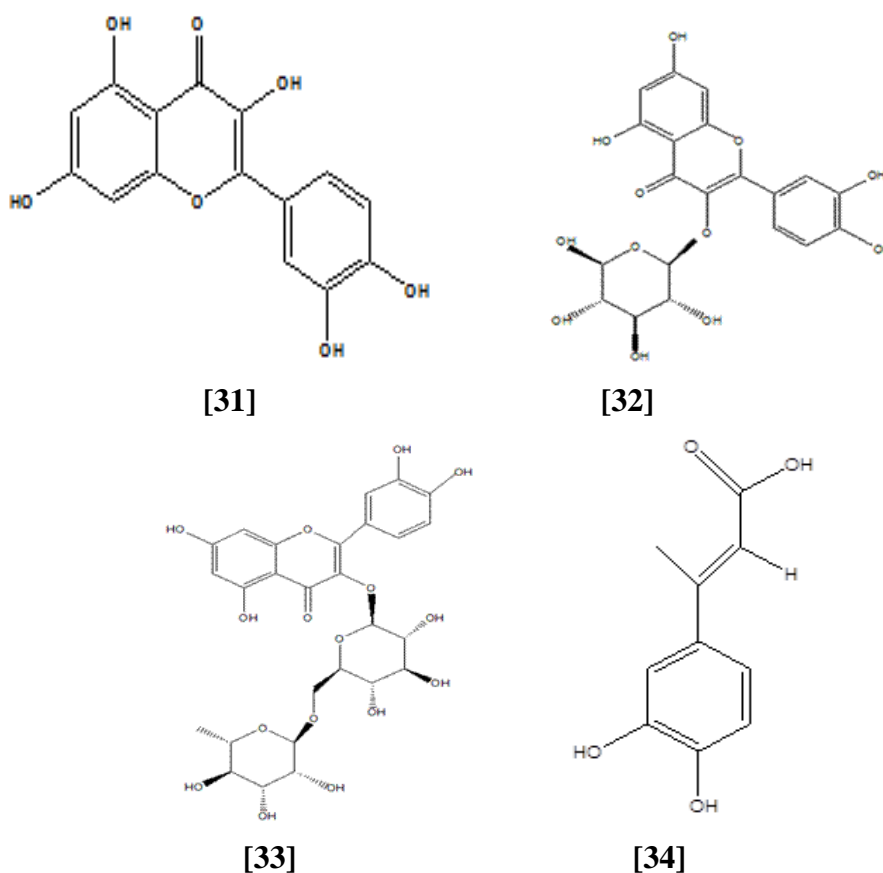
Ogbole *et al.*, (2014) reported the chemical composition as ethanolic extract from *A. annua* with the presence of contained terpenes, steroids, cardiac glycosides, carbohydrates and flavonoids.

Islamuddin *et al.*, (2014) reported that the major chemical constituent identified was represented in *A.annua* by camphor (52.06%) [2]. The other constituents present in the essential oil were β -caryophyllene (10.95 %) [7], 1,8-cineole (5.57 %) [14], β -caryophyllene oxide (4.21 %) [28], β -farnesene (3.83 %) [13], α -copaene (2.91 %) [29], β -cyperone (1.93 %) [30], α -selinene (1.54 %) [24], and *trans*-pinocarveol [5] (1.22 %).



Camphor (17.74%) [2], α -pinene (9.66%), germacrene D (7.55%) [3], 1,8-cineole (7.24%) [14], β -caryophyllene (7.02%) [7], artemisia ketone (6.26%) [1] have been reported in aerial parts *A.annua* from Romania (Marinas *et al.*, 2015).

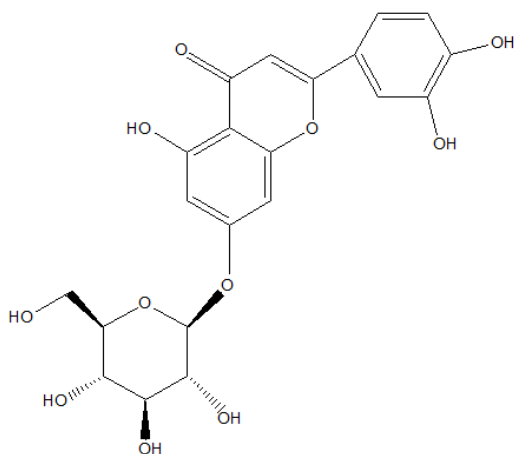
Flavonoids like quercetin (0.01 mg/g DW) [31], isoquercetin (0.107 mg/g DW) [32], rutin (0.31 mg/g DW) [33] and caffeic acid (0.03 mg/g DW) [34] have also been reported in *A.annua* using HPLC methods (Dilshad *et al.*, 2016).



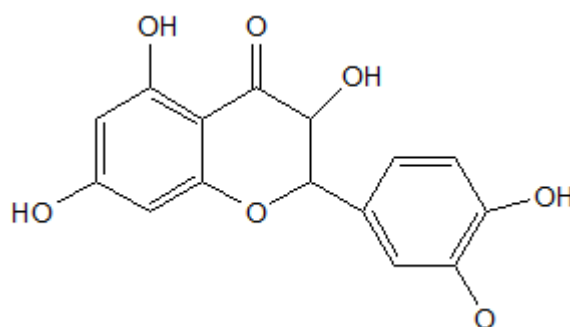
The chemical derivatives of artemisinin [4], a sesquiterpene lactone produced by *Artemisia annua*, was found to be the active ingredient in the most effective treatment for malaria. Detailed phytochemical analysis of two contrasting chemotypes a low-artemisinin production (LAP) chemotype and a high-artemisinin production (HAP) chemotype of *A. annua* resulted in the characterization of over 80 natural products by NMR (Czechowski *et al.*, 2018).

Artemisia annua has been declared as a therapeutic herb by the World Health Organisation (WHO), for its artemisinin content, an active ingredient against malarial infections. This drug is used by over 300 million people worldwide. In context to low artemisinin content in past few years work has been reported for the optimization of new extraction methods for better yield of artemisinin[4] (Zarrelli *et al.*, 2019).

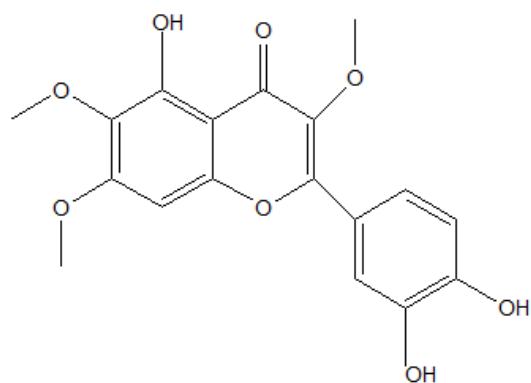
Fu *et al.*, (2020) reported the presence of eleven bioactive components, including five flavonoids i.e. rutin [33], cynaroside [35], isorhamnetin [36], chrysoptanol D [37] and casticin [38], four sesquiterpenes viz.. arteannuin B [20], dihydroartemisinic acid [21] artemisinic acid [15] and artemisinin [4] and two coumarins scopoletin [39] and scopolin [40] in *A.annua*.



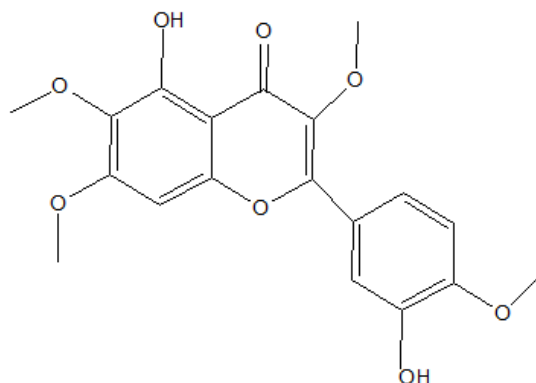
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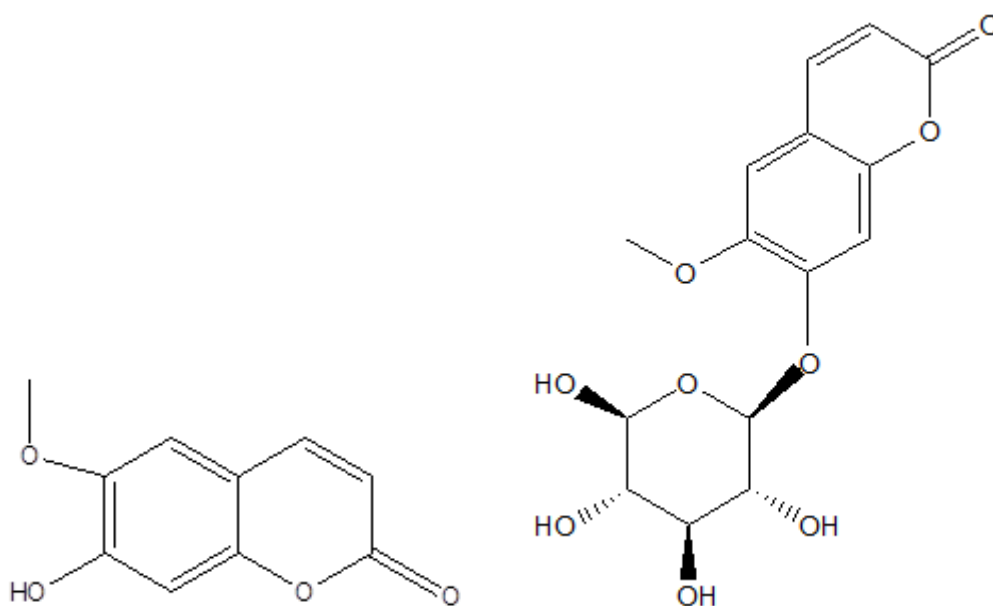
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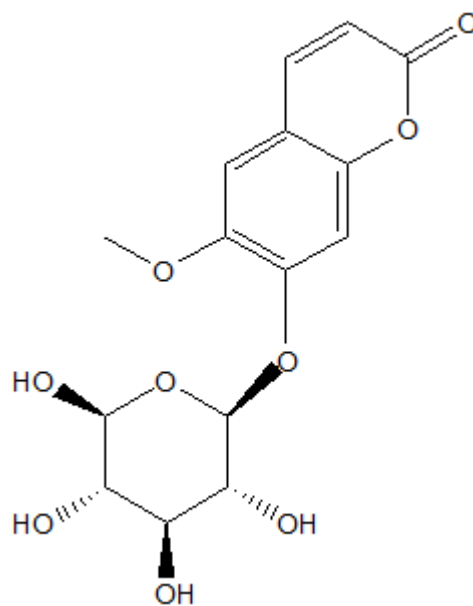
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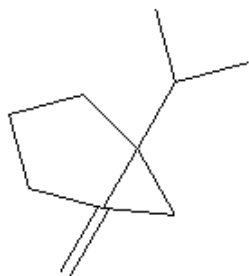
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A brief review of literature on phytochemical composition in different species of *Artemisia* besides *A. annua* reported from Uttarakhand is being given in **table 2.5**.

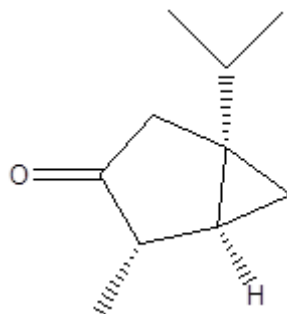
Table 2.5 Phytochemical composition of different Species of *Artemisia* reported in Uttarakhand

SN.	Plant Species	Major essential oil components	Reference
1.	<i>Artemisia gmelinii</i> Weber ex Stechm.	artemisia ketone (28.2%)[1], 1,8-cineole (13.0%)[14], sabinene (6.6%)[41]	Mathela et al., (1994)
		α -thujone (9.9%)[42], ar-curcumene (8.5%)[43]artemisia ketone (53.3%)[1], 1,8-cineole (6.6%)[14]	Haider et al., (2012)
2.	<i>Artemisia japonica</i> Thunb.	linalool (27.5%) [44], germacrene-D[3] (11.2%), (E)- β -ocimene (6.5%)[45], 1,8-cineole (5.5%)[14], (Z)- β -ocimene (5.5%)[46]	Joshi (2015)
3.	<i>Artemisia maritima</i> L.	α -thujone (63.3%)[42], sabinene (7.8%) [41], 1,8-cineole (6.5%) [14]	Mathela et al., (1994)
		1,8-cineole (23.6%)[14], chrystanthenone (25.7%)[47], germacrene-D (6.7%)[3]	Sah et al., (2010)
4.	<i>Artemisia nilagirica</i> (C.B. Clarke) Pamp.	α -thujone (36.4%)[42], β -thujone (9.4%) [48],germacrene-D (6.3%) [3],terpinen-4-ol (6.3%) [49]	Sati et al., (2013)
		α -thujone (36.9%)[42], β -thujone (8.2%)[48], terpinen-4-ol (7.1%)[49], β -eudesmol (12.4%)[50], β -caryophyllene (7.4%)[7],sabinene (6.6%) [41]	Badoni et al., (2009)
		artemisia ketone (55.07%)[1], germacrene-D (5.8%) [3], perillene (3.2%)[51],camphene (5.47%)[18] and borneol (4.12%)[12]	Joshi (2020)
5.	<i>Artemisia scoparia</i> Waldst. and Kit	capillene (42.1%)[52], β -caryophyllene (12.5%)[7], myrcene (9.2%)[53], β -pinene (8.6%)[22], p-cymene (6.8%)[54], γ -terpinene (5.3%)[55], 1-phenyl-2,4-pentadiyne (1.1%) [56].	Semwal et al., (2015)

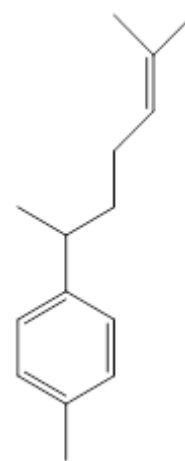
6.	<i>Artemisia indica</i> Willd.	davanone (30.8%)[57], β -pinene (15.3%)[22], β -pinene (15.30%)---[22], germacrene-D(5.82%)[3], β -elemene (4.93%), cymene (4.30%)[54], <i>trans</i> -caryophyllene (3.81%)[7], and linalool (3.60%)	Haider et al., (2014)
7.	<i>Artemisia roxburghiana</i> Bess.	β -thujone (65.3%)[48]	Haider et al., (2009)
		borneol (21.2%)[12], linalyl acetate (7.4%), α -humulene (6.7%)[58], β -caryophyllene (16.3%)[7], α -thujone (12.0%)[42], eugenol (16.2%)	Mathela et al., (1994)
8.	<i>Artemisia vulgaris</i> L.	artemisia ketone (6.77–8.64 %)[1], <i>trans</i> -caryophyllene (6.22–6.94 %)[7] and 1,8-cineole (4.75–5.13 %)[14]	Lohani et al., (2016)
		chrysanthenone (0.1-26.6 %)[47], vulgarole (0.1-20.6 %)[59], artemisia ketone (0.01-19.8 %)[1], α -thujone (0.01-19.0 %)[42], 1,8-cineole (1.6-13.5 %)[14], β -thujone (0.2-13.2 %)[48], caryophyllene oxide (1.4-11.2 %) [28] and camphor (0.9-11.1 %) [2]	Haider et al., (2019)
9.	<i>Artemisia parviflora</i> Buch.-Ham. ex D. Don	germacrene D (41.01%)[3], α -humulene (7.86%) [58], β -caryophyllene (10.58%) [7]	Tewan et al., (2015)
		β -caryophyllene (15.3%) [7] camphor (11.4%)[2], artemisia ketone (7.8%)[1], 1,8-cineole (5.8%)[14],germacrene D (14.7%) [3]	Rana et al., (2003)
10.	* <i>Artemisia elegantissima</i> var. <i>Kumaunensis</i> Pamp	davanone (5.5%)[57]	Shah (2010)



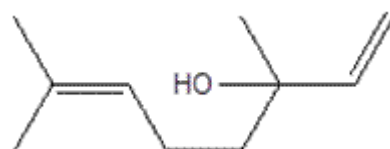
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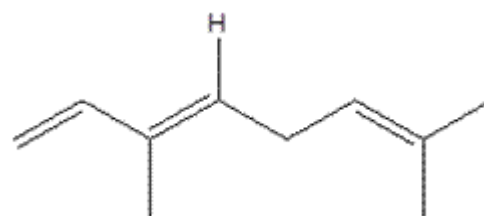
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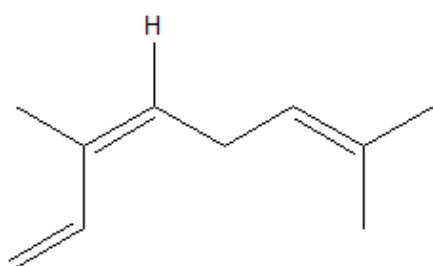
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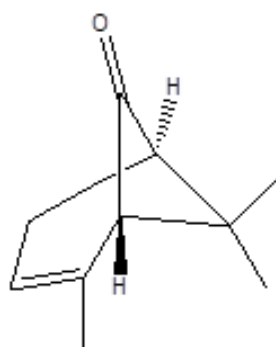
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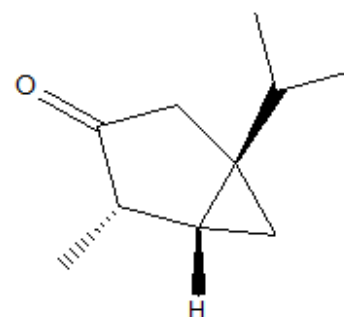
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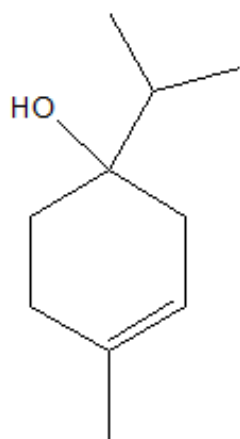
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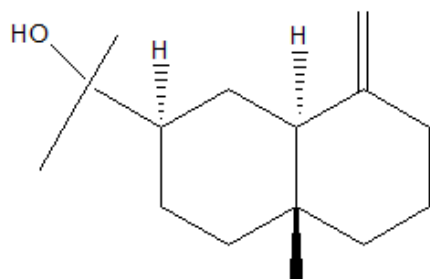
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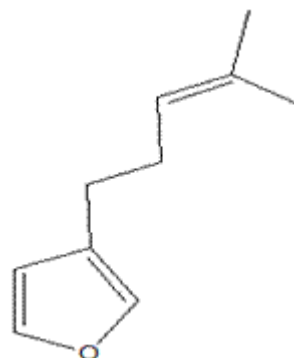
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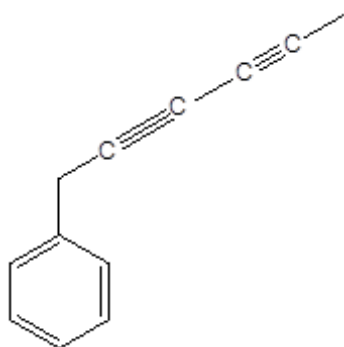
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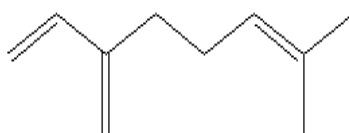
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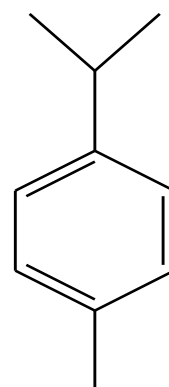
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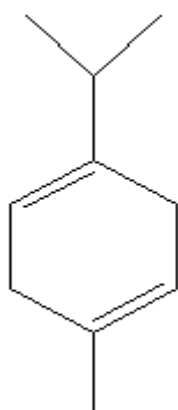
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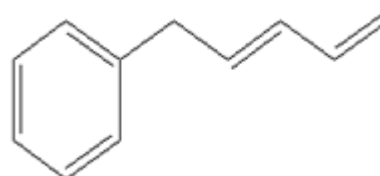
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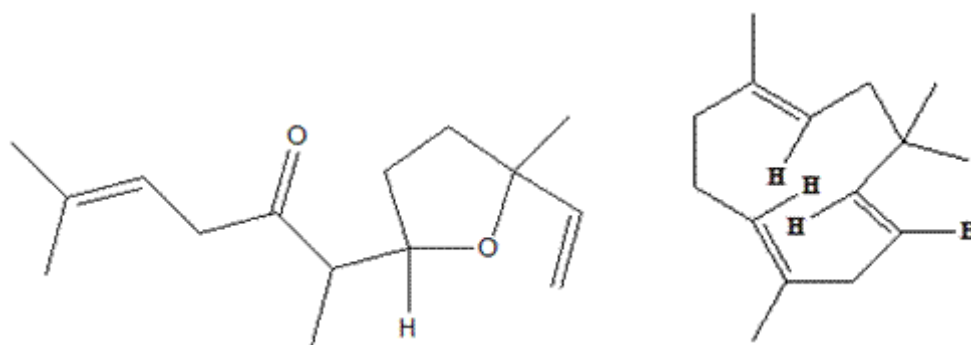
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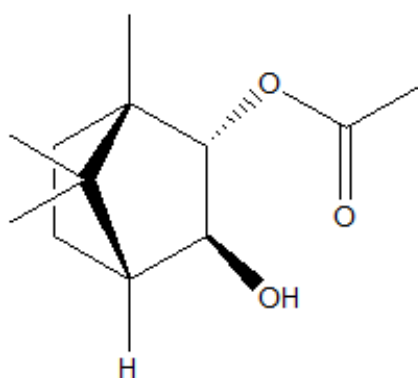


[56]



[57]

[58]



[59]

2.3.2 Biological activities

A detailed updated review of literature on biological activities of *A. annua* is being given in chronological order as under:

It has been reported that flavonoids and essential oil content present in *A. annua* are known to impart antioxidant activity. These studies ranked *A. annua* among those medicinal plants which are at the top of list, based on their highest antioxidant potential (**Juteau *et al.*, 2002**).

The documentation revealed antifeedant activity in crude extracts of *A. annua*. Due to deterency, growth regulatory effect and ovicidal potential it has also been recommended it as a good antifeedant herb (**Haghighian *et al.*, 2008**).

Due to notable activity of artemisinin, the anti-leishmanial activity of *A. annua* has been reported (**Malebo *et al.*, 2009**).

Ryu et al., (2011) reported that the anti-oxidant activity of the *A. annua* essential oil has been evaluated by five testing methods, DPPH, ABTS, reducing power, ORAC, and metal-chelating assays.

Anticancerous activity of various organic extracts of *A. annua* has been analysed by determining their cytotoxic potential in *Trypanosoma b. brucei* (TC221 cells) and HeLa cancer cells (**Efferth et al., 2011**).

The antimicrobial activity of *A. annua* essential oil have been evaluated against gram-positive, gram-negative bacteria, and one fungus. (**Cavar et al., 2012**).

Lubbe et al., (2012) documented in the literature, the first *in vitro* evidence of anti-HIV activity of the *Artemisia annua* tea infusion.

Artemisinic acid was reported to be the first ever found component obtained from *A. annua*. The compound has been which known to possess *in vitro* anti-adipogenic activities (**Lee et al., 2012**).

It has been reported that essential oil of *A.annua* had a good antibacterial activity against gram-positive and gram-negative bacteria (**Massiha et al., 2013**).

It has been documented in the literature that *A. annua* contains essential component known as artemisinin which provides structural chemical base for combinatorial treatment therapy for world antimalarial program (**Sadiq et al., 2014**).

It has been reported that artemisinin and its derivatives from *A.annua* exhibits anti-cancerous effect by causing cancer cell growth cycle arrest, promoting apoptosis, and preventing angiogenesis and tissue invasion of tumour (**Ho et al., 2014**).

It has been reported that the essential oil from *A.annua* possess exciting antibacterial and antifungal activities. Studies show that artemisia ketone is the component of the oil that possess the maximum antimicrobial activity. (**Bilia et al., 2014**).

The camphor-rich essential oil of *A. annua* has been reported to exhibit a dose dependent antileishmanial activity against the promastigotes as well as the intracellular amastigotes of *L. donovani* (**Islamuddin et al., 2014**).

The antimicrobial activity of artemisinin and precursor derived from the *in vitro* plantlets of *A. annua* has been reported to be antimicrobial in nature similar to that of the commercially available antibiotic, streptomycin. (Appalasamy *et al.*, 2014).

Artemisinin, the active component found in *A. annua* has been reported to have pronounced anti-malarial activity. It has also been reported that artemisinin and its derivatives could inhibit *T. gondii* infection (Ho *et al.*, 2014).

Kim *et al.*, (2015) evaluated the anti-inflammatory, antioxidant, and antimicrobial properties of artemisinin derived from water, methanol, ethanol, or acetone extracts of *Artemisia annua* L.

The anti-fungi activity of *A. annua* essential oil has been reported against *C. albicans* and *C. dubliniensis* and *C. parapsilosis* (Santomauro *et al.*, 2016).

It has been reported that the synergistic effects of artemisinin and its flavonoids compounds and their biological interaction between malaria and cancer can be a rich source to anti-malarial, immunosuppressive, anti-inflammatory, and anti-cancer properties of this plant (Alesaeidi and Miraj, 2016).

Huang *et al.*, (2017) reported the *in vitro* anti-asthmatic activities of *A. annua* by using tracheal rings (TRs) and acute isolated airway smooth muscle cells (ASMCs) of mice.

The *in vitro* and *in vivo* anti leishmanial activity and cytotoxicity of *A. annua* L. has been reported (Mesa *et al.*, 2017).

The published data revealed the antifungal activity of *Artemisia annua* synthesized AgNPs against three *Candida* species (*C. albicans* ATCC 90028, *C. tropicalis* ATCC 750 and *C. glabrata* ATCC 90030) (Khatoon *et al.*, 2019).

The *in vivo* hepatoprotection, antihyperlycemic and antioxidant activities of the polar fraction of the 70% ethanolic extract of *A. annua* leaves in model rats has been reported (El-Askary *et al.*, 2020).

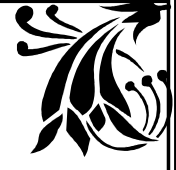
It has been reported that casticin and chrysopterol D, the flavonoid components present in *A. annua* possess anti-cancerous activities (**Fu et al., 2020**).

Artemisia extracts have been reported to have a strong bactericidal activity against Mtb (*Mycobacterium tuberculosis*), suggesting that *A. annua* extracts kill Mtb through a combination of Artemisinin and other additional compounds (**Martini et al., 2020**).

The leaves powder extraction and the crude extraction of the whole plant of *A. annua* have been reported to show anti-microbial activities, making *A. annua* a promising source of new anti-microbial agents (**Feng et al., 2020**).

Studies revealed that artemisinin and arteannuin B which are found in *A. annua*, possessed anti-osteoporotic activities by blocking receptor activator of nuclear factor kappa-B ligand (RANKL)-induced osteoclast differentiation by reducing the expression of the two transcription factors, c-Fos and NFATc1 (**Feng et al., 2020**).

Rolta et al., (2021) reported that the methanolic extract of *A. annua* showed higher antioxidant potential ($IC_{50} = 37.075 \pm 0.34 \mu\text{g mL}^{-1}$) in comparison to petroleum ether extract. Whereas in antimicrobial analysis, methanolic and petroleum ether extracts of *A. annua* have been reported to possess strong inhibitory activity against the strains of *Candida* as compared to bacterial strains.



*Materials
and
Methods*



The present investigation was implemented in Govind Ballabh Pant University of Agriculture and Technology, Pantnagar (Uttarakhand). Extraction and isolation of essential oils from the plant, Evaluation of its different activities like antioxidant, anti-inflammatory, anti-diabetic, insect-repellent was performed in phytochemistry Laboratory of Department of Chemistry, College of Basic Science and Humanities, Pantnagar. The insect repellent activity was performed in Department of Entomology, College of Agriculture, G.B.P.U.A.T, Pantnagar. Chemical Investigation of Essential Oil was done with the help of GC and GC/MS analysis in Advanced Instrument Research Facility (AIRF), Jawahar Lal Nehru University, New Delhi.

3.1 Materials

3.1.1 Collection of Plant material

The herb *Artemisia annua* was collected from Pantnagar, Uttarakhand in the month of September, 2020 and January 2021. The collected plant was identified by Taxonomist Dr. D.S. Rawat, (Assistant professor). Department of Biological Sciences, College of Basic Sciences and Humanities, Pantnagar. The herbarium specimen of *Artemisia annua* L. with the herbarium number GBPUH-1033 was deposited at herbarium of Department of Biological Sciences, C.B.S&H., G. B. Pant University of Agriculture & Technology, Pantnagar.

3.1.2 Chemicals

The chemicals and solvents used for the study were of laboratory grade and analytical grade and were acquired from Merk, Himedia and SD fine. Chemicals such as ferrozine, DPPH, BHT, acarbose, catechin used were of sigma Aldrich and standard drug diclofenac sodium was purchased from the shop.

3.1.3 Glass wares and plastic wares

The glass wares and plastic wares used during the whole work were either Borosil or Thermo Fisher Scientific made.

3.1.4 Source of insect anti-feedant activity

The egg mass (**fig 3.1**) of *Spodoptera litura* was collected from the castor field which was grown in Pantnagar and was acclimatized until 3rd instar larvae, in the laboratory condition of department of Entomology, College of Agriculture, G.B. Pant University and Technology, Pantnagar.

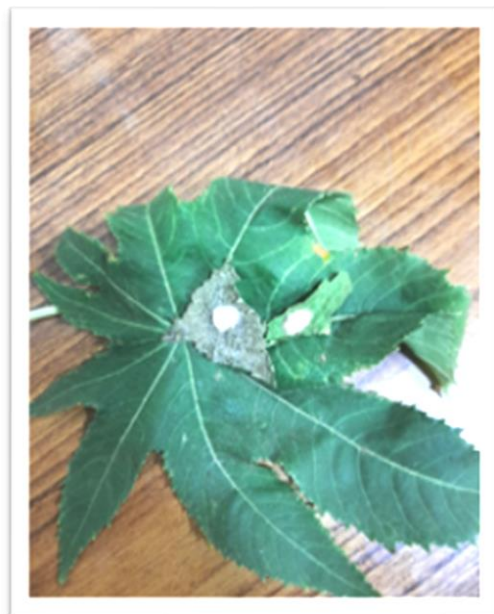


Fig 3.1 Egg masses of *Spodoptera litura*

3.1.5 Isolation and Identification of the Root-Knot nematode (*Meloidogyne incognita*)

3.1.5.1 Collection of Root knot nematode infected sample

Capsicum was taken as the Root knot infected plant collected from the Vegetable Research Center, G. B. Pant University of Agriculture and Technology, Pantnagar. The sample was collected on the basis of the visual symptoms of Root-knots or galls formed in the capsicum plant.

3.2 Phytochemical analysis

Essential oils extraction from the aerial parts of *A.annua* were isolated by hydro-distillation method in a Clevenger apparatus (**Clevenger,1928**) as **fig.3.2(a)** **and (b)**.The plant was cut into small pieces, crushed and hydrodistilled for 3-4 hours.

The whole process of hydrodistillation was repeated for three times in order to obtain desirable amount of essential oil. The oil was extracted with hexane and desiccated over anhydrous sodium sulphate. The solvent was evaporated under vacuum and the essential oils so obtained were stored at low temperature (4° C in refrigerator) for further use.

The % yield of (v/w) is being record in **table 3.1** below:

Table 3.1 Yield of essential oil from *A.annua* from two seasons i.e *Artemisia annua* rainy essential oil (AAREO) and *Artemisia annua* winter essential oil (AAWEO).

SN.	Plant part	Method of extraction	Amount of plant material	Duration of heating	Yield % (V/W)
1.	AAREO	Hydrodistillation	600g	3-4 hours	0.7%
2.	AAWEO	Hydrodistillation	500g	3-4 hours	0.5%



(a)



(b)

Fig. 3.2 *Artemisia annua* from two different seasons

3.3 GC-MS Analysis

For the phytochemical composition of essential oil, an analytical technique GC-MS was carried out using GC MS-QP 2010 plus equipment with following experimental conditions:

3.3.1 GC-MS parameters

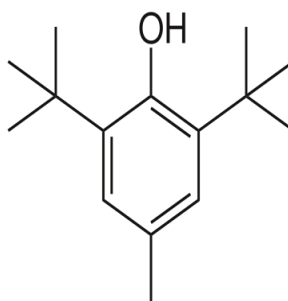
Parameters	Conditions
Carrier gas	Helium
Column flow rate	1.21 mL/min
Temperature programme: -	Initial temperature at 50°C RAMP@ 3°C/min upto 210°C (isotherm for 2 min) then 6°C/min upto 280°C (isotherm for 2 min), finally hold for 11 min
Injection volume	1 µL
Injection temperature	260°C
Injection mode	Split
Split ratio	22.0
Flow control mode	Linear Velocity
Pressure	69.0 kPa
Total flow	30.8 mL/min
Linear velocity	39.9 cm/sec
Purge flow	3.0 mL/min
High Pressure Injection	OFF
Carrier Gas Saver	OFF
Splitter Hold	OFF

Identification of the chemical constituents of the essential oils were done by comparing their mass spectral fragmentation pattern and their RI with that of the MS library (NIST14.lib, FFNSC2.lib, WILEY8.LIB) and comparing the spectra with literature data (Adams, 2007).

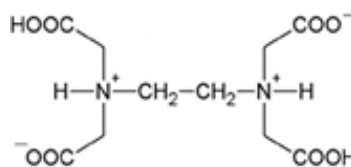
3.4 Biological activities

3.4.1 Evaluation of antioxidant activity:

Investigation of the *in-vitro* antioxidant potential of the essential oils of *A.annua* (AAREO & AAWE0) was done by generally practiced and reported methods. Standard antioxidants used were BHT (butylated hydroxyl toluene) and Na₂-EDTA (disodium salt of ethylenediaminetetraacetic acid) **fig 3.3**.



BHT



Na₂EDTA

Fig 3.3 Structures of standard antioxidants

3.4.2 Metal chelating activity

The metal chelating activity of Fe^{2+} was determined by spectrophotometric method. Based on the principle of Fe^{2+} chelating ability of the antioxidant by measuring the absorbance of the ferrous ion-ferrozine complex formed at 560 nm (**Kunwar *et al.*, 2013**). 0.1 mL of 2mM $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$, 0.2 mL of 5mM ferrozine and 4.7 mL of methanol was added to different concentrations (20-100 $\mu\text{l}/\text{mL}$). the solutions were mixed and incubated for about 10 min. The absorbance of test sample was measured at 562 nm using UV Visible spectrophotometer. Na_2EDTA (0.01 mM) was used as the standard. The metal-chelating activity of samples, was calculated using the following formula:

$$\text{IC \%} = \frac{(A_0 - A_t)}{A_0} \times 100$$

where, A_0 = absorbance value of control sample

A_t = absorbance value of test sample

The percent of chelating activity was plotted against concentrations and the standard curve was drawn using standard antioxidant (Na_2EDTA) to calculate the IC_{50} values for standard and different samples of essential oil, column fraction and solvent extracts. A lower IC_{50} values valued indicates greater metal chelating ability.

3.4.3 Hydrogen peroxide (H_2O_2) scavenging activity

Hydrogen peroxide activity of the essential oils was evaluated by method given (**Leyton *et al.*, 2015**). The reaction mixture comprised of 0.4 ml of different concentration of essential oils in methanol (20-100 $\mu\text{l}/\text{mL}$) and 0.6 ml of 40mM hydrogen peroxide in phosphate buffer pH (7.4). After keeping the reaction mixture at room temperature for 10 minutes. , the absorbance was taken at 230 nm against the blank, i.e., methanol. Here BHT was taken as a positive control.

The percentage of H_2O_2 scavenging was determined by using the given formula:

$$\text{IC \%} = \frac{(A_0 - A_t)}{A_0} \times 100$$

where, A₀= Absorbance value of control sample

A_t= Absorbance value of test sample

IC = Inhibitory concentration

IC₅₀ values of all the samples were calculated with regression method analysis.

3.4.4 DPPH (2,2-diphenyl-2-picrylhydrazyl) radical scavenging activity

Estimation of DPPH radical scavenging activity was based on the method given by (Prakash *et al.*, 2011). Various concentrations of essential oil (20-100µl/mL) were mixed with 5 mL of freshly prepared methanol solution of DPPH (0.004%). The reaction mixture was thoroughly mixed then their absorbance was measured by using UV-Visible spectrophotometer at 517 nm. All observations were taken in triplicate and the standard antioxidants used were butylated hydroxytoluene (BHT). The ability to scavenge DPPH radical (fig 3.4) was calculated by the following equation:

$$IC \% = \frac{(A_0 - A_t)}{A_0} \times 100$$

where, A₀ = absorbance value of control sample

A_t = absorbance value of test sample

IC = inhibitory concentration

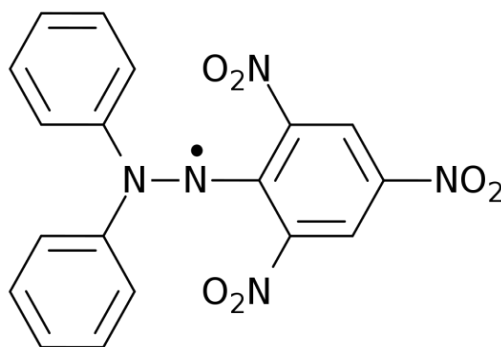


Fig 3.4 DPPH (free radical)

3.4.5 *In-vitro* anti-inflammatory activity

The in-vitro anti-inflammatory activity of the essential oil was estimated by using inhibition of albumin denaturation technique (Singh *et al.*, 2020). The sample consisted of 200µL of egg albumin, 3 mL of saline phosphate buffer solution (PBS) (pH= 6.4) and different amount of essential oil (20-100µl/mL). Double distilled water was used as control. The prepared mixtures were incubated at 37±2°C (for 15 min and then heated for 5 min at 72°C in water bath) . After the cooling of the mixture, the absorbance was recorded at 660nm. Diclofenac sodium (fig 3.5) was used as the standard.

The percentage inhibition of protein denaturation was calculated using the formula:

$$IC\% = (A_0 - A_t) / A_0 \times 100$$

A₀ = absorbance value of control sample

A_t = absorbance value of test sample

IC = inhibitory concentration

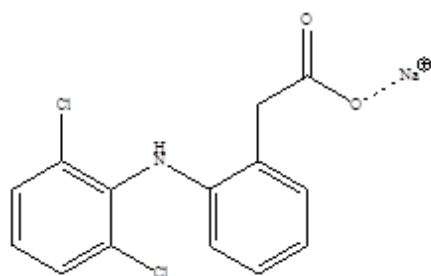


Fig 3.5 Diclofenac sodium

3.4.6 *In-vitro* antidiabetic activity

The antidiabetic activity of the essential oil was evaluated by using α-amylase assay as given by Nazir *et al.*, (2021) with slight modifications. Different concentrations of the oil and the standard drug, acarbose (20-100µl/mL) (fig 3.6) were added to a reaction mixture containing 100 µL of 2mM phosphate buffer (pH 6.9) and α-amylase (200 µL). The reaction mixture was then incubated at 25°C for 20 min.

After that 100 μ L of 1starch solution was added to the above solution and then incubated at room temperature for 5 min. Then 1 mL of 3,5- Dinitrosalicylic acid (DNSA) solution was added to the reaction mixture and placed in a boiling water bath for 5 min and cooled at room temperature. Further the absorbance was recorded at 540nm using a UV spectrophotometer.

The percentage inhibition was calculated using the following formula:

$$\text{Inhibition of } \alpha\text{-amylase activity (\%)} = (A_0 - A_t)/A_0 \times 100$$

Where,

A_0 = absorbance value of control sample

A_t = absorbance value of test sample

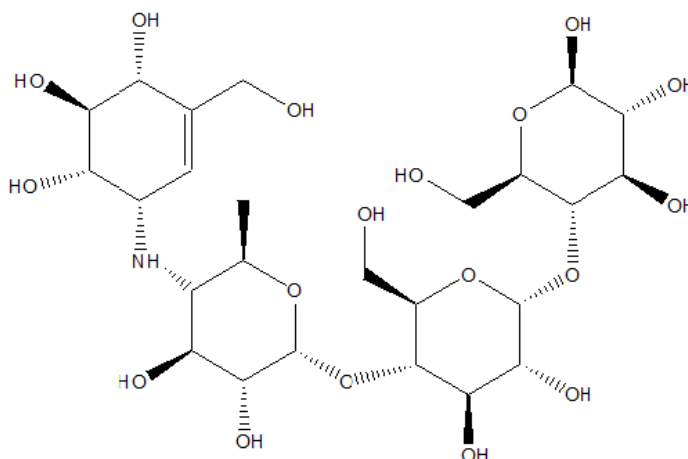


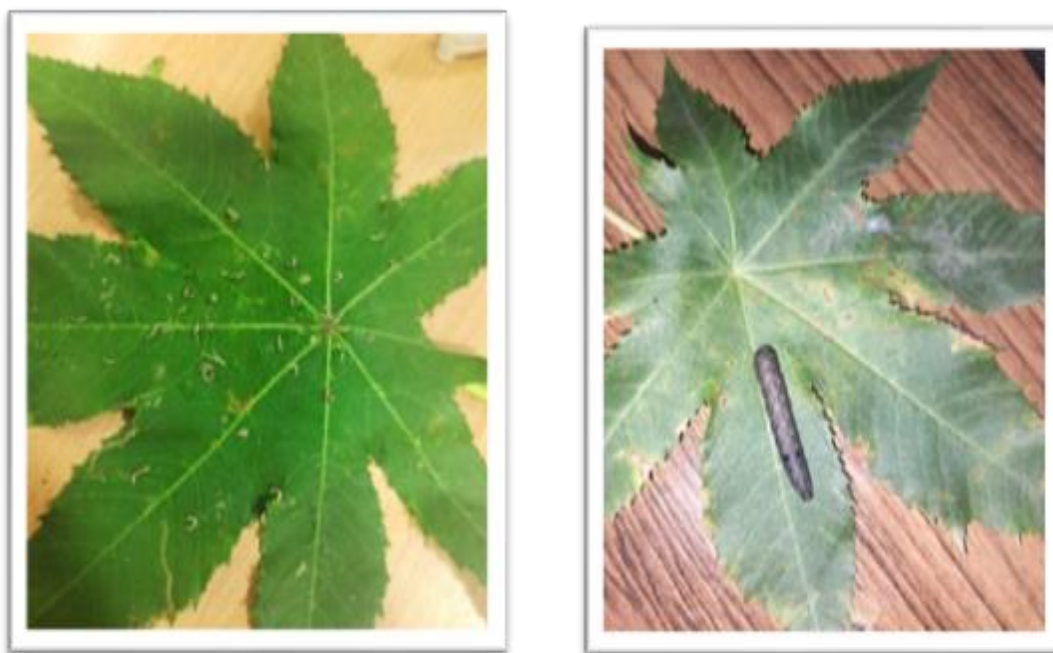
Fig 3.6 structure of acarbose

3.4.7 Evaluation of anti-feedent activity

3.4.7.1 Test insect

The antifeedant activity of the essential oils was tested against the second instar larvae of *Spodoptera litura* **Fig 3.7 (a)**, *Spodoptera litura* (Fab). (Lepidoptera: Noctuidae) is one of the major polyphagous pest which attacks economically essential crops. viz., cotton, groundnut, chilly, tobacco, castor and pulses in India, China and Japan thereby causing extensive damage (Arivoli and Tennyson, 2012). In severe

infestation, this pest can cause severe damage to the plant which results into stunted growth and reduced yield of the plant (Javier *et al.*, 2017).



(a) Second instar larvae of *Spodoptera litura*

(b) Third instar larvae of *Spodoptera litura*

Fig 3.7 Picture showing (a) second and (b) third instar larvae of *Spodoptera litura*

3.4.7.2 Collection of Larvae and its maintenance

The egg mass of *Spodoptera litura* (Fig. 3.1) was collected from the castor crop (*Ricinus communis*) from C.R.C. (Crop Research Center), G.B.P.U.A.T. Pantnagar, Uttarakhand, India in the month of April. The rearing of insect was done in laboratory conditions maintained at 27°C temperature in a glass jar packed with muslin cloth from the top. The insect larvae were fed on fresh castor leaves collected on daily basis. The fully grown 2nd instar larvae kept starved for 2-3 hours in insect growth chamber (Fig 3.8) and were then tested for the study of anti-feedant activity.



Fig 3.8 Second instar larvae kept in Insect growth chamber

3.4.7.3 Experimental procedure Of Anti-feedant activity

The experiment was conducted as per the developed protocol. The efficacy was tested against *Spodoptera litura* using leaf dipping method. Essential oils (AAREO & AAWE0) in 0.5% concentration were prepared by dissolving different concentrations of oils (2 μ L, 4 μ L, 8 μ L) in the solvent which is 0.5 mL of tween 20 (stabilizing agent) in 99.5 mL of water i.e., 0.5% Tween 20 (V/V). Leaf discs (4 \times 4 sq cm) of castor leaves were then dipped in each concentration of oils for few seconds and then were air dried. The leaf discs were then kept in glass petri dishes and one 2nd instar larvae (starved for 2-3 hours) of test insect was then released in each petri dish. Solvent without oil was taken as control. Three replications were maintained for each

oil concentration. The consumed area of leaf after 24 and 48 hours was then observed and the percent antifeedant activity was then calculated using the following formula:

$$\% \text{ Antifeedant activity} = \frac{\text{Leaf disc consumed in control} - \text{Leaf disc consumed in treated}}{\text{Leaf disc consumed in control} + \text{Leaf disc consumed in treated}} \times 100$$

3.4.8 Evaluation of nematocidal activity

3.4.8.1 Identification of the Root-knot nematode species

- ❖ Infected capsicum roots with root-knot nematode were washed and stained with acid fuschin lactophenol solution.
- ❖ Then after the roots were allowed to cool at room temperature followed by destaining via acidified glycerine.
- ❖ Females were dissected out from the well-developed galls of root under stereobinocular microscope and placed on lactophenol solution. To get the perineal section, the posterior portion of the female was cut and the body contents were cleaned off.
- ❖ The cleaned posterior portion of female was trimmed and transferred to drop of lactophenol on a microscopic slide.
- ❖ The confirmation of species was done by examining the pattern of posterior section as described by **Eisenback, 1985**.

3.4.8.2 Preparation of Acid Fuschin stain and procedure for root staining

Acid Fuschin stain: 3.5g of acid Fuschin added in 250 mL of acetic acid, later dissolved in 750mL of distilled water and mixed thoroughly. This can be used as stock stain by storing.

Staining of roots (**Eisenback, 1985**)

- ❖ The infected roots were soaked in 2% chlorine bleach solution for 2 min followed by rinsing in running water for 45 seconds and soaking in tap water for 15 min to remove the residual bleach that interferes with stain.

- ❖ The drained roots were then transferred to a beaker containing 30-45mL of tap water added with 1ml of acid Fuschin stain.
- ❖ The roots in solution were boiled for 30 seconds on hot plate and cooled to cooled to room temperature followed by draining and rinsing with water.
- ❖ Stained roots were then transferred to 20-30ml of acidified glycerin (few drops of 5N HCl were added) and heat to destain.
- ❖ After staining, the roots were stored in acidified glycerin until their examination.

3.4.8.3 Pathogenicity test

The seedling of 10-12 days old cucumber seedlings raised in autoclaved soil in pots were inoculated with the fresh second stage juveniles (J2) emerged from the egg mass of *Meloidogyne incognita* maintained in pure culture. Inoculation was done by removing 1-2cm of top soil surrounding the seedlings and the nematode juveniles in water were poured on the exposed root system. Exposed roots after inoculation were covered with soil and lightly watered. After 45-50 days of inoculation, roots were examined with the presence of root galls and microscopically studied by preparing perineal pattern of female for the confirmation of pathogen (*Meloidogyne incognita*).

3.4.8.4 Nematode culture preparation

The pure culture of nematode to be used in the experiment was obtained from a single egg mass of *M. incognita* infected roots.

- ❖ A single egg mass was transferred on to a petri plate containing a little amount of water and the incubated at room temperature of 25-27°C for 3 days for emergence of second stage juveniles (**Den Ouden, 1958**).
- ❖ 3 weeks old transplanted tomato seedlings raised in autoclaved soil were inoculated with second stage juveniles. Inoculation was done by removing top soil of 1-2 cm around the seedlings and the juveniles in water were poured on partly exposed root system.

- ❖ Nematodes were then allowed to multiply in the pot for upto 2-month duration and inoculum for the experiment was obtained by extracting eggs and second stage juveniles from it using 2% sodium hypochlorite solution as described by **Hussy and Baker, 1973**.

3.4.8.5 *In vitro* evaluation of AAREO and AAWE0 on egg hatching of *Meloidogyne incognita*

In vitro experiment was conducted to evaluate the efficacy of different essential oils on egg hatching of *M. incognita*. The eggs of *M. incognita* were obtained from the nematode pure culture.

Gridded Petri dishes were used for carrying out the experiment. Total 3 treatments were there, each one replicated three times and 3 different doses respectively. Out of the three treatments, 2 were oils of *A. annua* collected in two different seasons and the rest was distilled water treated as control. 2 egg masses of *M. incognita* were suspended in 2, 6 and 10 µl/mL concentration of essential oils taken in gridded Petri dishes. All the treatments were arranged in completely randomized design and kept at ambient temperature of 27±1°C in BOD incubator. Observations on egg hatchability of eggs were recorded at 24hr, 48hr and 72hr exposer period by counting the number of eggs hatched, under microscope at 40x magnification.

3.4.8.6 *In vitro* evaluation of essential oil samples on mobility of second stage larvae of *M. incognita*

The experiment was conducted under *in vitro* condition to evaluate the efficacy of essential oils on the mobility of second stage larvae of *M. incognita*. The second stage larvae were isolated from the galled roots with egg masses from pure culture of nematodes. After removal of galled roots from pot, they were washed in tap water to remove the soil adhered to it. Then after it was cut into small pieces of 2 cm and transferred into a 2% sodium hypo chloride solution and shaken for two minutes to separate the organic debris from eggs. The suspension was poured through series of sieves and the eggs sieved on 38µm- pore were collected and washed cautiously in the tap water. Egg suspension was then poured and incubated at 28±1°C, so that the juveniles of second stage from the hatched eggs were collected within 48 hrs.

100 second stage juveniles were counted and placed on the gridded Petri dishes containing stock solution of essential oils. In total, there were 3 treatments replicated three times and three different doses of 1, 3 and 5 μ l/mL. The juveniles placed in distilled water were treated as control. All the treatments were arranged in randomized manner. Observation on the count of dead juveniles was done at different time periods of 24hrs, 48hrs and 72hrs using stereo-binocular microscope. The nematodes appearing almost straight in position (not showing any movement) were trapped by picking instrument to check the aliveness of larvae and when they were completely not responding, they were picket out of the Petri dish and placed in distilled water to confirm the larval death.

3.5 Statistical analysis

All the experiments were conducted in three replicates and the data were expressed in terms of mean \pm standard deviation. Data illustrated in the tables and the graphs were subjected to ANOVA at 1% level of significance.



*Results
and
Discussion*



This chapter deals with the result of the present study in accordance to the objectives mentioned in chapter one. The plant *Artemisia annua* was collected widely from Tarai Region of Pantnagar, Uttarakhand, India having altitude of 270 m, For the study of seasonal variation in essential oil components. The plant material was collected in the month of September 2020 and January 2021. The essential oils were studied by GC/MS and various biological activities were carried out in EO's. The results are being described as follows.

4.1 GC/MS analysis of AAREO

The essential oils isolated in rainy and winter season for their chemical composition and various biological activities were designated as AAREO & AAWE0 respectively. Seventy five constituents contributing 91.7% of the total oil were identified in AAREO. Camphor (14.1%) was found to be the major component of the oil followed by germacrene-D (9.0%), *trans*- β -caryophyllene (8.7%), eucalyptol (4.2%), *cis*-cadin-4en-7-ol (3.9%), isoborneol (3.2%), p-cymene (2.9%), (-)-aromadendrene (2.6%), lavandulyl acetate (2.1%), α -cadinol (2.0%), camphene (2.0%), (-)-globulol (1.8), γ -cadinene (1.7%), spathulenol (1.5%), amyl vinyl carbinol (1.5%), germacra-4 (15), 5,10 (14)-trien-1- α -ol (1.3%), bicyclogermacrene (1.3%), selina-4,11-diene (1.2), (+)- α -terpineol (1.2%), isogermacrene- D (1.1%), α -humulene (1.0%), dihydro- β -ionone (1.0%), β - *trans*- farnesene (1.0%), *trans*-3-hexenyl acetate (1.0%), and β -pinene (1.0%). The minor components (<1.0%) were identified as α -copaene (0.9%), naphthalene (0.8%), bornyl angelate (0.8%), phytol (0.8%), benzyl 2-methylbutyrate (0.7%), *trans*-carveol (0.6%), 2- δ -carene (0.5%), epicedrol (0.5%), α -Selinene (0.5%), sobrerol 8-acetate (0.5%), β - costol (0.5%), (+)-spathulenol (0.4%), sabinene (0.4%), α -hydroxy-terpenyl acetate (0.4%), *cis*-carvyl acetate (0.4%), *cis*-jasnone (0.4%), *cis*-3-hexen-1-ol (0.3%), thymol (0.3%), bicycloelemene (0.3%), *neo*-alloocimene (0.3%), γ -gurjunenepoxide- (2) (0.2%), isogeranial (0.2%), caryophyllene oxide (0.2%), eugenol (0.2%), *cis*-carveol (0.2%),

pinocarvone (0.2%), ethanone (0.2%), linalool (0.2%), and *cis*-sabinene hydrate (0.2%). The compound which was present in trace amount was myrtenal i.e its composition was (<0.1%). The detailed representation of chemical composition of aerial parts of AAREO along with their chemical formula and Kovatt index (KI) value are being recorded in **table 4.1**. The class composition of essential oil constituents has been depicted in **table4.2** and **fig.4.2**, and its gas chromatogram in **fig. 4.1**. The structures of major compounds of AAREO are given in **fig. 4.5**.

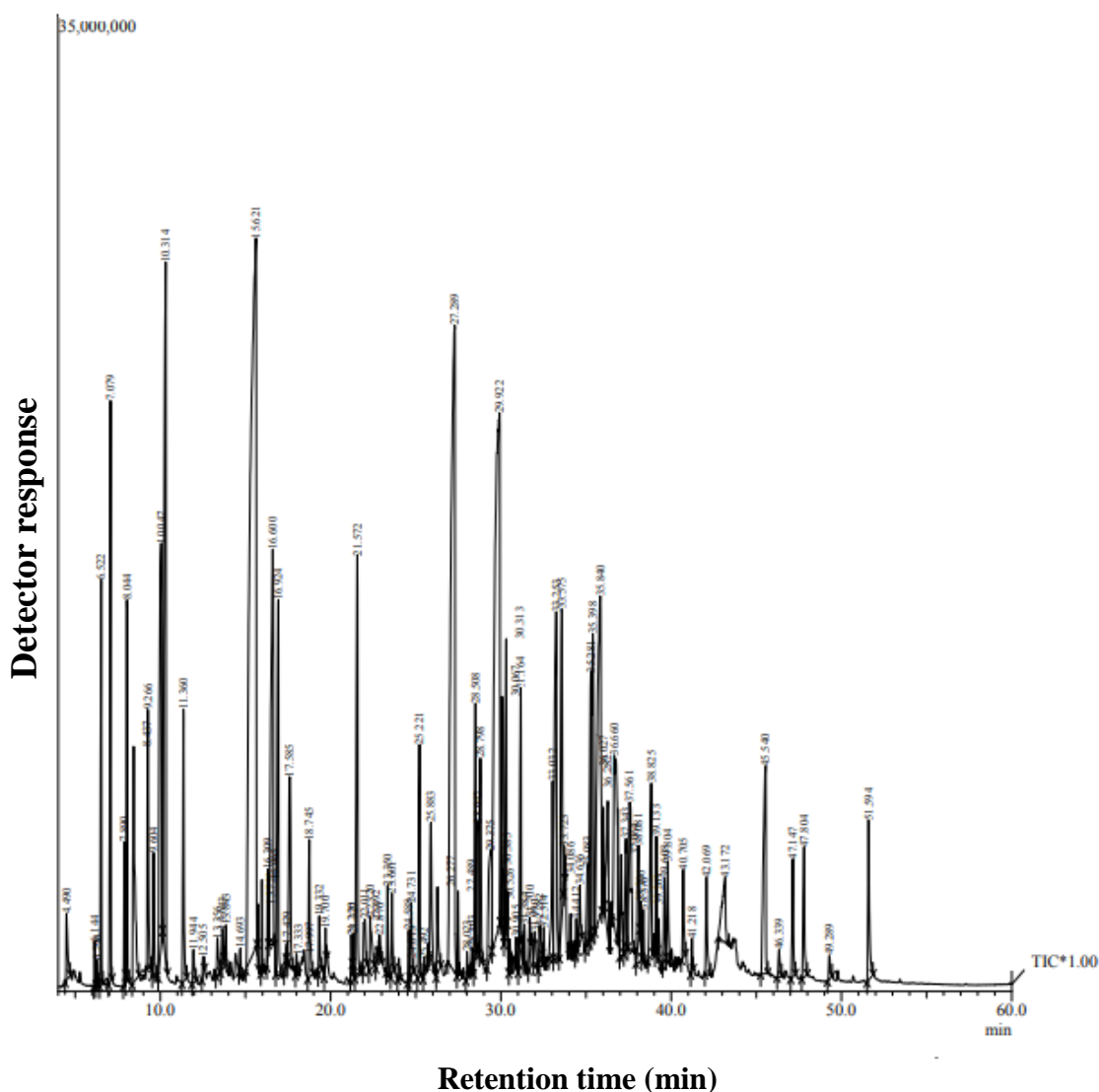


Fig 4.1 Gas Chromatogram of AAREO

Table 4.1 Chemical composition of AAREO

S.N.	Compound name	K.I	% Composition	Molecular Formula	Method of Identification (MS) (NIST-MS; Adams, 2007)
1.	<i>cis</i> -3-Hexen-1-ol	868	0.3	C ₆ H ₁₂ O	M ⁺ = 100, m/z; 100,82,67,55,41 (100%), 27
2.	tricyclene (MH)	923	0.1	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 79,67
3.	α -thujene (MH)	927	0.1	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 77,65
4.	α -pinene (MH)	933	1.2	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 92,91,77,67,41
5.	camphene (MH)	954	2.0	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,107,93 (100%), 79,67,53,41
6.	amyl vinyl carbinol (OM)	969	1.5	C ₁₀ H ₁₆ O	M ⁺ = 128, m/z; 127,99,85,72,57 (100%), 43,29,27
7.	sabinene (MH)	972	0.4	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 77,69,41
8.	β -pinene (MH)	978	1.0	C ₁₀ H ₁₆	M ⁺ = 136 m/z; 136,121,107,93 (100%), 79,69,41
9.	2- δ -carene (MH)	1000	0.5	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 79,77,65
10.	<i>trans</i> -3-hexenyl acetate	1001	1.0	C ₈ H ₁₄ O ₂	M ⁺ = 142, m/z; 112,98,82,67,43 (100%), 41
11.	p-cymene (MH)	1025	2.9	C ₁₀ H ₁₄	M ⁺ = 134, m/z; 134,119 (100%), 91,77,65
12.	eucalyptol (OM)	1031	4.2	C₁₀H₁₈O	M⁺= 154, m/z; 154,139,108,84,81,69,43 (100%), 41

13.	γ -terpinene (MH)	1058	1.0	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,,105,93 (100%), 77,43,41
14.	<i>cis</i> - sabinene hydrate (OM)	1069	0.2	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 154,136,121,111,93,71,59,43 (100%), 41
15.	hotrienol (OM)	1072	0.1	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 119,107,91,83,71 (100%), 67,43
16.	linalool (OM)	1096	0.2	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 136,121,93,79,71,69,43 (100%), 41
17.	(-)-terpinen-4-ol (OM)	1137	1.8	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 154,136,111,93,71 (100%), 69,43,41
18.	p-menth-2-en-1-ol (OM)	1140	0.1	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 154,140,139,121,111,94,79,71,69,43 (100%), 41
19.	camphor (OM)	1149	14.1	C₁₀H₁₆O	M⁺= 152, m/z; 152,137,108,95 (100%), 81,69,55,41
20.	camphene hydrate (OM)	1156	0.1	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 136,121,111,96,71 (100%), 69,43
21.	ethanone (OM)	1161	0.2	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,109 (100%), 93,81,67
22.	Pinocarvone (OM)	1164	0.2	C ₁₀ H ₁₄ O	M ⁺ = 150, m/z; 150,135,122,108,81,69,53 (100%), 41
23.	isoborneol (OM)	1165	3.2	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 121,110,95 (100%), 81,67,41
24.	naphthalene (SH)	1181	0.8	C ₁₅ H ₂₄	M ⁺ = 204, m/z;204,189,175,162,147,133,119 (100%), 105,93,79,67,55,41
25.	(+)- α -terpineol (OM)	1188	1.2	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 136,121,93,81,59 (100%), 43,41
26.	myrtenal (OM)	1197	t	C ₁₀ H ₁₄ O	M ⁺ = 150, m/z; 135,107,91,79 (100%), 66,53,41

27.	<i>trans</i> -carveol (OM)	1223	0.6	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,119,109 (100%), 84,83,69,55
28.	<i>cis</i> -Carveol (OM)	1232	0.2	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 134,119,109,84 (100%), 83,69,55,41
29.	sobrerol 8-acetate (OM)	-	0.5	C ₁₂ H ₂₀ O ₃	M ⁺ = 212, m/z; 152,137,109 (100%), 95,69,43
30.	isogeranial (OM)	-	0.2	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,119,109,94 (100%), 81,67,53,43,41
31.	carvone (OM)	1246	0.1	C ₁₀ H ₁₄ O	M ⁺ = 150, m/z; 150,108,93,82 (100%), 58,54,41,40
32.	lavandulyl acetate (OM)	1284	2.1	C ₁₂ H ₂₀ O ₂	M ⁺ = 196, m/z; 136,121,107,93,80,69 (100%), 43,41,40
33.	bornyl acetate (OM)	1285	0.1	C ₁₂ H ₂₀ O ₂	M ⁺ = 196, m/z; 196,136,121,108,95 (100%), 80,43,41
34.	thymol (OM)	1293	0.3	C ₁₀ H ₁₄ O	M ⁺ = 150, m/z; 150,135 (100%), 115,91,77,65,51,41
35.	4-vinyl Guaiacol	1309	0.1	C ₉ H ₁₀ O ₂	M ⁺ = 150, m/z; 150 (100%), 135,107,77,51,40
36.	<i>cis</i> -carvyl acetate (OM)	1346	0.4	C ₁₂ H ₁₈ O ₂	M ⁺ = 194, m/z; 152,134,119,109,84,79,67,43 (100%), 41
37.	α -Copaene (SH)	1375	0.9	C ₁₅ H ₂₄	M ⁺ = 204,m/z; 204,189,161,147,133,119,105 (100%), 93,81,
38.	(-)- <i>cis</i> -beta Elemene (SH)	1390	0.1	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 189,175,161,147,121,107,93 (100%), 68,55,41
39.	eugenol (OM)	1392	0.2	C ₁₀ H ₁₂ O ₂	M ⁺ = 164, m/z; 164 (100%), 149,131,121,103,91,77,65,55,40
40.	benzyl 2-methylbutyrate	1394	0.7	C ₁₂ H ₁₆ O ₂	M ⁺ =192, m/z; 192,174,108,91 (100%), 57,41

41.	<i>cis</i> -jasmone (MH)	1394	0.4	C ₁₁ H ₁₆ O	M ⁺ = 164, m/z; 164,149,135,122,110,93,79 (100%), 67,55,41,40
42.	<i>trans</i> -β-caryophyllene (SH)	1419	8.7	C ₁₅ H ₂₄	M ⁺ = 204, m/z;204,189,175,161,147,,120,105,93,79,55,41 (100%)
43.	ε-Muurolene (SH)	1430	0.1	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,176,161 (100%), 148,133,105,91,81,67,41
44.	β-Copaene (SH)	1433	0.3	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161 (100%), 147,133,119,105,91,81,67,55
45.	dihydro-β-ionone (OS)	1436	1.0	C ₁₃ H ₁₈ O	M ⁺ = 190, m/z; 175,147,131,105,91,77,43 (100%)
46.	β-trans-Farnesene (SH)	1440	1.0	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 161,133,120,93,79,69 (100%), 55,41
47.	(-)-aromadendrene (SH)	1441	2.6	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 161,147,133,119,105,91,81,55,41 (100%)
48.	isogermacrene- D (SH)	1447	1.1	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,161 (100%), 119,105,91,81,41
49.	α -humulene (SH)	1454	1.0	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161,147,121,107,93 (100%), 80,67,41,40
50.	δ-Cadinene (SH)	1469	0.1	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161,145,119,105 (100%), 91,81,55,41
51.	α-Selinene (SH)	1474	0.5	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,175,161,147,135,121,107,105,93,79,67,53,43,41 (100%)
52.	selina-4,11-diene (SH)	1476	1.2	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189 (100%), 175,161,147,133,119,105,91,41
53.	Germacrene-D (SH)	1485	9.0	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,161 (100%), 147,133,119,105,91,81,67,41

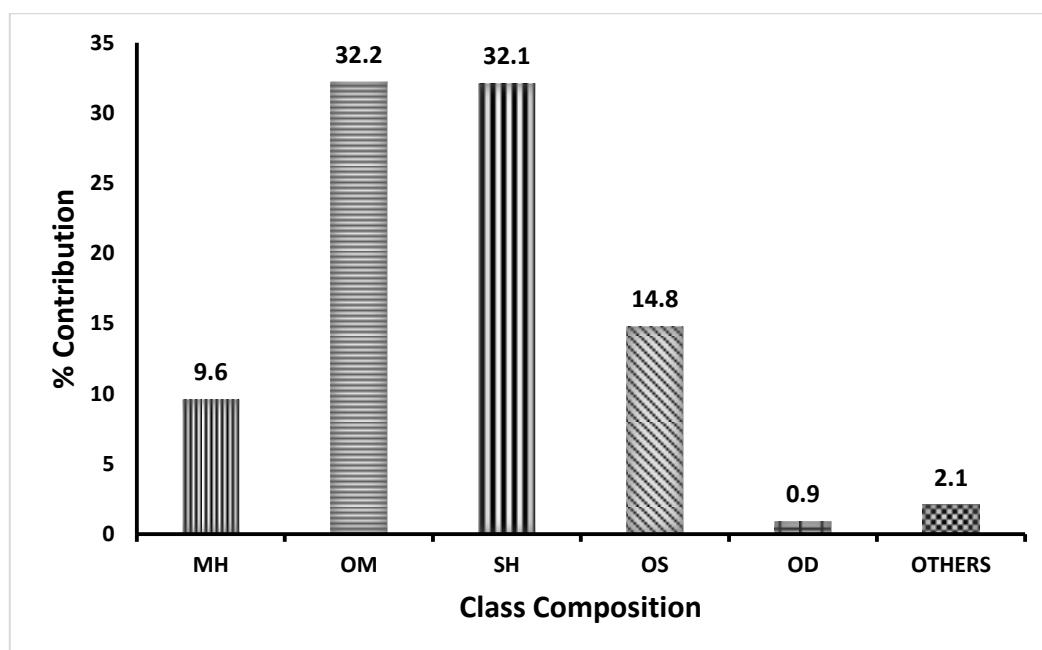
54.	(-)- β -Selinene (SH)	1489	1.0	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,175,161,147,133,121,93 (100%), 67,55,41
55.	bicyclogermacrene (SH)	1497	1.3	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161,147,136,121 (100%), 107,93,67,41
56.	γ -cadinene (SH)	1512	1.7	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 161 (100%), 147,133,119,105,91,79,67,55,41
57.	bicycloelemene (SH)	-	0.3	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 189,161,136,121 (100%), 107,93,79,67,43,41
58.	epicedrol (OS)	-	0.5	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 222,207,189,177,150,135,121,109,95 (100%), 81,69
59.	nealloocimene (SH)	-	0.3	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161 (100%), 147,133,119,105,81,69,55,
60.	α - Hydroxy-terpenyl acetate (OM)	1521	0.4	C ₁₂ H ₂₀ O ₃	M ⁺ = 212, m/z; 119,109,91,79,43 (100%), 41
61.	himbaccol (OS)	1530	0.1	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 189,147,133,121,109,95,81,69,43 (100%), 41
62.	(-)-globulol (OS)	1530	1.8	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 204,161,147,133,121,109,93,81,69,67,43 (100%), 41
63.	α -Cadinene (SH)	1538	0.1	C ₁₅ H ₂₄	M ⁺ = 204, m/z;204,289,161,147,133,119,105 (100%), 81,69,41
64.	γ -Gurjunenepoxide- (2) (OS)	1558	0.2	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,189,161,147,133,119,107,91,81 (100%), 67,55,41
65.	bornyl angelate (OS)	1569	0.8	C ₁₅ H ₂₄ O ₂	M ⁺ = 236, m/z; 236,136,109,93,83 (100%), 69,55,41
66.	spathulenol (OS)	1576	1.5	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 187,159,147,131,105,93,79,67,43 (100%), 41

67.	α -Cadinol (OS)	1580	2.0	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 204,161,137,121,95,79,69,43 (100%), 41
68.	caryophyllene oxide (OS)	1587	0.2	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 149,135,121,109,107,96,79,69,55,41 (100%)
69.	<i>cis</i> -cadin-4en-7-ol (OS)	1638	3.9	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 222,204,179,161,147,134,119 (100%), 105,93,71,43,41
70.	germacra-4 (15), 5,10 (14)- trien-1- α -ol (OS)	1683	1.3	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,177,159,145,133,121,109,91,79,67,43,41 (100%)
71.	8- α -11- elemodiol (OS)	1740	0.9	C ₁₅ H ₂₆ O ₂	M ⁺ = 238, m/z; 162,147,133,121,107,93,59 (100%), 43,41
72.	β - costol (OS)	1763	0.5	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,202,189,145,133,121 (100%), 105,91,79,67,55,41
73.	deoxyqinghaosu (OS)	1794	0.1	C ₁₅ H ₂₂ O ₄	M ⁺ = 266, m/z; 195,165,151,135,124,107,93,81,43 (100%), 41
74.	phytol (OD)	2045	0.8	C ₂₀ H ₄₀ O	M ⁺ = 296, m/z; 126,123,111,95,71 (100%), 57,43,41
75.	thunbergol (OD)	2211	0.1	C ₂₀ H ₃₄ O	M ⁺ = 290, m/z; 161,147,135,123,121,107,93,81,67,43 (100%), 41
	Total				

t= trace >0.1%, MS= Mass Spectrometry, KI= Kovatt Index, AAREO= *Artemisia annua* rainy Essential oil, AAWEO= *Artemisia annua* winter Essential oil, - = not present, MH= Monoterpenoids Hydrocarbons, SH= Sesquiterpenoids Hydrocarbons, OM=Oxygenated monoterpenoid, OS=Oxygenated sesquiterpenoids, OD= Oxygenated diterpenoid

Table 4.2 Classes of compounds present in essential oil of AAREO

S. No.	Classes of composition	% Composition
1.	Monoterpenoid hydrocarbon (MH)	9.6
2.	Oxygenated monoterpenoid (OM)	32.2
3.	Sesquiterpenoid hydrocarbon (SH)	32.1
4.	Oxygenated sesquiterpenoid (OS)	14.8
5.	Oxygenated diterpenoid (OD)	0.9
6.	others	2.1
	Total	91.7%



MH= monoterpenoids hydrocarbons, SH= sesquiterpenoids hydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoids

Fig 4.2 Class composition of AAREO

4.2 GC/MS analysis of AAWEO

A total of fifty nine components accounting for 94.1% of the total oil composition have been identified in AAWEO as presented in **Table 4.3** with structures of major constituents in **fig 4.5** .and the gas chromatogram in **Fig4.3** respectively.

Camphor (17.9%) was found to be the major component of the oil followed by eucalyptol (14.6%), (*E,E*)-2,6-dimethyl-3,5,7-octatrien-2-ol (6.3%), (*5E*)-2,6-dimethyl-1,5,7-octatrien-3-ol (5.2%) artemisia alcohol (3.5%), terpinene-4-ol (3.4%) isoborneol (3.2%), p-cymene (3.0%), caryophyllene oxide (3.1%), *trans* β -caryophyllene (2.7%), mycrene (2.6%), (+)- α -terpineol (2.5%), artemisia ketone (2.5%), camphene (2.1%), (-) β - farnesene (2.0%), yomogi alcohol (1.8%), longifolenaldehyde (1.5%), intermedeol (1.3%), sabinene (1.3%), α - pinene (1.3%) and *cis*-sabinene hydrate (1.1%) etc. The minor components (<1.0%) were α -terpinene (0.7%), α -copaene (0.7%), β -selinene (0.7%), (*3Z*)-2-Methyl-4- (2,6,6-trimethyl-1-cyclohexen-1-yl)-3-butenal (0.7%), germacra-4 (15), 5,10 (14)-trien-1- α -ol (0.6%), isodene (0.5%), β -pinene (0.5%), santolinatriene (0.4%), *neo* intermedeol (0.4%), 8- α -11- elemodiol (0.4%), *cis* -carvyl angelate (0.4%), *trans*-carveol (0.4%), p-menth-2-en-1-ol (0.3%), sobrerol 8-acetate (0.3%), (-)-spathulenol (0.3%), α -selinene (0.3%), camphene hydrate (0.3%), sorbyl butyrate (0.3%), α -humulene (0.2%), eugenol (0.2%), carvacrol (0.2%), (-)-carvone (0.2%), phellandral (0.2%), nerolidol (0.2%), t-butyl isobutyl ketone (0.2%), benzyl valerate (0.2%), (-)- β -bourbonene (0.2%), 3-furanpropanol (0.1%), *cis*-linalool oxide (0.1%), phytone (0.1%), alloaromadendrenoxide- (1) (0.1%), epoxymyrcene (0.1%), α - thujene (0.1%), tricyclene (0.1%) etc. The compounds which were present in trace amount (<0.1%) were filifolone and indipone . In term of their class composition the identified constituents in AAWEO have been categorized as mono, sesqui and diterpenoids and are presented in **table 4.4** and **fig. 4.4** respectively.

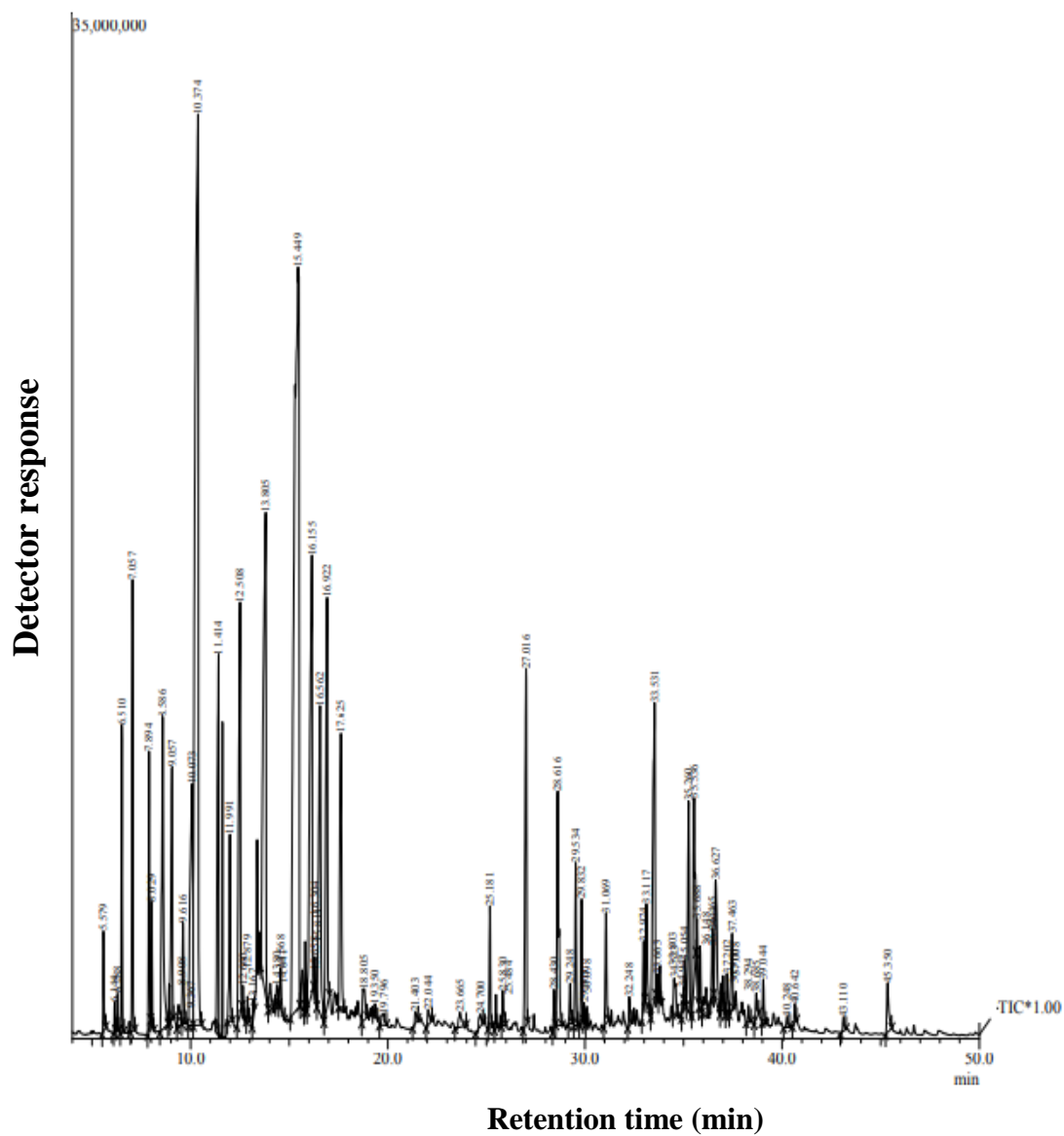


Fig 4.3: Gas chromatogram of AAWEO

Table 4.3 Chemical composition of AAWEO

S.N.	Compound Name	K.I	% Composition	Molecular Formula	Method of Identification (MS) (NIST-MS; Adams, 2007)
1.	santolinatriene (MH)	902	0.4	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 79,67
2.	t-Butyl isobutyl ketone	903	0.2	C ₉ H ₁₈ O	M ⁺ = 142, m/z; 142,85,57 (100%), 43,41,27
3.	tricyclene (MH)	923	0.1	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 79,67,43,41
4.	α- thujene (MH)	927	0.1	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136, 121,105,93 (100%), 77
5.	α- pinene (MH)	933	1.3	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136, 121,105,93 (100%), 77
6.	camphene (MH)	954	2.1	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,107,93 (100%), 79,67,53,41
7.	sabinene (MH)	972	1.3	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,105,93 (100%), 77,69
8.	β-pinene (MH)	978	0.5	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,107,93 (100%), 79,69
9.	mycrene (MH)	991	2.6	C ₁₀ H ₁₆	M⁺ = 136, m/z;136,121,107,93,79,69,53,41 (100%)
10.	yomogi alcohol (OM)	999	1.8	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 139,121,111,85,81,59,43 (100%), 41

11.	α -terpenine (MH)	1018	0.7	C ₁₀ H ₁₆	M ⁺ = 136, m/z; 136,121,107,93 (100%), 77,65,43,41
12.	p-cymene (MH)	1025	3.0	C ₁₀ H ₁₄	M ⁺ = 134, m/z; 134,119 (100%), 103,91,77,65
13.	eucalyptol (OM)	1031	14.6	C₁₀H₁₈O	M⁺ = 154, m/z; 154,139,108,84,81,69,43 (100%), 41,39
14.	artemisia ketone (OM)	1056	2.5	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,136,121,105,93,83 (100%), 55
15.	artemisia alcohol (OM)	1068	3.5	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 139,121,85 (100%), 70,67,55,41
16.	cis-sabinene hydrate (OM)	1069	1.1	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 154,121,111,93,71,69,43 (100%), 41
17.	cis-linalool Oxide (OM)	1072	0.1	C ₁₀ H ₁₈ O ₂	M ⁺ = 170, m/z; 155,112,11,94,81,59 (100%), 43,41
18.	(E,E)-2,6-dimethyl-3,5,7-octatrien-2-ol (OM)	1090	6.3	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,109,91,81,67,43 (100%), 41
19.	epoxymyrcene (OM)	1096	0.1	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 132,117,109,85,79 (100%), 59,43,41
20.	filifolone (OM)	-	t	C ₁₀ H ₁₄ O	M ⁺ = 80, m/z; 150,122,107,91,80 (100%)
21.	(5E)-2,6-Dimethyl-1,5,7-octatrien-3-ol (OM)	1120	5.2	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,119,105,84,71,67,41 (100%), 39

22.	terpinene-4-ol (OM)	1137	3.4	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 154,136,111,93,71 (100%), 69,43
23.	p-Menth-2-en-1-ol (OM)	1140	0.3	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 154,139,121,111,93,71,69,43 (100%)
24.	camphor (OM)	1149	17.9	C₁₀H₁₆O	M⁺ = 152, m/z; 152,108,95 (100%), 81,69,55,41
25.	camphene hydrate (OM)	1156	0.3	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 136,121,111,96,71 (100%), 69,43
26.	sabine ketone	1159	0.4	C ₉ H ₁₄ O	M ⁺ = 138, m/z; 138,123,96,81 (100%), 67,55,41
27.	isoborneol (OM)	1165	3.2	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 136,121,110,95 (100%), 67,43,41
28.	(+)- α -terpineol (OM)	1188	2.5	C ₁₀ H ₁₈ O	M ⁺ = 154, m/z; 136,121,93,81,59 (100%), 43
29.	(-)-carvone (OM)	1190	0.2	C ₁₀ H ₁₄ O	M ⁺ = 150, m/z; 150,135,108,93,82 (100%), 54,39
30.	<i>trans</i> -carveol (OM)	1223	0.4	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,119,109,84,83,69,55,43 (100%), 41
31.	sorbyl butyrate (OM)	1252	0.3	C ₁₀ H ₁₆ O ₂	M ⁺ = 168, m/z; 168,139,135,97,71 (100%), 43,41
32.	phellandral (OM)	1277	0.2	C ₁₀ H ₁₆ O	M ⁺ = 152, m/z; 152,137,109 (100%), 95,79,67,55,41

33.	carvacrol (OM)	1299	0.2	C ₁₀ H ₁₄ O	M ⁺ = 150, m/z; 150,135 (100%), 115,107,91,77
34.	(-)-β-bourbonene (SH)	1339	0.2	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161,123,81 (100%), 79,41
35.	eugenol (OM)	1357	0.2	C ₁₀ H ₁₂ O ₂	M ⁺ = 164, m/z;164 (100%), 149,131,121,103,91,77,65,55
36.	cis Carvyl angelate (OM)	-	0.4	C ₁₅ H ₂₂ O ₂	M ⁺ = 234, m/z; 191,152,134,119,107,93,83 (100%)
37.	longifolenaldehyde (OS)	-	1.5	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,205,187,176,163,135,109,95,81,67,55,43 (100%)
38.	α-copaene (SH)	1375	0.7	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 161,119,105 (100%), 93,81,69,55,41
39.	isolekene (SH)	1376	0.5	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161 (100%), 147,119,105,91,81
40.	<i>trans</i> β-caryophyllene (SH)	1419	2.7	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,161,133,120,105,93,79,69 (100%), 41
41.	β-farnesene (SH)	1452	2.0	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,161,133,120,93,69 (100%), 41
42.	α-humulene (SH)	1454	0.2	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,161,147,121,107,93 (100%), 80

43.	benzyl valerate	1458	0.2	C ₁₂ H ₁₆ O ₂	M ⁺ = 192 m/z; 192,174,145,108,91 (100%), 57,44
44.	β-Selinene (SH)	1469	0.7	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189,175,161,147,133,105 (100%), 93,79,67
45.	germacrene-D (SH)	1485	0.1	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,161 (100%), 133,119,105,91,81,67
46.	indipone (OM)	1492	T	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 149 (100%), 121,107,93,81,69,41
47.	sobrerol 8-acetate (OM)	-	0.3	C ₁₂ H ₂₀ O ₃	M ⁺ = 212, m/z; 152,137,119,109 (100%), 95,43
48.	3-furanpropanol (OM)	-	0.1	C ₁₀ H ₁₄ O ₂	M ⁺ = 166, m/z; 166,148,133,119,105,95,82 (100%)
49.	α-selinene (SH)	1501	0.3	C ₁₅ H ₂₄	M ⁺ = 204, m/z; 204,189 (100%), 175,161,147,133,119,107,93
50.	caryophyllene oxide (OS)	1507	3.1	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,187,177,161,149,138,121,109,93,79,69,43 (100%)
51.	(3Z)-2-methyl-4-(2,6,6-trimethyl-1-cyclohexen-1-yl)-3-butenal	1545	0.7	C ₁₄ H ₂₂ O	M ⁺ = 206, m/z; 206,191,163,135,123,107,95,81 (100%), 69,55
52.	nerolidol (SH)	1564	0.2	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 136,121,107,93,71,69 (100%), 43,41

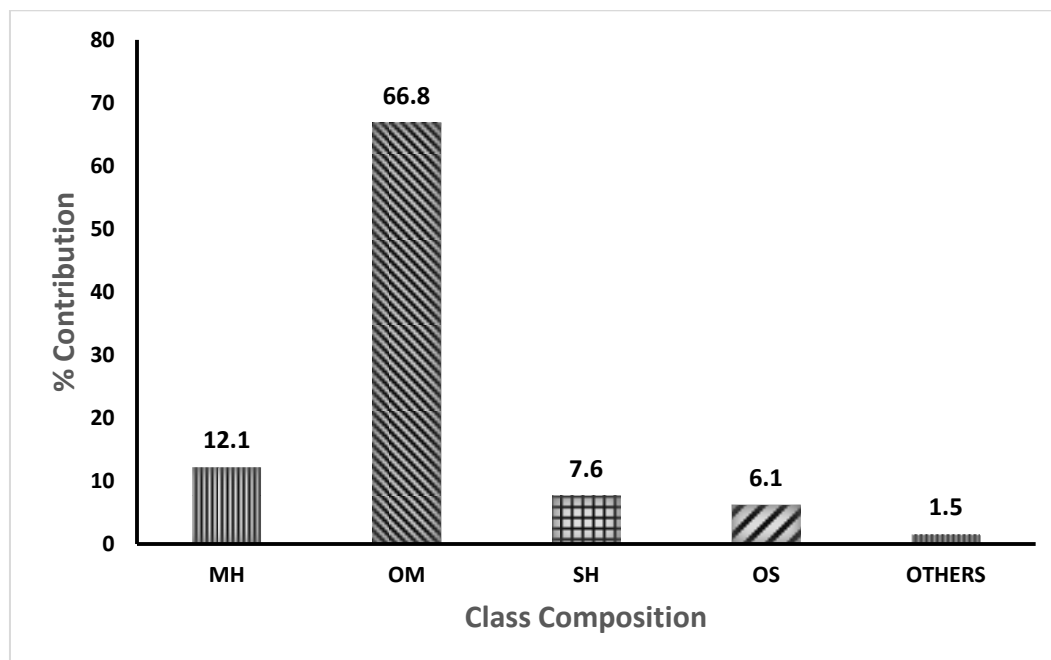
53.	(-)-spathulenol (OS)	1578	0.3	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 205,187,159,147,131,119,91,79,43 (100%), 41
54.	alloaromadendrenoxid- (1) (OS)	1641	0.1	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,189,177,159,147,133,119,105,94 (100%), 81
55.	neo intermedeol (OM)	1661	0.4	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 204,189,161,147,135,121,109,93,81,67,43 (100%)
56.	intermedeol (OM)	1668	1.3	C ₁₅ H ₂₆ O	M ⁺ = 222, m/z; 204,189,161,133,123,107,93,81,67,43 (100%)
57.	germacra-4 (15), 5,10 (14)-trien-1- α -ol (OS)	1683	0.6	C ₁₅ H ₂₄ O	M ⁺ = 220, m/z; 220,177,159,145,133,121,109,91,79,41 (100%)
58.	8- α -11- elemodiol (OS)	1740	0.4	C ₁₅ H ₂₆ O ₂	M ⁺ = 238, m/z; 162,147,121,107,93,79,59,43 (100%)
59.	phytone (OS)	1841	0.1	C ₁₈ H ₃₆ O	M ⁺ = 268, m/z; 250,179,124,109,85,71,58,43 (100%)
			94.10%		

t= trace >0.1%, MS= Mass Spectrometry, KI= Kovatt Index, AAREO= *Artemisia annua* rainy Essential oil, AAWEO= *Artemisia annua* winter Essential oil, - = not present, MH= monoterpenoids hydrocarbons, SH= sesquiterpenoids hydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoid

Table 4.4 Classes of compounds present in AAWEO

S. N.	Classes of composition	% composition
1.	Monoterpenoid hydrocarbon (MH)	12.1
2.	Oxygenated monoterpenoid (OM)	66.8
3.	Sesquiterpenoid hydrocarbon (SH)	7.6
4.	Oxygenated sesquiterpenoid (OS)	6.1
5.	Others	1.5
	Total	94.1%

MH= monoterpenoids hydrocarbons, SH= sesquiterpenoids hydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoids



MH= monoterpenoids hydrocarbons, SH= sesquiterpenoids hydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoids

Fig. 4.4 Class composition of AAWEO

4.3 Comparative study of AAREO and AAWE0

Comparing the identified constituents of AAREO and AAWE0 by GC-MS analysis, It was observed that the monoterpene hydrocarbon in AAREO contributed 9.6 % while in AAWE0, there contribution was 12.1%, being mainly represented by α -pinene, camphene, sabinene, p-cymene, myrcene, γ -terpinene etc. The oxygenated monoterpenoids in both the oils contributed 32.2% and 66.8 % respectively which dominated overall composition in both the oils. The major oxygenated monoterpenoids identified in AAREO & AAWE0 were eucalyptol (1,8 cineole), *cis*-sabinene hydrate, yomogi alcohol, artemisia ketone, artemisia alcohol, amyl-vinyl carbinol, linalool, hotrineol etc. However the major constituent identified was camphor with 14.1 % and 17.9 % contribution in AAREO and AAWE0 respectively. The sesquiterpene hydro carbon were dominated in AAREO with 32.1% while 7.6 % in AAWE0. The constituents identified as sesquiterpene hydrocarbons were *trans*- β -caryophyllene, α -humulene, germacrene-D, β -selinene, aromadendrene etc. besides other minor and trace constituents. Similarly the oxygenated sesquiterpenoids were dominated in AAREO with 14.8% contribution over AAWE0 with 6.1% contribution only. The major representative oxygenated sesquiterpenoids identified were caryophyllene oxide, longifolenaldehyde, spathulenol, *cis*-cadin-4-en-1-ol, α -cadinol, globulol etc. besides other minor and trace constituents. The contribution of oxygenated diterpene identified in AAREO contributed only 0.9% being represented by phytol (0.8%) and thunbergol (0.1%). However none of the oxygenated diterpene could be identified in AAWE0. The other constituents identified in both the oils contributed 2.1 and 1.5% in AAREO & AAWE0 respectively. In terms of constituents identified in AAREO & AAWE0 both the oils showed different qualitative and quantitative makeup viz..santolinatriene (0.4%), t-Butyl isobutyl ketone (0.2%), myrcene (2.6%), yomogi alcohol (1.8%), α -Terpinene (0.7%), artemisia ketone (2.5%) artemisia alcohol (3.5%), *cis*-linalool Oxide (0.1%), (*E,E*)-2,6-dimethyl-3,5,7-octatrien-2-ol (6.3%), epoxy myrcene (0.1%), (*5E*)-2,6-dimethyl-1,5,7-octatrien-3-ol (5.2%), sorbyl butyrate (0.3%), phellandral (0.2%), carvacrol, (0.2%), (-)- β -bourbonene (0.2%), isodene (0.5%), benzyl valerate (0.2%), (3*Z*)-2-methyl-4- (2,6,6-trimethyl-1-cyclohexen-1-yl)-3-butenal (0.7%),

alloaromadendrenoxid- (1) (0.1%), *neo*-intermedeol (0.4%), intermedeol (1.3%), phytone (0.1%), *cis*-carvyl angelate (0.4%), longifolenaldehyde (1.5%), filifolone ($t > 0.1\%$) and indipone ($t > 0.1\%$) were identified in AAWEO and were missing in AAREO, Similarly the constituents like *cis*-3-Hexen-1-ol (0.3%), amyl vinyl carbinol (1.5%), 2- δ -carene (0.5%), *trans*-3-hexenyl acetate (1.0%), Υ -terpinene (1.0%), hotrienol (0.1%), linalool (0.2%), ethanone (0.2%), pinocarvone (0.2%), naphthalene (0.8%), *cis*-Carveol (0.2%), lavandulyl acetate (2.1%), bornyl acetate (0.1%), thymol (0.3%), 4-vinyl guaiacol (0.1%), *cis*-carvyl acetate (0.4%), (-)-*cis*- β elemene (0.1%), benzyl 2-methylbutyrate (0.7%), *cis*-jasmane (0.4%), ϵ -muurolene (0.1%), β -Copaene (0.3%), dihydro- β -ionone (1.0%), (-)-aromadendrene (2.6%), isogermacrene- D (1.1%), δ -cadinene (0.1%), selina-4,11-diene (1.2%), bicyclogermacrene (1.3%), Υ -cadinene (1.7%), α - hydroxy-terpenyl acetate (0.4%), (-)-globulol (1.8%), himbaccol (0.1%), Υ -gurjunenepoxide- (2) (0.2%), bornyl angelate (0.8%), α -Cadinol (2.0%), *cis*-cadin-4en-7-ol (3.9%), β - costol (0.5%), deoxyqinghaosu (0.1%), phytol (0.8%), bicycloelemene (0.3%), epicedrol (0.5%), isogeranial (0.2%), and neoalloocimene (0.3%) were identified in AAREO and were found to be missing in AAWEO respectively.

The detailed comparative composition in terms of identified constituents is being depicted in **table 4.5** and the class composition in **table 4.6 & fig 4.6** respectively and their mass spectra of major compounds in **appendix -1**.

It has been reported that the chemical composition of the essential oil of different species of aromatic plants might differ mainly due to harvesting season and geographical conditions arising from seasonal variations (**Verma *et al.*, 2010**).

Padalia *et al.*, (2011) have reported the presence of camphor (22.8%-42.6%), 1,8 cineole (3.7- 8.4%), β -caryophyllene (2.0-9.2%), germacrene-D (0.5-7.3%) in essential oils of *A.annua* collected at different stages.

Camphor (48.00%), 1,8 – cineole (9.39%), camphene (6.98%) and spathulenol (4.695%) have been reported as major constituents in the essential oils of *A.annua* (**Mohammadreza, 2008**).

Presence of artemisia ketone (68.5%) as a major component along with camphor (27.5%), 1,8-cineole (22.8%), α -pinene (16.0%), camphene hydrate (12.0%), and artemisia alcohol (5.2%) have been reported by (Charles *et al.*, 1991). Artemisia ketone (35.7%), α -pinene (16.5%), 1,8-cineole (5.5%) have been reported in the aerial parts of *A.annua* from Serbia (Radulovic *et al.*, 2013). Camphor (48%) including 1,8 cineole (9.39%), camphene (6.98%) have been reported in essential oil of *A.annua* (Verdian-rizi, 2009).

Islamuddin *et al.*, (2014) reported the chemical constituent identified in essential oil of *A.annua* with camphor (52.06%). The study also revealed the presence of the other constituents like β -caryophyllene (10.95 %), 1,8-cineole (5.57 %), β -caryophyllene oxide (4.21 %), β -farnesene (3.83 %), α -copaene (2.91 %), β -cyperone (1.93 %), α -selinene (1.54 %), and *trans*-pinocarveol (1.22 %). Bagchi *et al.*, (2003) reported the major changes in the chemical constituents of *A.annua* due to the change in growing season. It has been reported that camphor (10.5–44.4%) tends to be the major constituent of oil instead of usually dominated constituent i.e. artemisia ketone. Artemisia ketone (35.7%), The major constituent followed by 1,8-cineole (31.5%), α -pinene (11.2%), artemisia alcohol (5.2%), and myrcene (4.6%) have been reported in the essential oils of *A.annua* (Libbey and Sturtz, 1989). Omer *et al.*, (2013) have reported 1-camphor, camphene, β -camphene, isoartemisia ketone, β -caryophyllene and α -pinene in the essential oils of *A.annua*.

Comparing the chemical composition of essential oils isolated from *A.annua* in two different seasons with that of previous reports, It was observed that AAREO and AAWEEO have vastly different qualitative and quantitative variations. It has also been observed that camphor and 1,8 cineole, germacrene-D with different quantitative make up has been reported as the major constituent in earlier as well as in the present study along with other constituents. Under stress conditions, plants biosynthesize secondary metabolites such as terpenoids, alkaloids, polyphenolics, and other compounds in response to their need for defence. It could be possible that the qualitative and quantitative differences in chemical composition observed in AAREO and AAWEEO are due to changing weather conditions.

Table 4.5 Comparative chemical composition of AAREO and AAWE0:

S.N.	Name of compound	% Contribution		KI	Method of Identification (MS) (NIST-MS; Adams, 2007)
		AAREO	AAWE0		
1.	<i>cis</i> -3-hexen-1-ol	0.3	-	868	M ⁺ = 100, m/z; 100,82,67,55,41 (100%), 27
2.	santolinatriene (MH)	-	0.4	902	M ⁺ = 136, m/z; 136,121,105,93 (100%), 79,67
3.	t-butyl isobutyl ketone	-	0.2	903	M ⁺ = 142, m/z; 142,85,57 (100%), 43,41,27
4.	tricyclene (MH)	0.1	0.1	923	M ⁺ = 136, m/z;136,121,105,93 (100%), 79,67
5.	α -thujene (MH)	0.1	0.1	927	M ⁺ = 136, m/z; 136,121,105,93 (100%), 77,65
6.	α -pinene (MH)	1.2	1.3	933	M ⁺ = 136, m/z; 136,121,105,93 (100%), 92,91,77,67,41
7.	camphene (MH)	2.0	2.1	954	M ⁺ = 136, m/z; 136,121,107,93 (100%), 79,67,53,41
8.	amyl vinyl carbinol (OM)	1.5	-	969	M ⁺ = 128, m/z; 127,99,85,72,57 (100%), 43,29,27
9.	sabinene (MH)	0.4	1.3	972	M ⁺ = 136, m/z; 136,121,105,93 (100%), 77,69,41
10.	β -pinene (MH)	1.0	0.5	978	M ⁺ = 136 m/z; 136,121,107,93 (100%), 79,69,41

11.	myrcene (MH)	-	2.6	991	M ⁺ = 136, m/z; 136,121,107,93,79,69,53,41 (100%)
12.	yomogi alcohol (OM)	-	1.8	999	M ⁺ = 154, m/z; 139,121,111,85,81,59,43 (100%), 41
13.	2- δ -carene (MH)	0.5	-	1000	M ⁺ = 136, m/z; 136,121,105,93 (100%), 79,77,65
14.	<i>trans</i> -3-hexenyl acetate	1.0	-	1001	M ⁺ = 142, m/z; 112,98,82,67,43 (100%), 41
15.	α -terpinene (MH)	-	0.7	1018	M ⁺ = 136, m/z; 136,121,107,93 (100%), 77,65,43,41
16.	p-cymene (MH)	2.9	3.0	1025	M ⁺ = 134, m/z; 134,119 (100%), 91,77,65
17.	eucalyptol (OM)	4.2	14.6	1031	M ⁺ = 154, m/z; 154,139,108,84,81,69,43 (100%), 41
18.	artemisia ketone (OM)	-	2.5	1056	M ⁺ = 152, m/z; 152,136,121,105,93,83 (100%), 55
19.	γ -terpinene (MH)	1.0	-	1058	M ⁺ = 136, m/z; 136,121,,105,93 (100%), 77,43,41
20.	artemisia alcohol (OM)	-	3.5	1068	M ⁺ = 154, m/z; 139,121,85 (100%), 70,67,55,41
21.	<i>cis</i> -sabinene hydrate (OM)	0.2	1.1	1069	M ⁺ = 154, m/z; 154,121,111,93,71,69,43 (100%), 41
22.	hotrienol (OM)	0.1	-	1072	M ⁺ = 152, m/z; 119,107,91,83,71 (100%), 67,43

23.	<i>cis</i> -linalool Oxide (OM)	-	0.1	1072	M ⁺ = 170, m/z; 155,112,11,94,81,59 (100%), 43,41
24.	(<i>E,E</i>)-2,6-dimethyl-3,5,7-octatrien-2-ol (OM)	-	6.3	1090	M ⁺ = 152, m/z; 152,137,109,91,81,67,43 (100%), 41
25.	linalool (OM)	0.2	-	1096	M ⁺ = 154, m/z; 136,121,93,79,71,69,43 (100%), 41
26.	epoxymyrcene (OM)	-	0.1	1096	M ⁺ = 152, m/z; 132,117,109,85,79 (100%), 59,43,41
27.	(<i>5E</i>)-2,6-dimethyl-1,5,7-octatrien-3-ol (OM)	-	5.2	1120	M ⁺ = 152, m/z; 152,137,119,105,84,71,67,41 (100%), 39
28.	p-menth-1-en-4-ol/ terpinen-4-ol (OM)	1.8	3.4	1137	M ⁺ = 154, m/z; 154,136,111,93,71 (100%), 69,43
29.	p-menth-2-en-1-ol (OM)	0.1	0.3	1140	M ⁺ = 154, m/z; 154,140,139,121,111,94,79,71,69,43 (100%), 41
30.	camphor (OM)	14.1	17.9	1149	M ⁺ = 152, m/z; 152,137,108,95 (100%), 81,69,55,41
31.	camphene hydrate (OM)	0.1	0.3	1156	M ⁺ = 154, m/z; 136,121,111,96,71 (100%), 69,43
32.	sabine ketone	-	0.4	1159	M ⁺ = 138, m/z; 138,123,96,81 (100%), 67,55,41
33.	ethanone (OM)	0.2	-	1161	M ⁺ = 152, m/z; 152,137,109 (100%), 93,81,67
34.	Pinocarvone (OM)	0.2	-	1164	M ⁺ = 150, m/z; 150,135,122,108,81,69,53 (100%), 41

35.	isoborneol (OM)	3.2	3.2	1165	M ⁺ = 154, m/z; 121,110,95 (100%), 81,67,41
36.	naphthalene (SH)	0.8	-	1181	M ⁺ = 204, m/z;204,189,175,162,147,133,119 (100%), 105,93,79,67,55,41
37.	(+)- α -terpineol (OM)	1.2	2.5	1188	M ⁺ = 154, m/z; 136,121,93,81,59 (100%), 43,41
38.	myrtenal (OM)	t	-	1197	M ⁺ = 150, m/z; 135,107,91,79 (100%), 66,53,41
39.	<i>trans</i> -carveol (OM)	0.6	0.4	1223	M ⁺ = 152, m/z; 152,137,119,109,84,83,69,55,43 (100%), 41
40.	<i>cis</i> -carveol (OM)	0.2	-	1232	M ⁺ = 152, m/z; 134,119,109,84 (100%), 83,69,55,41
41.	carvone (OM)	0.1	0.2	1246	M ⁺ = 150, m/z; 150,135,108,93,82 (100%), 54,39
42.	sorbyl butyrate (OM)	-	0.3	1252	M ⁺ = 168, m/z; 168,139,135,97,71 (100%), 43,41
43.	phellandral (OM)	-	0.2	1277	M ⁺ = 152, m/z; 152,137,109 (100%), 95,79,67,55,41
44.	lavandulyl acetate (OM)	2.1	-	1284	M ⁺ = 196, m/z; 136,121,107,93,80,69 (100%), 43,41,40
45.	bornyl acetate (OM)	0.1	-	1285	M ⁺ = 196, m/z; 196,136,121,108,95 (100%), 80,43,41

46.	thymol (OM)	0.3	-	1293	M ⁺ = 150, m/z; 150,135 (100%), 115,91,77,65,51,41
47.	carvacrol (OM)	-	0.2	1299	M ⁺ = 150, m/z; 150,135 (100%), 115,107,91,77
48.	4-vinyl guaiacol	0.1	-	1309	M ⁺ = 150, m/z; 150 (100%), 135,107,77,51,40
49.	(-)- β -bourbonene (SH)	-	0.2	1339	M ⁺ = 204, m/z; 204,189,161,123,81 (100%), 79,41
50.	<i>cis</i> -carvyl acetate (OM)	0.4	-	1346	M ⁺ = 194, m/z; 152,134,119,109,84,79,67,43 (100%), 41
51.	α -copaene (OM)	0.9	0.7	1375	M ⁺ = 204, m/z; 189,161,147,133,119,105 (100%), 93,81,69,55,41
52.	isolekene (SH)	-	0.5	1376	M ⁺ = 204, m/z; 204,189,161 (100%), 147,119,105,91,81
53.	(-)- <i>cis</i> -beta Elemene (SH)	0.1	-	1390	M ⁺ = 204, m/z; 189,175,161,147,121,107,93 (100%), 68,55,41
54.	eugenol (OM)	0.2	0.2	1392	M ⁺ = 164, m/z;164 (100%), 149,131,121,103,91,77,65,55
55.	benzyl 2-methylbutyrate	0.7	-	1394	M ⁺ =192, m/z; 192,174,108,91 (100%), 57,41
56.	<i>cis</i> -jasmone (MH)	0.4	-	1394	M ⁺ = 164, m/z; 164,149,135,122,110,93,79 (100%), 67,55,41,40
57.	<i>trans</i> β -caryophyllene (SH)	8.7	2.7	1419	M ⁺ = 204, m/z; 204,189,161,133,120,105,93,79,69 (100%), 41

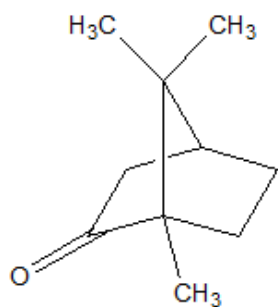
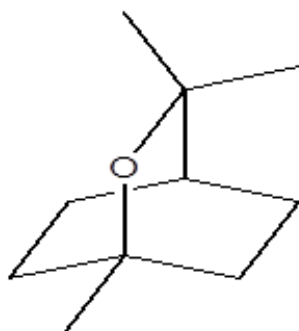
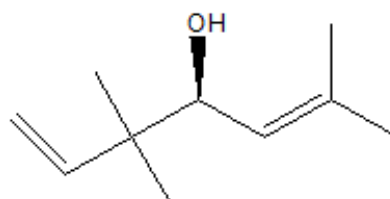
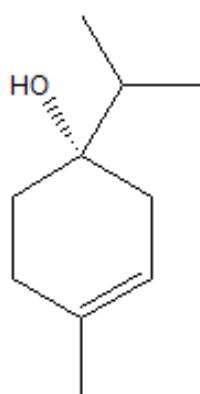
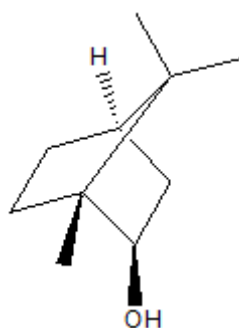
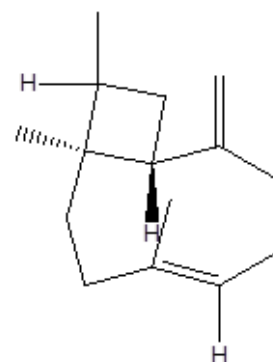
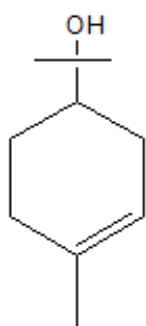
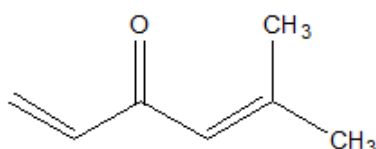
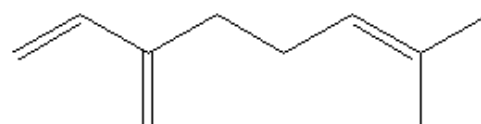
58.	ϵ -Muurolene (SH)	0.1	-	1430	M ⁺ = 204, m/z; 204,176,161 (100%), 148,133,105,91,81,67,41
59.	β -copaene (SH)	0.3	-	1433	M ⁺ = 204, m/z; 204,189,161 (100%), 147,133,119,105,91
60.	dihydro- β -ionone (OS)	1.0	-	1436	M ⁺ = 190, m/z; 175,147,131,105,91,77,43 (100%)
61.	(-)-aromadendrene (SH)	2.6	-	1441	M ⁺ = 204, m/z; 161,147,133,119,105,91,81,55,41 (100%)
62.	isogermacrene- D (SH)	1.1	-	1447	M ⁺ = 204, m/z; 204,161 (100%), 119,105,91,81,41
63.	β -farnesene (SH)	1.0	2.0	1452	M ⁺ = 204, m/z; 204,161,133,120,93,69 (100%), 41
64.	α -humulene (SH)	1.0	0.2	1454	M ⁺ = 204, m/z; 204,161,147,121,107,93 (100%), 80
65.	benzyl valerate	-	0.2	1458	M ⁺ = 192 m/z; 192,174,145,108,91 (100%), 57,44
66.	δ -cadinene (SH)	0.1	-	1469	M ⁺ = 204, m/z; 204,189,161,145,119,105 (100%), 91,81,55,41
67.	α -selinene (SH)	0.5	0.3	1474	M ⁺ = 204, m/z;204,189 (100%), 175,161,147,133,119,107,93
68.	selina-4,11-diene (SH)	1.2	-	1476	M ⁺ = 204, m/z; 204,189 (100%), 175,161,147,133,105,91,41
69.	germacrene-d (SH)	9.0	0.1	1485	M ⁺ = 204, m/z; 204,161 (100%), 147,133,119,105,91,81,67,41

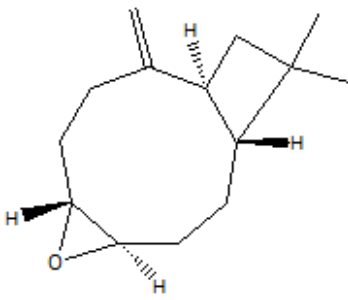
70.	β -Selinene (SH)	1.0	0.7	1489	M ⁺ = 204, m/z; 204,189,175,161,147,133,121,93 (100%), 67,55,41
71.	indipone (OM)	-	t	1492	M ⁺ = 220, m/z; 149 (100%), 121,107,93,81,69,41
72.	bicyclogermacrene (SH)	1.3	-	1497	M ⁺ = 204, m/z; 204,189,161,147,136,121 (100%), 107,93
73.	Υ -cadinene (SH)	1.7	-	1512	M ⁺ = 204, m/z; 161 (100%), 147,133,119,105,91,79,67,55,41
74.	α - hydroxy-terpenyl acetate (OM)	0.4	-	1521	M ⁺ = 212, m/z; 119,109,91,79,43 (100%), 41
75.	(-)-globulol (OS)	1.8	-	1530	M ⁺ = 222, m/z; 204,161,147,133,121,109,93,81,69,67,43 (100%),
76.	himbaccol (OS)	0.1	-	1530	M ⁺ = 222, m/z; 189,147,133,121,109,95,81,69,43 (100%), 41
77.	spathulenol (OS)	1.5	0.3	1536	M ⁺ = 220, m/z; 205,187,159,147,131,119,91,79,43 (100%), 41
78.	α -cadinene (SH)	0.1	-	1538	M ⁺ = 204, m/z;204,289,161,147,133,119,105 (100%), 81,69,41
79.	(3Z)-2-methyl-4- (2,6,6-trimethyl-1-cyclohexen-1-yl)-3-butenal	-	0.7	1545	M ⁺ = 206, m/z; 206,191,163,135,123,107,95,81 (100%), 69,55
80.	Υ -gurjunenepoxide- (2) (OS)	0.2	-	1558	M ⁺ = 220, m/z; 220,189,161,147,133,119,107,91,81 (100%), 67,55,41

81.	bornyl angelate (OS)	0.8	-	1569	M ⁺ = 236, m/z; 236,136,109,93,83 (100%), 69,55,41
82.	α -cadinol (OS)	2.0	-	1580	M ⁺ = 222, m/z; 204,161,137,121,95,79,69,43 (100%), 41
83.	caryophyllene oxide (OS)	0.2	3.1	1587	M ⁺ = 220, m/z; 220,187,177,161,149,138,121,109,93,79,69,43 (100%)
84.	<i>cis</i> -cadin-4-en-7-ol (OS)	3.9	-	1638	M ⁺ = 222, m/z; 222,204,179,161,147,134,119 (100%), 105,93,71,43,41
85.	alloaromadendrenoxid- (1) (OS)	-	0.1	1641	M ⁺ = 220, m/z; 220,189,177,159,147,133,119,105,94 (100%), 81
86.	<i>neo</i> intermedeol (OM)	-	0.4	1661	M ⁺ = 222, m/z; 204,189,161,147,135,121,109,93,81,67,43 (100%)
87.	intermedeol (OM)	-	1.3	1668	M ⁺ = 222, m/z; 204,189,161,133,123,107,93,81,67,43 (100%)
88.	germacra-4 (15), 5,10 (14)-trien-1- α -ol (OS)	1.3	0.6	1683	M ⁺ = 220, m/z; 220,177,159,145,133,121,109,91,79,41 (100%)
89.	8- α -11- elemodiol (OS)	0.9	0.4	1740	M ⁺ = 238, m/z; 162,147,121,107,93,79,59,43 (100%)
90.	β - costol (OS)	0.5	-	1763	M ⁺ = 220, m/z; 220,202,189,145,133,121 (100%), 105,91,79,67,55,41

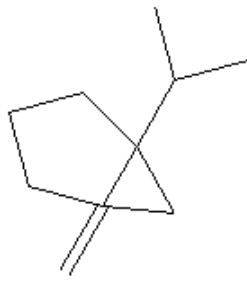
91.	deoxyqinghaosu (OS)	0.1	-	1794	M ⁺ = 266, m/z; 195,165,151,135,124,107,93,81,43 (100%), 41
92.	phytone (OS)	-	0.1	1841	M ⁺ = 268, m/z; 250,179,124,109,85,71,58,43 (100%)
93.	phytol (OD)	0.8	-	2045	M ⁺ = 296, m/z; 126,123,111,95,71 (100%), 57,43,41
94.	thunbergol (OD)	0.1	-	2211	M ⁺ = 290,m/z; 161,147,135,123,107,93,67,43 (100%), 41
95.	cis-carvyl angelate (OM)	-	0.4	-	M ⁺ = 234, m/z; 191,152,134,119,107,93,83 (100%)
96.	longifolenaldehyde (OS)	-	1.5	-	M ⁺ = 220, m/z; 220,205,187,176,163,135,109,95,81,67,55,43 (100%)
97.	sobrerol 8-acetate (OM)	0.5	0.3	-	M ⁺ = 212, m/z; 152,137,119,109 (100%), 95,43
98.	3-furanpropanol (OM)	-	0.1	-	M ⁺ = 166, m/z; 166,148,133,119,105,95,82 (100%)
99.	bicycloelemene (SH)	0.3	-	-	M ⁺ = 204, m/z; 189,161,136,121 (100%), 107,93,79,67,43,41
100.	epicedrol (OS)	0.5	-	-	M ⁺ = 222, m/z; 222,207,189,177,150,135,121,109,95 (100%), 81,69,41
101.	filifolone (OM)	-	t	-	M ⁺ = 80, m/z; 150,122,107,91,80 (100%)
102.	isogeranial (OM)	0.2	-	-	M ⁺ = 152, m/z; 152,137,119,109,94 (100%), 81,67,53,43,41
103.	nealloocimene (SH)	0.3	-	-	M ⁺ = 204, m/z; 204,189,161 (100%), 47,133,119,105,81,69,55,41
	Total	91.7%	94.10%		

t= trace >0.1%, MS= Mass Spectrometry, KI= Kovatt Index, AAREO= *Artemisia annua* rainy Essential oil, AAWEO= *Artemisia annua* winter Essential oil, - = not present, MH= monoterpenoids hydrocarbons, SH= sesquiterpenoids hydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoids

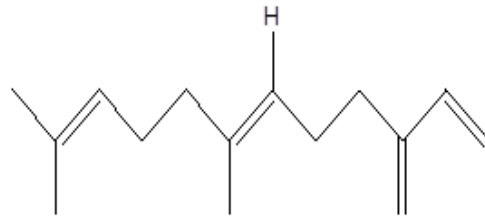
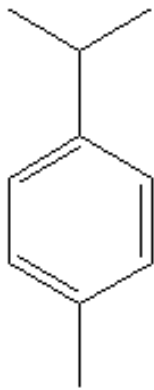
**camphor****eucalyptol****artemisia alcohol****terpinene-4-ol****isoborneol*****trans* β-caryophyllene****(+)- α-terpineol****artemisia ketone****myrcene**



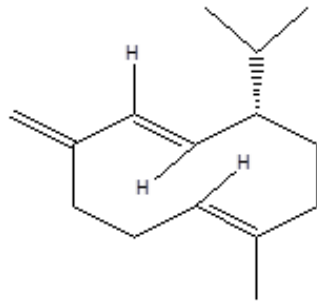
caryophyllene oxide



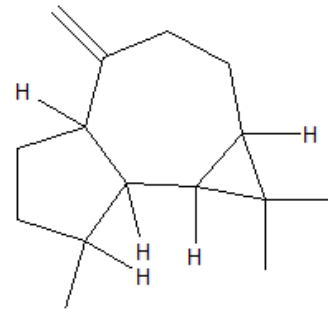
sabinene

 β -Farnesene

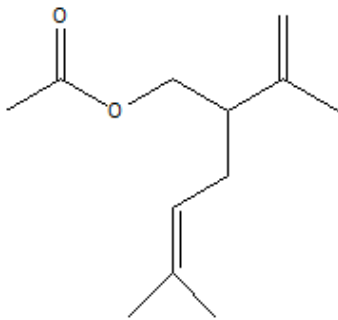
p-cymene



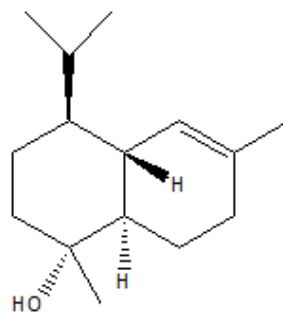
germacrene-D



(-)-aromadendrene



lavandulyl acetate

 α -Cadinol

camphene

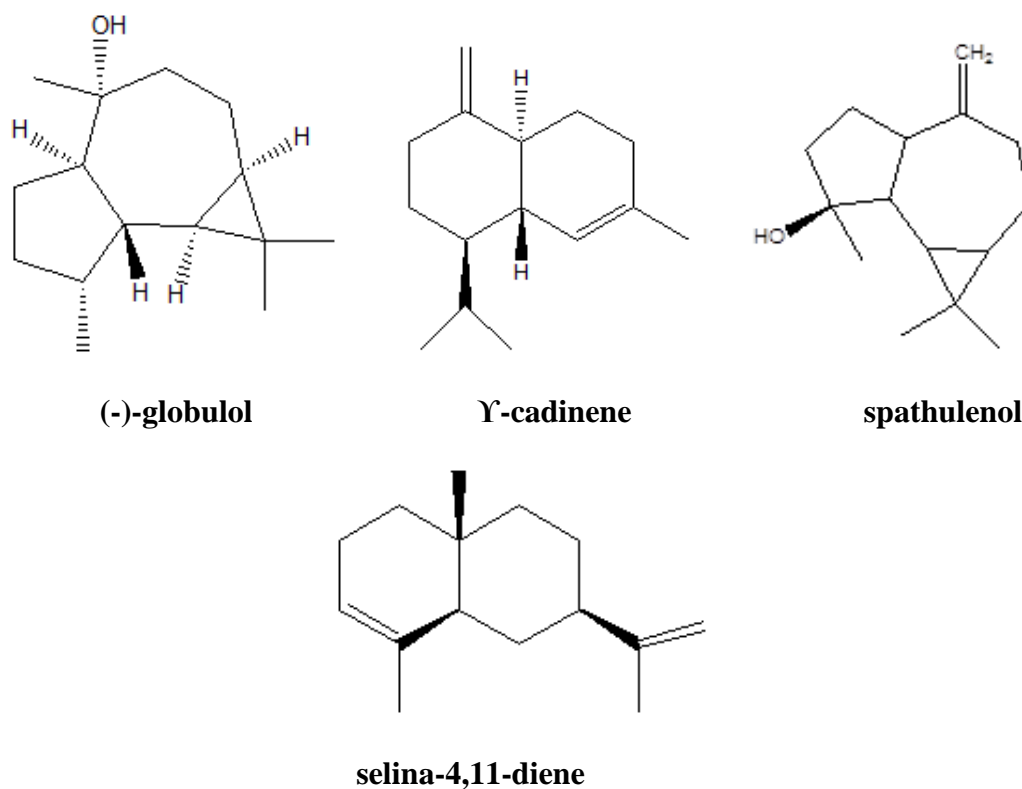
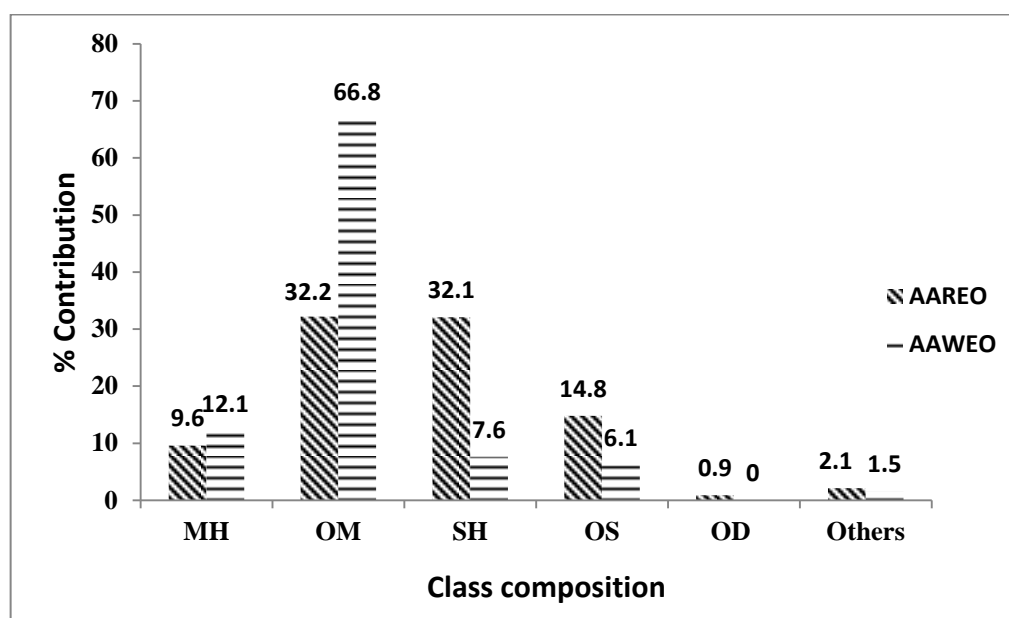


Fig 4.5 Structures of major compounds present in AAREO & AAWE0

Table: 4.6 Comparison of classes of compounds present in AAREO and AAWE0

S.N.	Classes of composition	% Composition	
		AAREO	AAWE0
1.	monoterpenoid hydrocarbon (MH)	9.6	12.1
2.	oxygenated monoterpenoid (OM)	32.2	66.8
3.	sesquiterpenoid hydrocarbon (SH)	32.1	7.6
4.	oxygenated sesquiterpenoid (OS)	14.8	6.1
5.	oxygenated Diterpenoid (OD)	0.9	-
6.	Others	2.1	1.5
	Total	91.7%	94.10%

MH= monoterpenoids hydrocarbons, SH= sesquiterpenoidshydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoids



MH= monoterpenoids hydrocarbons, SH= sesquiterpenoidhydrocarbons, OM= oxygenated monoterpenoid, OS= oxygenated sesquiterpenoids, OD= oxygenated diterpenoids

Fig 4.6 : Comparative class composition of AAREO and AAWEO

4.4 Biological activities

Various biological activities like antioxidant, *in-vitro* anti-inflammatory, *in-vitro* antidiabetic, anti-feedant, nematocidal activity anti bacterial activity and anti fungal activity of AAREO & AAWEO are being presented as follows.

4.4.1 Antioxidant activity

Antioxidants are the compounds that are known to be capable of retarding the oxidation of an oxidizable material, despite used in very moderate amount. Most common synthetic antioxidants which are found to be damaging to human health are butylated hydroxyl anisole (BHA) or butylated hydroxyl toluene (BHT) (Amorati *et al.*, 2013). Natural antioxidant compounds possess antioxidant activity through different mechanisms which includes chain breaking by donation of hydrogen atoms or electrons which convert free radicals into stable species (Iqbal *et al.*, 2013). Biological redox reactions leads to the production of reactive oxygen and nitrogen free radicals, which can be triggered by a variety of environmental factors such as pollution, smoke, and sunlight. Reactive oxygen and nitrogen species (ROS/RNS) are

formed when electron acceptors, such as molecular oxygen and nitrogen, react readily with free radicals to become radicals themselves. There's a lot of evidence that these free radicals produce oxidative stress by causing oxidative damage to biomolecules like lipids, proteins, and nucleic acids (Anthony *et al.*, 2012). Oxidative stress influences the development and of many diseases, such as cardiovascular diseases, inflammation, neurodegenerative diseases and aging processes also arterial hypertension can be developed from oxidative stress and is believed to result from systemic damage in different target tissues by oxygen free radicals (Khaleel and Buchbauer 2018).

Plants produce a plethora of secondary products with potent antioxidant properties. Some natural ingredients, such as essential oils, have strong biological, antioxidant, therapeutic, and pharmaceutical applications (Mata *et al.*, 2007).

Fig. 4.7 depicts the generation of free radicals and their scavenging by antioxidants where as mechanism of hydrocarbon autoxidation and antioxidant protection is shown in Fig.4.8.

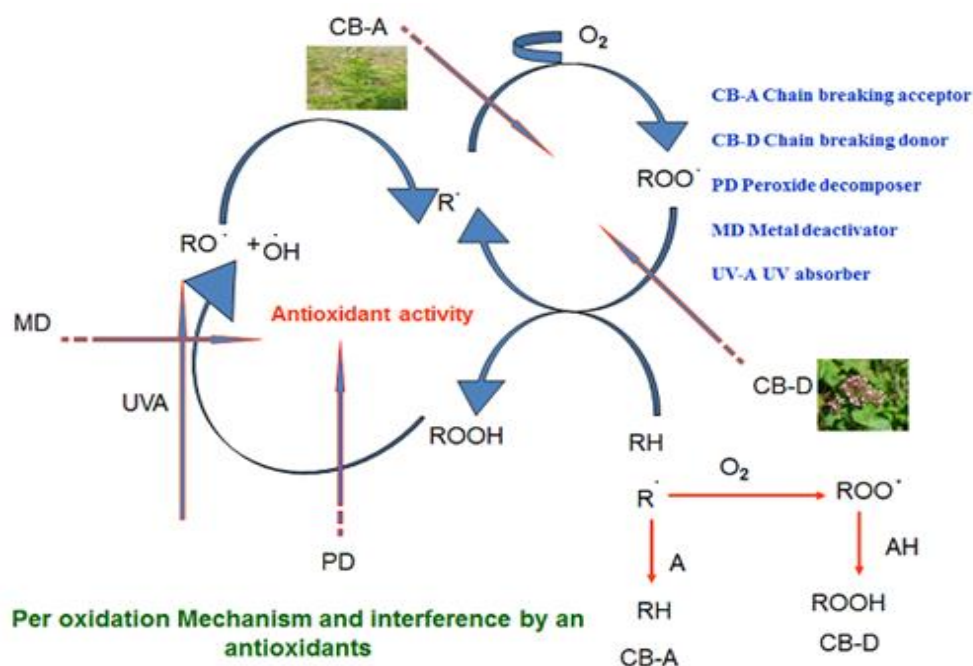


Fig. 4.7 Generation of free radicals and their scavenging by antioxidants

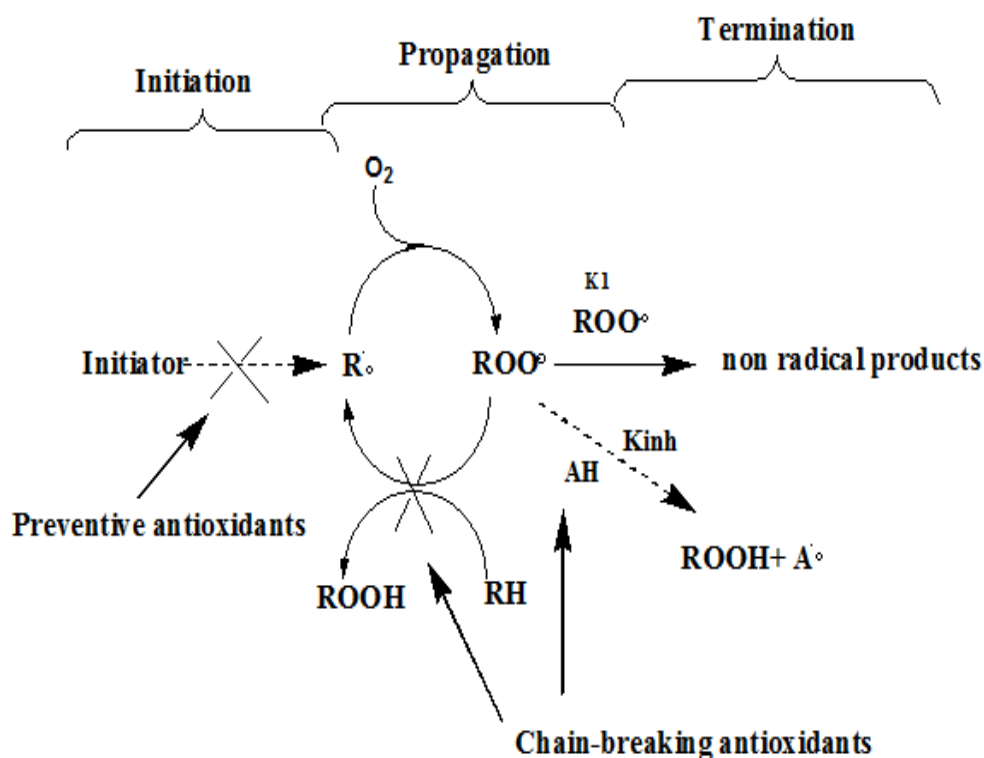


Fig. 4.8 Mechanism of hydrocarbon autoxidation and antioxidant protection

In present study, the *in vitro* antioxidant activity of AAREO and AAWEEO was evaluated by DPPH radical scavenging method, H_2O_2 radical scavenging activity and metal chelating activity in comparison to standard antioxidants, BHT, Na_2 -EDTA. The results obtained are being described as under.

4.4.1.1 DPPH free radical scavenging activity

DPPH radical scavenging assay is the most renowned and widely used for the of antioxidant activity of essential oils. DPPH is known as a stable radical in solution and appears purple colour absorbing at 515 nm in methanol. This assay is based on the principle that DPPH on accepting a hydrogen (H) atom from the scavenger molecule i.e. antioxidant, resulting into reduction of DPPH to 1,1-diphenyl-2-picrylhydrazine (DPPH-H) or DPPH₂ and characterized by changing colour from purple to yellow with decrease in absorbance **fig 4.9**. Hence DPPH provides a simple way to analyse anti oxidant activity spectrophotometrically (Mishra *et al.*, 2012).

Table 4.7 DPPH radical scavenging antioxidant activity of AAREO and AAWEO

S.N	Sample Name	R	20µl/mL		40 µl/mL		60 µl/mL		80 µl/mL		100 µl/mL		IC50
			O.D	%I	O.D	%I	O.D	%I	O.D	%I	O.D	%I	
1.	AAREO	R1	0.385 0.386 0.387	20.24	0.292 0.290 0.295	39.66	0.278 0.276 0.281	42.56	0.242 0.240 0.245	50	0.197 0.195 0.200	59.29	77.29
		Avg	0.386±0.001		0.292 ±0.002		0.278±0.002		0.242 ±0.002		0.197±0.002		
		R2	0.382 0.383 0.382		24.63		0.298 0.296 0.301		38.42		0.272 0.270 0.275		
		Avg	0.382±0.005	0.298±0.002		0.272±0.002	0.248±0.002	0.199±0.002					
		R3	0.385 0.386 0.387	20.24		0.298 0.296 0.301	38.42	0.282 0.280 0.285		41.7	0.244 0.242 0.247	49.58	
Avg	0.385±0.001	21.27±2.534	0.298±0.002		39.12±0.715	0.282±0.002		47.01±1.055	0.244±0.002		57.085±0.630		0.197±0.002
2.	AAWEO	R1	0.303 0.302 0.302	37.60	0.289 0.288 0.287	40.49	0.208 0.207 0.208	57.23	0.107 0.107 0.108	77.89	0.077 0.076 0.077	84.29	45.471
		Avg	0.302±0.0005		0.288±0.001		0.207±0.0005		0.107±0.005		0.076±0.005		
		R2	0.302 0.301 0.302		37.80		0.287 0.287 0.288		40.70		0.209 0.208 0.207		
		Avg	0.301±0.0005	0.287±0.0005		0.208±0.001	0.107±0.005	0.075±0.001					
		R3	0.303 0.302 0.302	37.60		0.287 0.286 0.287	40.90	0.209 0.207 0.209		57.02	0.108 0.109 0.108	77.68	
Avg	0.302±0.0005	37.33±0.642	0.286±0.0005		40.69±0.205	0.208±0.001		57.09±0.121	0.108±0.005		77.82±0.121		0.078±0.005
3.	BHT	R1	0.299 0.297 0.302	38.22	0.232 0.230 0.235	52.06	0.211 0.209 0.214	56.40	0.186 0.184 0.189	61.5	0.168 0.166 0.171	65.28	45.23
		Avg	0.299±0.0025		0.232±0.0025		0.211±0.0025		0.186±0.005		0.168±0.005		
		R2	0.292 0.290 0.295		39.66		0.225 0.223 0.227		53.51		0.215 0.213 0.217		
		Avg	0.292±0.0025	0.225±0.0025		0.215±0.0025	0.181±0.005	0.164±0.005					
		R3	0.292 0.290 0.295	39.66		0.230 0.228 0.233	52.47	0.209 0.207 0.212		56.81	0.186 0.184 0.189	61.5	
Avg	0.292±0.0025	34.38±0.831	0.230±0.0025		49.51±0.715	0.209±0.747		57.19±0.631	0.186±0.0025		66.40±0.635		0.168±0.005

R= Replica, O.D= optical density, %I= percent inhibition AAREO= *Artemisia annua* rainy essential oil, AAWEO= *Artemisia annua* winter essential oil

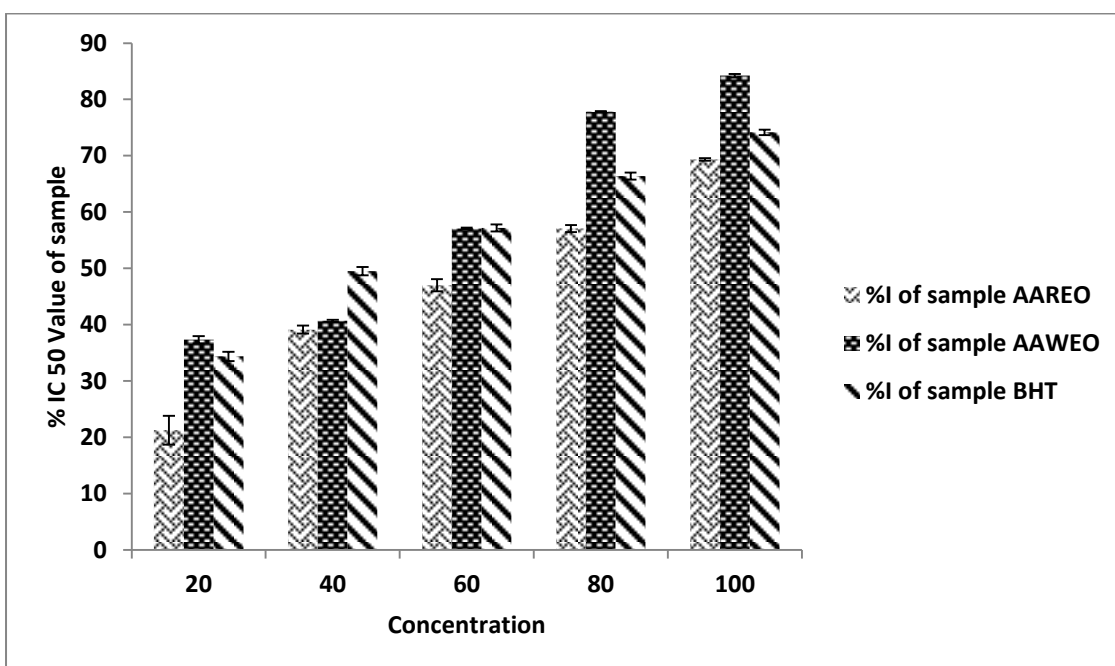


Fig. 4.10 DPPH radical scavenging activity of AAREO and AAWE0

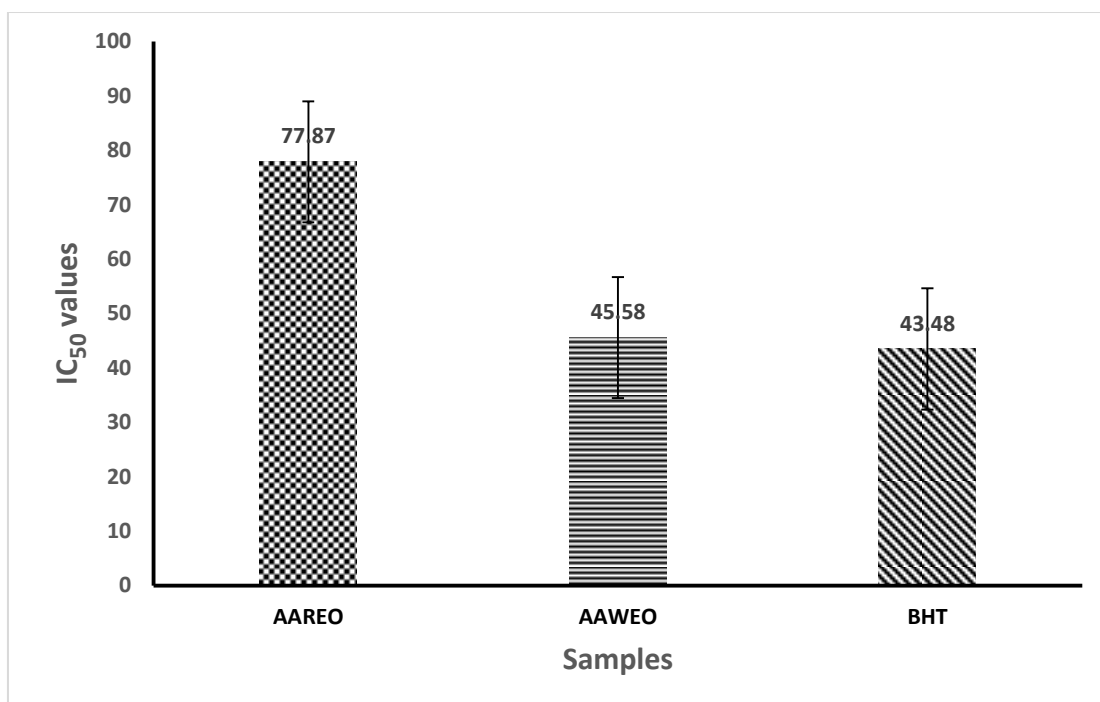


Fig. 4.11 IC₅₀ Values of AAREO & AAWE0

4.4.1.2 Metal chelating activity

Chelation of transition metals has been reported to be one of the important mechanisms of antioxidative action. Transition metal ions can promote lipid peroxidation through generation of initiating species, and accelerating peroxidation by decomposing lipid hydroperoxides into other components which are capable of abstracting hydrogen (Miguel, 2010).

The transition metal ion, Fe^{2+} exhibits the ability to move single electrons by virtue of which it can allow the formation and propagation of many radical reactions even starting with relatively non-reactive radicals. The main strategy to avoid generation of reactive oxygen species is associated with redox active metal catalysis involves chelating of the metal ions. Ferrozine can quantitatively chelate with Fe^{2+} and form a complex with a red color. This reaction is limited in the presence of other chelating agents and results in a decrease of the red color of the ferrozine- Fe^{2+} complexes. Measurement of the color reduction estimates the chelating activity to compete with ferrozine for the ferrous ions (Ćavar *et al.*, 2012).

The effect on metal chelating activity of Fe^{2+} by AAREO and AAWE0 was evaluated in presence of FeCl_2 and ferrozine in dose dependent manner. The metal chelating activity of AAREO and AAWE0 was observed in the following order of AAWE0 ($\text{IC}_{50}=49.50\pm 0.234$)>AAREO ($\text{IC}_{50} =51.40\pm 0.158$) compared to EDTA ($\text{IC}_{50}=41.68\pm 0.396$) which indicated that AAWE0 possess maximum metal chelating activity with minimum IC_{50} value followed by AAREO. The detailed data of percent metal chelating activity and IC_{50} values of oils has been depicted in **table 4.8 and fig 4.12** while the IC_{50} values of AAREO and AAWE0 are being presented in **Fig 4.13**

Table 4.8 Metal chelating activity of AAREO and AAWEO

S.N	Sample	R	20µl/mL		40 µl/mL		60 µl/mL		80 µl/mL		100 µl/mL		IC50	
			O.D	%I	O.D	%I	O.D	%I	O.D	%I	O.D	%I		
1.	AAREO	R1	1.285 1.283 1.288	15.01	0.826 0.824 0.829	45.37	0.586 0.584 0.589	61.24	0.324 0.322 0.327	78.57	0.184 0.187 0.182	87.83	51.49	
		Avg	1.285±0.0025		0.826±0.0025		0.586±0.0025		0.324±0.0025		0.184±0.0025			
		R2	1.279 1.282 1.277		15.41		0.819 0.817 0.822		45.83		0.586 0.584 0.589			61.24
		Avg	1.279±0.0025		0.819±0.0025		0.586±0.0025		0.321±0.0025		0.186±0.0025			
		R3	1.282 1.28 1.285	15.34	0.826 0.824 0.829	45.37	0.589 0.592 0.587	61.04	0.324 0.322 0.327	78.57	0.192 0.19 0.195	87.30	51.5	
		Avg	1.282±0.0025	16.44±0.213	0.826±0.0025	44.14±0.265	0.589±0.0025	60.88±0.115	0.324±0.0025	78.97±0.109	0.192±0.0025	90.70±0.274	51.40±0.158	
2.	AAWEO	R1	1.342 1.345 1.34	11.24	0.624 0.622 0.627	58.73	0.548 0.551 0.546	63.75	0.418 0.416 0.421	72.35	0.176 0.174 0.179	88.35	49.41	
		Avg	1.342±0.0025		0.624±0.0025		0.548±0.0025		0.418±0.0025		0.176±0.0025			
		R2	1.348 1.346 1.351		10.84		0.626 0.624 0.629		58.59		0.552 0.55 0.555			63.49
		Avg	1.348±0.0025		0.626±0.0025		0.552±0.0025		0.422±0.0025		0.171±0.0025			
		R3	1.342 1.345 1.34	11.24	0.621 0.619 0.624	58.92	0.548 0.551 0.546	63.75	0.416 0.414 0.419	72.48	0.176 0.174 0.179	88.35	49.33	
		Avg	1.342±0.0025	11.10±0.23	0.621±0.0025	58.74±0.165	0.548±0.0025	63.66±0.150	0.416±0.0025	72.30±0.204	0.176±0.0025	88.46±0.196	49.50±0.234	
3.	Na ₂ EDTA	R1	1.306 1.305 1.306	13.69	0.466 0.465 0.464	69.24	0.391 0.392 0.390	74.14	0.301 0.301 0.302	80.09	0.273 0.274 0.272	81.94	41.23	
		Avg	1.305±0.0005		0.465±0.001		0.391±0.001		0.301±0.0005		0.273±0.001			
		R2	1.316 1.314 1.316		13.02		0.478 0.477 0.478		68.45		0.394 0.393 0.393			74.00
		Avg	1.315±0.001		0.477±0.0005		0.393±0.0005		0.301±0.001		0.275±0.0005			
		R3	1.312 1.311 1.313	13.22	0.479 0.478 0.479	68.38	0.396 0.395 0.395	73.87	0.302 0.301 0.302	80.09	0.275 0.276 0.274	81.81	41.89	
		Avg	1.312±0.001	13.31±0.343	0.478±0.005	68.69±0.477	0.395±0.0005	74.00±0.135	0.301±0.0005	80.09±0	0.275±0.001	81.85±0.075	41.68±0.396	

R= Replica, O.D= optical density, %I= percent inhibition AAREO= *Artemisia annua* rainy essential oil, AAWEO= *Artemisia annua* winter essential oil

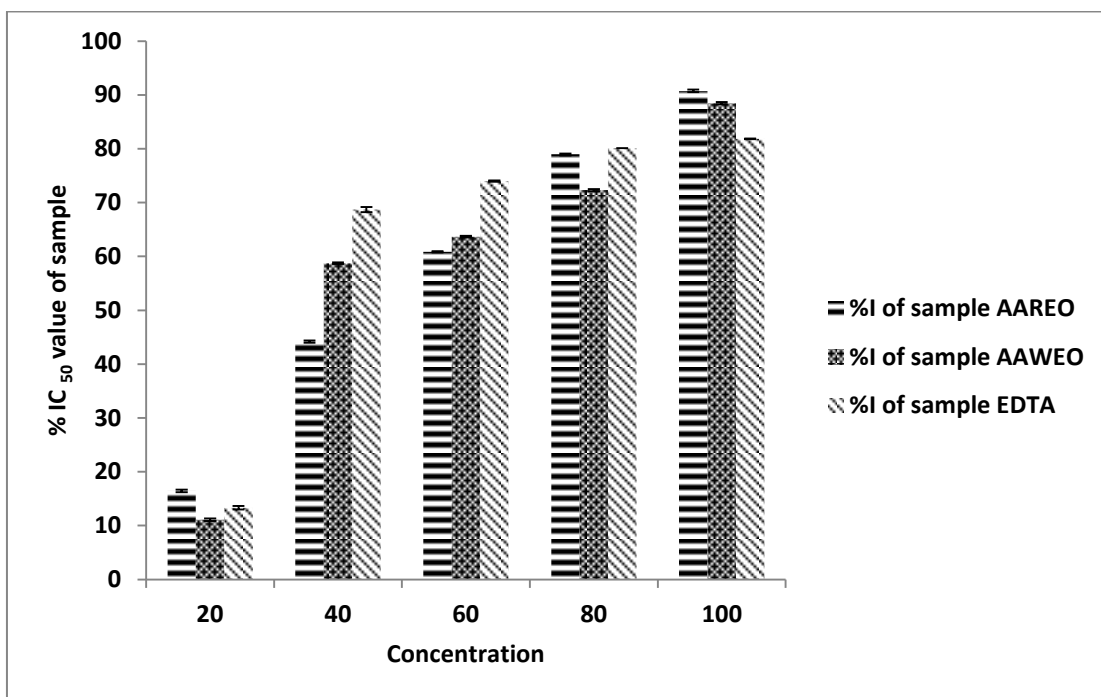


Fig.4.12 Metal chelating activity of AAREO and AAWE0

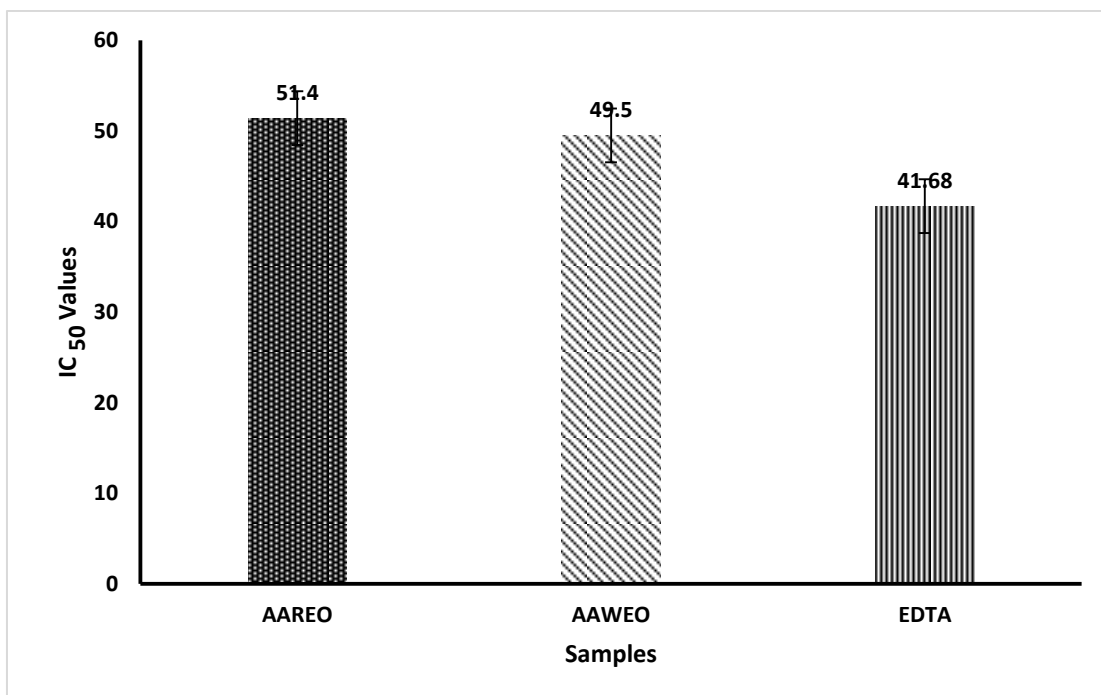
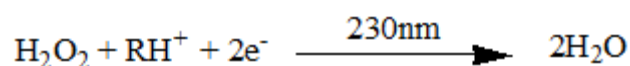


Fig. 4.13 IC₅₀ Values of AAREO and AAWE0

4.4.1.3 H₂O₂ radical scavenging activity

Free radical generation has a direct relation with oxidation in foods and biological systems. Biological systems may produce hydrogen peroxide (H₂O₂). The generation of hydrogen peroxide (H₂O₂) through the activation of phagocytes is known to play an important part in the killing of various bacterial and fungal strains (Sánchez-Moreno, 2002). Hydrogen peroxide (H₂O₂) belongs to the reactive oxygen species (ROS), known as oxidants that can react with various cellular targets thereby causing cell damage or even cell death. Therefore it is necessary to remove H₂O₂ radicals from the body as it can attack cellular energy producing systems. The antioxidant compound has the ability to scavenge H₂O₂ through the donation of e⁻ and neutralizes it to a water molecule.



The ability of oil to scavenge hydrogen peroxide was determined spectrophotometrically and compared with that of BHT as a standard. The H₂O₂ free radical scavenging activity of AAREO, AAWE0 and standard was observed in dose dependent manner (20 µg/mL, 40 µg/mL, 60 µg/mL, 80 µg/mL and 100 µg/mL). The order of activity was observed as AAWE0 (IC₅₀ = 34.79 ± 0.419) > AAREO (IC₅₀ = 44.83 ± 1.304) compared to BHT (IC₅₀ = 27.96 ± 1.195). The detailed % H₂O₂ radical scavenging activity with IC₅₀ values at 230nm of AAREO and AAWE0 are being presented in **Table 4.9** and **Fig 4.14** and the IC₅₀ values of AAREO and AAWE0 are being presented in **Fig 4.15**.

Table 4.9 H₂O₂ radical scavenging activity of AAREO and AAWEO

S.N	Sample	R	20 µl/mL		40 µl/mL		60 µl/mL		80 µl/mL		100 µl/mL		IC ₅₀
			O.D	%I	O.D	%I	O.D	%I	O.D	%I	O.D	%I	
1.	AAREO	R1	0.429	43.10	0.398	47.21	0.328	47.21	0.262	65.25	0.184	75.59	46.33
			0.427		0.396		0.326		0.265		0.182		
			0.432		0.401		0.331		0.26		0.187		
		Avg	0.429±0.0025		0.398±0.005		0.398±0.0025		0.262±0.025		0.184±0.0025		
		R2	0.448	40.15	0.406	46.15	0.316	58.09	0.261	65.38	0.179	76.25	44.23
			0.446		0.404		0.314		0.264		0.177		
0.451	0.409		0.319		0.259		0.182						
Avg	0.448±0.0025		0.406±0.0025		0.316±0.0025		0.261±0.0025		0.179±0.0025				
R3	0.452	40.05	0.406	46.15	0.310	58.88	0.262	65.25	0.188	75.06	43.94		
	0.450		0.404		0.308		0.265		0.186				
	0.455		0.409		0.313		0.26		0.191				
Avg	0.452±0.0025	41.1±1.732	0.406±0.0025	46.50±0.611	0.310±0.0025	54.72±6.521	0.262±0.0025	65.29±0.075	0.188±0.0025	75.63±0.596	44.83±1.304		
2.	AAWEO	R1	0.443	41.24	0.324	57.02	0.310	58.88	0.298	60.47	0.252	66.57	34.72
			0.441		0.322		0.308		0.296		0.255		
			0.447		0.327		0.313		0.301		0.250		
		Avg	0.443±0.003		0.324±0.0025		0.310±0.0025		0.298±0.0025		0.252±0.003		
		R2	0.444	41.11	0.323	57.16	0.309	59.01	0.292	61.27	0.253	66.31	34.41
			0.442		0.321		0.307		0.290		0.251		
0.447	0.327		0.312		0.295		0.255						
Avg	0.444±0.0025		0.324±0.0025		0.309±0.0025		0.292±0.0025		0.253±0.002				
R3	0.445	40.98	0.325	56.89	0.309	59.01	0.298	60.44	0.251	66.71	35.24		
	0.448		0.323		0.307		0.296		0.249				
	0.443		0.328		0.312		0.301		0.254				
Avg	0.443±0.0025	41.11±0.130	0.324±0.0025	57.02±0.135	0.309±0.0025	58.96±0.075	0.298±0.0020	60.72±0.470	0.251±0.0025	66.53±0.202	34.79±0.419		
3.	BHT	R1	0.438	41.90	0.315	58.22	0.272	63.92	0.218	71.08	0.196	74.00	29.31
			0.436		0.318		0.27		0.216		0.194		
			0.441		0.312		0.275		0.221		0.199		
		Avg	0.438±0.0025		0.315±0.0025		0.272±0.0025		0.218±0.0025		0.196±0.0025		
		R2	0.432	42.70	0.308	59.12	0.27	64.19	0.212	71.88	0.198	73.74	27.02
			0.436		0.306		0.268		0.21		0.196		
0.429	0.311		0.273		0.215		0.201						
Avg	0.432±0.0035		0.308±0.0025		0.270±0.0025		0.212±0.0025		0.198±0.0025				
R3	0.438	41.90	0.315	58.22	0.272	63.92	0.218	71.08	0.196	74.00	29.31		
	0.436		0.318		0.27		0.216		0.194				
	0.441		0.312		0.275		0.221		0.199				
Avg	0.438±0.0025	42.43±0.461	0.315±0.0025	58.75±0.478	0.272±0.0025	63.92±0.265	0.218±0.0025	71.61±0.461	0.196±0.0025	73.82±0.150	27.96±1.195		

R= Replica, O.D= optical density, %I= percent inhibition, AAREO= *Artemisia annua* rainy essential oil, AAWEO= *Artemisia annua* winter essential oil

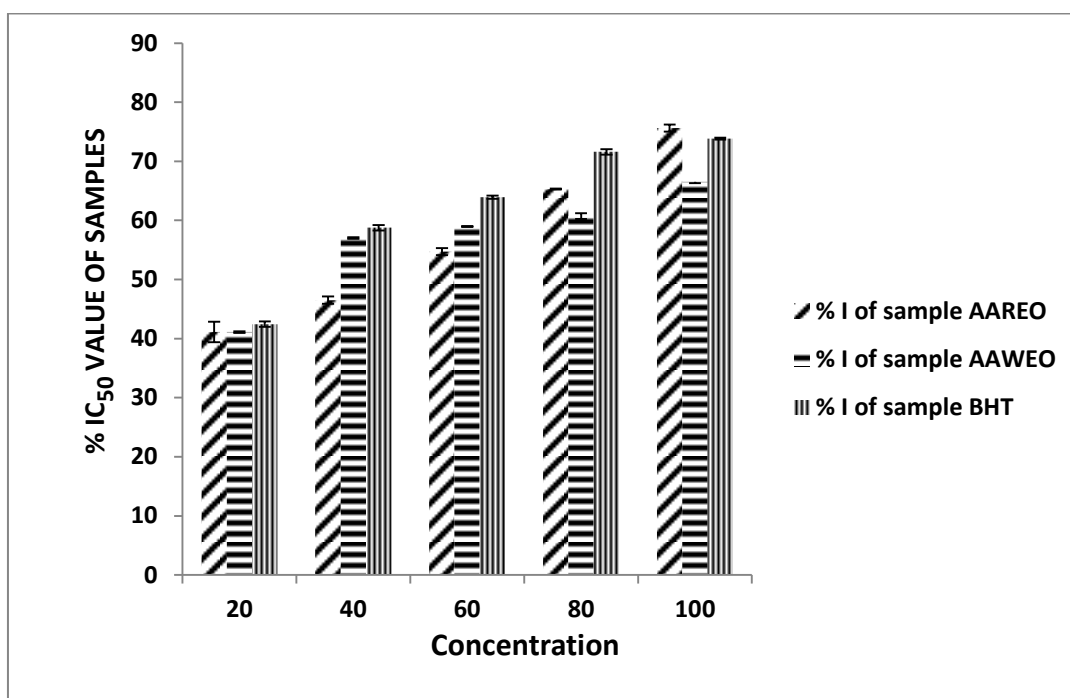


Fig.4.14 H_2O_2 radical scavenging activity of AAREO and AAWE0

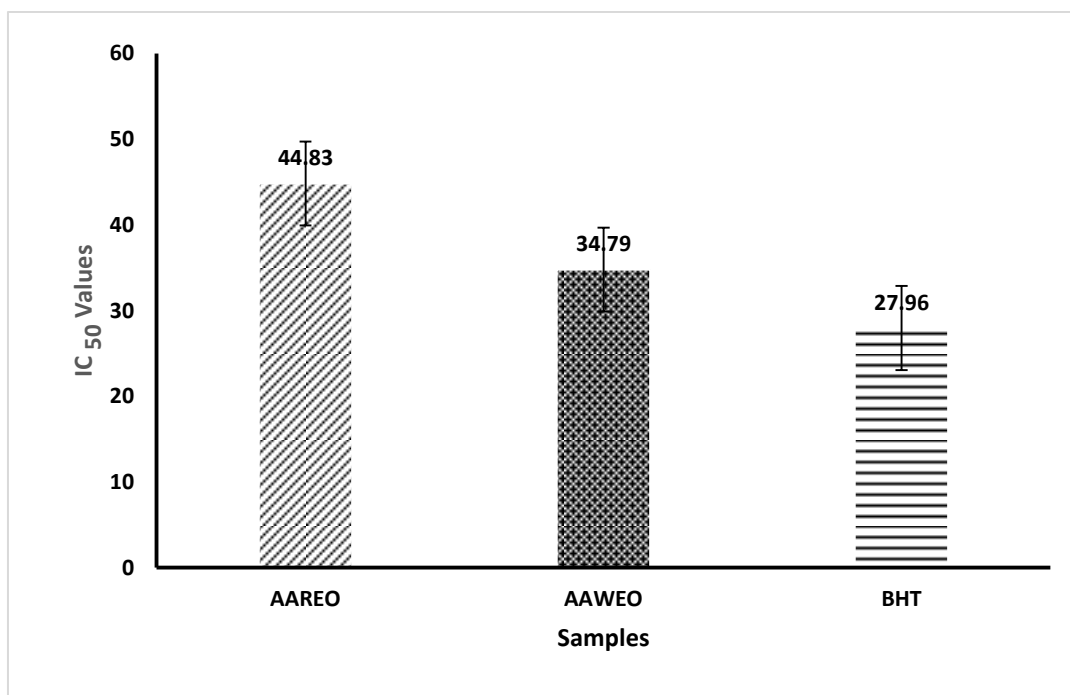


Fig. 4.15 IC₅₀ Values of AAREO and AAWE0

Cavar et al., (2012) reported the antioxidant activity of the *A. annua* essential oil by five testing methods viz. DPPH, ABTS, reducing power, ORAC, and metal-chelating assays. The oil sample was capable of reducing the stable violet DPPH radical to the yellow DPPH-H, reaching 50% of reduction also the essential oil of *A. annua* interrupted with the process of formation of ferrous and ferrozine complex, concluding that it has chelating activity and captures ferrous ion before ferrozine.

Wang et al., (2008) have analysed the DPPH scavenging activity of 1,8-cineole, α -pinene, β -pinene, camphene. **Da Silva et al., (2014)** have also reported the anti-oxidant activity evaluated by means of DPPH free radical scavenging activity because of the presence of β -selinene, α -selinene, α -humulene, germacrene-D, spathulenol, α -pinene, α -copaene, α -terpineol, *neo*-intermedol.

Keskes et al., (2014) reported that due incorporation of terpenes there is high antioxidant activity in alloxan-induced diabetic rats. α -terpinene and p-cymene have been reported to possess antioxidant activity (**Masotti et al., 2003**). It has been reported that artemisia ketone exhibited higher anti oxidant activity than camphor and 1,8-cineole in DPPH assay (**Zhigzhitzhapova et al., 2020**).

A thorough search of literature revealed that essential oils obtained from plants belonging to the *Artemisia* genus are markedly rich in non-phenolic constituents therefore *A. annua* essential oil is found to be a relatively weak anti oxidant (**Juteau et al., 2002; Lopes-Lutz et al., 2008**).

However in the present study of AAREO & AAWEO revealed the notable antioxidant activity which might be due to presence of 1,8-cineole, α -pinene, β -pinene, camphene, artemisia ketone, germacrene-D, spathulenol etc as the previous reports support the scientific rational antioxidant activity of *A.annua* essential oil or due to synergic effects of other major, minor or trace constituents of the essential oils.

4.4.1 In-vitro anti-inflammatory activity

Inflammation is defined as a complex biological response of vascular tissues against following agents viz pathogens, irritants, or damaged cells. It can be classified as either acute or chronic, and involves a surge of biochemical processes consisting of

the local vascular system, the immune system, and different cell types found in the injured tissue (Andrade and De Sousa, 2013). In an inflammatory infection, macrophages and mast cells release inflammatory mediators, such as cytokines (e.g., interleukin-1 β (IL-1 β), IL-6, IL-12, and the chemokine IL-8), tumor necrosis factors (e.g., TNF- α and TNF- β) or interferons (e.g., IFN- γ) (Andrade *et al.*, 2014). It has been inferred from the reported that medicinal plants and their isolated compounds are employed worldwide in folk herbal medicine for the treatment of different inflammatory conditions, such as lung and skin inflammations. In the continuous search for natural products against inflammation, essential oils are increasingly being regarded to as a rich source of such products. This activity was based on the inhibition percent of denaturation of the albumin proteins, as a measure of *in vitro* anti-inflammatory activity. In presence of saline salt phosphate buffer solution (at physiological pH), the denaturation reaction of albumin protein (Hen's egg) takes place. The inhibition of denaturation of albumin protein is because of the presence of compounds having anti-inflammatory property, which is measured spectrophotometrically by change in absorbance at 660nm (Kar *et al.*, 2012).

In present study, the *in vitro* anti-inflammatory activity of AAREO and AAWE0 was investigated. Both the essential oils exhibited dose-dependent anti-inflammatory activity. The order in which AAREO and AAWE0 possessed anti-inflammatory activity was observed as AAWE0 (IC₅₀=66.90 \pm 0.040) > AAREO (IC₅₀=76.55 \pm 0.218) compared to the standard anti-inflammatory drug diclofenac sodium tablets (IC₅₀=63.41 \pm 0.037). AAREO was found to exhibit maximum activity with minimum IC₅₀ value followed by AAWE0 with higher IC₅₀ value. The detailed presentation of *in vitro* anti-inflammatory activity observed spectrophotometrically in terms of OD (optical density), percent inhibition and IC₅₀ values of AAREO and AAWE0 are being presented in **Table 4.10** and **Fig.4.16** while the IC₅₀ values of AAREO and AAWE0 are being shown in **Fig 4.17**.

Camphor has been reported to possess a potent anti-inflammatory activity (Sharma, 2021).

Essler *et al.*, (2013) reported that eucalyptol (1,8-cineole), camphor, β -caryophyllene exhibit an effective *in vitro* and *in vivo* anti inflammatory activity.

Table 4.10 *In vitro* anti-inflammatory activity of AAREO and AAWEO

S.N	Sample	R	20 µl/mL		40 µl/mL		60 µl/mL		80 µl/mL		100 µl/mL		IC50
			O.D	%I	O.D	%I	O.D	%I	O.D	%I	O.D	%I	
1.	AAREO	R1	1.276	19.03	1.151	26.96	0.934	40.73	0.892	43.40	0.464	70.55	76.51
			1.274		1.149		0.932		0.889		0.462		
			1.279		1.154		0.937		0.895		0.46		
		Avg	1.276±0.0025		1.151±0.0025		0.934±0.0025		0.892±0.003		0.464±0.0025		
		R2	1.278	18.9	1.146	27.28	0.942	40.22	0.896	43.14	0.465	70.49	76.79
			1.281		1.144		0.94		0.899		0.463		
			1.276		1.149		0.945		0.894		0.468		
		Avg	1.278±0.0025		1.146±0.0025		0.942±0.0025		0.896±0.0025		0.465±0.0025		
		R3	1.276	19.03	1.142	27.53	0.934	40.73	0.891	43.46	0.464	70.55	76.36
			1.274		1.14		0.932		0.889		0.462		
			1.279		1.145		0.937		0.894		0.467		
		Avg	1.276±0.0025	18.98±0.0750	1.142±0.0025	27.25±0.285	0.934±0.0025	40.56±0.294	0.891±0.0025	43.33±0.170	0.464±0.0025	70.53±0.034	76.55±0.218
2.	AAWEO	R1	1.307	17.06	1.198	23.98	0.885	43.84	0.594	62.3	0.382	75.76	66.95
			1.305		1.196		0.888		0.592		0.38		
			1.31		1.201		0.882		0.597		0.385		
		Avg	1.307±0.0025		1.198±0.0025		0.885±0.03		0.594±0.0025		0.382±0.0025		
		R2	1.306	17.13	1.195	24.17	0.883	43.97	0.585	62.88	0.391	75.19	66.88
			1.304		1.193		0.881		0.583		0.389		
			1.309		1.198		0.886		0.588		0.394		
		Avg	1.306±0.0025		1.195±0.0025		0.883±0.0025		0.585±0.0025		0.391±0.0025		
		R3	1.3	17.51	1.198	23.98	0.885	43.84	0.591	62.5	0.385	75.57	66.88
			1.298		1.196		0.888		0.589		0.383		
			1.303		1.201		0.882		0.594		0.388		
		Avg	1.300±0.0025	17.23±0.242	1.198±0.0025	24.04±0.109	0.885±0.03	43.88±0.075	0.591±0.0025	62.56±0.294	0.385±0.0025	75.50±0.290	66.90±0.040
3.	Diclofenac	R1	1.423	9.70	1.085	31.15	0.767	51.33	0.556	64.72	0.338	78.55	63.39
			1.421		1.083		0.765		0.554		0.336		
			1.425		1.088		0.77		0.559		0.341		
		Avg	1.423±0.0025		1.085±0.0025		0.767±0.0025		0.556±0.0025		0.338±0.0025		
		R2	1.422	9.77	1.090	30.83	0.761	51.71	0.554	64.84	0.342	78.29	63.40
			1.420		1.093		0.759		0.552		0.345		
			1.425		1.088		0.764		0.557		0.34		
		Avg	1.422±0.0025		1.090±0.0025		0.767±0.0025		0.554±0.0025		0.342±0.0025		
		R3	1.421	9.83	1.081	31.40	0.767	51.33	0.554	64.84	0.348	77.91	63.46
			1.419		1.084		0.765		0.552		0.346		
			1.424		1.079		0.77		0.557		0.351		
		Avg	1.421±0.0025	9.76±0.065	1.081±0.0025	31.12±0.285	0.767±0.0025	51.45±0.219	0.554±0.0025	64.80±0.069	0.348±0.0025	78.25±0.321	63.41±0.037

R= Replica, O.D= optical density, %I= percent inhibition AAREO= *Artemisia annua* rainy essential oil, AAWEO= *Artemisia annua* winter essential oil

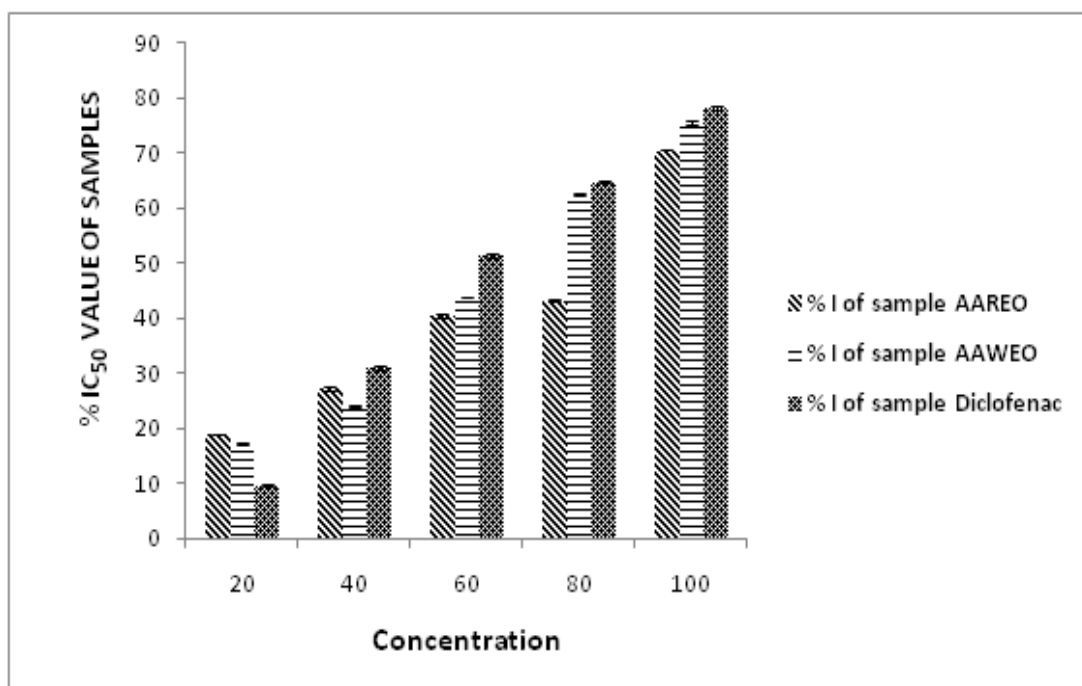


Fig.4.16 *In- vitro* anti-inflammatory activity of AAREO and AAWEO

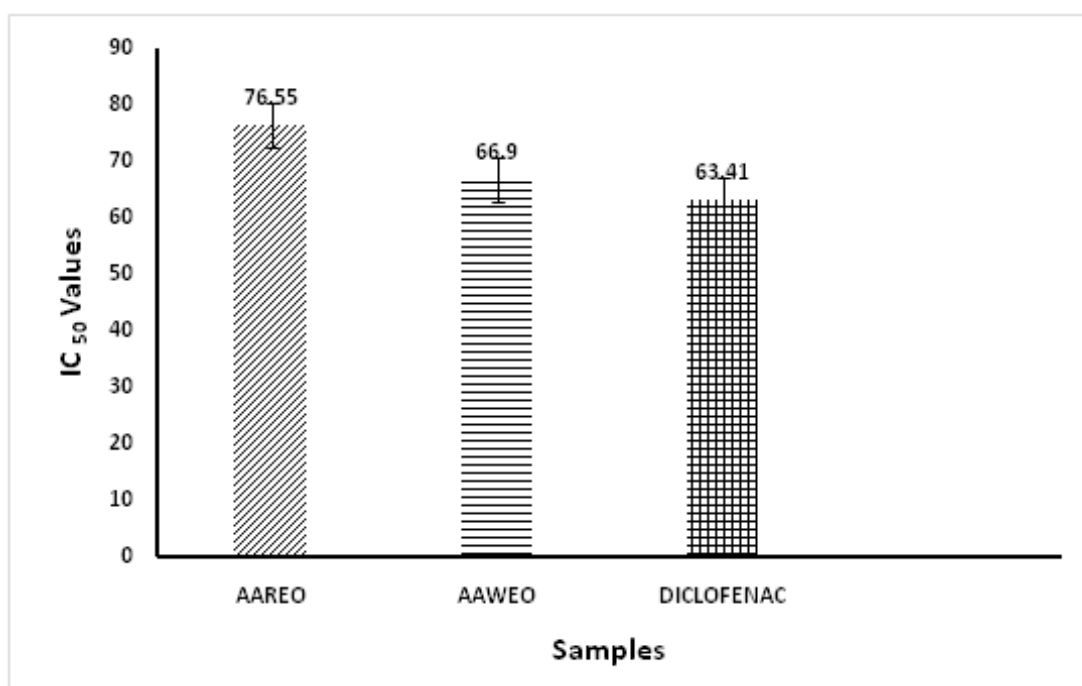


Fig. 4.17 IC₅₀ Values of AAREO and AAWEO

The major components of *Artemisia annua* like artemisinin and scopoletin, have been reported to possess potent anti-inflammatory effects (**Verma *et al.*, 2020**). The aqueous methanolic extract of *Artemisia annua* has been reported to possess anti-inflammatory activity when studied using carrageenan and egg albumin induced rat paw edema in acute, and cotton pellets and grass pith induced chronic inflammation models (**Das, 2012**).

The terpenoids *viz.*, 1,8-cineole, spathulenol, α -cadinol, γ -cadinene, α -pinene, terpinen-4-ol, borneol, β -pinene, sabinene, camphene etc have been reported to exhibit anti-inflammatory activity (**Martins *et al.*, 2008; E Sa *et al.*, 2013; Ludwiczuk *et al.*, 2017**).

Compounds like borneol α - β -pinenes, have been previously reported to possess anti-inflammatory properties (**Asanova *et al.*, 2003**). **Rungqu *et al.*, (2016)** reported that Artemisia ketone and δ -cadinene could be associated with the anti-inflammatory effect and to reduce pain and inflammation.

β -caryophyllene, a bicyclic sesquiterpene has been reported to exhibit anti-inflammatory action through the inhibition of the main inflammatory mediators, like inducible nitric oxide synthase (iNOS), Interleukin 1 beta (IL-1 β), Interleukin-6 (IL-6), tumor necrosis factor- α (TNF- α), nuclear factor kappa B (NF- κ B), cyclooxygenase 1 (COX-1), cyclooxygenase 2 (COX-2) (**Francomano *et al.*, 2019**).

Most of the compounds reported above like camphor, 1,8 cineole, camphor, Artemisia ketone, α - cadinol, γ -cadinene, β -caryophyllene, caryophyllene oxide, α -pinene, terpinen-4-ol, borneol, β -pinene and sabinene etc have been identified both in AAREO & AAWEEO with different quantity hence the results are in support that the *in-vitro* anti-inflammatory activity of AAREO & AAWEEO might be possibly due to these constituents or synergistic effects of other major/minor constituents present in the essential oils. Based on the observed data it can be inferred that the herb, *Artemisia annua* can be a good source for the herbal drug development program against inflammation after proper clinical trials.

4.4.2 *In vitro* Anti diabetic activity

Diabetes mellitus or type-2 diabetes, is known to be one of the most common endocrine metabolic disorders has caused significant morbidity and mortality due to

microvascular (retinopathy, neuropathy, and nephropathy) and macrovascular (heart attack, stroke and peripheral vascular disease) complications. Human bodies undergo enzymatic and non-enzymatic antioxidative mechanisms which reduce the formation of reactive oxygen species, responsible for many degenerative diseases including diabetes (Patel *et al.*, 2011). This disease is rapidly increasing worldwide and affecting all parts of the world. Deficiency of insulin leads to an increase in blood sugar levels (Ponnusamy *et al.*, 2011). Some medicinal plants have been reported to be useful in diabetes worldwide and have been used empirically in antidiabetic and anti-hyperlipidemic remedies. It has been reported that plants showing hypoglycemic potential mainly belong to the families Asteraceae, Leguminosae, Lamiaceae, Liliaceae, Cucurbitaceae, Moraceae, Rosaceae and Araliaceae. Antihyperglycemic activity of the plants is basically due to their capability to restore the function of pancreatic tissues by causing an increase in insulin output or inhibit the intestinal absorption of glucose or to the facilitation of metabolites in insulin-dependent processes (Malviya *et al.*, 2010). The cause of diabetes might be because of the increase in oxidative stress which results in irregular production of free radicals. This irregular formation of free radicals and sudden degradation in the mechanism of antioxidant defence in the body leads to the destruction of enzymes and different cell organelles, increase in lipid peroxidation and irregular production of insulin (Maritim *et al.*, 2003) and thus; there is a need of herbal and medicinal anti-diabetic agent to inhibit the action of oxidative stress.

Thus; there is a need of herbal and medicinal anti-diabetic agent to inhibit the action of oxidative stress. In the present study, the *in vitro* anti-diabetic activity of AAREO and AAWEO has been investigated. The observation shows that both the oils possess anti-diabetic activity in a dose-dependent manner. As the concentration of essential oils increases, %inhibition also increases with an increase in absorbance. The order of anti-diabetic activity of AAREO and AAWEO has been observed in the order of AAWEO (47.45 ± 0.195) > AAREO (74.26 ± 3.706) compared to the standard acarbose (38.21 ± 0.03) while the detailed results of anti-diabetic activity observed spectrophotometrically in terms of OD (optical density), percent inhibition and IC_{50} values of AAREO and AAWEO are being presented **Table 4.11** and **Fig 4.18 and Fig. 4.19**.

Table 4.11 *In vitro* anti-diabetic activity of AAREO and AAWEO

S.N.	Sample	R	20 µl/mL		40 µl/mL		60 µl/mL		80 µl/mL		100 µl/mL		IC50	
			O.D	%I	O.D	%I	O.D	%I	O.D	%I	O.D	%I		
1.	AAREO	R1	1.198	17.54	0.942	35.16	0.896	38.33	0.718	50.58	0.418	71.23	78.54	
			1.196		0.94		0.894		0.716		0.416			
			1.201		0.945		0.899		0.721		0.421			
		Avg	1.198±0.0025	0.942±0.0025	0.896±0.0025	0.718±0.0025	0.418±0.0025							
		R2	1.192	17.96	0.939	35.37	0.891	38.67	0.721	50.37	0.422	70.95		72.12
			1.19		0.937		0.889		0.719		0.425			
1.195	0.942		0.894		0.724		0.42							
Avg	1.192±0.0025	0.939±0.0025	0.891±0.0025	0.721±0.0025	0.422±0.0025									
R3	1.194	17.82	0.942	35.16	0.892	38.60	0.718	50.58	0.422	70.95	72.12			
	1.192		0.94		0.89		0.716		0.425					
	1.197		0.945		0.895		0.721		0.42					
Avg	1.192±0.0025	0.942±0.0025	0.892±0.0025	0.718±0.0025	0.422±0.0025	71.04±0.161	74.26±3.706							
2.	AAWEO	R1	0.917	36.88	0.722	50.3	0.659	52.37	0.542	62.69		0.344	76.32	47.47
			0.915		0.725		0.657		0.54			0.342		
			0.921		0.72		0.662		0.545		0.347			
		Avg	0.917±0.003	0.722±0.0025	0.659±0.0025	0.542±0.0025	0.344±0.0025							
		R2	0.922	36.54	0.726	50.03	0.661	54.5	0.545	62.49	0.345	76.25	47.64	
			0.92		0.724		0.659		0.548		0.343			
0.925	0.729		0.664		0.543		0.348							
Avg	0.922±0.0025	0.726±0.0025	0.661±0.0025	0.545±0.0025	0.345±0.0025									
R3	0.921	36.61	0.722	50.3	0.661	54.5	0.562	61.32	0.353	75.70	47.25			
	0.919		0.725		0.659		0.56		0.351					
	0.924		0.72		0.664		0.565		0.356					
Avg	0.921±0.0025	0.722±0.0025	0.661±0.0025	0.562±0.0025	0.353±0.0025	76.09±0.339	47.45±0.195							
3.	acarbose	R1	0.976	32.82	0.607	58.27	0.512	64.76	0.345	76.25		0.121	91.67	38.42
			0.975		0.607		0.513		0.346			0.123		
			0.977		0.606		0.512		0.344		0.121			
		Avg	0.976±0.001	0.606±0.0005	0.512±0.0005	0.345±0.001	0.121±0.001							
		R2	0.975	32.89	0.606	58.29	0.513	64.69	0.346	76.25	0.122	91.60	38.21	
			0.975		0.607		0.514		0.345		0.123			
0.977	0.607		0.513		0.345		0.122							
Avg	0.975±0.001	0.606±0.0005	0.513±0.0005	0.345±0.0005	0.122±0.0005									
R3	0.976	32.89	0.606	58.29	0.512	64.83	0.346	76.18	0.121	91.67	38.18			
	0.975		0.605		0.511		0.347		0.122					
	0.974		0.607		0.512		0.347		0.122					
Avg	0.975±0.001	0.606±0.001	0.511±0.0005	0.346±0.0005	0.121±0.0005	91.64±0.040	38.21±0.03							

R= Replica, O.D= optical density, %I= percent inhibition AAREO= *Artemisia annua* rainy essential oil, AAWEO= *Artemisia annua* winter essential oil

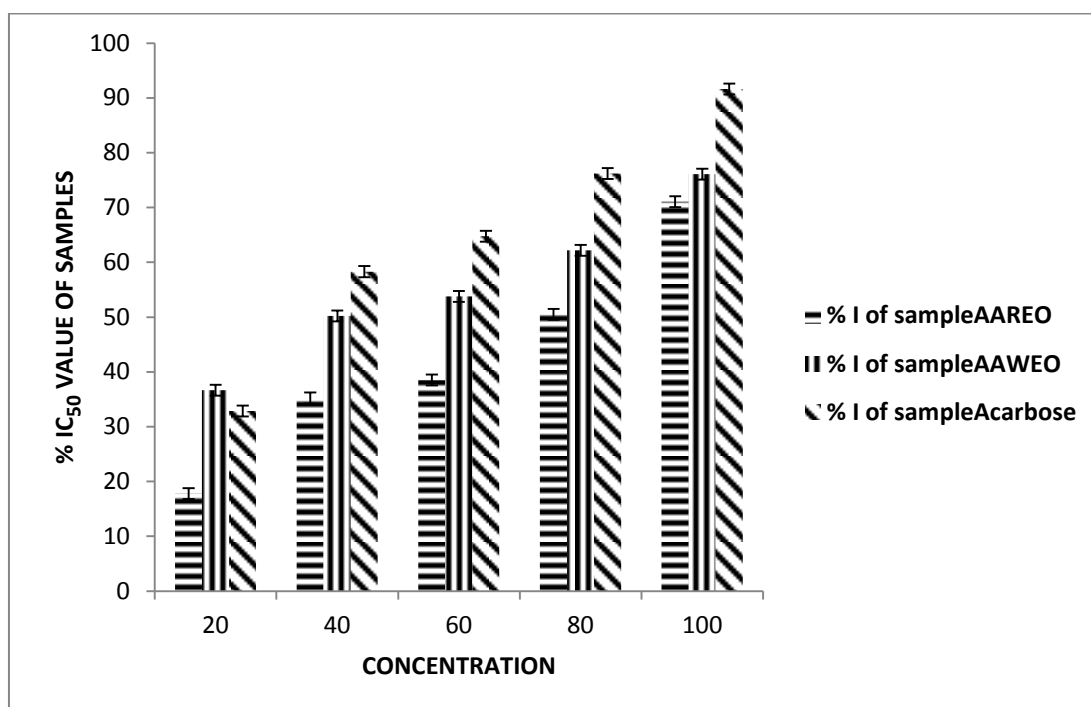


Fig.4.18 *In- vitro* anti-diabetic activity of AAREO and AAWEO

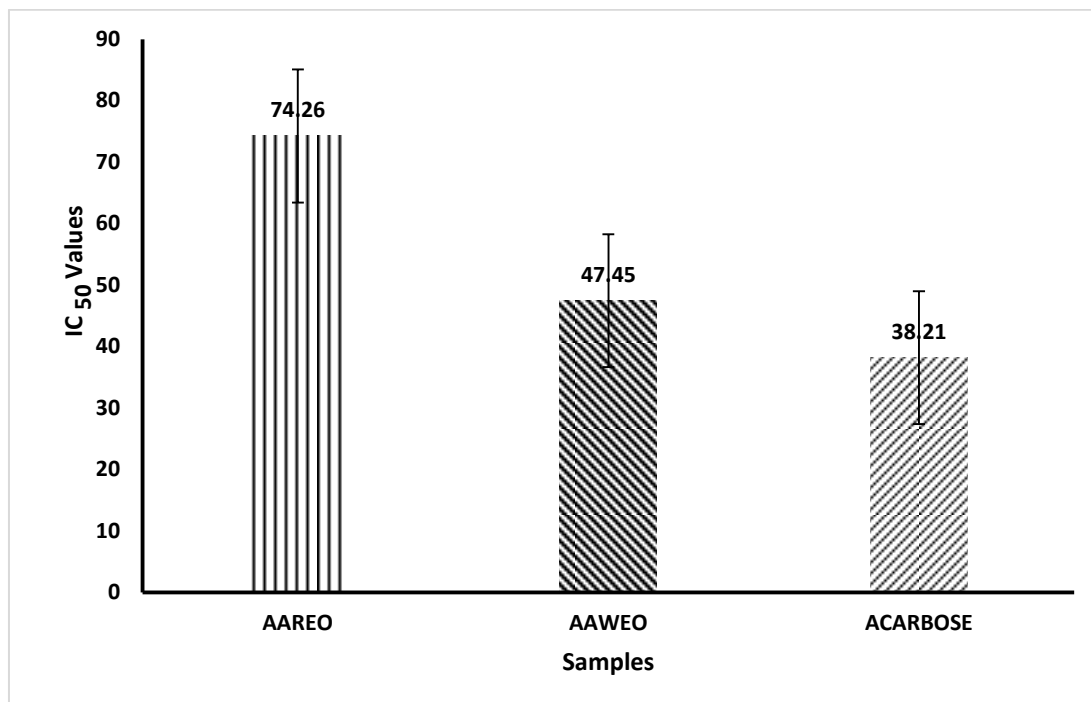


Fig. 4.19 IC₅₀ Values of AAREO and AAWEO

It has been documented in the literature that the diabetic rats when treated with aqueous Extract of *A. annua*, exhibited significant anti-hyperglycemic activity in diabetic animals (Verma *et al.*, 2020).

Camphor has been reported to exhibit a significant antidiabetic activity (Sharma, 2021).

The terpenoids *viz.*, α -pinene, caryophyllene oxide, β -pinene, 1,8-cineole, terpinen-4-ol, α -humulene, β -selinene, β -caryophyllene oxide, borneol and γ -cadinene have been reported to possess anti-diabetic activity (Tahir *et al.*, 2016; Bouyahya *et al.*, 2020).

Literature search revealed that *Artemisia annua* extracts play a crucial role in inhibiting hyperglycemia and improving metabolic abnormalities induced by diabetes through its antioxidant advantage. It has also been reported in the studies that essential oils components such as (linalool, carvacrol, eugenol,, sabinene hydrate, α -terpineol, camphor, germacrene-D, artemisia ketone and 1,8-cineole) have a modulatory effect on the values of HOMA_IR, which may be attributed to enhanced peripheral uptake of glucose and clarify its hypoglycemic effect in addition to the increase in serum insulin level (Helal *et al.*, 2014).

It has been reported that germacrene-D and eucalyptol (1,8 cineole) had a higher α -glucosidase inhibition activity suggesting that these compounds are responsible for a significant anti-diabetic effect in EO of *Artemisia gmelinii* weber ex Stechm (Xu *et al.*, 2021).

Keskes *et al.*, (2014) reported that the terpene such as α -pinene which exist in *Juniperus phoenicea* L. essential oil might inhibited key enzymes related to Type 2 diabetes principally α -amylase. It was reported that due administration of terpenes to diabetic exhibits blood glucose lowering effect in alloxan-induced diabetic rats.

On the basis of the above mentioned literature, it can be concluded that the anti-diabetic activity of AAREO and AAWEO might be possibly due to the presence of eucalyptol (1,8-cineole), germacrene-D, camphor, Artemisia ketone, terpinen-4-ol, isoborneol α -humulene, β -selinene, β -caryophyllene oxide, etc or synergetic effects of other major, minor or trace constituents of essential oils.

4.4.3 Anti-feedant activity

Human beings suffer extensively due to havoc created by insect populations in agriculture as well as in health. In agriculture, an insect directly affect the crop and causes severe damage to the growing crop which results in revenue loss. Crop loss due to insect pests has been estimated between 10-30% percent for major crops (Arivoli and Tennyson, 2013).

To control the increasing population of pests in the crop fields, due to which the use of synthetic insecticides is drastically increasing to protect the crops from getting damaged by pests (Montenergo *et al.*, 2018). Excessive utilization of synthetic insecticides has resulted in to various problems such as the resurgence of primary pests, secondary pest outbreaks, resistance development, insecticide residue, health hazards, environmental contamination, and increased costs of insect control (Roy and Mukhopadhyay, 2010).

Plants' bioactive molecules are being used now a days for controlling pests. As plants produce phytochemicals in the environment, including secondary metabolites that are used by the plant in defense against phytophagous insects. These secondary metabolites are basically alkaloids, polyphenols, terpenoids, polyacetylenes, flavonoids, unusual amino acids, sugars, phenylpropanoids, quinines, essential oils, etc., which usually have a wide variety of insecticidal properties, including, repellent, antifeedant, and insect growth inhibitory activities (Ahmad, 2007; Dhaliwal and Koul 2011).

Amongst the different group of agricultural pest, *Spodoptera litura* (Fab). (Lepidoptera: Noctuidae) is known as the one of the major polyphagous pest attacking economically important crops thereby causing major damage. Various economically important crops such as cotton, groundnut, chilly, tobacco, castor and pulses are being damaged on a large scale by *Spodoptera litura*. commonly known as tobacco bud worm (Ayyangar and Rao, 1989). In context to these facts in the present study, the anti-feedant activity was evaluated based on no-choice experiment (Zandi-Sohani *et al.*, 2012). It was observed that both AAREO and AAWEEO possessed significant anti-feedant activity in a dose dependent manner as a function of concentration. At 8µl/mL

concentration both AAREO and AAWE0 showed 90.85% and 87.92% anti-feedant activity respectively. The detailed results of AAREO and AAWE0 exhibiting the anti-feedant activity are being presented in **Table 4.12 & Table 4.13** respectively and **fig 4.20 & fig 4.21** while the comparison of antifeedant activity between AAREO & AAWE0 is being presented in **fig 4.22**. The photographs of the present experimental data layout are being shown in **fig 4.23 & fig 4.24**.

Tripathi et al., (2001) reported that the compound 1,8-cineole drastically reduced the hatching of *T. castaneum* eggs and the successive survival rate of the larvae. Adult emergence was also reported to be hampered by 1,8-cineole.

It has been inferred from the earlier study that the methanolic extract *A. annua* critically affects feeding pattern in larva by acting as an antifeedant (**Khosravi et al., 2010**).

The essential oil of *Artemisia annua* L., has been studied for its toxicity and physiological aspects on 4th instar larva of the cotton bollworm *Helicoverpa armigera*. It has been suggested that essential oil of *A. annua* can be used as an active agent in controlling important pest of field crops (**Mojarab-Mahboubkar et al., 2015**).

The compounds like α -pinene, β -pinene, camphene, sabinene, 1,8-cineole, α -copaene, germacrene-D, borneol, α -selinene, and, γ -cadinene, have been reported to exhibit an effective larvicidal and anti-feedant activity (**Mondal et al., 2009; Espinoza et al., 2018**).

α -pinene, β -caryophyllene, has been reported to possess insect repellent activity against *S. oblique* (**Kumar et al., 2019**). α -pinene, camphene, sabinene, D-camphor and eucalyptol have been reported to exhibit antifeedant and insecticidal activity against the 4th larval instar of *S. littoralis* (**Ali and Ibrahim, 2018**).

on the basis of afore mentioned facts it can be concluded that the anti-feedant activity of AAREO and AAWE0 might be possibly due to the presence of major constituents like camphor, 1,8 cineole, germacrene-D, β -caryophyllene or due to the synergic effect of other major or minor/trace components of essential oils which show both qualitative and quantitative variations in their make-up.

Table 4.12 Antifeedant activity of AAREO

Conc. $\mu\text{L}/\text{mL}$	24 hrs	48 hrs	MLAC	Feeding %	Antifeedant activity	Feeding inhibition	Preference index	Antifeedant category
2 $\mu\text{L}/\text{mL}$	1.61	2.19	1.9 \pm 0.410	11.875	47.36842	31.03448	0.689655	Slightly antifeedant
4 $\mu\text{L}/\text{mL}$	0.59	0.96	0.77 \pm 0.261	4.8125	78.67036	64.84018	0.351598	Moderately antifeedant
8 $\mu\text{L}/\text{mL}$	0.2	0.46	0.33 \pm 0.183	2.0625	90.85873	83.24873	0.167513	Extremely antifeedant
Control	2.02	5.21	3.61 \pm 2.255	22.5625	0	0	1	
CD1	0.67		CD2	0.96		CD3	1.35	
CV	47.10							

CD1=critical difference with respect to treatment CD2= critical difference with respect to concentration, CD3= critical difference with respect to interaction between concentration and treatment, CV= Coefficient of variation AAREO= *Artemisia annua* rainy essential oil MLAC= mean leaf square area

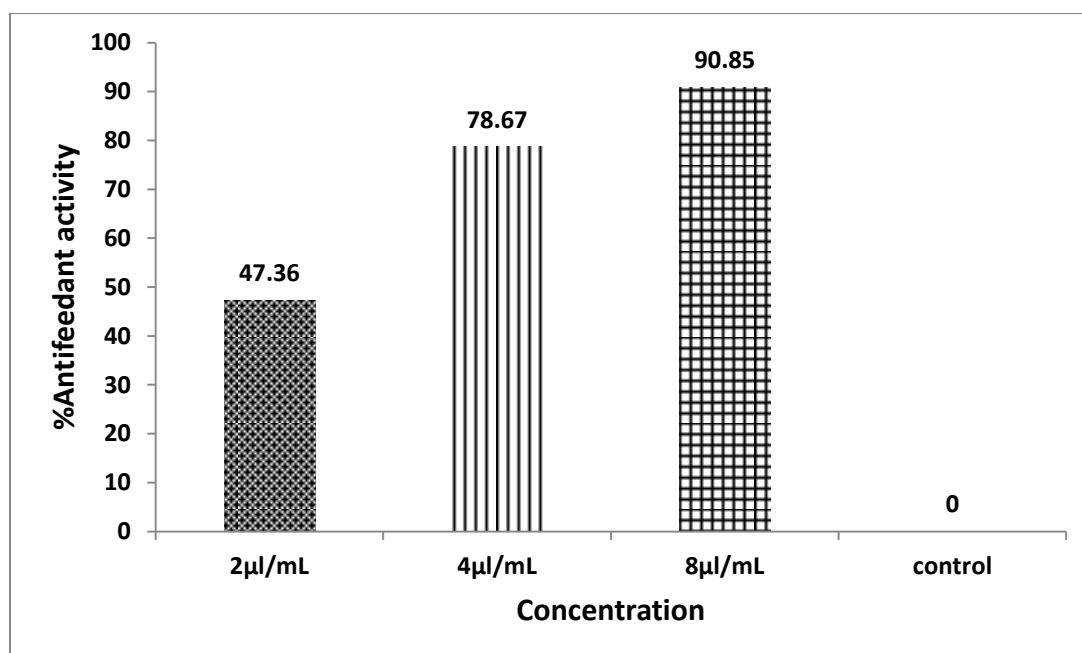
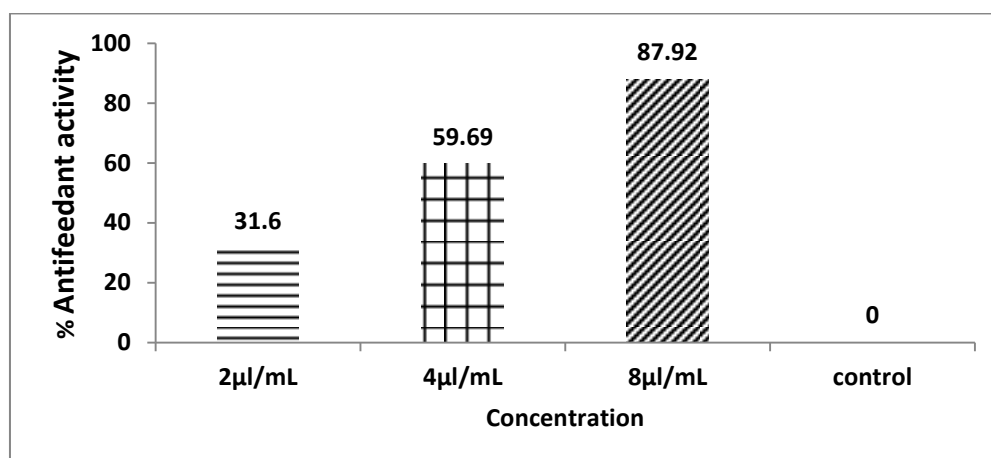
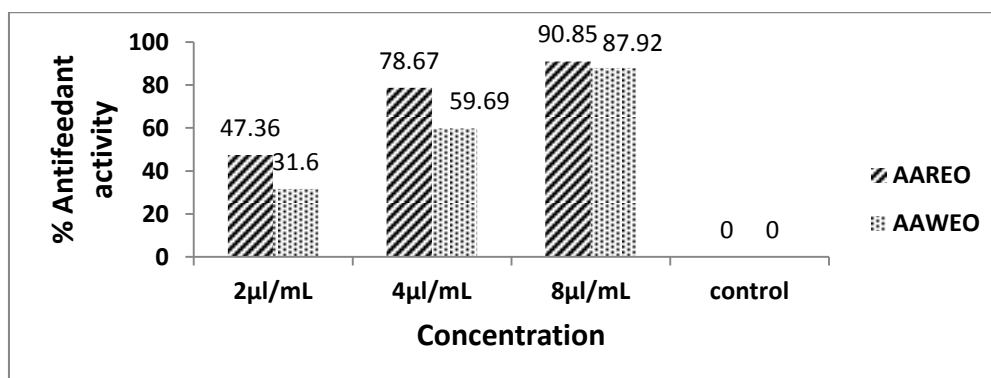


Fig. 4.20 Antifeedant activity of AAREO

Table. 4.13 Antifeedant activity of AAWEO

Conc. $\mu\text{L/mL}$	24 hrs	48 hrs	MLAC	Feeding %	Antifeedant activity	Feeding inhibition	Preference index	Antifeedant category
2 $\mu\text{L/mL}$	3.20	1.67	2.435 \pm 1.081	15.21875	31.60112	18.76564	0.812344	Slightly antifeedant
4 $\mu\text{L/mL}$	1.63	1.24	1.435 \pm 0.275	8.96875	59.69101	42.54254	0.574575	Moderately antifeedant
8 $\mu\text{L/mL}$	0.24	0.62	0.43 \pm 0.268	2.6875	87.92135	78.44612	0.215539	Extremely antifeedant
Control	3.84	3.28	3.56 \pm 0.395	22.25	0	0	1	
CD1	0.65		CD2	0.92		CD3	1.30	
CV	37.87							

CD1=critical difference with respect to treatment, CD2= critical difference with respect to concentration, CD3= critical difference with respect to interaction between concentration and treatment, CV= Coefficient of variation
AAWEO= *Artemisia annua* winter essential oil MLAC= mean leaf square area

**Fig. 4.21 Antifeedant activity of AAWEO****Fig. 4.22 Comparison between antifeedant activity of AAREO & AAWEO**

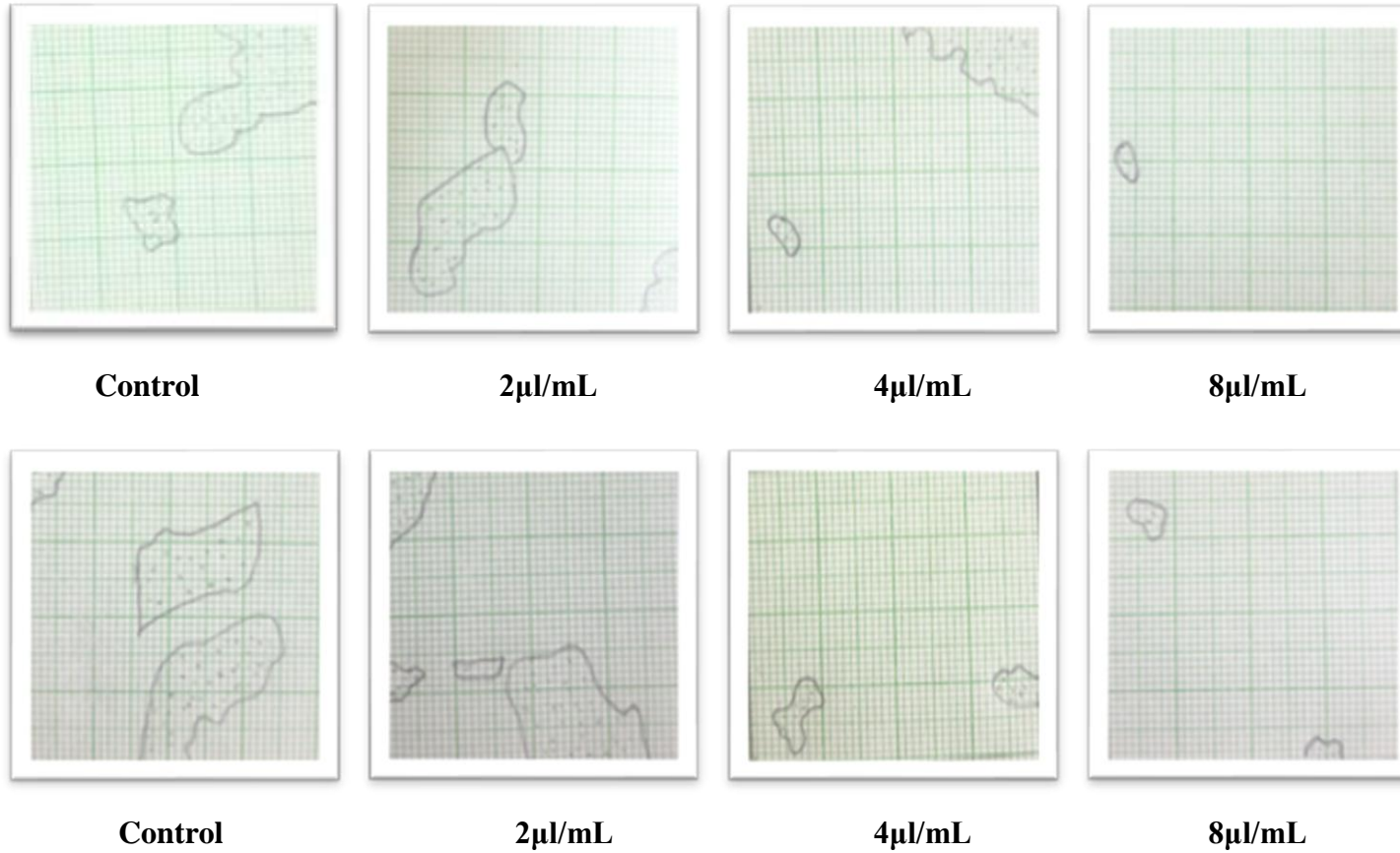


Fig 4.23 Graph paper method analysis of anti-feedant activity of AAREO & AAWEO against *S. litura*.

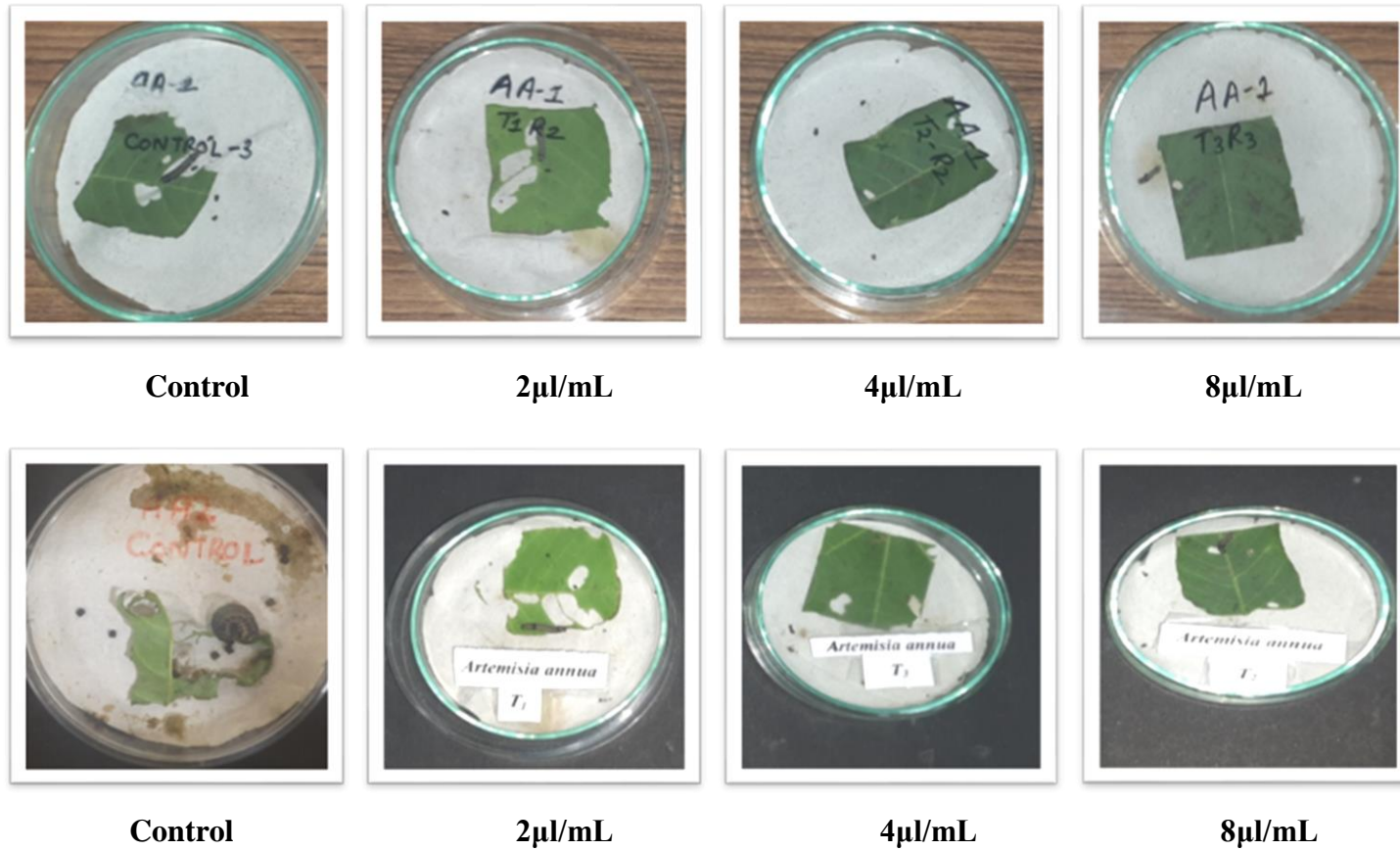


Fig.4.24 Experimental layout of antifeedant activity in (a) AAREO & (b) AAWEO against *Spodoptera litura*

4.4.4 NEMATOCIDAL ACTIVITY

Andrés *et al.*, (2012) reported that plant parasitic nematodes are the known to be one of the most destructive group of plant pathogens all over the world and their control is highly challenging. Plant parasitic nematodes mainly attack their host by means of various strategies. They can be ectoparasites, which feed on the plant's outer tissues, or endoparasites, which feed and live within the plant's tissues. Endoparasites include a number of important plant parasitic nematodes. *Meloidogyne* spp., root-knot nematodes, are one of the most economically damaging genera of plant parasitic nematodes on horticultural and field crops. Root-knot nematodes. *Meloidogyne* sp., attacking over 2,000 different plant species are widely distributed. Root-knot nematodes spend a portion of their lives in the soil as eggs or as second-stage larvae. The latter enter the roots and develop feeding sites in susceptible hosts, causing roots to expand and take on a distinctive "knotty" appearance. Root galling limits water and nutrient uptake, resulting in serious symptoms such as malnutrition, chlorosis, and stunting, leading to significant quantitative and qualitative losses in crop plants. In tropical and subtropical environments, four species, *M. incognita*, *M. javanica*, *M. arenaria*, and *M. hapla*, have been reported to be responsible for nearly 90% of the agronomical damages (**Echeverrigaray *et al.*, 2010**). Phytochemicals present in essential oils have been reported to possess great potential in nematode control (**Chitwood, 2002**).

Essential oils from various families, such as Asteraceae, Lamiaceae, and Rutaceae, have thoroughly been examined for nematicidal activity against *Meloidogyne* sp. with special reference to genus *Artemisia*, *Lavandula*, *Mentha* etc (**Andrés *et al.*, 2012**). The presence of volatile monoterpenes in essential oils have been reported to provide an important defense strategy to the plant against insect pests and pathogenic organisms. These terpenoids have also been reported to play a role in plant parasitic interactions, acting as signaling molecules (**Batish *et al.*, 2008**).

4.4.4.1 Effect of AAREO & AAWE0 on mobility of second stage larvae of *Meloidogyne. incognita*

In present investigation both AAREO & AAWE0 have been examined against second stage juvenile larvae of *M. incognita* (root knot nematode). Both the samples

were tested at three concentration (1µl/mL, 3 µl/mL, 5 µl/mL) and were observed for the duration of 12, 48 and 72 hour.

A dose dependent and time dependent mortality/immobility of second instar larvae of *M. incognita* was observed both in AAREO & AAWE0 with respect to negative control. AAREO was observed to be more effective than AAWE0. The immobility of larvae at a dose level of 5µl/mL was observed 21.37%, 24.08% and 27.71% respectively after the time interval of 24, 48 & 72 hours respectively. In AAREO with final average value of 24.38% at the end of the experiment. Similarly the immobility in AAWE0 was observed in the order of 18.16%, 23.84% and 27.25% with the time interval of 24, 48 & 72 hours respectively at a dose level of 5µl/mL. However the order of activity was observed as AAREO (24.38±3.181) > AAWE0 (23.08±4.591%) > Control (7.91±4.902) respectively. The detailed experimental observation in terms of percentage mortality of *M. incognita* 2nd instar larvae has been depicted in **table 4.14**.

4.4.4.2 Effect of AAREO & AAWE0 on egg hatching of *M. incognita*

A concentration dependent and time dependent hatching of egg masses of *M. incognita* was observed in AAREO & AAWE0 with respect to negative control. It was observed that AAWE0 was more effective than AAREO on the inhibition of egg hatching. The rate of egg hatching was directly proportional to the exposure of time period and was inversely proportional to the concentration of oil samples. The maximum inhibitory effect on egg hatching at a concentration of 2µl/mL was observed as 23.79% in AAWE0 after a time interval of 72 hour while the minimum inhibitory effect was observed as 5.38% at a concentration of 10 µl/mL after 72 hours. Similarly in AAREO the maximum rate of hatching 18.31% was observed at a concentration of 2µl/mL while the minimum 11.41% was observed at 10 µl/mL concentration after 72 hours. However the order of activity was observed as: AAWE0 (5.38±1.814) > AAREO (18.31±7.495) > control (32.29±16.61) respectively. The experimental layout with the observation in terms of percentage egg hatching of *M. incognita* egg masses are being depicted in **table 4.15** and **fig 4.25** while the morphological characters of *M. incognita* in different stages are being shown in **fig. 4.26**.

Table 4.14 Nematicidal activity of AAREO and AAWEO against 2nd instar larvae of *M.incognita*

Sample	Dose	Observation												Mean
		Percentage mortality of <i>M.incognita</i> 2 nd instar larvae												
		24 hour				48 hour				72 hour				
		R1	R2	R3	Avg	R1	R2	R3	Avg	R1	R2	R3	Avg	
AAREO	1 µl/mL	14.85	16.82	17.92	16.53±1.555	17.82	20.56	19.81	19.39±1.415	21.78	26.16	22.64	23.52±2.320	19.81±3.514
	3 µl/mL	18.44	19.62	20.90	19.65±1.230	22.33	21.49	24.54	22.78±1.575	28.15	27.1	27.1	27.45±0.606	23.29±3.925
	5 µl/mL	20.37	21.62	22.12	21.37±0.901	24.07	23.42	24.77	24.08±0.675	28.7	26.12	28.31	27.71±1.390	24.38±3.181
AAWEO	1 µl/mL	5.76	8.33	5.88	6.65±1.450	14.42	16.34	11.53	14.09±2.421	25.3	20.37	15.68	20.45±4.810	13.73±6.907
	3 µl/mL	12.72	13.27	15.25	13.74±1.330	16.36	16.81	20.33	17.83±2.173	20.9	22.12	23.72	22.24±1.414	17.93±4.251
	5µl/mL	15.51	20.33	18.64	18.16±2.44	20.68	24.57	26.27	23.84±2.865	25	27.96	28.81	27.25±2.00	23.08±4.591
control	In water	2.12	1.42	1.17	1.57±0.492	9.75	8.21	10.36	9.44±1.108	10.06	9.21	12.44	10.57±1.674	7.19±4.902
					CD (5%)				SEm±					
Hours (H)					1.13				0.387					
Concentration (C)					1.30				0.447					
Interaction (H×C)					2.26				0.775					
CV	7.19													

CD=critical difference at 5% CV= Coefficient of variation AAWEO= *Artemisia annua* winter essential oil

Table 4.15 Nematicidal activity of AAREO and AAWEO against *M. incognita* egg hatching from egg masses

Sample	Dose	Observation												Mean
		Percentage egg hatching of <i>M.incognita</i> from egg masses												
		24 hour				48 hour				72 hour				
		R1	R2	R3	Avg	R1	R2	R3	Avg	R1	R2	R3	Avg	
AAREO	2 µl/mL	17.85	12.65	8.48	10.24±4.694	23.33	27.36	25.96	19.66±2.046	34.53	33.65	30.02	25.05±2.390	18.31±7.495
	6 µl/mL	10.85	9.44	6.50	8.19±2.219	13.53	15.64	13.31	12.12±1.286	21.94	20.68	19.02	16.91±1.464	12.40±4.367
	10 µl/mL	7.00	5.21	6.50	7.17±0.923	11.76	13.96	12.87	12.14±1.100	17.66	15.44	16.66	14.94±1.111	11.41±3.935
AAWEO	2 µl/mL	11.76	13.53	15.23	13.50±1.735	25.32	26.12	23.84	25.09±1.156	35.39	33.26	29.73	32.79±2.858	23.79±9.710
	6 µl/mL	7.00	7.50	10.85	8.45±2.093	17.46	24.17	18.63	20.08±3.584	25.48	23.50	23.16	24.04±1.252	17.52±8.103
	10 µl/mL	3.89	3.36	4.43	3.89±0.535	4.95	5.73	3.89	4.85±0.923	7.50	8.97	5.73	7.40±1.622	5.38±1.814
control	In water	13.56	12.29	16.15	14.00±1.606	35.12	35.15	39.11	36.46±1.873	45.12	43.12	51.09	46.42±3.385	32.29±16.61
					CD (5%)				SEm±					
Hours (H)					2.04				0.699					
Concentration (C)					2.35				0.808					
Interaction (H×C)					4.08				1.399					
CV					11.755									

CD=critical difference at 5% CV= Coefficient of variation AAWEO= *Artemisia annua* winter essential oil

The *in-vitro* toxicity of *Artemisia annua* essential oil has been reported against J_2 of *M. incognita* with 100 percent mortality of the nematode *M. incognita* at the dose levels of in 500 and 250 ppm concentrations respectively which gradually decreased with the lower concentrations respectively (Shakil *et al.*, 2004). Nematotoxic effect of an aqueous extract of *Artemisia annua* and its components caffeic acid, chlorogenic acid (5-caffeoylquinic acid, 5-CQA), artemisinin and the related semi-synthetic artesunate, have been reported on the root-knot nematode *Meloidogyne incognita* (D'Addabbo *et al.*, 2013).

The effect of major compounds in the oils of AAREO &AAWEO cannot be neglected for attributing the nematocidal activity. Since the major compounds of both the oils are 1,8- cineole and camphor. These compounds have been reported to possess significant nematocidal activity in previous study against root knot nematode *M.incognita*. Compounds like 1,8-cineole, L-camphor, L-borneol, caryophyllene oxide α - and β -pinene, p-cymene, terpinen-4-ol, β -caryophyllene and carvacrol have been reported to possess a potent nematocidal activity against the *M.incognita* (Ntalli *et al.*, 2011). Nematocidal activity due to synergistic interaction of carvone with 1,8-cineole and D-menthol (Andrés *et al.*, 2012). Similarly anti-egg hatching and mortality of nematode larvae in α -terpineol and camphor has been reported (Echeverrigaray *et al.*, 2010).

Kumar and Singh, (2021) reported that camphene, pinene, myrcene, terpinene, terpinen- 4-ol, thymol and carvacrol) showed nematocidal activity against PWN32 (pine wood nematode).Camphor, 1,8- cineole, camphene from *A.sieberi* have been reported to possess nematocidal activity against *M. incognita*, *C. maculatus*, *S. oryzae*, *T. castaneum*, *B. tabaci* and *M. javanica* (Mahboubi, 2014). The chemical constituents identified in AAREO & AAWEO correlates with the above mentioned constituents. It can be thus inferred that the nematocidal activity of AAREO & AAWEO with respect to immobility of second instar larvae of *M.incognita* and egg hatching could be possibly due to the presence of camphor, p-cymene, 1,8-cineole, terpinen-4-ol, β -caryophyllene, carvacrol, L-borneol, caryophyllene oxide and α - and β -pinene or due to the synergistic effects of other major or minor/trace components of essential oils.

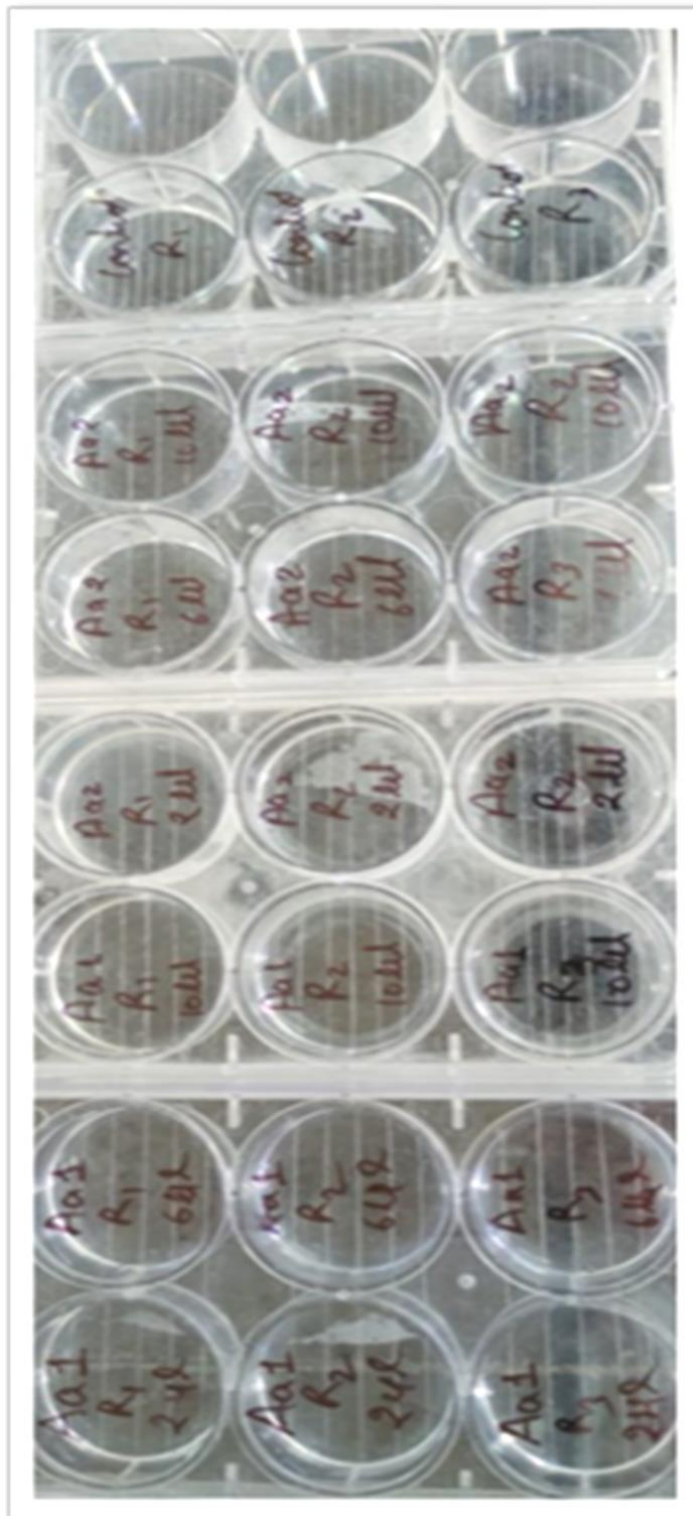


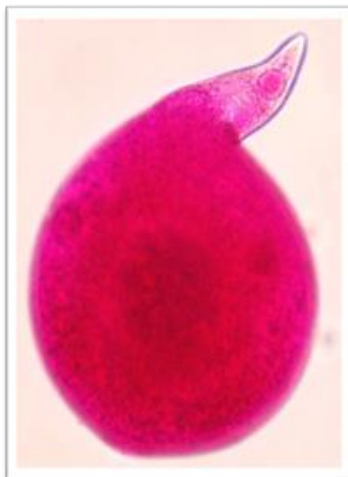
Fig.4.25 Experimental layout of egg hatching of *M.incognita*



(a) Second stage larvae of root knot (*Meloidogyne* spp.) nematode



(b) Posterior region of second stage larvae root knot (*Meloidogyne* spp.) nematode



(c) Adult female of root knot (*Meloidogyne* spp.) nematode



(d) Anterior region of second stage larvae of root knot (*Meloidogyne* spp.) nematode

Fig. 4.26 Morphological characters of *M. incognita* in different stages



The present research was investigated with the objective “Seasonal Variation in Chemical Composition and Biological Activities of Essential Oils from *Artemisia annua* L.” which was collected wildely from Tarai region of Kumaon viz., Pantnagar ,Uttarakhand ,India with an altitude of 243m. The plant material was collected in the month of September 2020 and January 2021 respectively. The essential oils isolated through hydrodistillation were studied by GC/MS and various biological activities were carried out .The essential oils isolated in rainy and winter season for their chemical composition and various biological activities were designated as AAREO & AAWE0 respectively

. The chemical composition of AAREO revealed the presence of seventy five essential oil constituents which contributed 91.7 % of the total oil. The presence of oxygenated monoterpenoids with 32.2% composition dominated over sesquiterpenoids hydrocarbon with 32.1% composition. Camphor (14.1%) was found to be the major component of AAREO followed by germacrene –D (9.0%), *trans*- β -caryophyllene (8.7%), eucalyptol (4.2%), *cis*-cadin-4en-7-ol (3.9%), isoborneol (3.2%), p-cymene (2.9%), (-)-aromadendrene (2.6%), lavandulyl acetate (2.1%), α –cadinol (2.0%), camphene (2.0%), The chemical composition of AAWE0 revealed the presence of fifty nine chemical constituents which contributed 94.1% of the total essential oil. Oxygenated monoterpenoids with 66.8% dominated over Monoterpenoid hydrocarbon and Sesquiterpenoid hydrocarbon with 12.1% and 7.6% respectively. The major chemical constituents of AAWE0 were Camphor (17.9%) followed by eucalyptol (14.6%), (E,E)-2,6-dimethyl-3,5,7-octatrien-2-ol (6.3%), (5E)-2,6-dimethyl-1,5,7-octatrien-3-ol (5.2%) artemisia alcohol (3.5%), terpinene-4-ol (3.4%) isoborneol (3.2%), p-cymene (3.0%), caryophyllene oxide (3.1%), *trans* β -caryophyllene (2.7%), mycrene (2.6%), (+)- α -terpineol (2.5%), artemisia ketone (2.5%), camphene (2.1%), (-) β -farnesene (2.0%), besides other constituents which were present in minor and trace amounts.

In present study, AAREO and AAWE0 were studied for anti-oxidant activity via following three methods viz..H₂O₂, DPPH scavenging and metal chelating activity. Both AAREO and AAWE0 exhibited DPPH radical scavenging activity at 515nm absorbance in dose dependent manner. The DPPH free radical scavenging activity of AAREO and AAWE0 was observed in the order of AAWE0 (IC₅₀=45.588±0.334) > AAREO (IC₅₀=77.87±0.528) compared to BHT (IC₅₀=43.48±1.55) which indicated that AAWE0 possess maximum DPPH free radical scavenging activity with minimum IC₅₀ value followed by AAREO.

The effect on metal chelating activity of Fe²⁺ by AAREO and AAWE0 was evaluated in presence of FeCl₂ and ferrozine in a dose dependent manner. The metal chelating activity of AAREO and AAWE0 was observed in the following order of AAWE0 (IC₅₀=49.50±0.234)> AAREO (IC₅₀ =51.40±0.158) compared to Na₂EDTA (IC₅₀ = 41.68±0.396)which indicated that AAWE0 possess maximum metal chelating activity with minimum IC₅₀ value followed by AAREO.

H₂O₂ radical savenging activity of AAREO and AAWE0 was conducted by measuring absorbance at 230nm in dose dependent manner. The H₂O₂ free radical scavenging activity of AAREO, AAWE0 and standard was observed in dose dependent manner (20µl/mL, 40 µl/mL,60 µg/mL,80 µl/mL and 100 µg/mL respectively). The order of activity was observed as : AAWE0 (IC₅₀ =34.79±0.419) > AAREO (IC₅₀ =44.83±1.304compared to BHT (IC₅₀ =27.96±1.195).

Both AAREO and AAWE0 were investigated for their *in-vitro* anti-inflammatory activity based on inhibition percentage of the Hen's egg albumin protein denaturation measured by absorbance at 660nm in a dose dependent manner. AAWE0 possess maximum anti-inflammatory activity with minimum IC₅₀ value followed by AAREO with higher IC₅₀ value. The order in which both AAREO and AAWE0 showed anti-inflammatory activity was observed as AAWE0 (IC₅₀=66.90±0.040) > AAREO (IC₅₀=76.55±0.218) compared to the standard anti-inflammatory drug diclofenac sodium tablets (IC₅₀=63.41±0.037).

A study was conducted to investigate *in-vitro* anti-diabetic potential of AAREO and AAWEEO spectrophotometrically by measuring absorbance at 540nm in dose dependent manner. AAWEEO exhibited maximum activity with minimum IC₅₀ value followed by AAREO with higher IC₅₀ value in comparison to the standard acarbose with IC₅₀= 38.21±0.03. The order in which both AAREO and AAWEEO exhibited anti-diabetic activity was observed as AAWEEO (47.45±0.195) > AAREO (74.26±3.706).

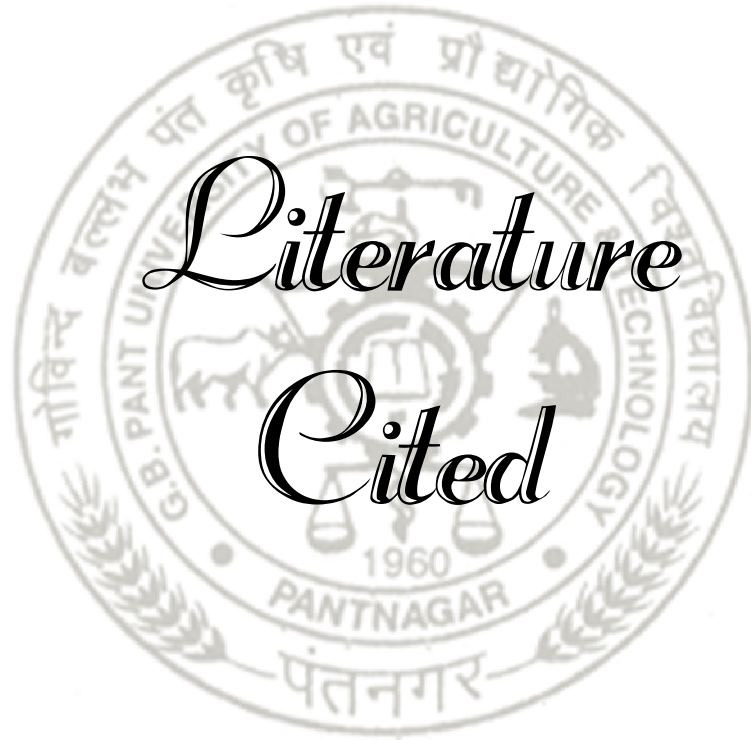
The essential oils were studied their anti-feedant activity. The activity was based on 'No-Choice Experiment' against *spodopteralitura*. Both AAREO and AAWEEO were observed to show strong anti-feedant activity in a dose dependent manner. AAREO exhibited 90.85 % anti-feedant activity while AAWEEO exhibited 87.92% anti-feedant activity at a dose level of 8µl/mL.

The present investigation of both AAREO & AAWEEO were examined for nematicidal activity against second stage juveniles of *M. incognita* (root knot nematode). It was observed that , Between the two samples, AAREO was found to be most effective at 5 µl/mL concentration with 24.38% inhibition in mobility of larvae, followed by AAWEEO (23.08%) inhibition. So from this analysis it was observed that AAREO is exhibiting more activity over AAWEEO at different concentration (1µl/mL, 3µl/mL,5µl/mL) respectively with the exposure time of 12, 48 and 72 hour, where as in the case of egg inhibition process AAWEEO was found to be more effective than AAREO as it exhibited minimum inhibition (5.38%) of egg hatching at a concentration of 10µl/mL, followed by AAREO with minimum (11.41%) inhibition respectively.

From various reports it has been observed that Camphor and 1,8 cineole is a strong bioactive natural compound and possess various biological activities. In present investigation, camphor has been found to be a major compound with 14.1% and 17.4% composition both in AAREO and AAWEEO respectively, and the oil yield has been observed as 0.70-0.50% in both the collections. Hence, it can be inferred that *Artemisia annua* grown in wild and cultivated in various part of Uttarakhand can also

be a good source of camphor and 1,8 cineole with revenue generation for local people of the state.

Similarly the essential oil components *viz.* camphor, α -pinene, 1,8-cineole, artemisia ketone, *trans*- β -caryophyllene, isoborneol, p-cymene, terpinen-4-ol, α , sabinene, germacrene-D, camphene etc. have been reported to exhibit strong to moderate anti-feedant, anti-inflammatory, anti-diabetic and nematicidal activities, In present investigation AAREO & AAWEEO have been observed to possess diversified chemical composition with varying chemical make-up along with the above mentioned compounds. Hence, it can be inferred that the essential oil of *Artemisia annua* can be used as a natural anti-feedant, nematicidal, anti-inflammatory or food preservative to protect it from larval stage of insects and oxidative deterioration, respectively and can also be used as a natural anti-diabetic agent in pure form or via formulation after proper clinical trials.



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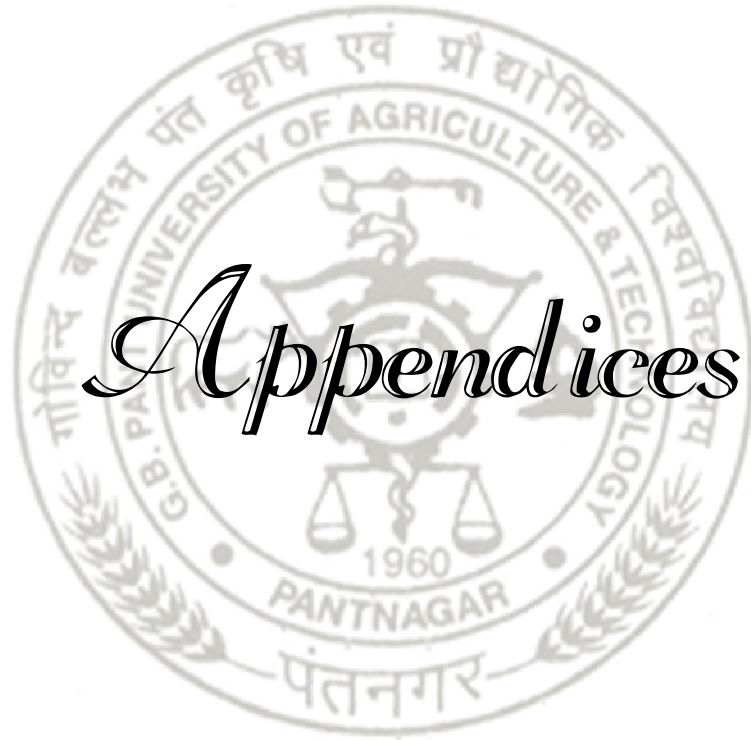
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Appendices

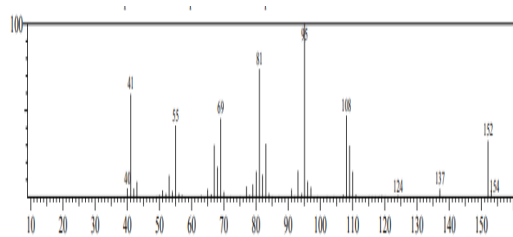


Appendix-1

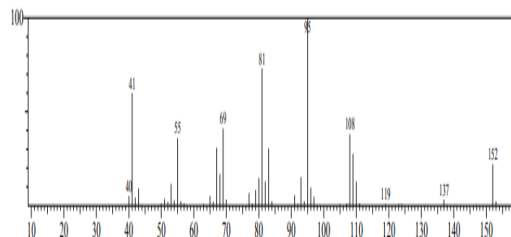
Mass spectra of identified constituents in the essential oils of *Artemisia annua* rainy essential oil (AAREO) and *Artemisia annua* winter essential oil (AAWEO)

1) Constituent name: camphor, Formula: $C_{10}H_{16}O$, Molecular weight: 152

Target

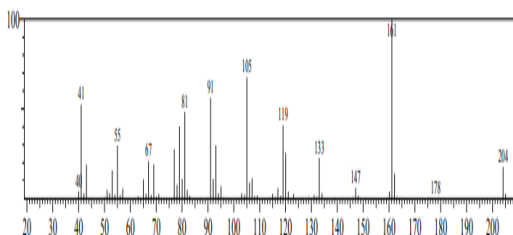


Constituent identified: camphor

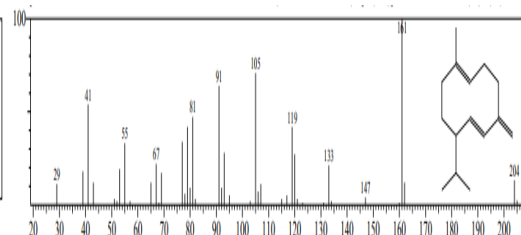


2) Constituent name: germacrene-D, Formula: $C_{15}H_{24}$, Molecular weight: 204

Target

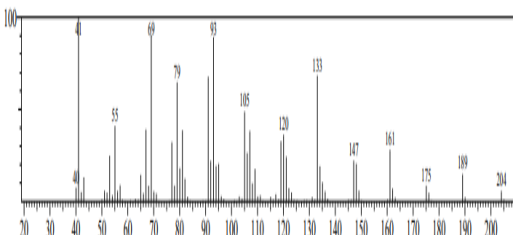


Constituent identified: germacrene-D

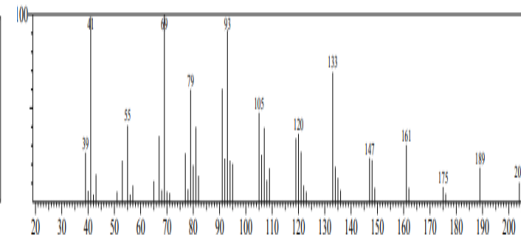


3) Constituent name: *trans*- β -caryophyllene, Formula: $C_{15}H_{24}$, Molecular weight: 204

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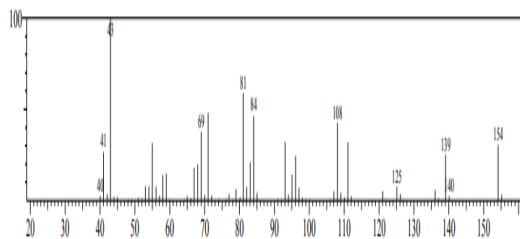


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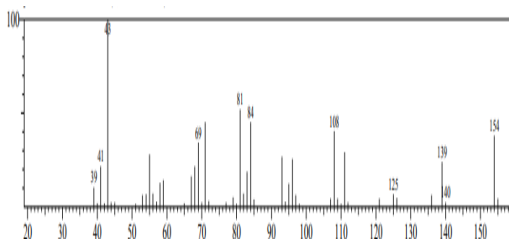


4) Constituent name: eucalyptol, Formula: C₁₀H₁₈O, Molecular weight: 154

Target

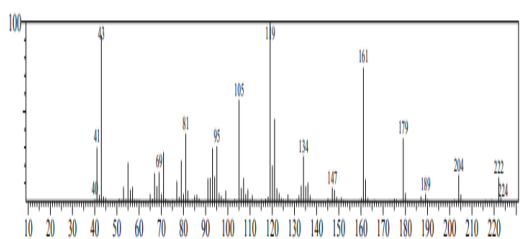


Constituent identified: eucalyptol

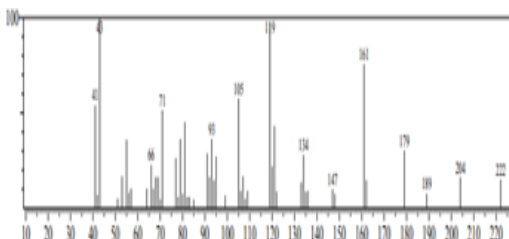


5) Constituent name: cis-cadin-4en-7-ol, Formula: C₁₅H₂₆O, Molecular weight: 222

Target

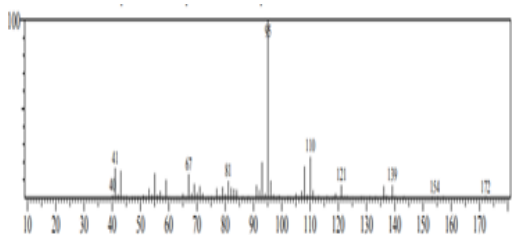


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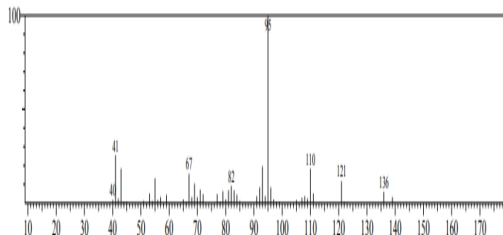


6) Constituent name: isoborneol, Formula: C₁₀H₁₈O, Molecular weight: 154

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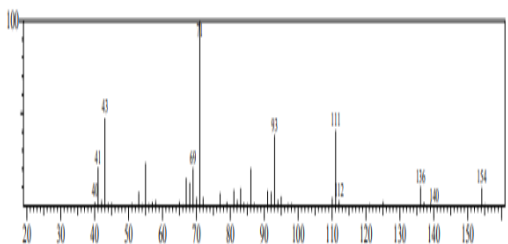


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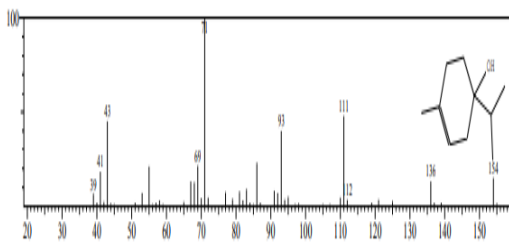


7) Constituent name: terpinene-4-ol, Formula: C₁₀H₁₈O, Molecular weight: 154

Target

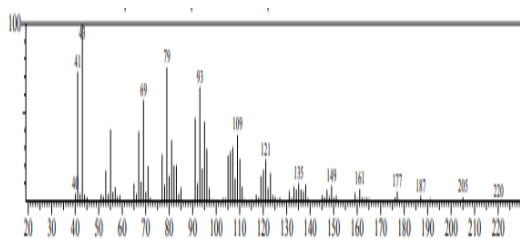


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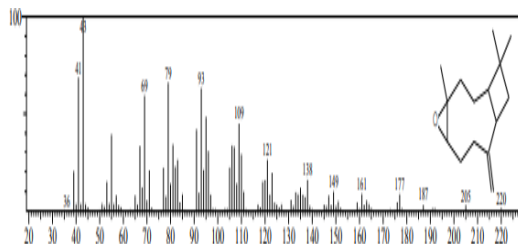


8) **Constituent name:** caryophyllene oxide, **Formula:** C₁₅H₂₄O, **Molecular weight:** 220

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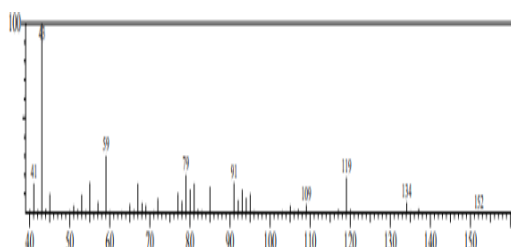


Constituent identified: caryophyllene oxide



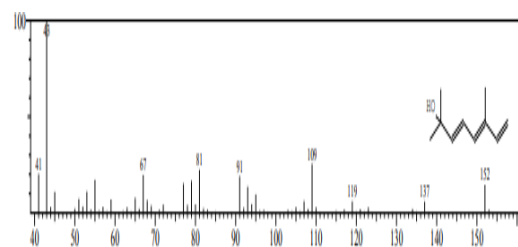
9) **Constituent name:** (E,E)-2,6-dimethyl-3,5,7-octatrien-2-ol, **Formula:** C₁₀H₁₆O, **Molecular weight:** 152

Target



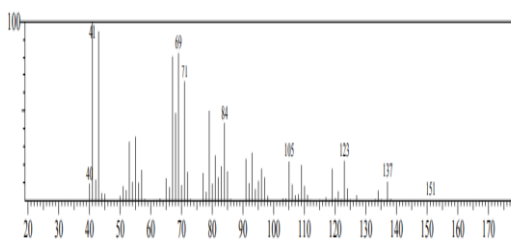
Constituent identified:

(E,E)-2,6-dimethyl-3,5,7-octatrien-2-ol



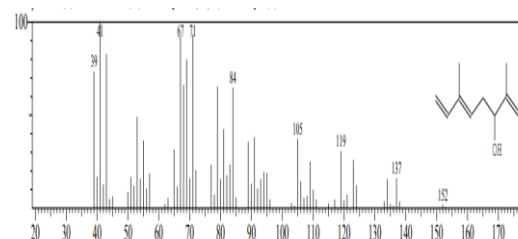
10. **Constituent name:** (5E)-2,6-dimethyl-1,5,7-octatrien-3-ol, **Formula:** C₁₀H₁₆O, **Molecular weight:** 152

Target



Constituent identified:

(5E)-2,6-dimethyl-1,5,7-octatrien-3-ol



CURRICULUM VITAE

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E-mail : Ashimalhotra764@gmail.com
Career Objective : To expand my learnings, knowledge and skills.

Educational Qualification :

S. N.	Examination Passed	Institution	Year	Percentage/CGPA
1.	M.Sc.	GBPUA&T, Pantnagar	2021	Pursuing
2.	B.Sc.	D.A.V. P.G. college, Dehradun	2019	6.98
3.	Intermediate	Shri guru nanak public senior secondary school	2015	80.2%
4.	High School	Shri guru nanak public senior secondary school	2013	8.6

Specialization: Major: Chemistry Minor: Nil

Publication: Nil

Conference/Seminars/Workshops/Training Attended: 2

List of papers presented in conference/seminar during degree programme: Nil.

Software Skills: MS Word, MS Excel, MS Power Point.

Professional Skills: Critical thinking and problem solving.

Professional Affiliations (Membership, etc.): Subject matter expert at chegg india

Awards/Honours/Achievements: Nil.

Place: Pantnagar

Date:


(Akanksha Malhotra)

Name : Akanksha Malhotra Id. No. : 55503
Sem. and year of admission : 1st Sem., 2019-20 Degree : M.Sc.
Department : Chemistry
Major : Chemistry
Thesis title : “Seasonal Variation in Chemical Composition and Biological Activities of Essential Oils from *Artemisia annua* L.”
No. of Pages : 126 Advisor : Dr. Om Prakash

ABSTRACT

The objectives of the present research study were to investigate the seasonal variation in phytochemical analysis and biological activities of *Artemisia annua* essential oils. The plants were collected from Tarai region of Uttarakhand, Pantnagar in the month of September, 2020 and January 2021 respectively. The essential oils of fresh aerial part of the collected plant materials were extracted by hydrodistillation using Clevenger apparatus and were studied by GC/MS for its phytochemical study and various biological activities. The essential oils isolated in rainy and winter season were designated as AAREO & AAWE0 respectively.

The phytochemical analysis of AAREO revealed the identification of 75 constituents which contributed 91.7% of the total essential oil. The major components of AAREO were camphor (14.1%) followed by germacrene -D (9.0%), *trans*- β -caryophyllene (8.7%), eucalyptol (4.2%), *cis*-cadin-4en-7-ol (3.9%), isoborneol (3.2%), p-cymene (2.9%) etc. Similarly, in AAWE0 59 constituents were identified which contributed 94.1% of the total essential oil. The major components of AAWE0 were camphor (17.9%) followed by eucalyptol (14.6%), (*E,E*)-2,6-dimethyl-3,5,7-octatrien-2-ol (6.3%), (*5E*)-2,6-dimethyl-1,5,7-octatrien-3-ol (5.2%), artemisia alcohol (3.5%), isoborneol (3.2%), caryophyllene oxide (3.1%), artemisia ketone (2.5%) etc besides other minor and trace constituents. Both AAREO & AAWE0 showed qualitative and quantitative differences in their phytochemical makeup.

Both AAREO and AAWE0 exhibited dose dependent *in-vitro* anti-oxidant activity, evaluated by DPPH scavenging, metal chelating assay and H₂O₂ radical scavenging as indicated by different IC₅₀ values. IC₅₀=77.87 \pm 0.528 to IC₅₀=45.588 \pm 0.334 μ l/mL for DPPH radical scavenging activity similarly IC₅₀=51.40 \pm 0.158 to IC₅₀=49.50 \pm 0.234 for metal chelating activity and IC₅₀=44.83 \pm 1.304 to IC₅₀=34.79 \pm 0.419 for H₂O₂ radical scavenging activity with respect to various standards like, BHT and Na₂EDTA respectively. With respect to diclofenac sodium, the standard anti-inflammatory drug both AAREO and AAWE0 exhibited anti-inflammatory activity with IC₅₀= AAWE0 (IC₅₀=66.90 \pm 0.040) > AAREO (IC₅₀=76.55 \pm 0.218) μ l/mL respectively. Similarly, significant anti-diabetic activity was observed in both AAREO and AAWE0 with IC₅₀= 74.26 \pm 3.706 μ l/mL to IC₅₀= 47.45 \pm 0.195 μ l/mL, respectively compared to standard acarbose (IC₅₀= 38.21 \pm 0.03 μ l/mL). Both AAREO and AAWE0 were observed to exhibit strong anti-feedant activity respectively against *Spodoptera litura* (90.85% & 87.92%) .Like wise AAREO & AAWE0 were subjected to nematicidal activity against root knot nematode, *Meloidogyne incognita*. It was observed that AAREO exhibited more mortality over AAWE0 at different concentration (1-5 μ l/mL), where as in the case of egg inhibition process AAWE0 was found to be more effective (5.38% egg hatching) than AAREO (11.41% egg hatching) at a concentration of 10 μ l/mL respectively.

Based on observations of above mentioned study, it can be inferred that *Artemisia annua* can be a good natural resource of camphor and 1,8 cineole and the essential oils can be good source of antioxidant, anti-inflammatory, anti-diabetic, anti-feedant and nematicidal activities after proper clinical trials.


(Om Prakash)
Advisor


(Akanksha Malhotra)
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षट्मास एवं प्रवेश वर्ष	: प्रथम, 2019-2020	डिग्री	: एम0एस0सी0
प्रमुख विषय	: रासायनिक विज्ञान	विभाग	: रासायनिक विज्ञान
शोध का शीर्षक	: आर्टिमिसिया एनुआ के सुगंधित तेलों का	मौसमी रासायनिक विश्लेषण तथा जैविक गतिविधियों	
पृष्ठ संख्या	: 126	सलाहकार	: डॉ0 ओम प्रकाश


सारांश

वर्तमान शोध का उद्देश्य आर्टिमिसिया एनुआ के सुगंधित तेलों की मौसमी संरचना पर प्रभाव तथा विभिन्न जैविक गतिविधियों का अध्ययन था। शोध हेतु पौधों को पंतनगर के तराई क्षेत्र से क्रमशः सितंबर 2020 तथा जनवरी 2021 के महीने में एकत्रित किया गया। सुगन्धित तेलों को हाइड्रोडिस्टिलेशन विधि द्वारा क्लेवेन्जर उपकरण के माध्यम से निकाला गया। सुगन्धित तेलों को गैस वर्ण लेखन तथा मास स्पेक्ट्रोमेट्री विधि द्वारा विश्लेषण किया गया। विभिन्न मौसमों में एकत्रित सुगन्धित तेलों को AAREO (बरसाती मौसम) तथा AAWE0 (सर्दी) नाम से नामित किया गया।

AAREO में 75 अवयव पाए गए जो कुल सुगन्धित तेल का 91.7% प्रतिशत थे। AAREO के प्रमुख घटक कैम्फर (14.1%), डी-जर्मैक्रिन (9.0%), ट्रांस- β -कैरीओफाइलीन (8.7%), सीस-कैडिन-4-एन-7-ओल (3.9%), युकालिपटोल (4.2%), आइसोबोर्नियोल (3.2%), पी-सायमीन (2.9%) आदि थे। इसी प्रकार, AAWE0 में 59 अवयव पहचाने गए जिनकी मात्रा 94.1% पायी गयी। AAWE0 के प्रमुख घटक कैम्फर (17.9%), युकालिपटोल (14.6%), 2इ, 6इ-डाइमिथाइल-3,5,7-ऑक्टेट्रियन-2-ओल (6.3%), 2इ,6इ -डाइमिथाइल-1,5,7-ऑक्टेट्रियन-3-ओल (5.2%), आर्टिमिसिया अल्कोहल (3.5%), आइसोबोर्नोल (3.2%), कैरियोफिलीन ऑक्साइड (3.1%), आर्टिमिसिया कीटोन (2.5%) आदि थे। AAREO तथा AAWE0 दोनों में गुणात्मक और मात्रात्मक अंतर पाया गया।

AAREO तथा AAWE0 के द्वारा इन विट्रो प्रति आक्षिकारक प्रतिक्रिया दर्शायी गयी, जिनका अध्ययन डी०पी०पी०एच मूलक उपमार्जक, धातु चिलेटिंग तथा हाइड्रोजन पेरोक्साइड मूलक उपमार्जक विधि से किया गया। प्रतिआक्षिकारक गतिविधि के रूप में डी०पी०पी०एच हेतु $IC_{50} = 77.87 \pm 0.528$ से $IC_{50} = 45.588 \pm 0.334$ माइक्रो ली०प्रति मी०ली०, धातु चिलेटिंग हेतु $IC_{50} = 51.40 \pm 0.158$ से $IC_{50} = 49.50 \pm 0.234$ तथा हाइड्रोजन पेरोक्साइड मूलक उपमार्जक विधि हेतु $IC_{50} = 44.83 \pm 1.304$ से $IC_{50} = 34.79 \pm 0.419$ माइक्रो ली०प्रति मी०ली० पायी गयी। जिनका प्रभाव मानक क्रमशः BHT तथा Na_2EDTA के सापेक्ष देखा गया। डाइक्लोफेनाक सोडियम के सापेक्ष AAREO तथा AAWE0 द्वारा एंटी इंफ्लेमेटरी गति विधि पायी गयी जिनका IC_{50} मान क्रमशः AAWE0 ($IC_{50} = 66.90 \pm 0.040$) > AAREO ($IC_{50} = 76.55 \pm 0.218$) माइक्रो ली०टी० था। इसी तरह $IC_{50} = 74.26 \pm 3.076$ से $IC_{50} = 47.45 \pm 0.195$ एकरबोस ($IC_{50} = 38.21 \pm 0.03$) माइक्रो ली०प्रति मी०ली० के सापेक्ष दोनों सुगन्धित तेलों द्वारा प्रति मधुमेह प्रतिक्रिया दर्शायी गयी। स्पेडोप्टेरा लिटुरा के प्रति AAREO तथा AAWE0 द्वारा क्रमशः (90.85% तथा 87.92%) की अच्छी एंटी फीडेंट प्रतिक्रिया पायी गयी। इसी तरह AAREO तथा AAWE0 द्वारा मेलोडोगाइनी इन्कॉगनीटा के प्रति निमेटोसाइडल प्रतिक्रिया दर्शायी गयी। AAREO द्वारा 1-5 माइक्रो ली० प्रति मी० ली० की सांद्रता में AAWE0 की तुलना में अधिक निमेटोसाइडल नश्वरता पायी गयी, जबकि 10 माइक्रो ली० प्रति मी० ली० की सांद्रता AAWE0 (5.38%) में AAREO (11.41%) की तुलना में ज्यादा अंडा निषेधक प्रतिक्रिया पायी गयी।

उपर्युक्त अध्ययन की टिप्पणियों के आधार पर, यह कहा जा सकता है कि आर्टिमिसिया एनुआ, कैम्फर और 1,8 सिनेओल का एक अच्छा प्राकृतिक संसाधन हो सकता है और उचित क्लीनिकल परीक्षणों के बाद आवश्यक तेल एंटी-ऑक्सिडेंट, एंटी-डायबिटीज, एंटी-फीडेंट और नेमाटीसाइडल का अच्छा स्रोत हो सकता है।


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